

Supporting Information

Synthesis and Characterization of New Organic Dyes Containing the Indigo Core

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Computational Details

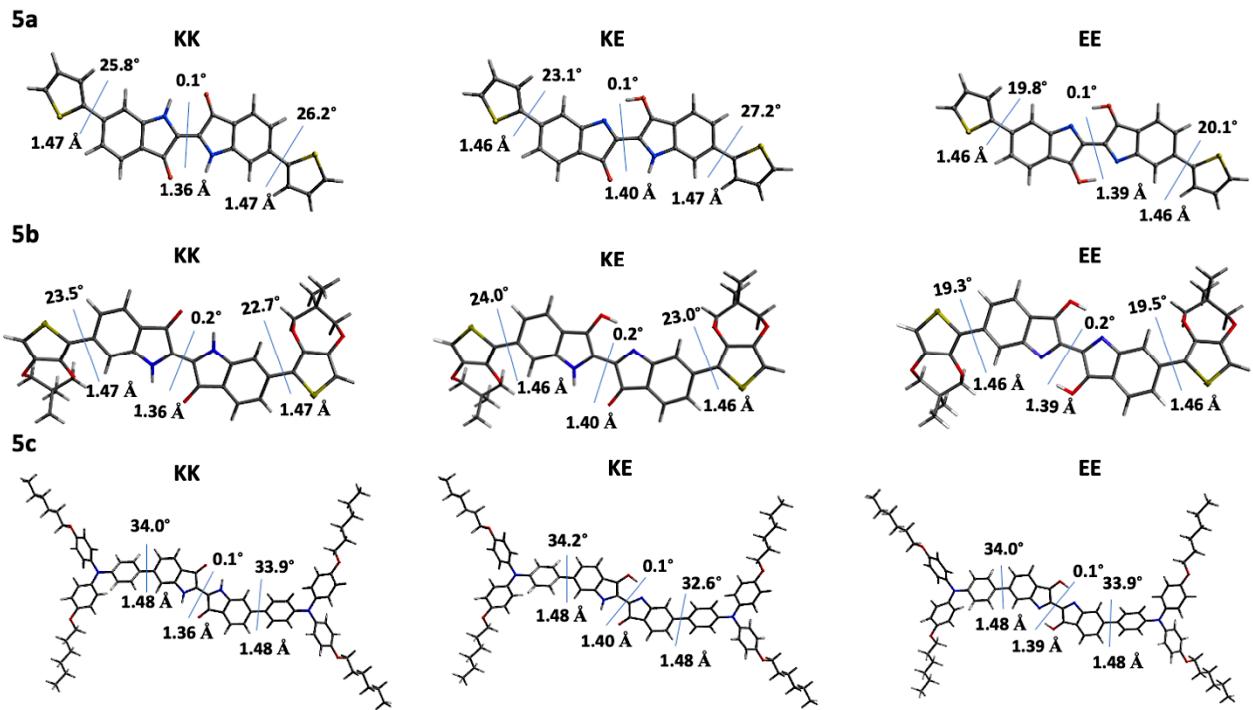


Figure S1. B3LYP/6-31G** optimized geometries in vacuo of KK, KE and EE tautomers of compounds **5a**, **5b** and **5c**.

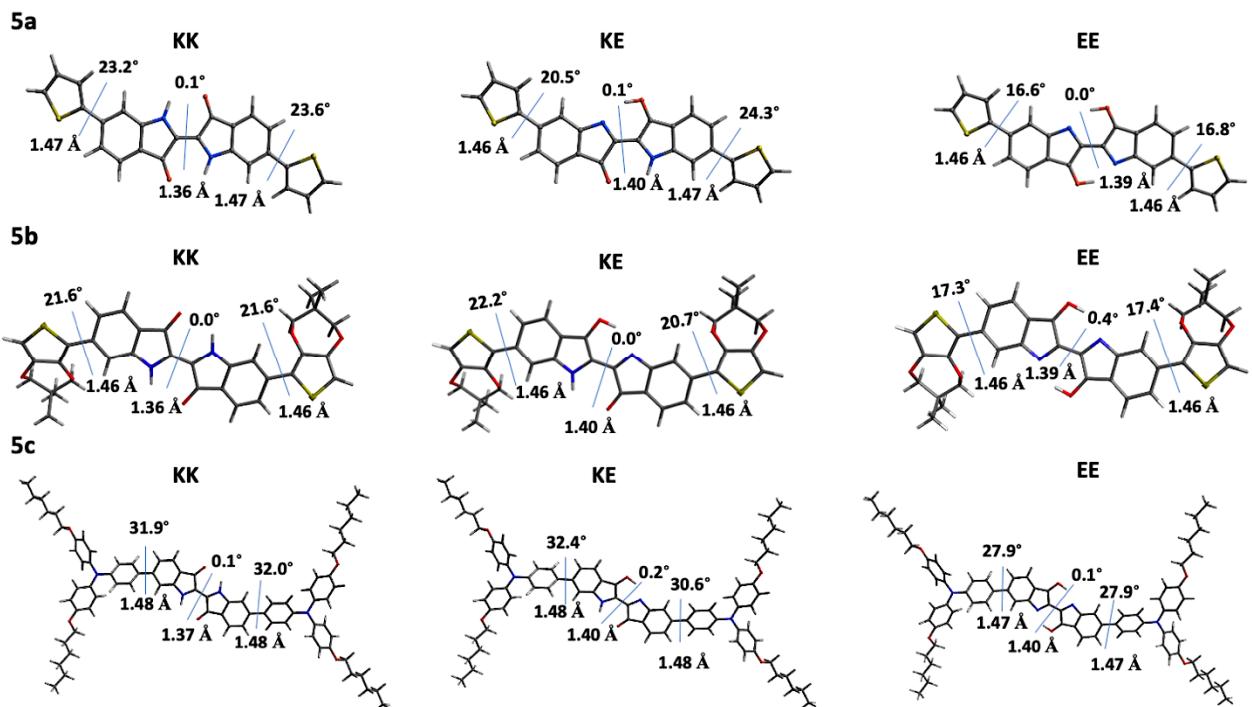
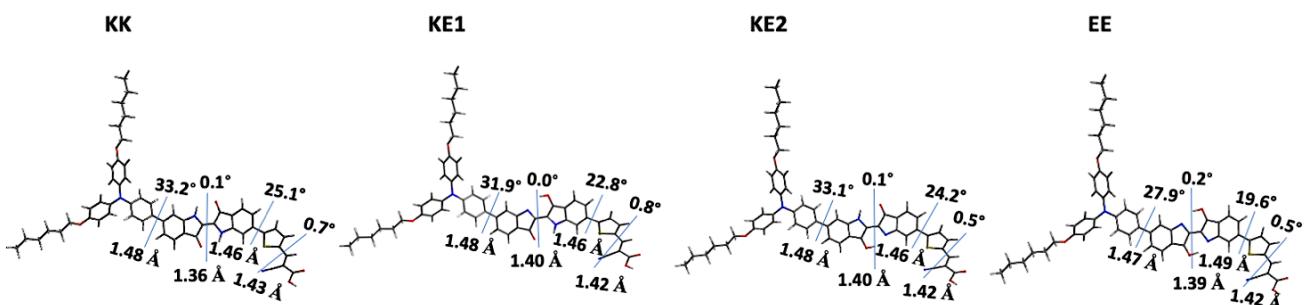


Figure S2. B3LYP/6-31G** optimized geometries in DCM of KK, KE and EE tautomers of compounds of **5a**, **5b** and **5c**.

