Supporting information

1. HPLC analysis of compounds.

HPLC analysis was performed using a C-18 column (150 \times 4.6 mm) with a 0.6 mL/min flow rate and checked with a detector ($\lambda = 380$ or 220 nm). The column was initially held at 10% CH₃CN - 90% H₂O. The concentration of CH₃CN was ramped to 100% in 25 min and this concentration was maintained for 5 min and returned to 10% within 9 min. Before each next injection, the column was allowed to equilibrate to the initial mobile phase conditions for 30 min.

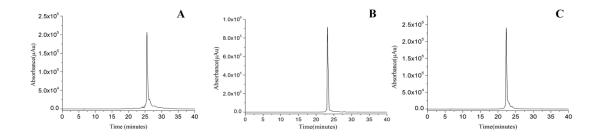


Fig. S1. HPLC traces (λ = 380 nm) of the conjugates (A) Pyro-MonoRGD, (B) Pyro-DiRGD and (C) Pyro-TriRGD.

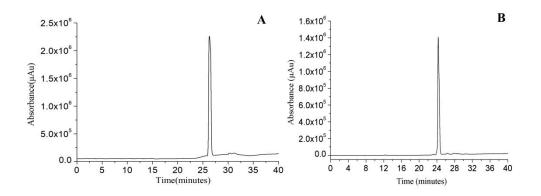


Fig. S2. HPLC traces ($\lambda = 220$ nm) of (A) cyclic RGD pentapeptide 1 and (B) amino-modified cyclic RGD peptapeptide 2 after purification.

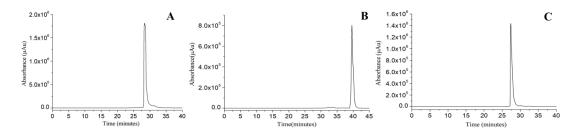


Fig. S3. HPLC traces ($\lambda = 380$ nm) of (A) Pyro-conjugated dimeric linker 7, (B) compound 10 and (C) Pyro-conjugated trimeric linker 11 after purification.

2. Mass spectrometry analysis of compounds.

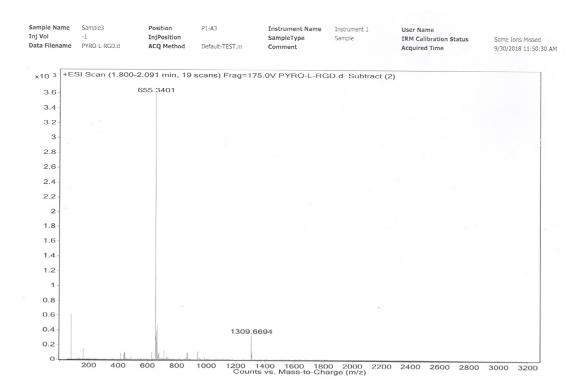


Fig. S4. Mass spectrometry analysis of Pyro-MonoRGD (ESI-HRMS: m/z = 1309.6694, calcd for C₆₈H₈₉N₁₄O₁₃ m/z = 1309.6734 [M+H]⁺).

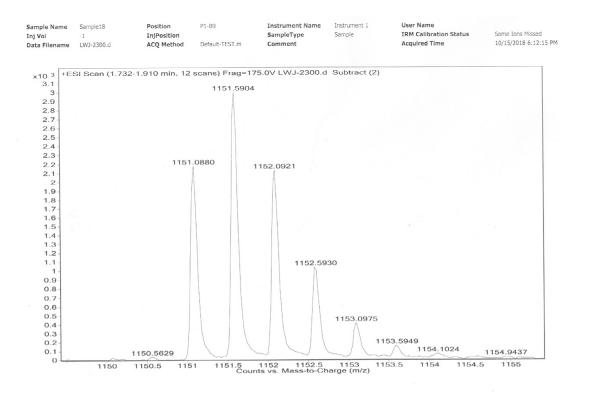


Fig. S5. Mass spectrometry analysis of Pyro-DiRGD (HRMS-ESI: m/z = 1151.0880, calcd for C₁₁₂H₁₅₉N₂₅O₂₈ m/z = 1151.0893 [M+2H]^{2+/2}).

