## Article

## A Reliable Enantioselective Route to Mono-Protected N1-Cbz Piperazic Acid Building Block

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## Optimization of the mono-deprotection reaction.

Table S1. Full optimization study of the selective deprotection reaction.


| Entry. | Starting <br> material | Base <br> (equiv) | Additives <br> (equiv) | Solvent <br> $(0.26 \mathrm{M})$ | Temperature <br> $\left({ }^{\circ} \mathrm{C}\right)$ | Time <br> (h) | Full consump- <br> tion of 13 (iso- <br> lated yield of 5) |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | 0.13 mmol <br> $(50 \mathrm{mg})$ | NaOH <br> $(1.2)$ | - | THF | 23 | 1 | - |
| 2 | 0.13 mmol <br> $(50 \mathrm{mg})$ | NaOH <br> $(1.2)$ | - | THF | 23 | 2 | - |
| 3 | 0.13 mmol <br> $(50 \mathrm{mg})$ | NaOH <br> $(1.2)$ | - | THF | 23 | $2^{1 / 2}$ | - |
| 4 | 0.13 mmol <br> $(50 \mathrm{mg})$ | NaOH <br> $(1.2)$ | - | THF | 23 | 18 | - |
| 5 | 0.13 mmol <br> $(50 \mathrm{mg})$ | NaOH <br> $(2.0)$ | - | THF | 23 | 1 | - |
| 6 | 0.13 mmol <br> $(50 \mathrm{mg})$ | NaOH <br> $(2.0)$ | - | THF | 23 | 2 | - |
| 7 | 0.13 mmol <br> $(50 \mathrm{mg})$ | NaOH <br> $(2.0)$ | - | THF | 23 | $2^{1 / 2}$ | - |
| 8 | 0.13 mmol <br> $(50 \mathrm{mg})$ | NaOH <br> $(2.0)$ | - | THF | 23 | 18 | - |
| 9 | 0.13 mmol <br> $(50 \mathrm{mg})$ | KOH <br> $(1.2)$ | - | THF | 23 | 1 | - |
| 10 | 0.13 mmol <br> $(50 \mathrm{mg})$ | KOH <br> $(1.2)$ | - | THF | 23 | 2 | - |
| 11 | 0.13 mmol <br> $(50 \mathrm{mg})$ | KOH <br> $(1.2)$ | - | THF | 23 | $2^{1 / 2}$ | - |
| 12 | 0.13 mmol <br> $(50 \mathrm{mg})$ | KOH <br> $(1.2)$ | - | THF | 23 | 18 | - |
| 13 | 0.13 mmol <br> $(50 \mathrm{mg})$ | KOH <br> $(2.0)$ | - | THF | 23 | 1 | - |
| 14 | 0.13 mmol <br> $(50 \mathrm{mg})$ | KOH <br> $(2.0)$ | - | THF | 23 | 2 | - |
| 15 | 0.13 mmol | KOH | - | THF | 23 | $2^{1 / 2}$ | - |