DF90 (in vacuo optimized geometries)



DF90 (in DCM optimized geometries)

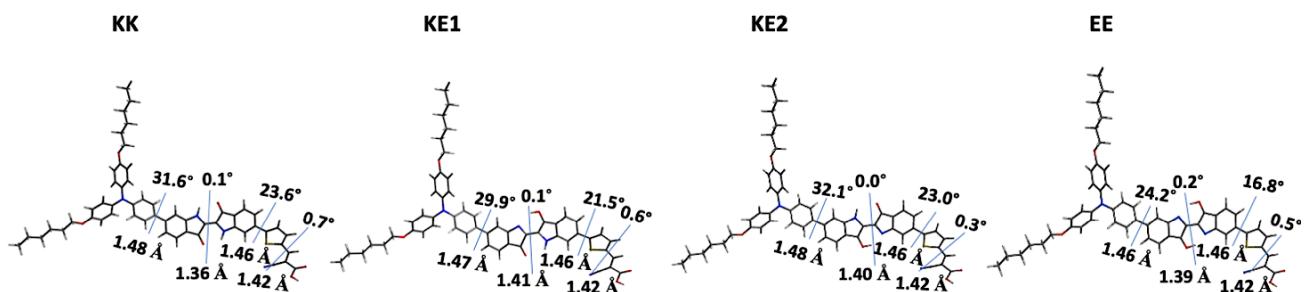
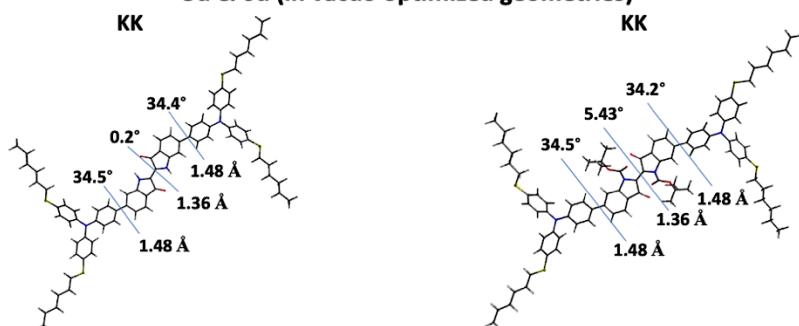


Figure S3. B3LYP/6-31G** optimized geometries in vacuo and in DCM of KK, KE and EE tautomers of compound **DF90**.

5d & 6d (in vacuo optimized geometries)



5d & 6d (in DCM optimized geometries)

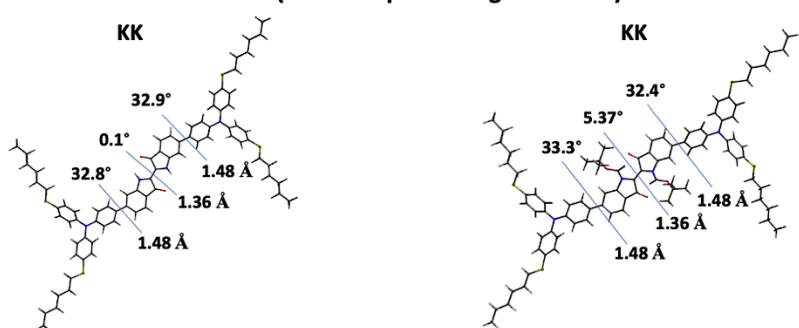


Figure S4. B3LYP/6-31G** optimized geometries in vacuo and in DCM of KK tautomer of compounds **5d** and **6d**.

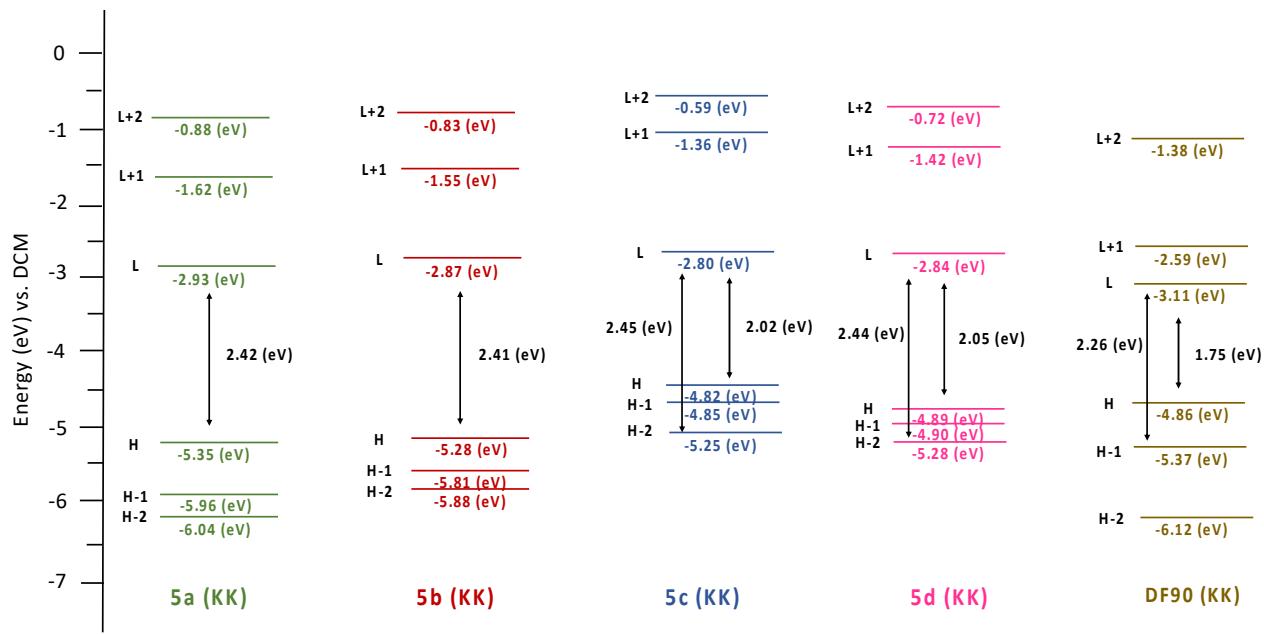


Figure S5. B3LYP/6-31G** FMOs energies in DCM solution of tautomer KK of **5a**, **5b**, **5c**, **5d** and **DF90**.

Table S1. B3LYP/6-31G** absolute and relative energies (kcal/mol) of KK, KE and EE tautomers of compounds **5a**, **5b**, **5c** and **DF90** computed *in vacuo* and DCM.

