

**Profiling Volatile Constituents of Homemade Preserved Foods Prepared in Early 1950s
South Dakota (USA) Using Solid-Phase Microextraction (SPME) with Gas
Chromatography-Mass Spectrometry (GC-MS) Determination**

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Supplementary Material

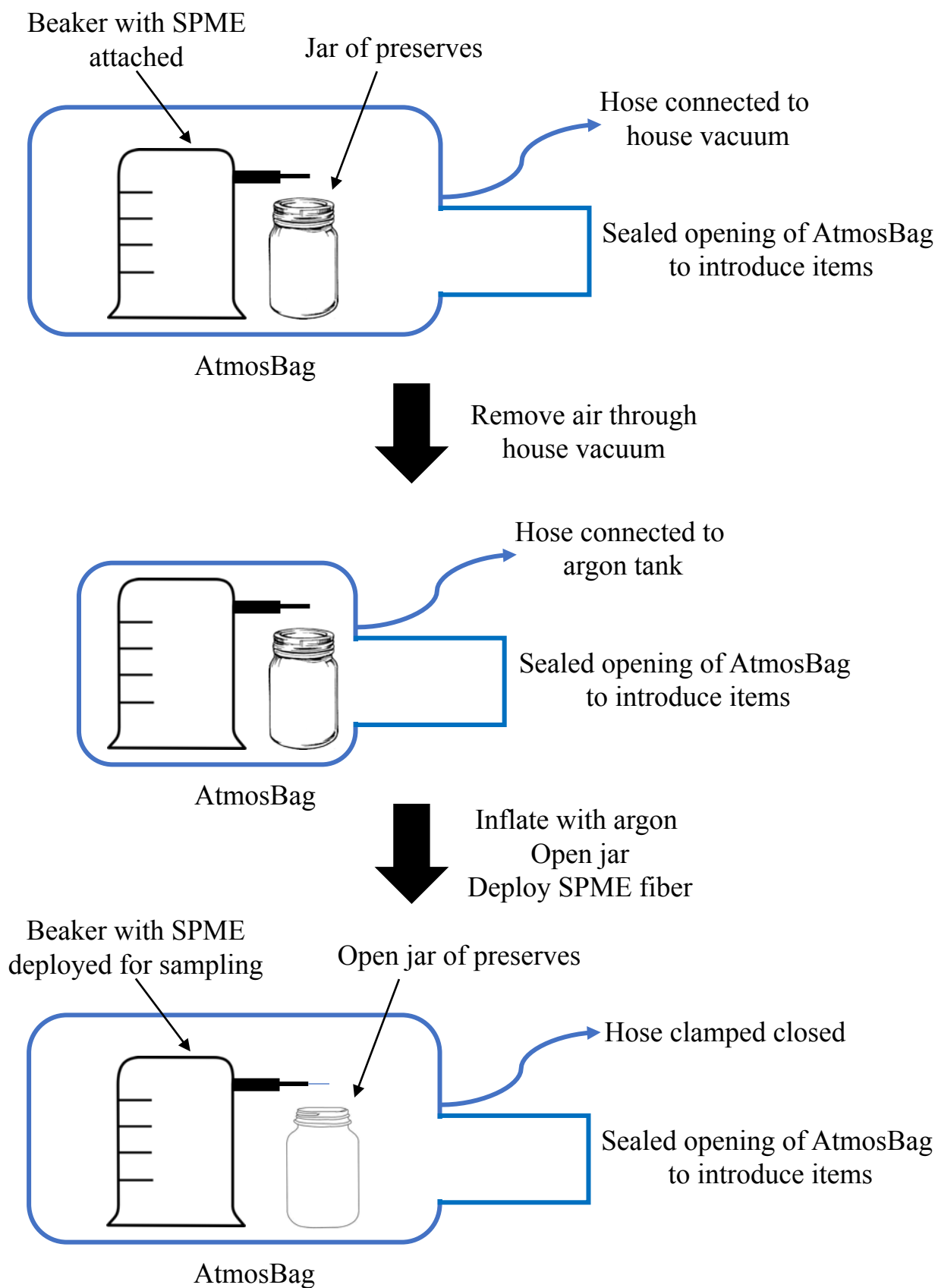
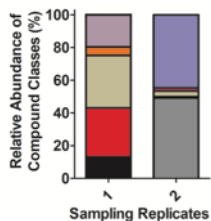
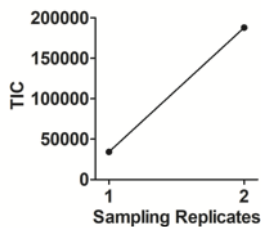


Figure S1. Schematic of the SPME sampling protocol used in this work.