|  | (50mg) | (2.0) |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 16 | $\begin{gathered} 0.13 \mathrm{mmol} \\ (50 \mathrm{mg}) \\ \hline \end{gathered}$ | $\begin{gathered} \hline \mathrm{KOH} \\ (2.0) \\ \hline \end{gathered}$ | - | THF | 23 | 18 | - |
| 17 | $\begin{gathered} 0.13 \mathrm{mmol} \\ (50 \mathrm{mg}) \\ \hline \end{gathered}$ | $\begin{gathered} \hline \mathrm{KOH} \\ (2.0) \\ \hline \end{gathered}$ | - | MeOH | 23 | 1 | - |
| 18 | $\begin{gathered} 0.13 \mathrm{mmol} \\ (50 \mathrm{mg}) \end{gathered}$ | $\begin{gathered} \mathrm{KOH} \\ (2.0) \end{gathered}$ | - | MeOH | 23 | 2 | - |
| 19 | $\begin{gathered} 0.13 \mathrm{mmol} \\ (50 \mathrm{mg}) \end{gathered}$ | $\begin{gathered} \mathrm{KOH} \\ (2.0) \\ \hline \end{gathered}$ | - | MeOH | 23 | $2^{1 / 2}$ | - |
| 20 | $\begin{gathered} 0.13 \mathrm{mmol} \\ (50 \mathrm{mg}) \\ \hline \end{gathered}$ | $\begin{gathered} \mathrm{KOH} \\ (2.0) \\ \hline \end{gathered}$ | - | MeOH | 23 | 18 | - |
| 21 | $\begin{gathered} 0.13 \mathrm{mmol} \\ (50 \mathrm{mg}) \\ \hline \end{gathered}$ | $\begin{gathered} \mathrm{NaOH} \\ (2.0) \\ \hline \end{gathered}$ | - | MeOH | 23 | 1 | - |
| 22 | $\begin{gathered} 0.13 \mathrm{mmol} \\ (50 \mathrm{mg}) \\ \hline \end{gathered}$ | NaOH <br> (2.0) | - | MeOH | 23 | 2 | - |
| 23 | $\begin{gathered} \hline 0.13 \mathrm{mmol} \\ (50 \mathrm{mg}) \\ \hline \end{gathered}$ | $\begin{gathered} \mathrm{NaOH} \\ (2.0) \\ \hline \end{gathered}$ | - | MeOH | 23 | $2^{1 / 2}$ | - |
| 24 | $\begin{gathered} 0.13 \mathrm{mmol} \\ (50 \mathrm{mg}) \end{gathered}$ | $\begin{gathered} \mathrm{NaOH} \\ (2.0) \end{gathered}$ | - | MeOH | 23 | 18 | - |
| 25 | $\begin{gathered} 0.13 \mathrm{mmol} \\ (50 \mathrm{mg}) \\ \hline \end{gathered}$ | $\begin{gathered} \mathrm{KOH} \\ (2.0) \\ \hline \end{gathered}$ | - | THF | 45 | 18 | (45\%) |
| 26 | $\begin{gathered} 0.13 \mathrm{mmol} \\ (50 \mathrm{mg}) \\ \hline \end{gathered}$ | $\begin{gathered} \mathrm{NaOH} \\ (2.0) \\ \hline \end{gathered}$ | - | THF | 45 | 18 | $(72 \%)$ |
| 27 | $\begin{gathered} 0.13 \mathrm{mmol} \\ (50 \mathrm{mg}) \\ \hline \end{gathered}$ | $\begin{gathered} \mathrm{NaOH} \\ (3.0) \\ \hline \end{gathered}$ | - | THF | 45 | 18 | (60\%) |
| 28 | $\begin{gathered} 0.13 \mathrm{mmol} \\ (50 \mathrm{mg}) \end{gathered}$ | $\begin{gathered} \hline \mathrm{NaOH} \\ (3.0) \end{gathered}$ | - | THF | 23 | 18 | - |
| 29 | $\begin{gathered} 0.13 \mathrm{mmol} \\ (50 \mathrm{mg}) \end{gathered}$ | $\begin{gathered} \mathrm{Rb}_{2} \mathrm{CO}_{3} \\ (2.0) \end{gathered}$ | - | THF | 45 | 18 | - |
| 30 | $\begin{gathered} 0.13 \mathrm{mmol} \\ (50 \mathrm{mg}) \\ \hline \end{gathered}$ | $\begin{gathered} \mathrm{Rb}_{2} \mathrm{CO}_{3} \\ (3.0) \\ \hline \end{gathered}$ | - | THF | 45 | 18 | - |
| 31 | $\begin{gathered} 0.13 \mathrm{mmol} \\ (50 \mathrm{mg}) \\ \hline \end{gathered}$ | $\begin{gathered} \hline \mathrm{LiOH} \\ (2.0) \\ \hline \end{gathered}$ | - | THF | 23 | 18 | - |
| 32 | $\begin{gathered} 0.13 \mathrm{mmol} \\ (50 \mathrm{mg}) \\ \hline \end{gathered}$ | $\begin{gathered} \hline \mathrm{LiOH} \\ (2.0) \\ \hline \end{gathered}$ | - | THF | 45 | 18 | - |
| 33 | $\begin{gathered} \hline 0.13 \mathrm{mmol} \\ (50 \mathrm{mg}) \\ \hline \end{gathered}$ | $\begin{gathered} \hline \mathrm{NaH} \\ (1.1) \end{gathered}$ | - | THF | 0 to 23 | 2 | - |
| 34 | $\begin{gathered} 0.13 \mathrm{mmol} \\ (50 \mathrm{mg}) \\ \hline \end{gathered}$ | $\begin{gathered} \hline \mathrm{NaH} \\ (1.1) \end{gathered}$ | - | THF | 0 to 23 | 18 | - |
| 35 | $\begin{gathered} 0.13 \mathrm{mmol} \\ (50 \mathrm{mg}) \end{gathered}$ | $\begin{gathered} \hline \mathrm{NaOH} \\ (2.0) \end{gathered}$ | - | THF | 23 | 18 | - |
| 36 | $\begin{gathered} 0.13 \mathrm{mmol} \\ (50 \mathrm{mg}) \\ \hline \end{gathered}$ | $\begin{gathered} \mathrm{NaOH} \\ (2.0) \\ \hline \end{gathered}$ | Crown ether (1.1) | THF | 23 | 18 | - |
| 37 | $\begin{gathered} 0.13 \mathrm{mmol} \\ (50 \mathrm{mg}) \\ \hline \end{gathered}$ | $\begin{gathered} \mathrm{NaOH} \\ (2.0) \\ \hline \end{gathered}$ | Crown ether (1.1) | THF | 45 | 18 | $(35 \%)$ |
| 38 | $\begin{gathered} 0.13 \mathrm{mmol} \\ (50 \mathrm{mg}) \end{gathered}$ | $\begin{gathered} \hline \text { KOH } \\ (2.0) \end{gathered}$ | Crown ether (1.1) | THF | 23 | 18 | - |
| 39 | $\begin{gathered} 0.13 \mathrm{mmol} \\ (50 \mathrm{mg}) \\ \hline \end{gathered}$ | $\begin{gathered} \mathrm{KOH} \\ (2.0) \\ \hline \end{gathered}$ | Crown ether (1.1) | THF | 45 | 18 | $30 \%$ |
| 40 | $\begin{gathered} \hline 0.26 \mathrm{mmol} \\ (100 \mathrm{mg}) \end{gathered}$ | $\begin{gathered} \mathrm{NaOH} \\ (2.0) \end{gathered}$ | - | THF | 45 | 18 | (61\%) |
| 41 | $\begin{gathered} 1.30 \mathrm{mmol} \\ (500 \mathrm{mg}) \end{gathered}$ | $\begin{gathered} \mathrm{NaOH} \\ (2.0) \end{gathered}$ | - | THF | 45 | 18 | (70\%) |
| 42 | $\begin{aligned} & 2.6 \mathrm{mmol} \\ & (1000 \mathrm{mg}) \end{aligned}$ | $\begin{gathered} \mathrm{NaOH} \\ (2.0) \\ \hline \end{gathered}$ | - | THF | 45 | 18 | $(60 \%)$ |