Molecule	Tautomer	Absolute energy in vacuo	ΔE	Absolute energy in DCM	ΔE
5a	KK	-1676035.507	0.00	-1676045.021	0.00
	KE	-1676025.916	+9.60	-1676034.216	+10.80
	EE	-1676002.483	+33.02	-1676010.483	+34.53
5b	KK	-1242060.895	0.00	-1242068.158	0.00
	KE	-1242051.572	+9.32	-1242057.493	+10.66
	EE	-1242028.210	+32.67	-1242033.516	+34.64
5c	KK	-2269806.326	0.00	-2269820.066	0.00
	KE	-2269796.788	+9.54	-2269808.638	+11.43
	EE	-2269773.874	+32.45	-2269785.628	+34.44
DF90	KK	-1980710.268	0.00	-1980725.508	0.00
	KE1	-1980701.877	+8.31	-1980715.832	+9.68
	KE2	-1980700.304	+9.96	-1980714.280	+11.23
	EE	-1980678.398	+31.87	-1980692.148	+33.36

Table S2. B3LYP/6-31G** absolute and relative^a stabilities (kcal/mol) of KK, KE and EE tautomers of compounds **5a** and **DF90** in vacuo.

Molecule	Tautomer	Absolute stability	Relative stability
5a	KK	-1241896.059	0.0
	KE	-1241886.569	+9.49
	EE	-1241863.825	+32.23
DF90	KK	-1980229.065	0.0
	KE1	-1980220.368	+8.70
	KE2	-1980219.076	+9.99
	EE	-1980197.583	+31.48

^aThe Gibbs free energy of the most stable tautomer is taken as the reference (0.0 kcal/mol).

Table S3. B3LYP/6-31G* orbital energies of KK, KE and EE tautomers of compounds **5a**, **5b**, **5c** and **DF90** computed in DCM.

Molecule	Tautomer	HOMO-2	HOMO-1	HOMO	LUMO	LUMO+1	LUMO+2	$\Delta(H-L)$
5a	KK	-6,045	-5,962	-5,350	-2,932	-1,620	-0,885	2.418
	KE	-6,089	-6,026	-5,290	-3,124	-1,587	-0,895	2.166
	EE	-6,137	-5,851	-5,072	-3,674	-1,588	-0,958	1.398
5b	KK	-5.875	-5.814	-5.276	-2.871	-1.547	-0.835	2.405
	KE	-5.960	-5.880	-5.201	-3.063	-1.505	-0.843	2.138
	EE	-5.987	-5.758	-4.990	-3.593	-1.484	-0.897	1.397
5c	KK	-5,253	-4,847	-4,824	-2,799	-1,355	-0.593	2.025
	KE	-5,312	-4,862	-4,757	-2,989	-1,333	-0,600	1.768
	EE	-5,148	-4,966	-4,707	-3,479	-1,291	-0,618	1.228
DF90	KK	-6,119	-5,374	-4,863	-3,114	-2,593	-1,382	1.749
	KE1	-6,165	-5,395	-4,894	-3,190	-2,647	-1,369	1.704
	KE2	-6,114	-5,405	-4,796	-3,304	-2,639	-1,378	1.492
	EE	-5,837	-5,224	-4,943	-3,704	-2,641	-1,345	1.239

Table S4. B3LYP/6-31G* orbital energies of KK tautomer of compounds **5d** and **6d** in DCM.

Molecule	Tautomer	HOMO-2	HOMO-1	HOMO	LUMO	LUMO+1	LUMO+2	$\Delta(H-L)$
5d	KK	-5.282	-4.902	-4.885	-2.836	-1.416	-0.719	2.049
6d	KK	-5.909	-4.918	-4.906	-2.926	-1.615	-0.838	1.980

Table S5. TD-DFT (\$: CAM-B3LYP/6-311++G** and #: B3LYP/6-311++G**) absorption maxima ($\lambda^{\text{a}}_{\text{max}}$), excitation energies (E_{exc}), oscillator strengths (f) and contribution (%) to the transition in DCM of KK, KE and EE tautomers of compound **5a**.

Molecule	Tautomer	Excited States		$\lambda^{\text{a}}_{\text{max}}$ (nm)		E _{exc} (eV)		f		Contribution (%)		
		\$	#	\$	#	\$	#	\$	#	\$	#	
5a	KK	1	1	532.68	604.24	2.32	2.05	0.54	0.54	98% H→L	100% H→L	
		4	3	363.03	473.55	3.41	2.61	1.46	0.71	82% H-2→L	99% H-2→L	
										30% H-7→L		
		8	10	286.89	330.10	4.32	3.75	0.29	0.82	5% H-2→L	81% H-1→L+1	
										4% H-2→L+6		
	KE	1	1	593.70	704.80	2.08	1.76	0.45	0.45	89% H→L	99% H→L	
		2	6	423.34	383.71	2.93	3.23	0.25	0.66	69% H-1→L	97% H→L+1	
		4	9	378.27	338.88	3.27	3.65	0.90	0.09	66% H-2→L	86% H-7→L	
										34% H-3→L		
	EE	5	339.62		312.15		3.65	0.20		H-2→L+1 4% H→L	43% H-1→L+1 40% H→L+2	
		11					3.97	0.39				
		6	312.79		3.96		0.36		58% H→L+1			
	EE	1	1	972.54	1042.0 7	1.27	1.19	0.52	0.46	95% H→L	100% H→L	
		3	3	489.64	599.36	2.53	2.06	1.49	1.10	85% H-2→L	96% H-2→L	
		11	334.82		3.70		0.17		43% H-1→L+1	64% H-9→L		
		14	288.21		4.30		0.18		15% H→L+6	56% H-1→L+1		

Table S6. TD-DFT (\$: CAM-B3LYP/6-311++G** and #: B3LYP/6-311++G**) absorption maxima ($\lambda^{\text{a}} \text{ max}$), excitation energies (E_{exc}), oscillator strengths (f) and contribution (%) to the transition in DCM of KK, KE and EE tautomers of compound **5b**.

Molecule	Tautomer	Excited States		$\lambda^{\text{a}} \text{ max (nm)}$		E _{exc} (eV)		f		Contribution (%)	
		\$	#	\$	#	\$	#	\$	#	\$	#
KK	1	1	537.19	609.29	2.30	2.03	0.55	0.58	98%	100%	H \rightarrow L
	3	3	375.43	495.28	3.30	2.50	0.14	0.70	79%	100%	H-6 \rightarrow L
	4	10	374.21	334.63	3.31	3.70	1.33	0.83	75%	84%	H-2 \rightarrow L
	1	1	600.60	715.12	2.06	1.73	0.46	0.48	88%	99%	H \rightarrow L
	2	3	430.28	518.36	2.88	2.39	0.36	0.39	66%	81%	H-1 \rightarrow L
	4	8	385.36	382.34	3.21	3.24	0.81	0.68	56%	96%	H-2 \rightarrow L
KE	5	12	344.76	314.21	3.59	3.94	0.15	0.28	23%	40%	H-4 \rightarrow L
	6		314.69		3.93		0.46		19%	4%	H-8 \rightarrow L
	1		1044.77		1.18		0.51		65%		H \rightarrow L+1
	1	9	975.26	622.60	1.27	1.99	0.55	1.10	95%	96%	H \rightarrow L
	3	9	502.66	392.89	2.47	3.15	1.51	0.09	84%	94%	H-2 \rightarrow L
	6	16	359.71	328.22	3.44	3.77	0.03	0.99	82%	73%	H-5 \rightarrow L
EE	1										H-1 \rightarrow L+1

Table S7. TD-DFT (\$: CAM-B3LYP/6-311++G** and #: B3LYP/6-311++G**) absorption maxima ($\lambda^{\text{a}}_{\text{max}}$), excitation energies (E_{exc}), oscillator strengths (f) and contribution (%) to the transition in DCM of KK, KE and EE tautomers of compound **5c**.