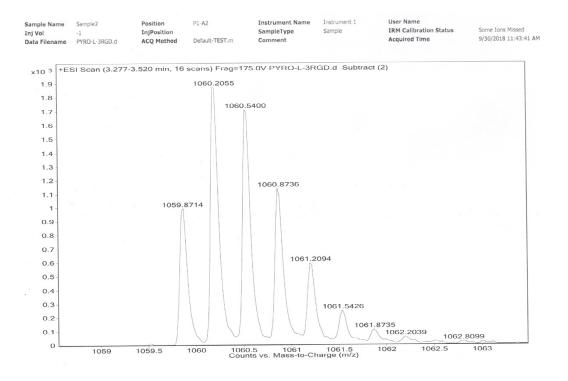


Fig. S6. Mass spectrometry analysis of Pyro-TriRGD (ESI-HRMS: m/z = 1059.8714, calcd for C₁₅₁H₂₂₀N₃₅O₄₁ m/z = 1059.8735 [M + 3H]^{3+/3}).

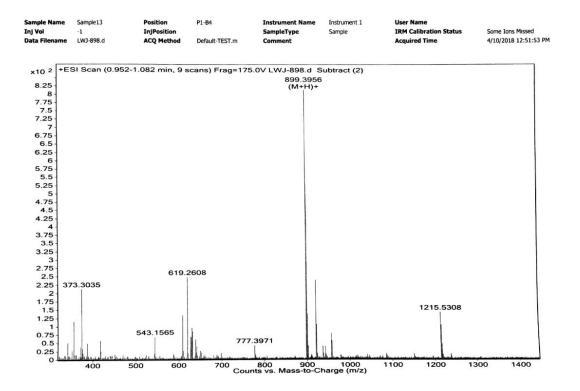


Fig. S7. Mass spectrometry analysis of protected pentapeptide cyclo(-Arg[Pbf]GlyAsp[tBu]-D-Phe-Asp-) **1** (HRMS-ESI: m/z = 899.3956, calcd for C₄₂H₅₉N₈O₁₂S m/z = 899.3973 [M+H]⁺).

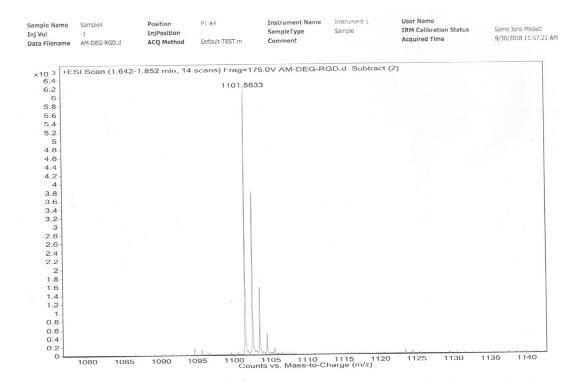


Fig. S8. Mass spectrometry analysis of amino-modifed cyclic pentapeptide **2** (ESI-HRMS: m/z = 1101.5633, calcd for C₅₂H₈₁N₁₀O₁₄S m/z = 1101.5654 [M+H]⁺).

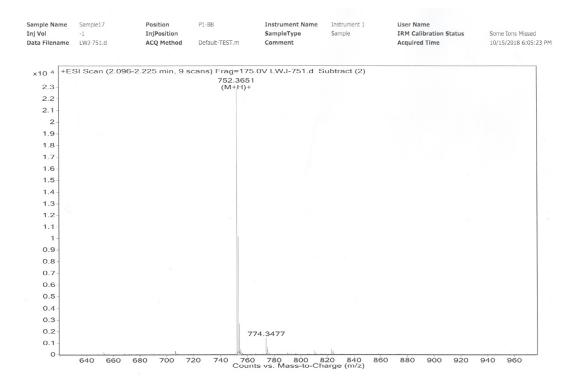


Fig. S9. Mass spectrometry analysis of Pyro-conjugated dimeric linker 7 (HRMS-ESI: m/z = 752.3651, calcd for C₄₂H₅₀N₅O₈ m/z = 752.3659 [M+H]⁺).