Figure S1. (A): Monitoring the reactions by TLC analysis. In lanes 2 and 4 where the sm is fully consumed the product was purified. Correlation between TLC lanes and Table S1: Lane $1 \rightarrow$ Entry 7, Lane $2 \rightarrow$ Entry 26, Lane $3 \rightarrow$ Entry 15, Lane $4 \rightarrow$ Entry 25, Lane $5 \rightarrow$ Entry 23, Lane $6 \rightarrow$ Entry 24, Lane $S \rightarrow$ starting material 13. (B): Reaction setup in a sand bath at $45^{\circ} \mathrm{C}$. (C): Reaction setup at ambient temperature.

## Synthetic protocols and characterization data

1-((Benzyloxy)carbonyl)hexahydropyridazine-3-carboxylic acid (5).


In a screw cap vial containing 13 ( $50 \mathrm{mg}, 0.13 \mathrm{mmol}, 1$ equiv.) in THF ( 0.5 mL ) was added solid NaOH ( $10 \mathrm{mg}, 0.26 \mathrm{mmol}, 2$ equiv.) and the mixture was heated to $45^{\circ} \mathrm{C}$ and left stirring for 18 h . After cooling to room temperature, the solvent was evaporated and the residue was dissolved in $\mathrm{H}_{2} \mathrm{O}(5 \mathrm{~mL})$, transferred to a separatory funnel and washed with $\mathrm{Et}_{2} \mathrm{O}(5 \mathrm{~mL})$. The organic layer was discarded to remove the byproduct [Figure S1(A), upper spot on TLC] and the aqueous phase was acidified with HCl 1 N to $\mathrm{pH} 4-5$ and then extracted thoroughly with ethyl acetate ( $4 \times 5 \mathrm{~mL}$ ). The combined organic layers were washed with brine, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, filtered and concentrated in vacuo to afford piperazic acid 5 ( $25 \mathrm{mg}, 72 \%$ ) as a white solid.
$R_{f} 0.2$ (9:1 chloroform/methanol, PMA stain). mp $158-160^{\circ} \mathrm{C}$, (Lit. ${ }^{3} \mathrm{mp} 166-167{ }^{\circ} \mathrm{C}$ ). For (R)-5 we found: $[\alpha]_{D}{ }^{23}=+22(c 1, \mathrm{MeOH}),\left\{\left[\mathrm{Lit}^{3}{ }^{3}[\alpha]_{D}{ }^{20}=-35(c 1, \mathrm{MeOH})\right.\right.$ for $(S)$-enantiomer $\} ;{ }^{1} \mathrm{H}-\mathrm{NMR}(400$ $\left.\mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right) \delta 7.39-7.29(\mathrm{br}, 5 \mathrm{H})$, $5.16(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 3.85(\mathrm{br}, 1 \mathrm{H}), 3.51-3.48(\mathrm{~m}, 1 \mathrm{H}), 3.35(\mathrm{br}, 1 \mathrm{H})$, $2.01(\mathrm{br}, 1 \mathrm{H}), 1.79-1.60(\mathrm{br}, 3 \mathrm{H}){ }^{13} \mathrm{C}$ NMR (101 MHz, CD ${ }_{3} \mathrm{OD}$ ) $\delta 174.7,157.5,137.9,129.5,129.3$, 129.2, 129.0, 128.3, 127.99, 68.7, 65.2, 59.3, 45.7, 28.4, 24.2. MS (ESI) m/z (\%) :265.2 [M+H, (100)] ${ }^{+}$.