Molecule	Tautomer	Excited States		$\lambda^{\text{a}}_{\text{max}}$ (nm)		E _{exc} (eV)		f		Contribution (%)	
		\$	#	\$	#	\$	#	\$	#	\$	#
KK	KK	1	1	759.60		1.63		0.86		99%	
		1	532.61		2.32		0.72		H-1 → L	H → L	
		3	3	593.75		2.08		0.32		99%	H-2 → L
		3	18	426.02	333.56	2.91	3.71	1.33	0.51	58%	H → L + 4
										27%	28%
	KE	9	24	311.70	316.43	3.97	3.91	1.48	0.21	H-1 → L + 1	H → L + 7
										17%	18%
	KE	15	25	280.55	316.40	4.41	3.91	0.32	0.26	H-1 → L + 13	H-1 → L + 7
										7%	18%
										H-1 → L + 10	H → L + 8
5c	KE	1	1	865.60		1.43		0.77		92%	
		1	599.40		2.06		0.56		H-2 → L	H → L	
		3	3	653.93		1.89		0.16		98%	H-2 → L
										26%	
		6	11	329.57	400.65	3.76	3.09	1.39	0.24	H-4 → L	H-1 → L + 1
	EE									7%	
										H → L	
		11		305.25		4.06		0.37		15%	
										H-1 → L	
				28		315.22		3.93		15%	
	EE	13		282.75		4.38		0.33		H-14 → L	
										23%	
										H → L + 13	
										12%	
		1	1	981.33	1205.0	1.26	1.02	0.80	1.51	H → L + 18	
	EE			9						62%	
										H → L	
				3		903.99		1.37		48%	
		3		597.41		2.07		1.38		H-2 → L	
		5		514.58		2.40		0.53		34%	
	EE									H → L	
										64%	
										H → L + 2	
		5	24	403.68	345.30	3.07	3.59	0.48	0.65	51%	H → L + 3
										66%	
	EE									86%	
										H → L + 4	
		14	29	308.08	320.16	4.02	3.87	1.05	0.46	37%	
	EE									H-1 → L + 1	
										8%	
	EE									H-1 → L + 7	
											62%
											H-3 → L + 1

Table S8 (to be continued). TD-DFT (\$: CAM-B3LYP/6-311++G** and #: B3LYP/6-311++G**) absorption maxima ($\lambda^{\text{a}} \text{ max}$), excitation energies (E_{exc}), oscillator strengths (f) and contribution (%) to the transition in DCM of KK, KE and EE tautomers of compound **DF90**.

Molecule	Tautomer	Excited States		$\lambda^{\text{a}} \text{ max (nm)}$		$E_{\text{exc}} (\text{eV})$		f		Contribution (%)	
		\$	#	\$	#	\$	#	\$	#	\$	#
KK	1	1		856.55		1.44		0.32		99%	
		1		542.90		2.28		0.66		H \rightarrow L	
		2		645.14		1.92		0.41		97%	
	3	10		413.86	405.06	2.99	3.06	1.03	0.42	67%	
										78%	
										H \rightarrow L+2	
	12	26		297.69	315.36	4.16	3.93	0.23	0.23	22%	
										H \rightarrow L+1	
										9%	
	12	26								H \rightarrow L+4	92%
										8%	H \rightarrow L+8
DF90	1	1		881.81		1.40		0.43		99%	
		1		582.82		2.12		0.55		H \rightarrow L	
		2		691.58		1.79		0.62		97%	
	4	10		406.29		3.05		0.38		86%	
		4		405.74		3.05		0.72		H \rightarrow L+1	
		11		399.63		3.10		0.17		88%	
	11	25		300.23	314.00	4.12	3.95	0.40	0.22	26%	
										H \rightarrow L+1	
										21%	
										H \rightarrow L+2	93%
											H \rightarrow L+7

Table S8 (continued). TD-DFT (\$: CAM-B3LYP/6-311++G** and #: B3LYP/6-311++G**) absorption maxima (λ^a max), excitation energies (E_{exc}), oscillator strengths (f) and contribution (%) to the transition in DCM of KK, KE and EE tautomers of compound **DF90**.

Molecule	Tautomer	Excited States		λ^a max (nm)		E_{exc} (eV)		f		Contribution (%)	
		\$	#	\$	#	\$	#	\$	#	\$	#
KE2	KE2		1			1019.03		1.21		0.42	
		1	2	633.30	724.02	1.96	1.71	0.57	0.11	H-1 → L	56%
			3			651.60		1.90		0.18	
		5	10	393.19	414.89	3.15	2.99	0.98	0.67	H-3 → L	46%
											100% H → L
	KE2	11		307.49		4.03		0.22			H → L+1
			25			318.42		3.89		0.22	H-9 → L
		14		281.77		4.40		0.29			18%
											86% H → L+2
											21%
DF90	DF90	1		1170.17		1.05		1.20			84% H → L+7
		1	2	981.47	944.89	1.26	1.31	0.75	0.18	H → L	51%
											95% H → L
		3	5	620.10	571.95	1.99	2.17	1.09	0.72	H-1 → L	40%
											92% H → L
	EE	4	9	457.18	449.22	2.71	2.76	0.84	0.29	H-3 → L	39%
			26		315.34		3.93		0.24	H-11 → L	54%
		18		294.14		4.21		0.19		H → L+4	15%
			28		309.56		4.00		0.15	H-14 → L	9%
											80% H → L+6

Table S9. TD-DFT (\$: CAM-B3LYP/6-311++G** and #: B3LYP/6-311++G**) absorption maxima ($\lambda^{\text{a}} \text{ max}$), excitation energies (E_{exc}), oscillator strengths (f) and contribution (%) to the transition in DCM of KK tautomer of compounds **5d** and **6d**.

Molecule	Tautomer	Excited States		$\lambda^{\text{a}} \text{ max (nm)}$		$E_{\text{exc}} (\text{eV})$		f		Contribution (%)	
		\$	#	\$	#	\$	#	\$	#	\$	#
5d	KK	1	1	532.68	746.44	2.32	1.62	0.69	0.76	93% $H-2 \rightarrow L$	99% $H \rightarrow L$
		3	3	419.97	595.84	2.95	2.08	1.35	0.36	74% $H \rightarrow L$	99% $H-2 \rightarrow L$
		7	10	436.81	315.62	3.92	2.83	1.45	0.62	20% $H-1 \rightarrow L+1$	93% $H-6 \rightarrow L$
										2% $H-1 \rightarrow L+3$	91% $H-6 \rightarrow L$
				410.25		3.02		0.50			
		12	16	305.43	345.48	4.05	3.58	0.60	0.51	32% $H-6 \rightarrow L$	47% $H-8 \rightarrow L$
										17% $H-1 \rightarrow L+1$	
										40% $H \rightarrow L+5$	42% $H \rightarrow L+5$
		13	20	299.98	339.94	4.13	3.64	0.21	0.18	29% $H-1 \rightarrow L+6$	15% $H-1 \rightarrow L+6$
		14	21	339.85	299.89	4.13	3.64	0.78	0.78	42% $H \rightarrow L+6$	51% $H \rightarrow L+6$
										27% $H-1 \rightarrow L+5$	27% $H-1 \rightarrow L+6$
										38% $H-16 \rightarrow L$	45% $H \rightarrow L+2$
		15	22	338.85	299.89	4.13	3.65	0.11	0.11	12% $H-7 \rightarrow L+1$	13% $H-8 \rightarrow L$
6d	KK	1	3	429.29	501.13	2.88	1.59	1.05	0.59	70% $H-4 \rightarrow L$	99% $H \rightarrow L$
											96% $H-4 \rightarrow L$
		7	21	349.88	317.14	3.90	3.54	1.31	0.71	25% $H-1 \rightarrow L+1$	60% $H \rightarrow L+2$
											52% $H \rightarrow L+6$