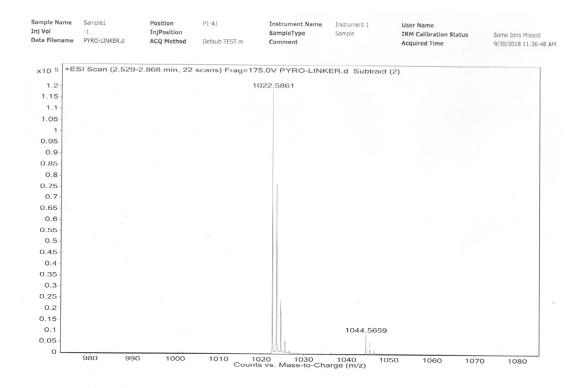


Fig. S10. Mass spectrometry analysis of compound **10** (ESI-HRMS: m/z = 1022.5861, calcd for C₅₈H₈₀N₅O₁₁ m/z = 1022.5854 [M+H]⁺).

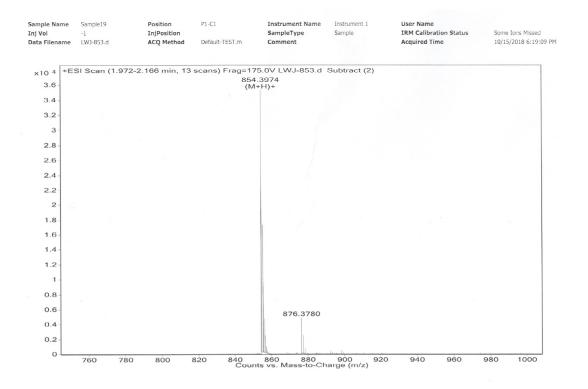


Fig. S11. Mass spectrometry analysis of Pyro-conjugated trimeric linker 11 (ESI-HRMS: m/z = 854.3974, calcd for C₄₆H₅₆N₅O₁₁ m/z = 854.3976 [M+H]⁺).

3. ¹H and ¹³C NMR spectra of compounds.

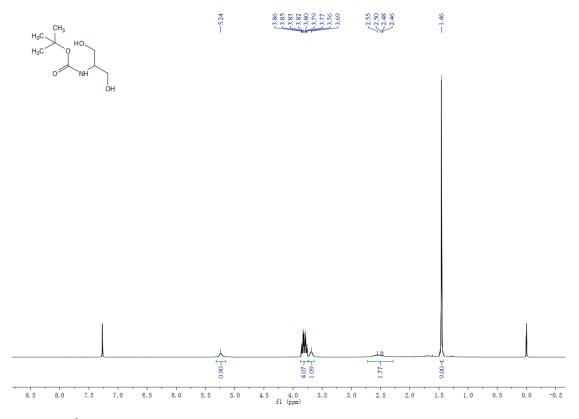


Fig. S12. ¹H NMR spectrum (400 MHz, CDCl₃) of compound 4.

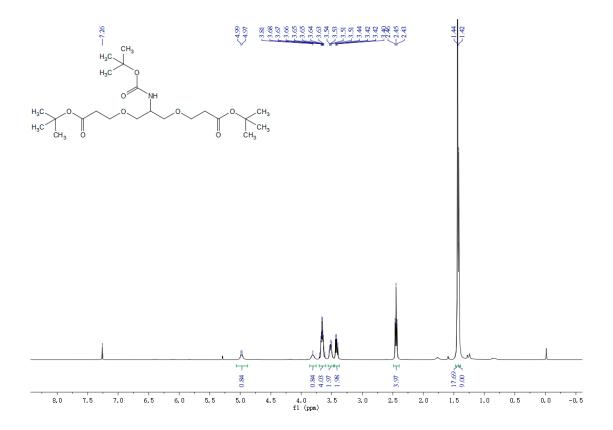


Fig. S13. ¹H NMR spectrum (400 MHz, CDCl₃) of compound **5**.