## Synthesis of aldehyde 7.



To a stirring solution of 1,5-pentanediol ( $2 \mathrm{~g}, 19.20 \mathrm{mmol}, 1$ equiv.) in toluene ( 58 mL ) was added HBr ( $48 \%$ aq., $2.6 \mathrm{~mL}, 23.04 \mathrm{mmol}, 1.2$ equiv.) and the mixture was heated at reflux for 24 hours. After cooling down to ambient temperature, the mixture was transferred to a separatory funnel and the layers were separated. The organic phase was washed with 1 N NaOH (aq., 15 mL ) and brine ( 30 mL ), dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, filtered and concentrated carefully at the rotavap (ATTENTION: alcohol $\mathbf{S 2}$ is volatile). The crude product was purified by FCC (pentane/Et ${ }_{2} \mathrm{O}$ 1:1 to 1:1.5) to furnish $\mathbf{S 2}(1.91 \mathrm{~g}, 60 \%)$ as a colorless liquid. NMR data matched those in the literature. ${ }^{1}$
 $\mathrm{Hz}, 2 \mathrm{H}), 3.42(\mathrm{t}, \mathrm{J}=6.7 \mathrm{~Hz}, 2 \mathrm{H}), 1.97-1.83(\mathrm{~m}, 2 \mathrm{H}), 1.67-1.43(\mathrm{~m}, 4 \mathrm{H})$.

To a stirring mixture of PCC ( $2.94 \mathrm{~g}, 13.65 \mathrm{mmol}, 1.2$ equiv.) and Florisil $(11 \mathrm{~g})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ( 35 $\mathrm{mL})$ at ambient temperature was added slowly via addition funnel a solution of alcohol $\mathbf{S 2}(1.90 \mathrm{~g}$, 11.37 mmol, 1 equiv.) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(20 \mathrm{~mL})$ and the reaction was monitored by TLC. After 2.5 hours the reaction was filtered through a pad of Celite and concentrated carefully at the rotavap (ATTENTION: aldehyde $\mathbf{7}$ is volatile). The crude product was purified by FCC (pentane/ $\mathrm{Et}_{2} \mathrm{O} 7: 3$ ) to furnish $7(1.32 \mathrm{~g}, 71 \%)$ as a colorless liquid. NMR data matched those in the literature. ${ }^{2}$
$R_{f} 0.6$ (7:3 pentane/Et ${ }_{2} \mathrm{O}, \mathrm{CAM}$ stain). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(200 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 9.73(\mathrm{t}, \mathrm{J}=1.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.37(\mathrm{t}, \mathrm{J}=$ $6.2,2 \mathrm{H}), 2.45(\mathrm{td}, J=7.0$ and $1.5 \mathrm{~Hz}, 2 \mathrm{H}), 1.88-1.64(\mathrm{~m}, 4 \mathrm{H}) ;{ }^{13} \mathrm{C} \mathrm{NMR}\left(50 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 201.8,42.8$, 33.1, 31.9, 20.6.

Synthesis of (S)-5 following the protocol by Ma and co-workers. ${ }^{3}$


To a stirring solution of aldehyde $7\left(1.30 \mathrm{~g}, 7.87 \mathrm{mmol}, 1.8\right.$ equiv.) in dry $\mathrm{CH}_{2} \mathrm{Cl}_{2}(11 \mathrm{~mL})$ at $0{ }^{\circ} \mathrm{C}$ was added $D$-proline ( $91 \mathrm{mg}, 0.78 \mathrm{mmol}, 0.18$ equiv.) followed by dibenzylazodicarboxylate ( $1.30 \mathrm{~g}, 4.37 \mathrm{mmol}, 1$ equiv.) and the mixture was left stirring at the same temperature for 15 hours. TLC analysis showed full consumption of the starting material (petroleum ether $40-60{ }^{\circ} \mathrm{C}$ /EtOAc 9:1, PMA stain). Sulfamic acid ( $764 \mathrm{mg}, 7.87 \mathrm{mmol}, 1.8$ equiv.) was added in one portion followed by $\mathrm{NaClO}_{2}\left(1.13 \mathrm{M}, 533 \mathrm{mg} 80 \% \mathrm{NaClO}_{2}\right.$ in 5.2 mL H O ) slowly while the color of the reaction turns green. After 10 minutes (TLC analysis showed disappearance of the intermediate aldehyde, petroleum ether $40-60^{\circ} \mathrm{C} / E t O A c 9: 1$, PMA stain) the reaction was quenched by addition of $\mathrm{Na}_{2} \mathrm{SO}_{3}$ (aq. saturated) and stirred for 1 hour at room temperature. Then, the mixture was transferred to a separatory funnel and the layers were separated. The aqueous layer was washed with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(2 \times 10 \mathrm{~mL})$ and the combined organic layers were washed with brine $(30 \mathrm{~mL})$, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, filtered and concentrated in vacuo. The crude product was purified by FCC (petroleum ether $40-60{ }^{\circ} \mathrm{C} / E t O A c 1: 1$ to remove the byproduct and then neat EtOAc) to furnish 4 ( 1.37 g , $66 \%$ ) as a white solid.