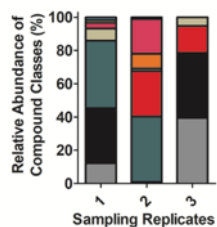
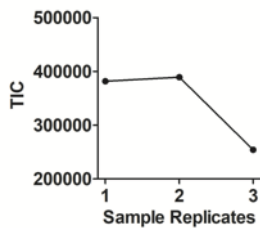


Figure S2. Photodocumentation of the 1950s preserved foods analyzed in this study after they were wiped down with 95% ethanol to remove dust and debris. Jars are aligned sequentially and the white label displays the sample number. Identifying markings on the jars are also included in the photographs for accuracy. An asterisk is included above jars that did not yield BPA above the limit of detection.

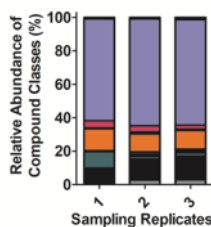
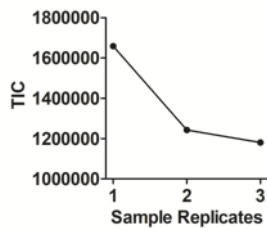
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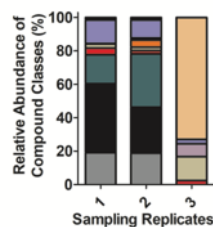
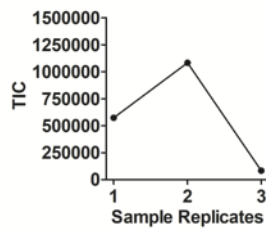
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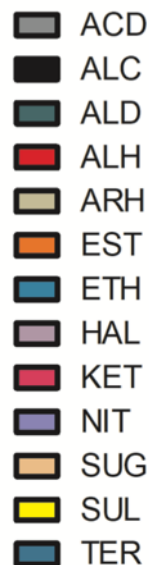
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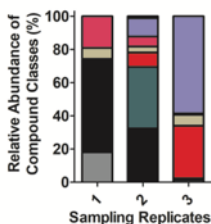
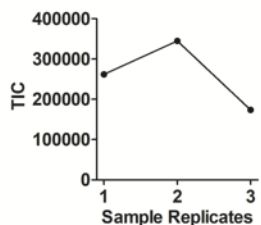
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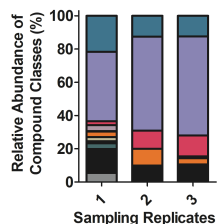
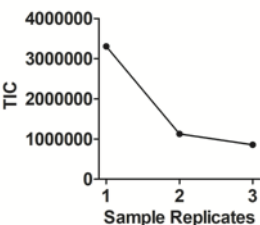
Lower graph legend