Spectroscopic and Electrochemical Measurements

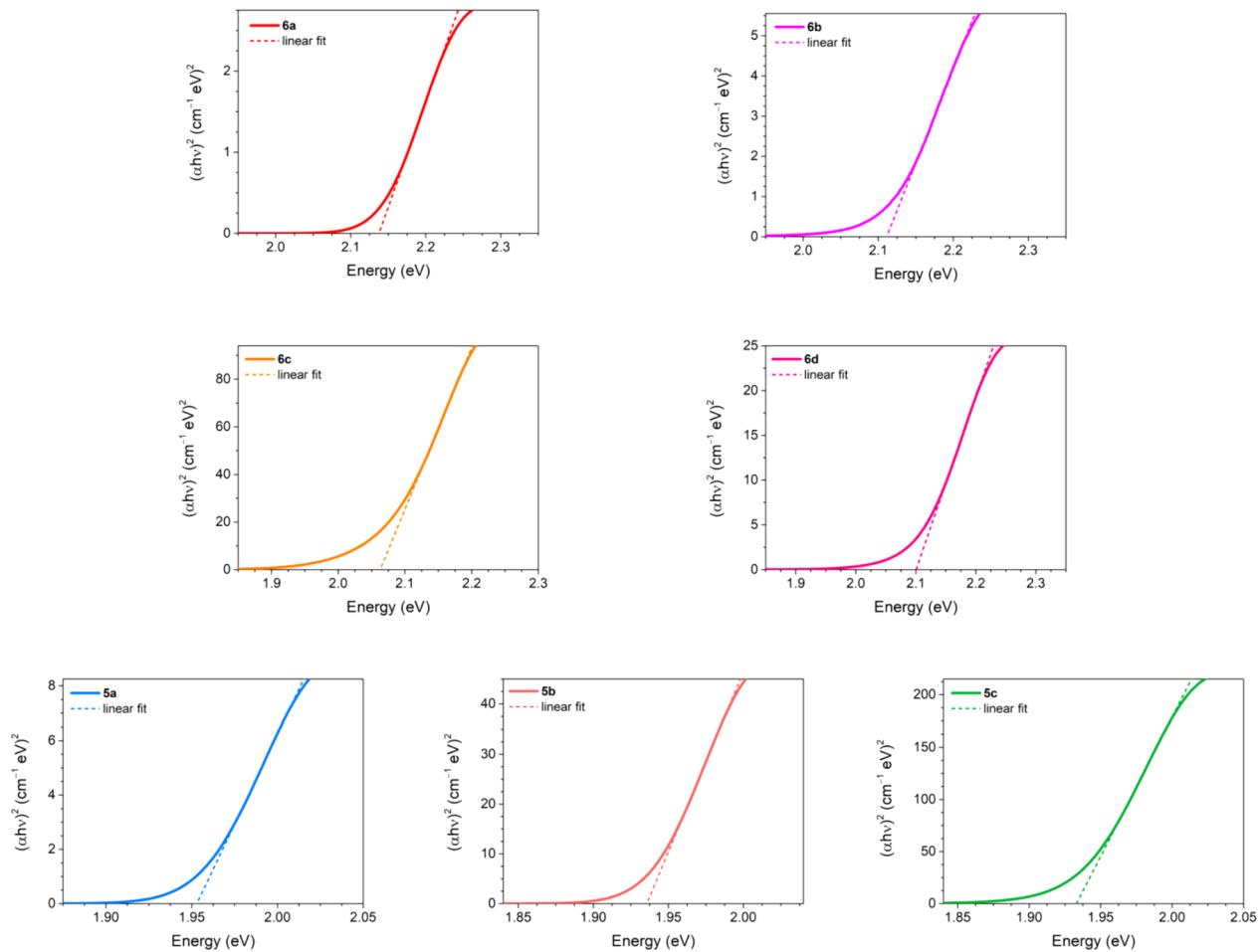


Figure S6. Tauc plots for the CH_2Cl_2 solutions of compounds **6a-d** and **5a-c**.