Figure S2. TLC analysis of the reaction sequence. A) TLC (petroleum ether $40-60{ }^{\circ} \mathrm{C} / E t O A c 9: 1$ ) of the organocatalytic reaction after 15 hours at $0^{\circ} \mathrm{C}$. In the left lane we have the starting material ( sm ), in the middle lane the co-spot of sm and reaction mixture (rm), and in the right lane the rm. We can see full consumption of sm. B) TLC (petroleum ether $40-60^{\circ} \mathrm{C} / E t O A c 9: 1$ ) after the Pinnick oxidation. C) A more polar TLC after the Pinnick oxidation (petroleum ether $40-60^{\circ} \mathrm{C} / E t O A c 1: 1$ ). The product 4 is the lower spot that "tails".

To a stirring solution of acid 4 ( $1.30 \mathrm{~g}, 2.71 \mathrm{mmol}, 1$ equiv.) in THF ( 9 mL ) at $0{ }^{\circ} \mathrm{C}$ was added solid NaOH ( $217 \mathrm{mg}, 5.42 \mathrm{mmol}, 2$ equiv.) and the mixture was stirred at the same temperature for 24 hours. Then, the reaction was allowed to reach ambient temperature and was left stirring for an additional 4 hours. Subsequently, NaOH 1 N (aq., 3 mL ) and sat. $\mathrm{NaHCO}_{3}$ (aq., 4 mL ) were added, the mixture was transferred to a separatory funnel and the layers were separated. The organic layer was washed with sat. $\mathrm{NaHCO}_{3}(5 \mathrm{~mL})$ and discarded. The aqueous layer was washed with petroleum ether $40-60^{\circ} \mathrm{C}(2 \times 5 \mathrm{~mL})$ and then was acidified to $\mathrm{pH} 4-5$ with
$\mathrm{HCl} 37 \%$. The aqueous phase was extracted with EtOAc ( $3 \times 5 \mathrm{~mL}$ ), dried and concentrated in vacuo to furnish a mixture of bis- and mono-Cbz Piz ( 307 mg ) as colorless glue.

## Synthesis of $(R)$-5 following the global deprotection and selective mono-protection sequence. ${ }^{4}$



To a stirring solution of acid 13 ( $1.56 \mathrm{~g}, 3.92 \mathrm{mmol}, 1$ equiv.) in dichloromethane ( 159 mL ) was added $10 \% \mathrm{Pd} / \mathrm{C}(1.4 \mathrm{~g})$ and trifluoroacetic acid ( $2.9 \mathrm{~mL}, 39.2 \mathrm{mmol}, 10$ equiv). The suspension was stirred under a hydrogen atmosphere (balloon) at $23^{\circ} \mathrm{C}$ for 12 h . The mixture was then filtered through a pad of Celite to remove the catalyst, the filter cake was washed thoroughly with MeOH and the filtrate was concentrated in vacuo to give the trifluroacetic acid salt $\mathbf{2}$ as a slurry oil. The product was advanced directly to the next step.
${ }^{1} \mathrm{H}-\mathrm{NMR}\left(200 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right) \delta 3.98(\mathrm{br}, 1 \mathrm{H}), 3.41-3.13(\mathrm{~m}, 2 \mathrm{H}), 2.25-1.91(\mathrm{~m}, 4 \mathrm{H}) .{ }^{4 \mathrm{a}}$
To a mixture of the above salt 2 ( $3.92 \mathrm{mmol}, 1$ equiv.) and $\mathrm{NaOH}(470 \mathrm{mg}, 11.76 \mathrm{mmol}, 3.0$ equiv.) in water ( 11.5 mL ) was added a solution of benzyl chloroformate ( $0.56 \mathrm{~mL}, 3.92 \mathrm{mmol}, 1.0$ equiv.) in toluene ( 8 mL ) at $+10^{\circ} \mathrm{C}$. After being stirred for 15 hours at room temperature, the reaction mixture was transferred to a separatory funnel and the layers were separated. The aqueous layer was washed with $\mathrm{Et}_{2} \mathrm{O}(10 \mathrm{~mL})$, acidified with 1 N HCl to pH 4 and extracted with AcOEt ( $3 \times 15 \mathrm{~mL}$ ). The combined organic layers were washed with brine, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated in vacuo. The residue was purified by $\mathrm{FCC}\left(\mathrm{CHCl}_{3} / \mathrm{MeOH}: 9 / 1\right)$ to give the monoprotected acid in a $15 \%$ yield ( 150 mg ) as yellowish oil. ${ }^{4 \mathrm{~b}}$