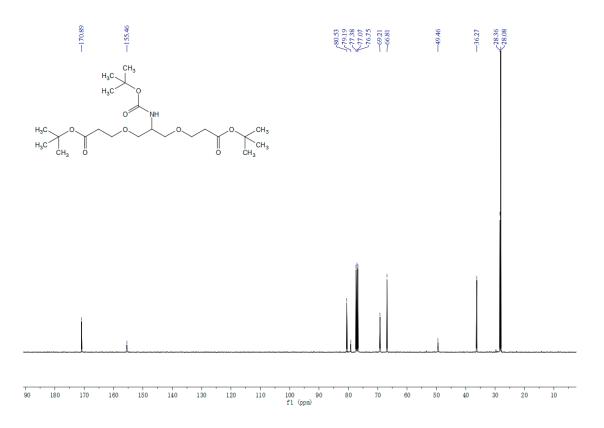


Fig. S14. ¹³C NMR spectrum (100.6 MHz, CDCl₃) of compound **5**.

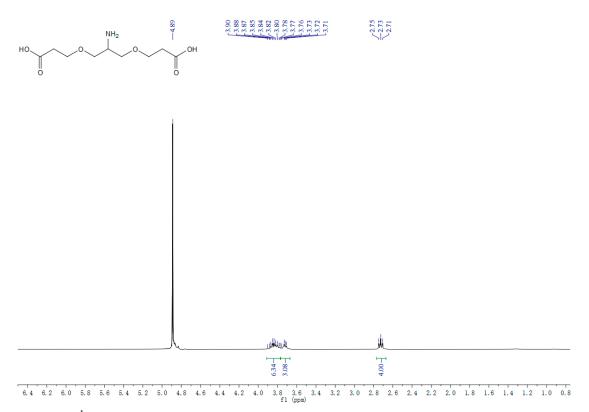


Fig. S15. ¹H NMR spectrum (300 MHz, D₂O) of dimeric linker **6**.

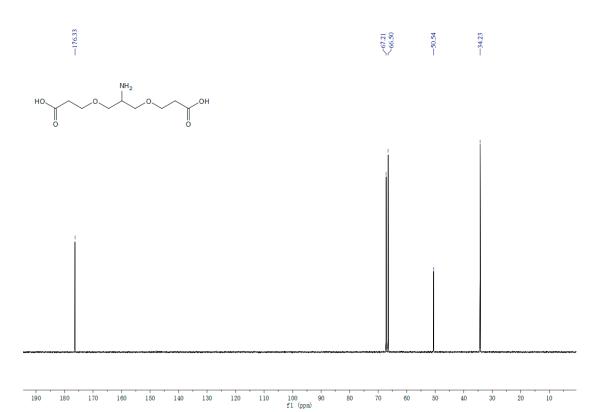


Fig. S16. ¹³C NMR spectrum (100.6 MHz, D₂O) of dimeric linker **6**.

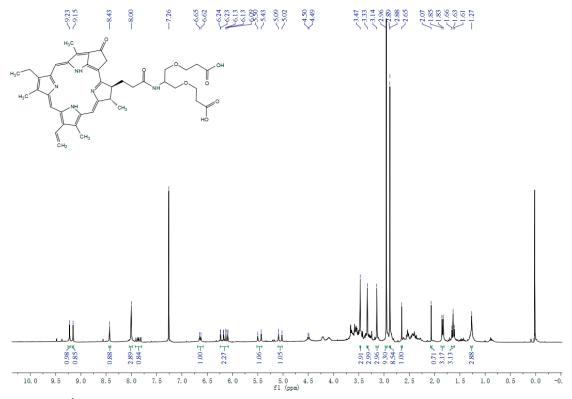


Fig. S17. ¹H NMR spectrum (300 MHz, CDCl₃) of Pyro-conjugated dimeric linker 7.

