

Supporting Material

Focus on chemistry of the 10-dioxane-*nido*- 7,8-dicarba-undecahydrido undecaborate zwitterion; exceptionally easy abstraction of hydrogen bridge and double-action pathways observed in ring cleavage reactions with OH⁻ as nucleophile.

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I. Crystallography

Table 1. Data collection and structure refinement for compounds **1** and **2**.

Theta range for data collection	2.37 to 27.51°	2.11 to 27.54°
Index ranges	-9<=h<=10, -14<=k<=14, -17<=l<=16	-9<=h<=9, -16<=k<=16, -19<=l<=19
Reflections collected	12334	32867
Independent reflections	2895 [R(int) = 0.0587]	3123 [R(int) = 0.0548]
Coverage of independent reflections	99.8%	99.8%
Absorption correction	Multi-Scan	Multi-Scan
Structure solution technique	direct methods	direct methods
Structure solution program	XT, VERSION 2014/5	XT, VERSION 2014/5
Refinement method	Full-matrix least-squares on F^2	Full-matrix least-squares on F^2
Refinement program	SHELXL-2014/7 (Sheldrick, 2014)	SHELXL-2014/7 (Sheldrick, 2014)
Function minimized	$\Sigma w(F_o^2 - F_c^2)^2$	$\Sigma w(F_o^2 - F_c^2)^2$
Data / restraints / parameters	2895 / 0 / 198	3123 / 0 / 208
Goodness-of-fit on F^2	1.079	1.171
Final R indices	2487 data; $I > 2\sigma(I)$, $R_1 = 0.0442$, $wR_2 = 0.0928$	2655 data; $I > 2\sigma(I)$, $R_1 = 0.0521$, $wR_2 = 0.1296$
	all data $R_1 = 0.0564$, $wR_2 = 0.0978$	all data $R_1 = 0.0637$, $wR_2 = 0.1349$
Weighting scheme	$w = 1/[\sigma^2(F_o^2) + (0.0501P)^2 + 0.1134P]$ where $P = (F_o^2 + 2F_c^2)/3$	$w = 1/[\sigma^2(F_o^2) + (0.0549P)^2 + 0.7643P]$ where $P = (F_o^2 + 2F_c^2)/3$
Largest diff. peak and hole	0.143 and -0.268 eÅ ⁻³	0.284 and -0.227 eÅ ⁻³
R.M.S. deviation from mean	0.051 eÅ ⁻³	0.056 eÅ ⁻³

II. NMR Spectra

Deprotonation of compound $\mathbf{1}^-$

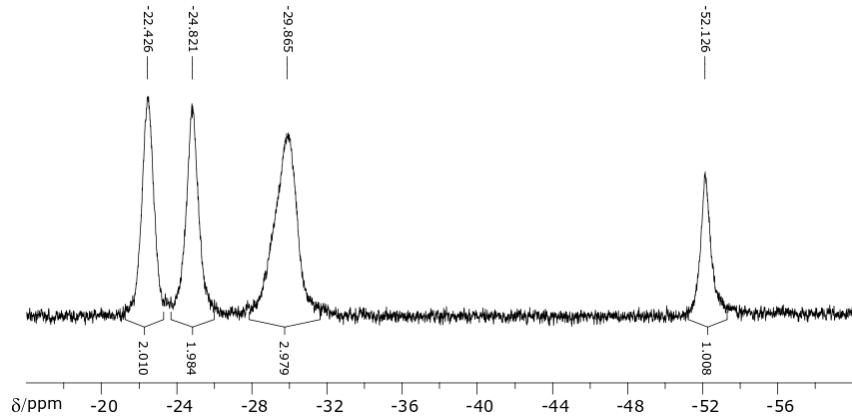


Fig S1a. ^{11}B NMR of deprotonated anion $\mathbf{1}^-$ (in $\text{CD}_3\text{CN}-\text{D}_2\text{O}$)

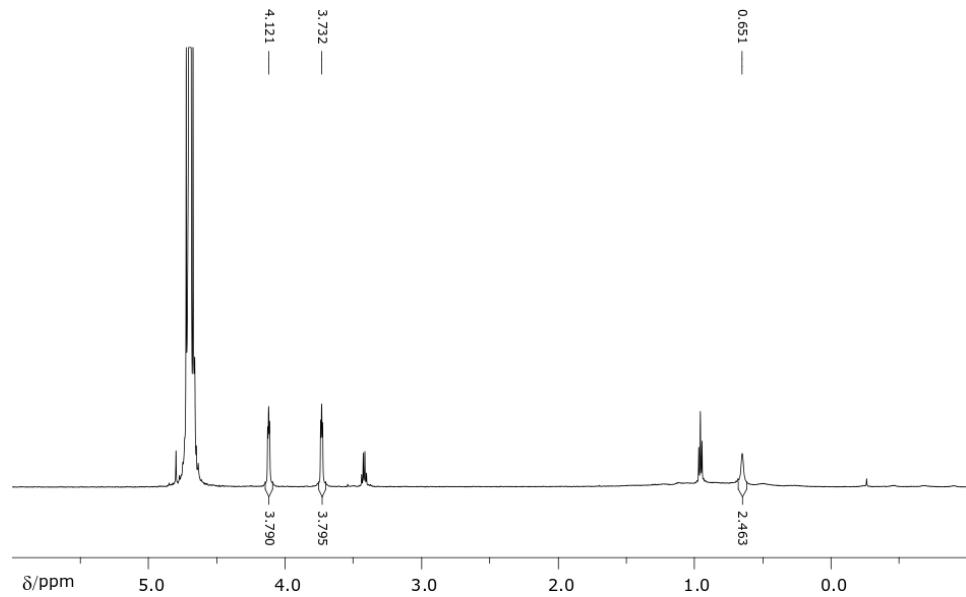


Fig S1b. ^1H NMR of deprotonated anion $\mathbf{1}^-$ (in $\text{CD}_3\text{CN}-\text{D}_2\text{O}$)