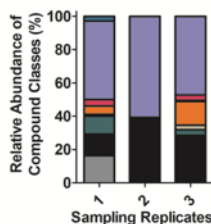
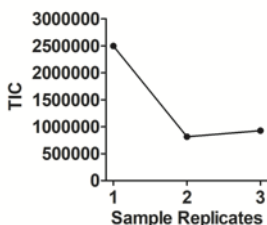
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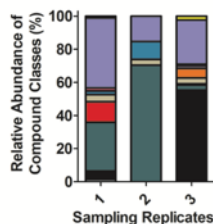
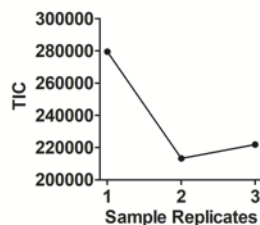
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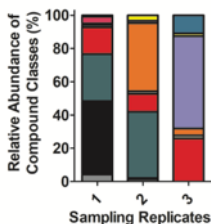
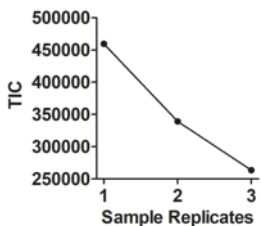
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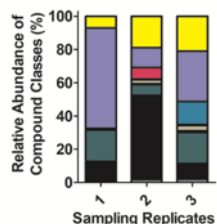
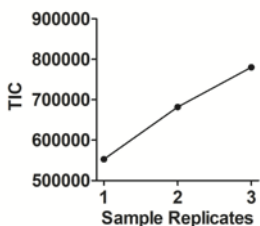
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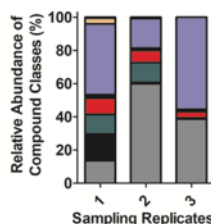
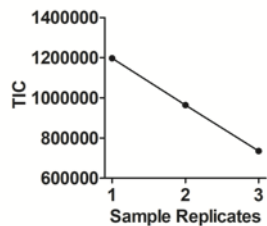
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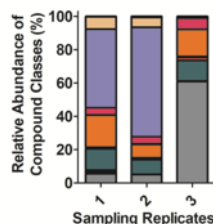
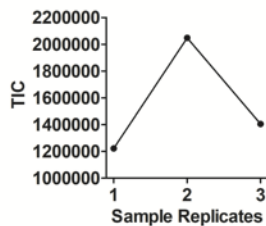
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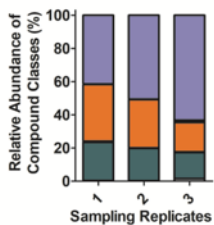
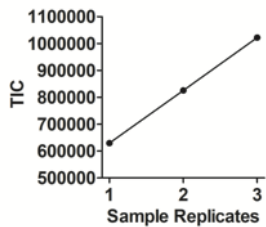
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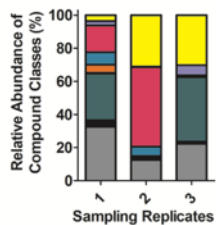
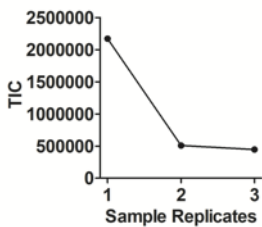
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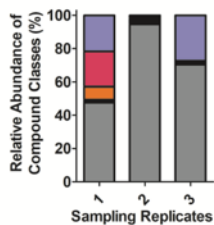
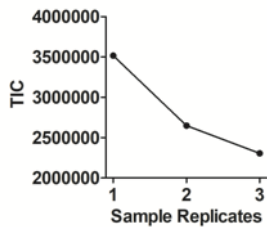