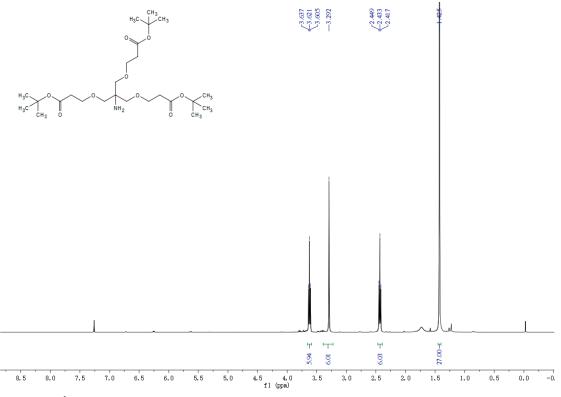


Fig. S18. ¹H NMR spectrum (400 MHz, CDCl₃) of compound **9**.

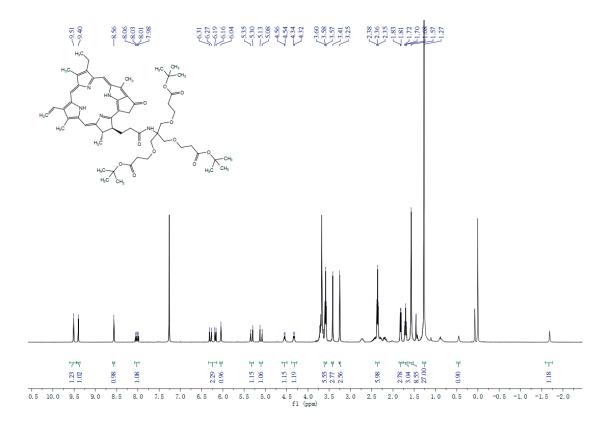


Fig. S19. ¹H NMR spectrum (400 MHz, CDCl₃) of compound **10**.

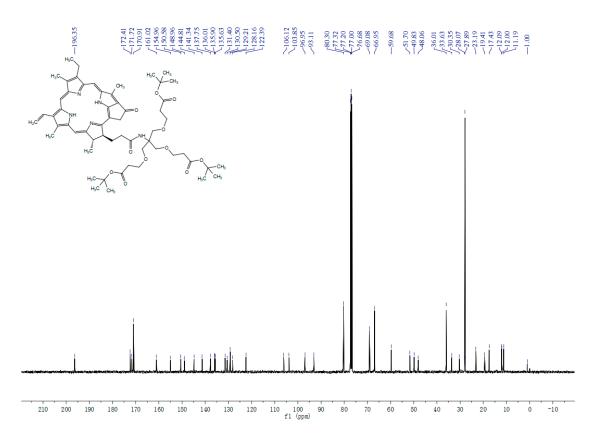


Fig. S20. ¹³C NMR spectrum (100.6 MHz, CDCl₃) of compound **10**.

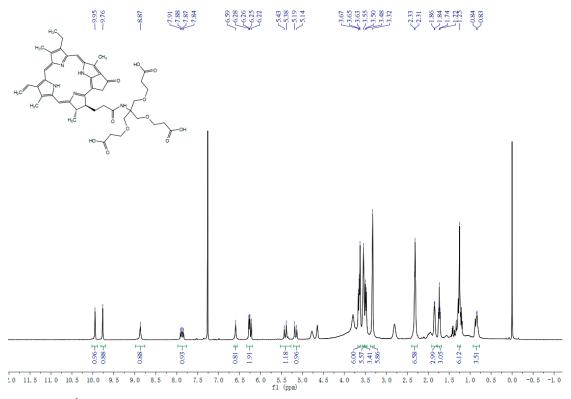


Fig. S21. ¹H NMR spectrum (400 MHz, CDCl₃) of Pyro-conjugated trimeric linker 11.