## Synthesis of (R)-13 following the Hamada protocol. ${ }^{4 a}$



To a stirring solution of aldehyde $7\left(1.10 \mathrm{~g}, 6.67 \mathrm{mmol}, 1.5\right.$ equiv.) in dry $\mathrm{CH}_{3} \mathrm{CN}(32 \mathrm{~mL})$ at $0^{\circ} \mathrm{C}$ was added dibenzylazodicarboxylate $6(1.33 \mathrm{~g}, 4.45 \mathrm{mmol}, 1$ equiv.) followed by L-proline ( 51 $\mathrm{mg}, 0.45 \mathrm{mmol}, 0.1$ equiv.) and the mixture was stirred at the same temperature for 20 hours. Then, $\mathrm{NaBH}_{4}$ ( $168 \mathrm{mg}, 4.45 \mathrm{mmol}, 1$ equiv.) and EtOH ( 13 mL ) were added sequentially and the reaction was left stirring at $0^{\circ} \mathrm{C}$ for an additional 1 hour before it was quenched by slow addition of $5 \%$ citric acid (aq., $5-6 \mathrm{~mL}$ ). After stirring 5 minutes at ambient temperature the reaction was concentrated in vacuo, re-dissolved in EtOAc ( 30 mL ) and transferred to a separatory funnel. The organic phase was washed with brine $(30 \mathrm{~mL})$ and the aqueous phase was back-extracted with EtOAc ( $2 \times 20 \mathrm{~mL}$ ). The combined organic layers were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, filtered and concentrated. The crude product was purified by FCC (petroleum ether $40-60^{\circ} \mathrm{C} / E t O A c 6: 4$ ) to give alcohol 9 as a white solid.
$R_{f} 0.3$ ( $7: 3$ petroleum ether $\left.40-60^{\circ} \mathrm{C} / E t O A c, ~ P M A ~ s t a i n\right) . ~{ }^{1} \mathrm{H}-\mathrm{NMR}\left(200 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.32(\mathrm{br}, 10 \mathrm{H})$, $6.53(\mathrm{br}, 1 \mathrm{H}), 5.17(\mathrm{br}, 4 \mathrm{H}), 3.56-3.24(\mathrm{~m}, 4 \mathrm{H}), 1.90(\mathrm{br}, 1 \mathrm{H}), 1.60-1.41(\mathrm{br}, 4 \mathrm{H}) . \mathrm{MS}(\mathrm{ESI}) \mathrm{m} / \mathrm{z}(\%):$ $487.2[\mathrm{M}+\mathrm{Na},(100)]^{+}, 489.2[\mathrm{M}+\mathrm{Na},(80)]^{+}$. The enantiomeric ratio of $(S)-9$ was determined to be $>99 \%$ ee by chiral HPLC (CHIRALPAK ${ }^{\circledR}$ OD-H, hexane/iPrOH/TFA 90:10:0.1, $0.6 \mathrm{~mL} / \mathrm{min}, 254 \mathrm{~nm}$, $32.4 \mathrm{~min})$.

To a stirring solution of the above alcohol 9 ( 4.45 mmol , 1 equiv.) in dry DMF ( 52 mL ) was added imidazole ( $1.51 \mathrm{~g}, 22.25 \mathrm{mmol}, 5$ equiv.) followed by TBS $-\mathrm{Cl}(805 \mathrm{mg}, 5.34 \mathrm{mmol}, 1.2$ equiv.) and the reaction was left stirring at ambient temperature until TLC analysis showed full conversion (approx. 3 hours). Then it was diluted with $\mathrm{Et}_{2} \mathrm{O}(50 \mathrm{~mL})$ and transferred to a separatory funnel. The organic layer was washed with $\mathrm{H}_{2} \mathrm{O}(2 \times 30 \mathrm{~mL})$ and the aqueous layers were back-extracted with $\mathrm{Et}_{2} \mathrm{O}(2 \times 25 \mathrm{~mL})$. The combined organic phases were then washed with brine ( 30 mL ), dried ( $\mathrm{Na}_{2} \mathrm{SO}_{4}$ ), filtered and concentrated in vacuo. The crude product was purified by FCC (petroleum ether $40-60^{\circ} \mathrm{C} / E t O A c 9: 1$ ) to furnish bromide $\mathbf{1 0}$ ( $2.31 \mathrm{~g}, 90 \%$ for three steps) as a white solid.
 $10 \mathrm{H}), 6.57(\mathrm{br}, 1 \mathrm{H}), 5.14(\mathrm{br}, 4 \mathrm{H}), 4.29(\mathrm{br}, 1 \mathrm{H}), 3.67-3.37(\mathrm{~m}, 4 \mathrm{H}), 1.88-1.47(\mathrm{~m}, 4 \mathrm{H}), 0.86(\mathrm{~s}, 9 \mathrm{H})$, 0.04 (s, 6H); ${ }^{13} \mathrm{C}$ NMR ( $50 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 157.0,156.1,135.7,128.6,128.4,128.2,67.8,62.9,29.0$, 26.7, 26.0, 25.7, 18.0, -3.5, -5.5. MS (ESI) $\mathrm{m} / \mathrm{z}(\%): 581.2[\mathrm{M}+\mathrm{H},(100)]^{+}, 579.4[\mathrm{M}+\mathrm{H},(90)]^{+}$.