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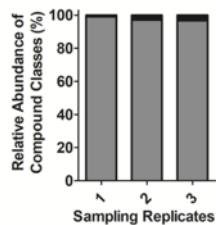
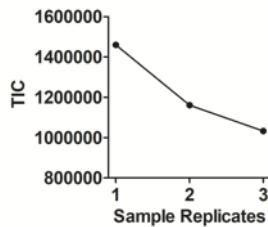
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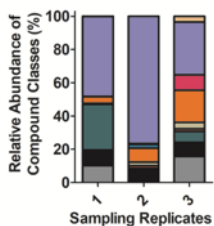
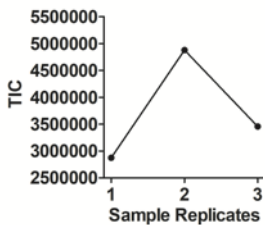
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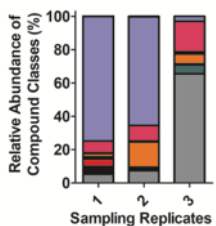
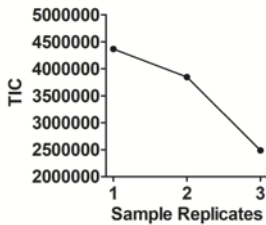
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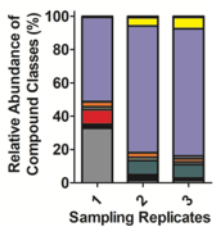
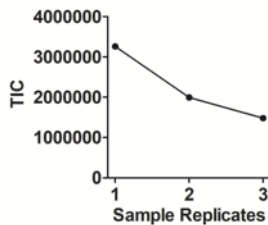
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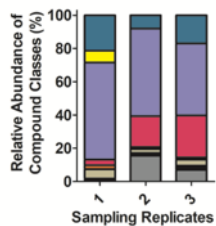
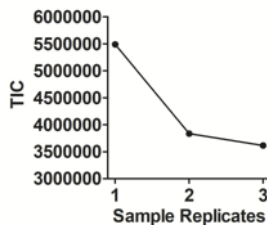
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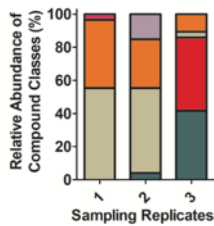
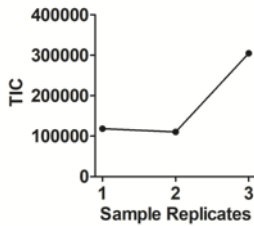
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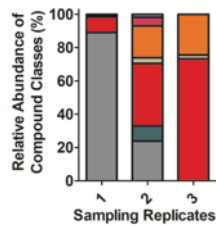
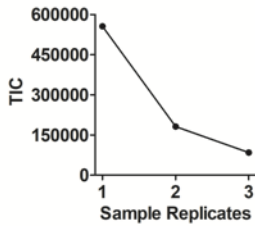
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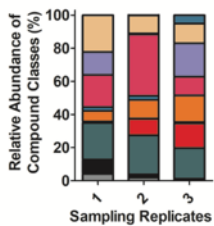
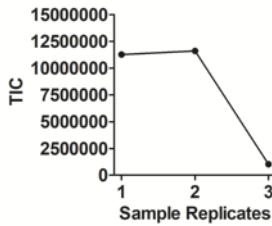
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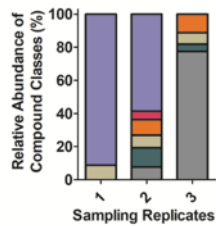
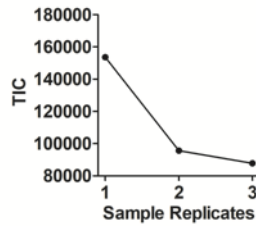
Sample 22