4. The detection of singlet oxygen quantum yield in DMSO.

The UV absorption spectra of the photosensitizers in the wavelength range from 300 nm to 800 nm were measured with DPBF ($3 \times 10^{-5} \text{ mol} \cdot \text{L}^{-1}$) and irradiated with the light of 680 nm. The absorbance at 680 nm was recorded, and the photobleaching rate of DPBF at 417 nm was recorded for different durations of light irradiation. The singlet oxygen quantum yield was obtained according to the following equation:

$$\Phi_{\Delta} = \Phi_{\Delta}^{\text{Std}} \frac{R_{\text{DPBF}}^{\text{s.m.}}(1-10^{-A_{680}})}{R_{\text{DPBF}}^{\text{Std}}(1-10^{-A_{680}})}$$

Where $\Phi_{\Delta}^{\text{Std}}$ is the singlet oxygen quantum yield of Pyro in DMSO, $\Phi_{\Delta}^{\text{Std}} = 0.52$. $R_{DPBF}^{\text{s.m.}}$ and R_{DPBF}^{Std} are photobleaching rates at 417 nm for each photosensitizer, respectively. A_{680}^{Std} and $A_{680}^{\text{s.m.}}$ represent the absorbance of Pyro and compounds at 680 nm, respectively.

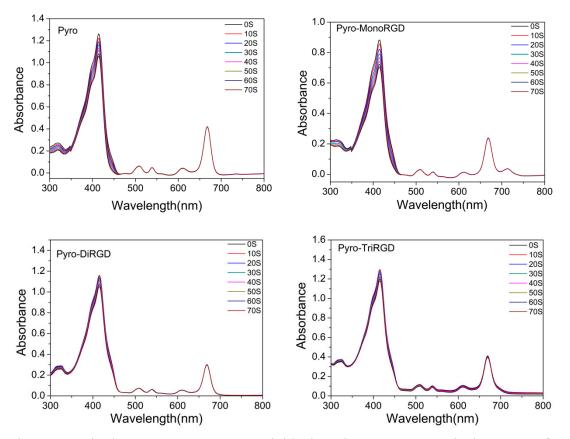


Fig. S22. singlet oxygen quantum yield data in DMSO. Typical spectra for determination of the singlet oxygen quantum yield of free Pyro, Pyro-MonoRGD, Pyro-DiRGD and Pyro-TriRGD in DMSO using DPBF as the scavenger. DPBF

concentration = 3×10^{-5} M.

5. The UV-Vis and Fluorescence spectra of Pyro-MonoRGD, Pyro-DiRGD, Pyro-TriRGD and free Pyro detected in DMSO.

The Cary 5000 spectrophotometer(Varian Co., USA) was used for UV-Vis spectra. Absorption spectra were recorded from 300 to 800 nm. Fluorescence emission and excitation spectra were recorded from 550 to 800 nm upon excitation at approximately 668 nm and emission at 672-673 nm using a Hitachi Model F-4500 FL spectrophotometer (Tokyo, Japan).

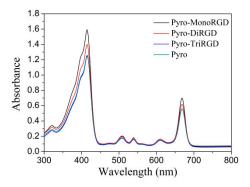
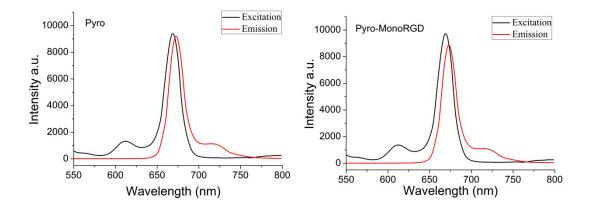


Fig. S23. UV-vis absorption spectra of Pyro-MonoRGD, Pyro-DiRGD, Pyro-TriRGD and Pyro in DMSO. The concentration for the absorption spectrum determination was $10 \mu M$.



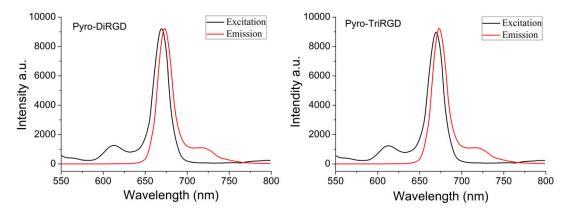


Fig. S24. Fluorescence absorption spectra of Pyro-MonoRGD, Pyro-DiRGD, Pyro-TriRGD and Pyro in DMSO. The concentration for the fluorescence determination was $2.0 \mu M$.