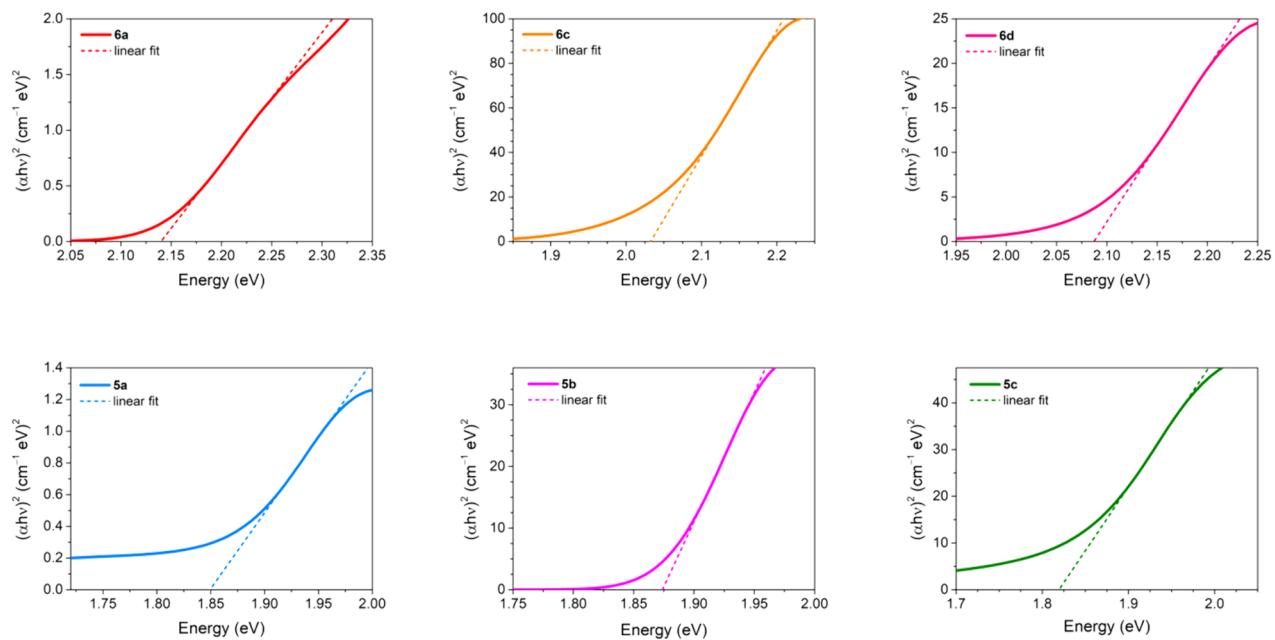
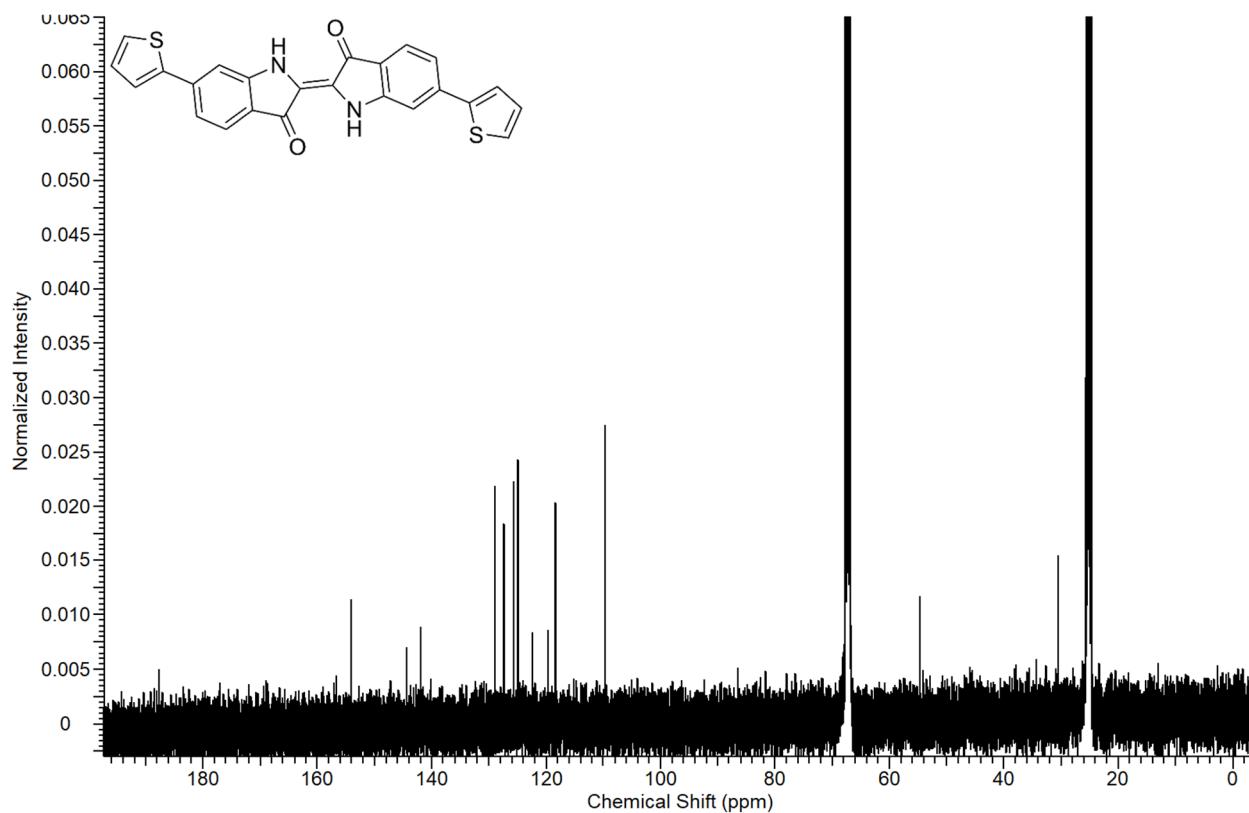
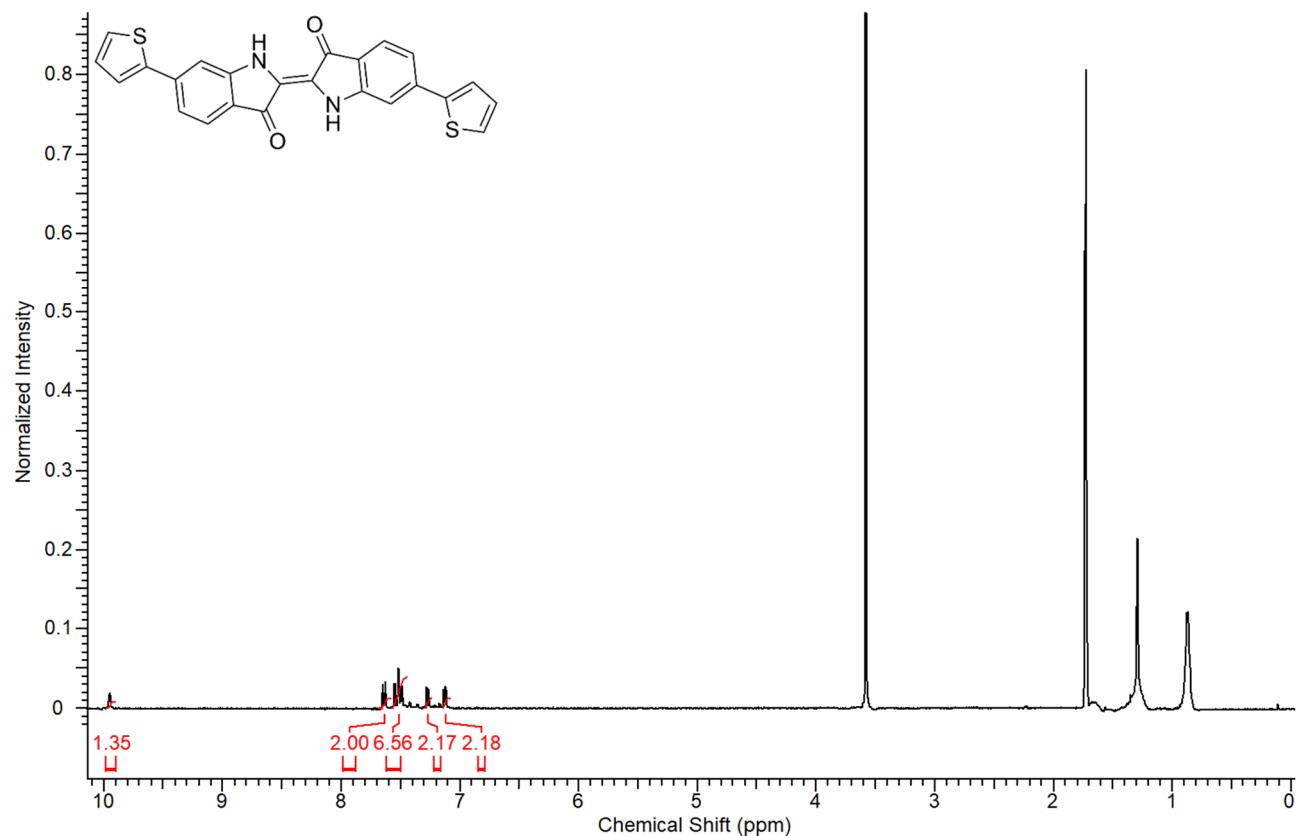