Sample 23



Sample 24



Lower graph legend



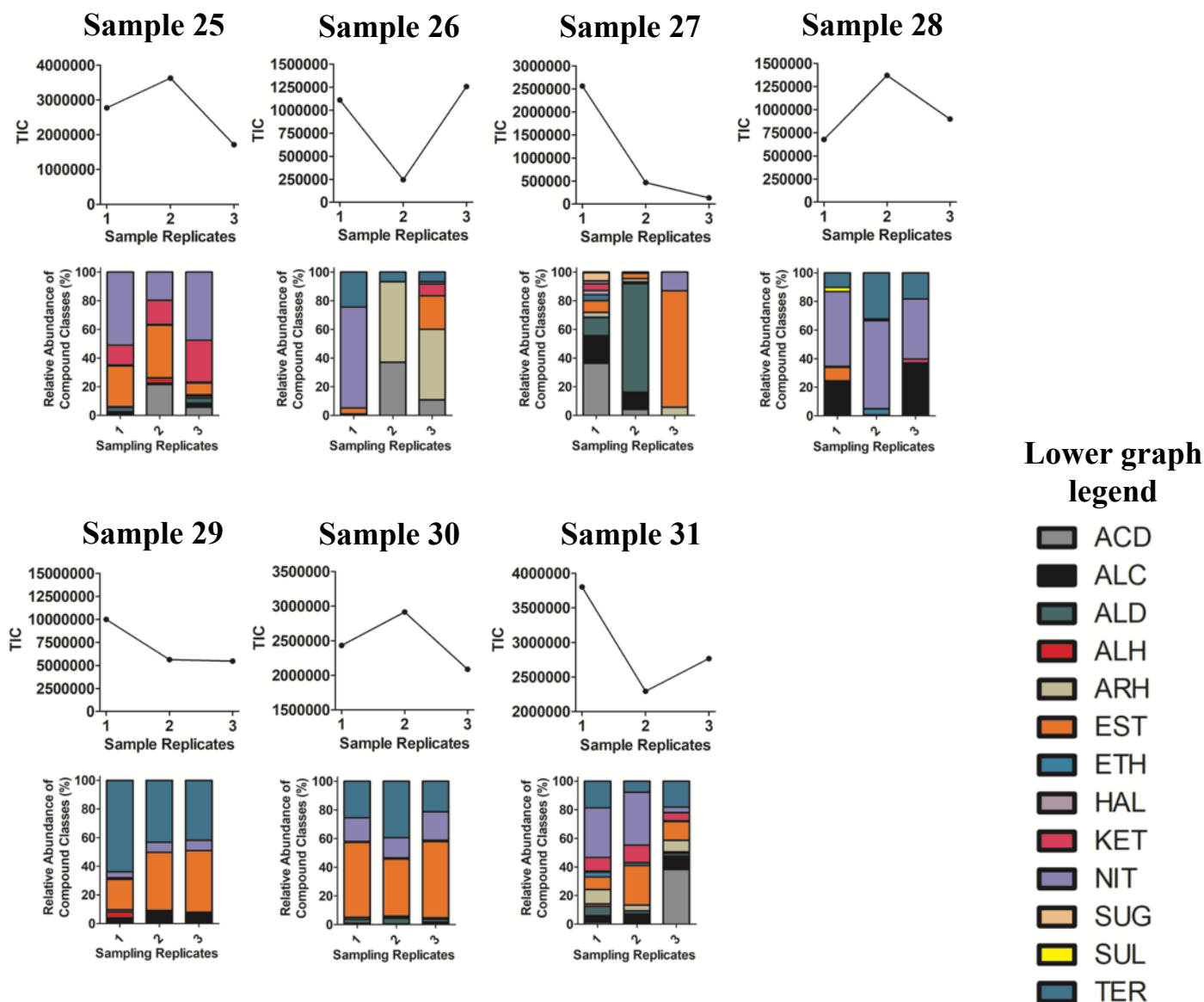


Figure S3. Assessment of the volatile profile of 31 historical preserve samples. The data from each of the three successive samplings of each historical preserve are presented as the total integrated area from the chromatograms of compounds with a NIST score over 80% (top graph), and the relative abundance of the compounds belonging to each compound class assessed here (bottom graph). The legend for the column colors used in the bottom graph for each sample is on the right side of the page. Compound classes are defined as follows: ACD = acids; ALC = alcohols; ALD = aldehydes; ALH = aliphatic hydrocarbons; ARH = aromatic hydrocarbons; EST = esters; ETH = ethers; HAL = halogen-containing; KET = ketones; NIT = nitrogen-containing; SUG = sugar alcohols; SUL = sulfur-containing; TER = terpenoids.