A stirring solution of bromide $10(2.30 \mathrm{~g}, 3.98 \mathrm{mmol}, 1$ equiv.) in dry DMF ( 25 mL ) under Ar was cooled to $0^{\circ} \mathrm{C}$ and $\mathrm{NaH}(60 \%, 280 \mathrm{mg}, 6.97 \mathrm{mmol}, 1.7$ equiv.) was added in three portions
over a period of 30 minutes. After stirring at the same temperature for an additional 2 hours TLC analysis showed full conversion. The reaction was quenched by slow addition of $5 \%$ citric acid (aq., 10 mL ) and then it was allowed to reach room temperature and it was transferred to a separatory funnel. The layers were separated and the aqueous phase was back-extracted with $\mathrm{Et}_{2} \mathrm{O}(50 \mathrm{~mL}$ and then $2 \times 25 \mathrm{~mL}$ ). The combined organic layers were washed with brine ( 30 mL ), dried ( $\mathrm{Na}_{2} \mathrm{SO}_{4}$ ), filtered and concentrated in vacuo. The crude product was purified by FCC (petroleum ether 40-60 $\left.{ }^{\circ} \mathrm{C} / E t O A c 9: 1\right)$ to furnish silyl ether 11 ( $1.95 \mathrm{~g}, 98 \%$ ) as a colourless oil. $R_{f} 0.5$ ( $85: 15$ petroleum ether $40-60{ }^{\circ} \mathrm{C} / E t O A c, C A M$ stain). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(200 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.33-7.28$ (br, 10H), 5.19-5.13 (m, 4H), 4.44-4.06 (m, 1H), 3.90-3.50 (m, 3H), $3.01(\mathrm{br}, 1 \mathrm{H}), 1.90-1.67(\mathrm{br}, 3 \mathrm{H})$, 1.59-1.43 (br, 1H), $0.87(\mathrm{~s}, 9 \mathrm{H}), 0.09(\mathrm{~s}, 6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(50 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 156.5,136.2,136.0,128.4$, 127.8, 67.7, 67.5, 67.3, 60.8, 54.2, 44.3, 25.7, 18.8, 18.1, -5.6. MS (ESI) m/z (\%): 499.4 [M+H, (100)] ${ }^{+}$.

A stirring solution of 11 ( $1.95 \mathrm{~g}, 3.91 \mathrm{mmol}, 1$ equiv.) in dry THF ( 30 mL ) under $\operatorname{Ar}$ was cooled to $0^{\circ} \mathrm{C}$ and TBAF ( 1 M in THF, $4.7 \mathrm{~mL}, 4.69 \mathrm{mmol}, 1.2$ equiv.) was added. After 1 hour (TLC analysis showed full conversion) the reaction was quenched by the addition brine ( 20 mL ) and the mixture was transferred to a separatory funnel and was extracted with EtOAc ( $3 \times 20 \mathrm{~mL}$ ). The combined organic layers were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, filtered and concentrated in vacuo. The crude product was purified by FCC (petroleum ether $40-60^{\circ} \mathrm{C} /$ EtOAc $1: 1$ ) to furnish primary alcohol 12 ( $1.39 \mathrm{~g}, 93 \%$ ) as a colourless oil.
$R_{f} 0.4$ (1:1 petroleum ether $40-60{ }^{\circ} \mathrm{C} / E t O A c, C A M$ stain). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(200 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.37$ (br, $10 \mathrm{H}), 5.22(\mathrm{br}, 4 \mathrm{H}), 4.48(\mathrm{br}, 1 \mathrm{H}), 4.30-4.06(\mathrm{~m}, 1 \mathrm{H}), 3.81-3.50(\mathrm{~m}, 2 \mathrm{H}), 3.10(\mathrm{br}, 1 \mathrm{H}), 1.85-1.50(\mathrm{~m}$, $4 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 156.9,155.0,135.9,135.8,135.5,128.8,128.7,128.6,128.5$, $128.4,128.3,128.1,127.9,68.6,68.5,68.3,60.7,60.1,19.8,19.5 . \mathrm{MS}(E S I) \mathrm{m} / \mathrm{z}(\%): 402.1\left[\mathrm{M}+\mathrm{NH}_{4}\right.$, $(100)]^{+}, 385.4[\mathrm{M}+\mathrm{H},(76)]^{+}$.