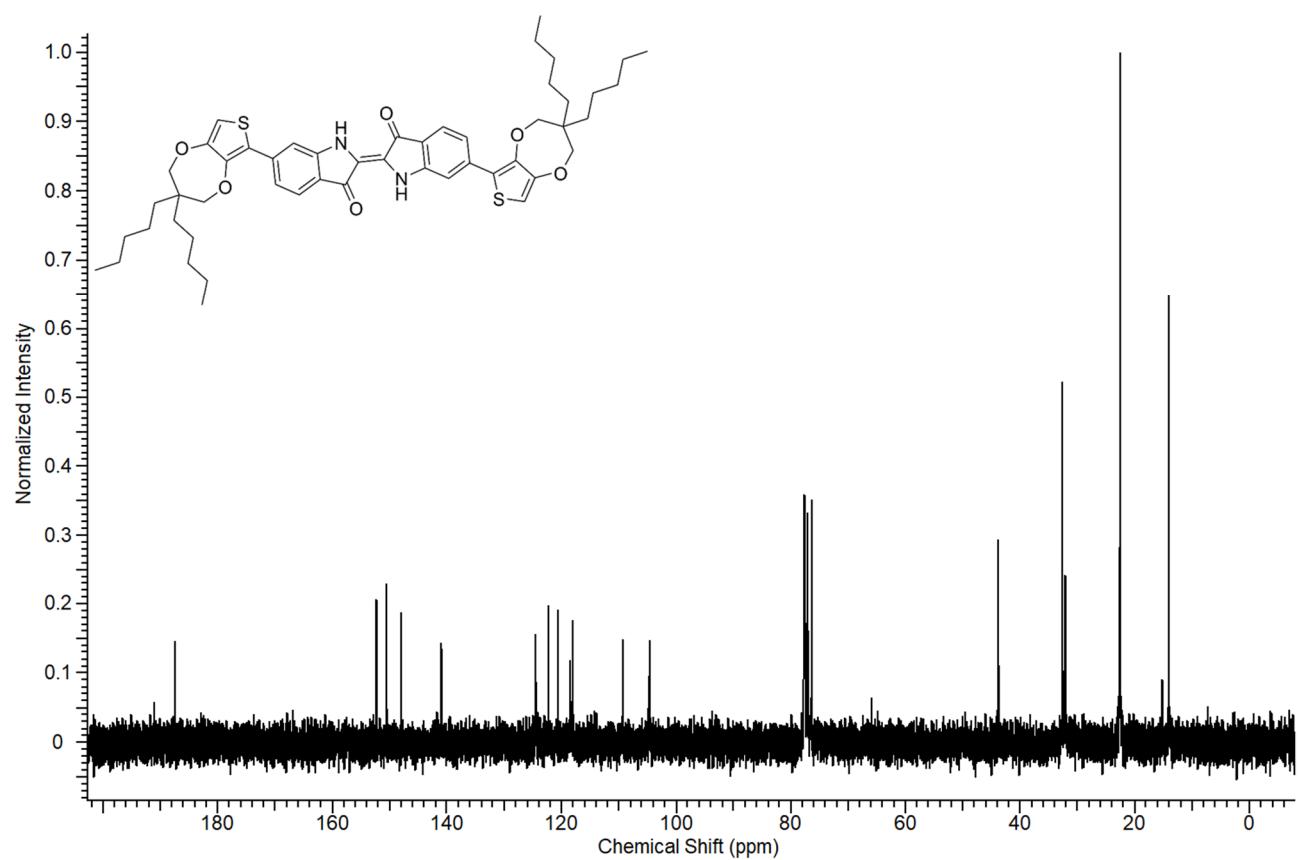
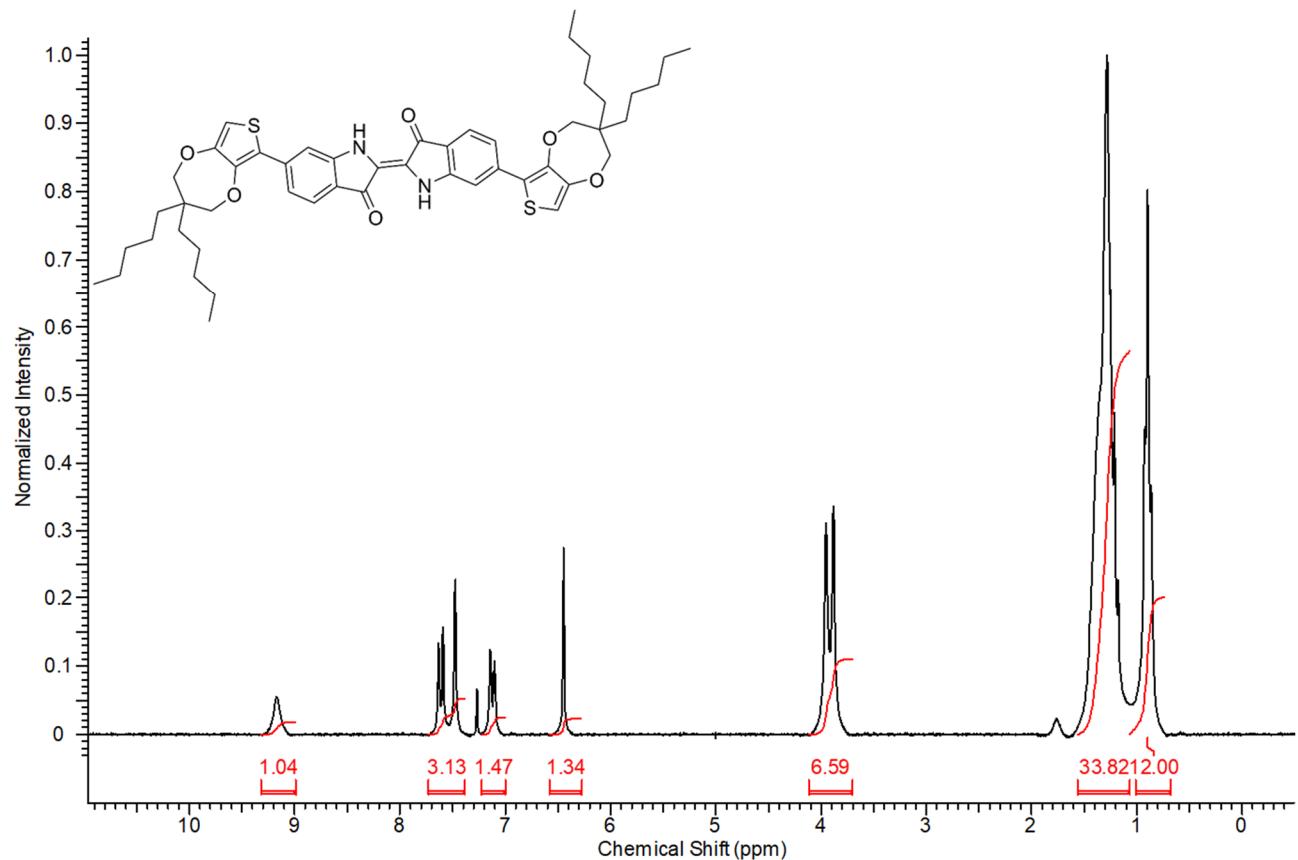
Figure S7. Tauc plots for the EtOH solutions of compounds **6a,c,d** and **5a-c**.