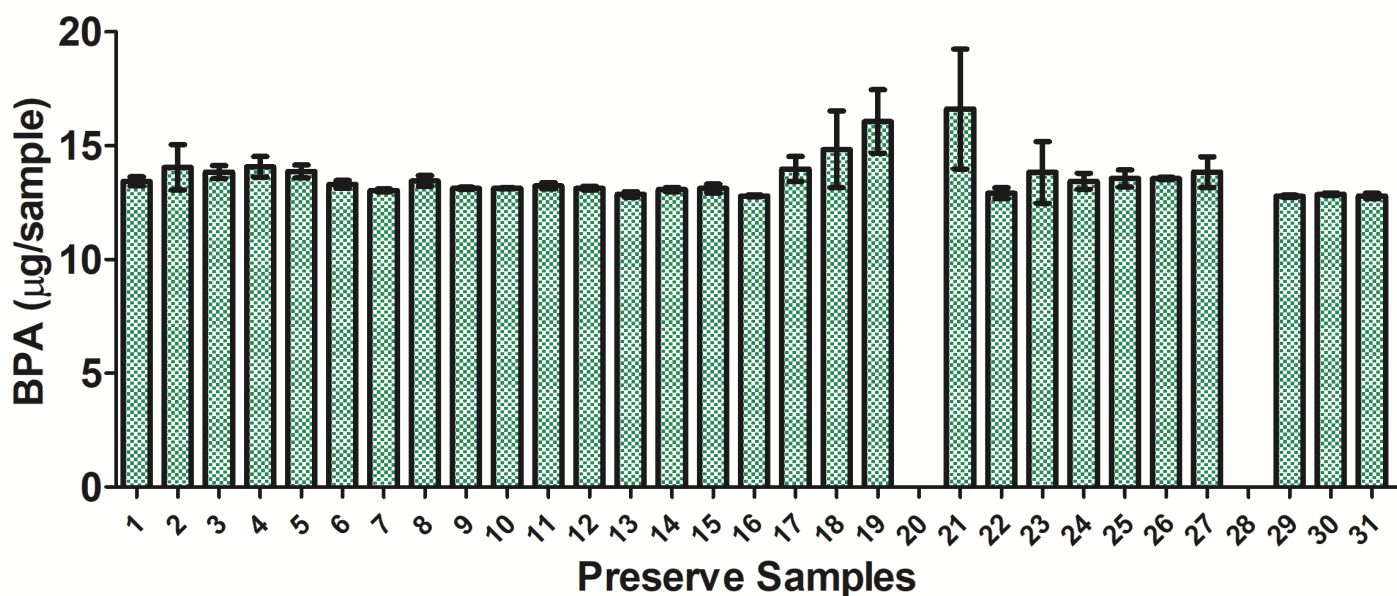


Figure S4. The concentration of BPA in historical preserves is presented as the average \pm standard deviation normalized to absolute mass detected per sample as measured from three successive samplings of each historical preserve.

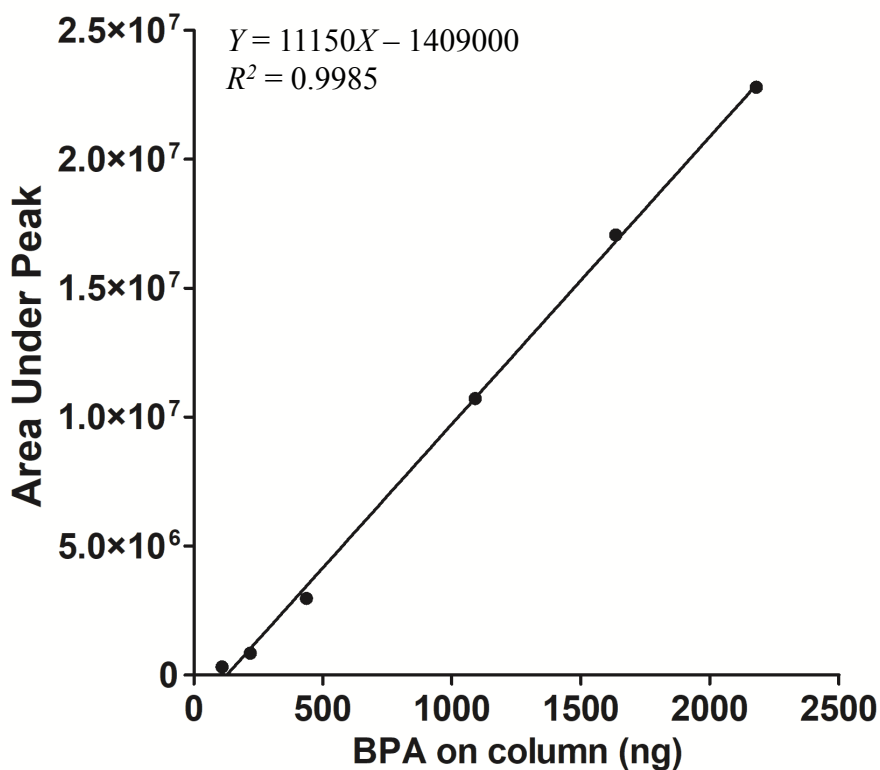


Figure S5. Quantitation curve created with the an authentic BPA standard generated over the concentration range of 100 to 2200 ng on-column in triplicate.