To a stirred solution of alcohol 12 ( $1.63 \mathrm{~g}, 4.24 \mathrm{mmol}$, 1 equiv.) in $\mathrm{CH}_{3} \mathrm{CN}(4.4 \mathrm{~mL})$ and $\mathrm{H}_{2} \mathrm{O}$ $(4.4 \mathrm{~mL})$ were added TEMPO ( $133 \mathrm{mg}, 0.85 \mathrm{mmol}, 0.2$ equiv.) and BAIB ( $3.0 \mathrm{~g}, 9.33 \mathrm{mmol}, 2.2$ equiv.) at room temperature. After stirring the mixture at the same temperature for 2 h , water was added and the mixture was extracted with ethyl acetate ( $3 \times 20 \mathrm{~mL}$ ). The combined organic layers were washed with brine ( 20 mL ), dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, filtered and concentrated in vacuo. The residue was purified by FCC (petroleum ether $40-60^{\circ} \mathrm{C} / E t O A c /$ formic acid; 1:1:0.01) to furnish 13 ( $1.57 \mathrm{~g}, 93 \%$ ) as a colorless glue.
$R_{f} 0.4$ (9:1 CHCl $3 / \mathrm{MeOH}, \mathrm{PMA}$ stain). For (S)-13: $[\alpha]_{\mathrm{D}}{ }^{23}=-17$ (c 1, $\mathrm{CHCl}_{3}$ ), $\left\{\mathrm{Lit}^{5}{ }^{5}[\alpha]_{\mathrm{D}}{ }^{23}=-19.6\right.$ (c 1, $\left.\left.\mathrm{CHCl}_{3}\right)\right\}$; ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(200 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.86(\mathrm{br}, 1 \mathrm{H}), 7.35-7.24(\mathrm{br} \mathrm{s}, 10 \mathrm{H}), 5.31-4.97(\mathrm{~m}, 5 \mathrm{H}), 4.30-$ $3.97(\mathrm{br}, 1 \mathrm{H}), 3.29-2.98(\mathrm{br}, 1 \mathrm{H}), 2.33-1.60(\mathrm{~m}, 4 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 177.1,171.2$, 137.6, 135.1, 130.4, 128.8, 128.6, 128.3, 128.0, 127.6, 94.5, 69.4, 69.3, 68.9, 68.3, 20.8, 20.5, 20.1. MS (ESI) m/z (\%): 397.0 [M-H, (100)].

## Determination of enantiopurity of 1-((Benzyloxy)carbonyl)hexahydropyridazine-3carboxylic acid (5)

The enantiomeric ratio of compound 5 was determined after derivatization to the corresponding allyl ester (S3).


To a stirring solution of acid 5 ( $80 \mathrm{mg}, 0.30 \mathrm{mmol}, 1$ equiv.) in dry DMF ( 1.6 mL ) was added sequentially $\mathrm{NaHCO}_{3}$ ( $102 \mathrm{mg}, 1.20 \mathrm{mmol}, 4.0$ equiv.) and allyl bromide ( $50 \mu \mathrm{~L}, 0.61 \mathrm{mmol}, 2.0$ equiv.) and the mixture was stirred at ambient temperature overnight. The reaction was quenched by addition of $\mathrm{H}_{2} \mathrm{O}(2 \mathrm{~mL})$ and the product was extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times 5 \mathrm{~mL})$. The combined organic layers were washed with brine $(5 \mathrm{~mL})$, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, filtered and concentrated in vacuo. The residue was purified by FCC (petroleum ether $40-60^{\circ} \mathrm{C} / \mathrm{Et}_{2} \mathrm{O} 1: 1$ ) to furnish the desired allyl ester $\mathbf{S 3}$ as a colorless liquid.
$R_{f} 0.2$ (1:1 petroleum ether $40-60{ }^{\circ} \mathrm{C} / \mathrm{Et}_{2} \mathrm{O}$, PMA stain). For (S)-S3: $[\alpha]_{\mathrm{D}}{ }^{23}=-26\left(c 1, \mathrm{CHCl}_{3}\right)$, $\left\{\right.$ Lit. $\left.^{6}[\alpha]_{\mathrm{D}}{ }^{23}=-30.2\left(c 1, \mathrm{CHCl}_{3}\right)\right\}{ }^{1} \mathrm{H}-\mathrm{NMR}\left(200 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.39-7.26(\mathrm{~m}, 5 \mathrm{H}), 5.99-5.80(\mathrm{~m}, 1 \mathrm{H})$, 5.36-5.17 (m, 4H), 4.61 (dt, J = 5.8 and $1.4 \mathrm{~Hz}, 2 \mathrm{H}), 4.07-3.90(\mathrm{~m}, 1 \mathrm{H}), 3.63-3.48(\mathrm{~m}, 1 \mathrm{H}), 3.25-3.00$ $(\mathrm{m}, 1 \mathrm{H}), 2.17-2.00(\mathrm{~m}, 1 \mathrm{H}), 1.85-1.46(\mathrm{~m}, 3 \mathrm{H}) . \mathrm{MS}(E S I) \mathrm{m} / \mathrm{z}(\%): 305.2[\mathrm{M}+\mathrm{H},(100)]^{+}$. The enantiomeric ratio of $(S)$-S3 was determined to be $94: 6$ ( $88 \%$ ee) by chiral HPLC (CHIRALPAK ${ }^{\circledR}$ ADH , hexane/ $\mathrm{iPrOH} 90: 10,1.0 \mathrm{~mL} / \mathrm{min}, 230 \mathrm{~nm}, 17.6$ minor and 23.4 major).


## For (R)-S3




Racemic S3 was prepared by mixing equal amounts of chiral (S)- and (R)-S3


Spectra


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 evpl_C13 ix Osserve
$-201.80$

(OTBS


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\underbrace{-\mathrm{OH}}_{12}
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