Copies of the NMR spectra of compounds 5a, 5b, 5c and DF90

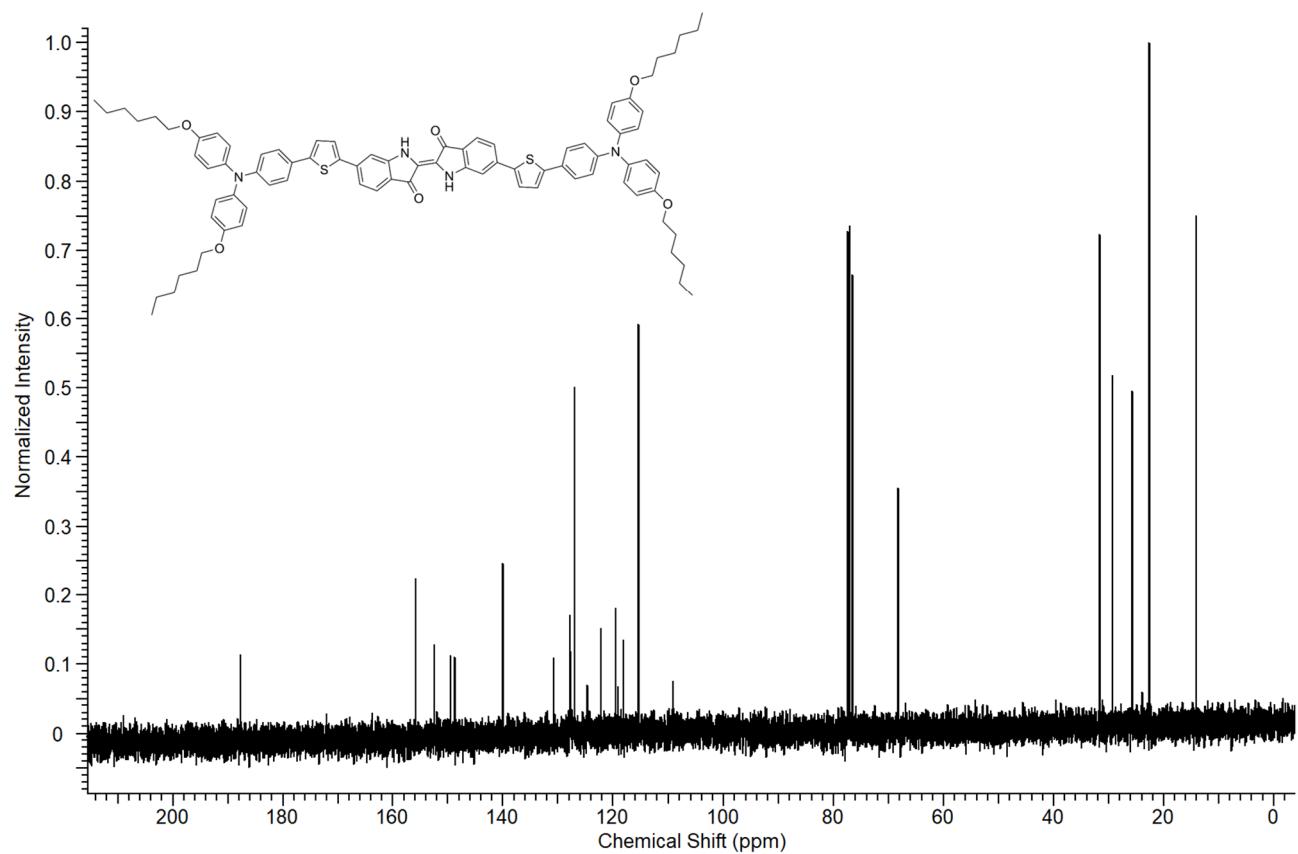
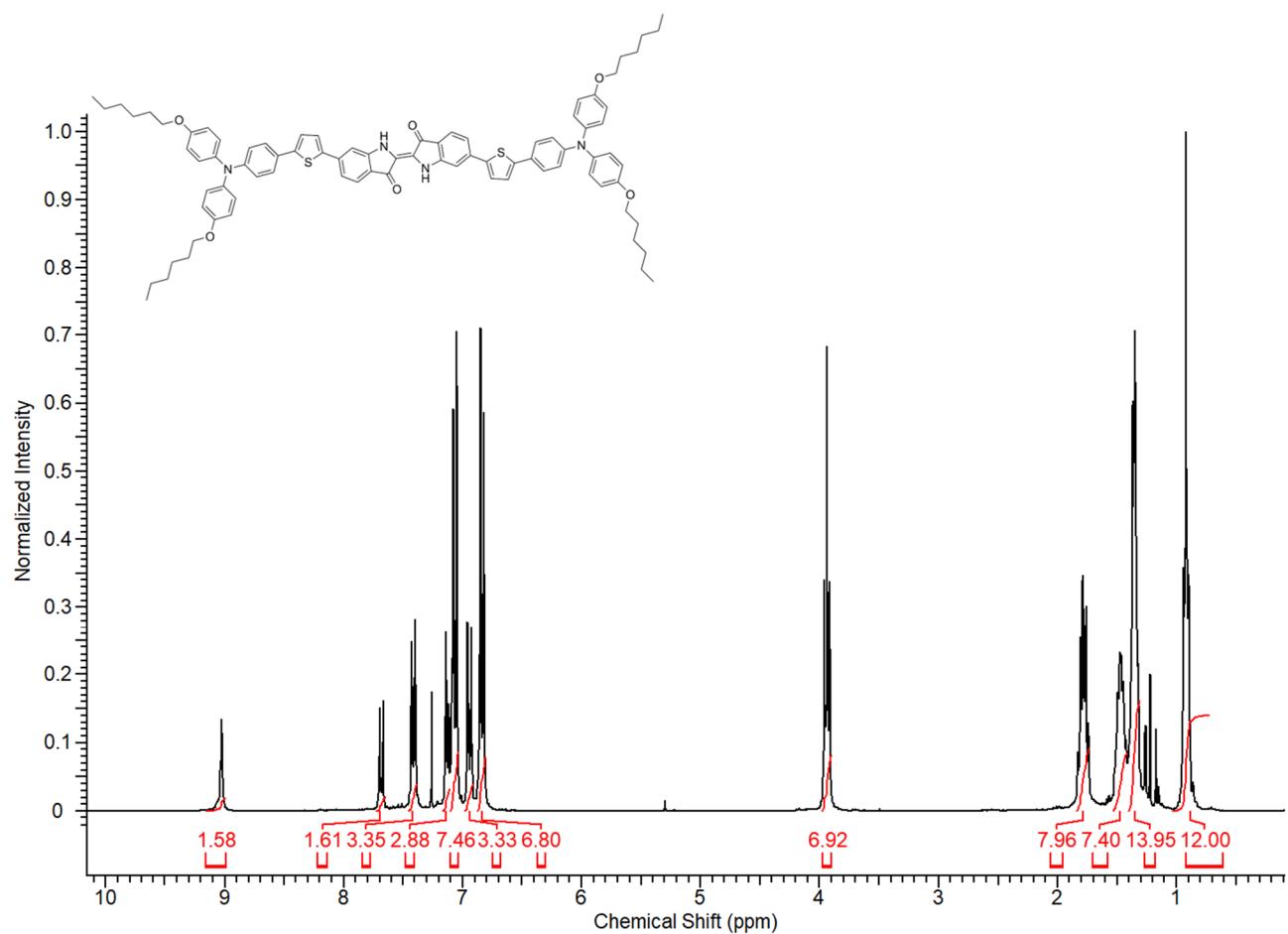
Compound 5a



Compound **5b**



Compound 5c



Compound DF90