Table S1. Quantification of the relative abundance of each compound class by sample used to create Figure 3. For multiple preserve types, the preserve name is followed by the sample number in parentheses, e.g. Apple 1 (1). The numbers in each compound class are the percentage of this compound class detected in the first (2-hour) SPME sampling of each jar. Percentages were computed by summing the total integrated peak areas for each of the compounds of a given class and dividing it by the total integrated peak area in the TIC. The absence of compounds in a given class is presented as an empty box.

Sample	ACD	ALC	ALD	ALH	ARH	EST	ETH	HAL	KET	NIT	SUG	SUL	TER
Apple 1 (1)		13.23		29.87	32.13	5.16		19.60					
Apple 1 (15)	47.58	1.29	0.15	0.16	0.22	7.69	0.12		21.08	21.70			
Apple 2 (8)	1.54	4.88	29.49	12.43	4.30		2.29		1.73	42.32		1.02	
Apricots (17)	10.13	9.39	27.53	0.10	0.44	4.01			0.16	48.09		0.13	
Apricots (27)	36.39	19.10	12.84	0.15	3.46	8.17	4.18	2.99	4.64	1.91	5.87	0.29	
Apricots (28)		24.64				9.32			0.75	52.29		2.92	10.08
Branded Fruit (4)	19.13	41.26	17.28	4.01	2.52				0.20	14.04		0.67	0.89
Branded Fruit (23)	4.13	8.66	22.18	0.49	0.26	6.48	2.31		19.46	13.86	22.18		
Crab Apple 1 (25)		2.52	3.27		0.27	28.58	0.60		13.92	50.84			
Crab Apple 2 (26)		0.17			0.97	4.10			0.04	70.43			24.28
Dill Pickle 1 (19)	32.84	2.16	0.16	8.85	1.66	3.08			0.07	50.58			0.61
Dill Pickle 2 (29)		4.03		3.94	1.70	21.28	0.57	0.19	0.29	4.13			63.88
Dill Pickle 2 (30)	0.42	0.34	2.99		1.34	52.28	0.32		0.31	16.54			25.46
Head Cheese (5)	18.08	56.17			6.61				19.13				
Mincemeat 1 (3)	0.96	8.91	9.85		0.66	13.00	0.10	0.41	4.24	61.05		0.55	0.27
Mincemeat 1 (12)	5.61	2.14	12.74	0.44	0.54	19.27	0.14		4.44	47.02	7.42	0.24	
Mincemeat 2 (13)	0.44		22.99		0.61	34.24			0.24	41.48			
Peaches (7)	16.36	12.88	10.91	0.67	0.21	5.19			3.78	47.18			2.82
Rhubarb 1 (9)	4.29	44.02	28.38	16.19	1.32			0.77	4.29			0.75	
Rhubarb 1 (11)	13.64	15.96	11.83	9.81	0.66	0.35		1.08		42.54	3.57	0.55	
Rhubarb 1 (21)					55.37	41.24			3.38				
Rhubarb 1 (22)	89.08			9.42	0.20	1.30							
Rhubarb 2 (24)	7.69		11.63		7.55	9.43			5.07	58.64			
Sweet Pickle 1 (6)	0.27	8.53		0.69	1.02	3.90	1.06		12.48	59.62			12.44
Sweet Pickle 1 (18)	5.29	3.45	1.16	4.58	1.30	1.95	0.34		7.17	74.71		0.05	
Sweet Pickle 2 (20)	0.32	0.32	0.74	0.44	5.63	2.26			3.56	58.32		7.13	21.29
Sweet Pickle 3 (31)	1.41	4.53	6.63	1.62	10.14	8.69	3.48	0.62	9.52	34.70		0.07	18.59
Tomato 1 (2)	12.32	33.04	40.58		7.30				3.48			1.56	1.72
Tomato 1 (10)		12.65	18.95		1.00					60.37		7.03	
Tomato 1 (16)	99.05				0.15	0.16			0.10	0.38	0.16		
Tomato 2 (14)	32.61	4.04	28.14	0.15	0.28	4.86	7.57		16.15	2.64		3.58	

Compound classes are defined as follows: ACD = acids; ALC = alcohols; ALD = aldehydes; ALH = aliphatic hydrocarbons; ARH = aromatic hydrocarbons; EST = esters; ETH = ethers; HAL = halogen-containing; KET = ketones; NIT = nitrogen-containing; SUG = sugar alcohols; SUL = sulfur-containing; TER = terpenoids.

Table S2. Comparison of the compounds tentatively identified during SPME analysis of Vlasic dill pickles, cucumbers, lab-made brine, and cucumber spears steeped in the lab-made brine. The compounds listed are those derived from three independent 2 h SPME samplings and aggregated to display the complexity of the volatile profile.

RT (min)	Compounds ^a	RI ^b	Vlasic Dill Pickles	Lab Prepared Constituents ^c		
				Cucumber ^d	Brine	Cucumber + Brine
1.8	Acetic acid	600	x	-	x	x
1.9	Ammonium acetate	-	x	-	x	x
2.0	Ethyl acetate	628	x	-	x	x
5.5	Diallyl sulfide	-	x	-	-	-
8.9	β -Myrcene	992	x	-	x	x
9.2	1-(2-methoxy-1-methylethoxy)-2-propanol	999	x	-	-	-
9.3	2-(2-ethoxyethoxy)-ethanol	-	x	-	-	-
9.4	α -Phellandrene	1006	x	-	x	x
9.6	1-(2-methoxypropoxy)-2-propanol	-	x	-	-	-
9.7	(+)-4-Carene	1009	x	-	x	x
9.9	p-Cymene	1027	x	-	x	x
10.0	Limonene ^c	1030	x	x	-	x
12.8	cis-1-methyl-4-(1-methylethyl)-2-cyclohexen-1-ol	1145	x	-	-	-
15.2	3,6-dimethyl-2,3,3a,4,5,7a-hexahydrobenzofuran	1187	x	x	-	x
15.5	α -Terpineol	1195	x	-	-	-
15.7	trans-2-methyl-5-(1-methylethenyl)-cyclohexanone	1200	x	-	x	x
17.2	D-Carvone	1253	x	-	x	x
17.3	(S)-2-methyl-5-(1-methylethyl)-2-cyclohexen-1-one	1256	x	-	-	-
17.9	Nonanoic acid	1275	x	-	-	-
21.8	1-Dodecanol	1577	x	-	-	-
	Total Compounds		20	2	9	11

“x” denotes compound detected.

“-” denotes compound not detected.

^aAll compounds identified at $\geq 80\%$ similarity to the NIST Mass Spectral Library.

^bKovats retention index

^cArtificial brine was prepared with as many ingredients as listed on the nutritional labeling of the Vlasic pickles. Cucumbers were purchased from a local supermarket, cut into spears of the same dimensions as the Vlasic pickles, and packed into washed and solvent-rinsed Vlasic pickle jars.

^dLimonene is presumed to be an artifact of chitosan coatings commonly used to extend cucumber shelf life, or perhaps the result of a metabolic stress response to preparation (Bruni et al. 2007). Bruni, R.; Bianchi, A.; Bellardi, M.G. Essential Oil Composition of *Agastache anethiodora* Britton (Lamiaceae) Infected by Cucumber Mosaic Virus (CMV). *Flavour and Fragrance Journal* 2007, 22, 66-70.

Table S3. Masses of the historical preserves were determined by weighing the jars prior to sampling but after they were wiped with 95% ethanol and allowed to dry in a fume hood.

Sample	Mass (g)
1	1319.9
2	1244.8
3	1400.1
4	1426.8
5	1382.0
6	1441.6
7	1466.2
8	1179.1
9	1471.6
10	1249.0
11	1393.3
12	1460.4
13	797.3
14	1259.3
15	1320.7
16	1269.1
17	1350.2
18	1388.2
19	1250.8
20	1421.7
21	1319.9
22	1385.8
23	1564.3
24	1387.5
25	1491.6
26	704.9
27	1305.3
28	716.8
29	3794.3
30	3490.9
31	3007.9

Table S4. Start times of the video documentation of the qualitative taste tests of representative historical preserve samples.

Sample	Start time in Video V1
1	0:01
2	0:39
3	1:22
4	2:39
5	3:46
6	4:58
7	5:55
11	7:09
19	8:20
20	9:09
25	10:00
26	11:30
29	12:26
31	13:15