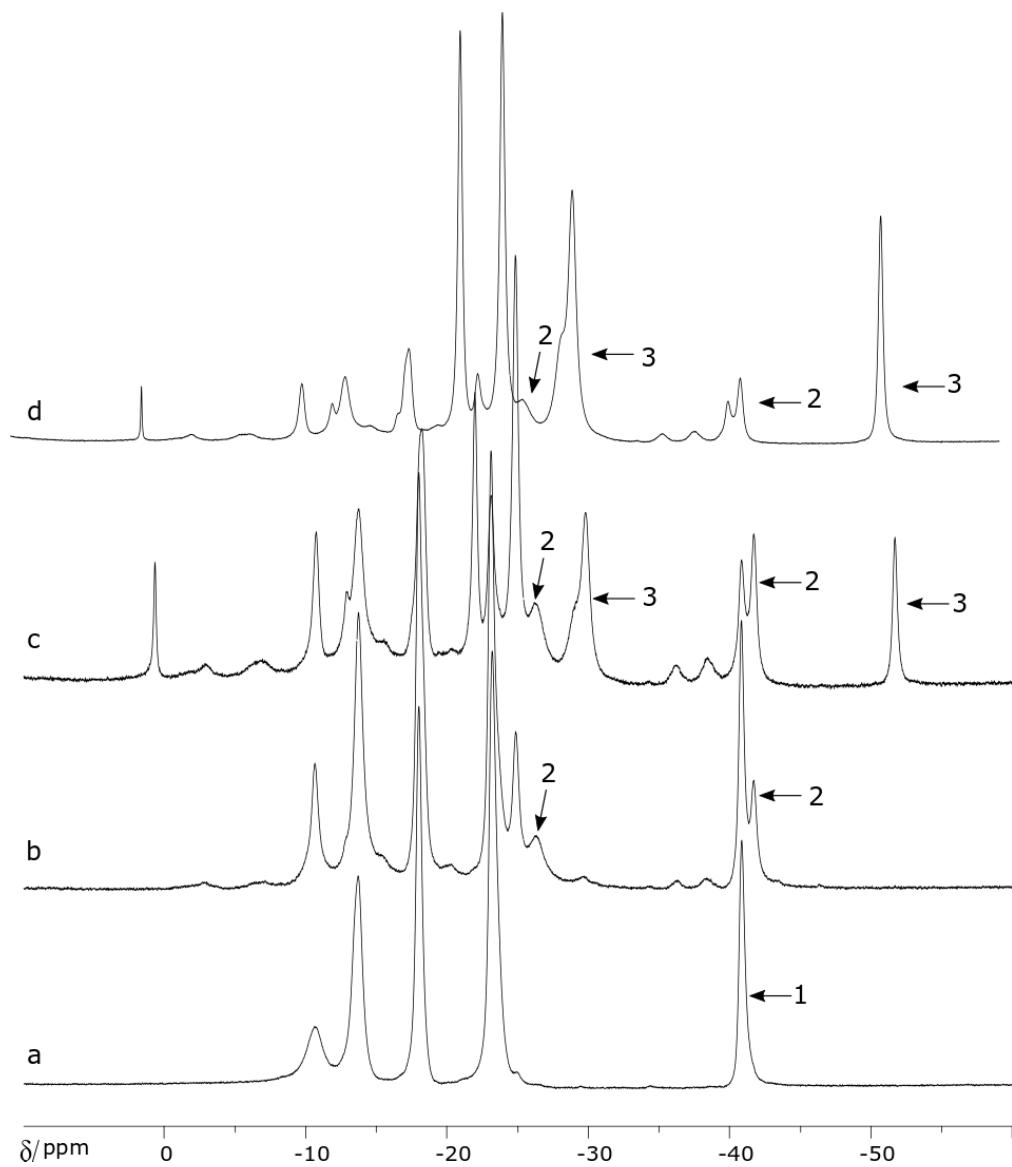


Fig S2. ^{11}B NMR spectrum of deprotonated anion $\mathbf{1}^-$ (in $\text{THF}_8\text{-D}_2\text{O}$ 1: 1 b.v.), from bottom to top: a compound $\mathbf{1}$, b $\mathbf{1} + \frac{1}{2}$ eq KOH, c $\mathbf{1} + 1$ eq KOH, d $\mathbf{1} + 1.5$ eq KOH, peaks marked with numbers 2 and 3 correspond: 2- exchange of extra hydrogen atom for deuterium in from D_2O , 3- to the deprotonated anionic form $\mathbf{1}^-$.

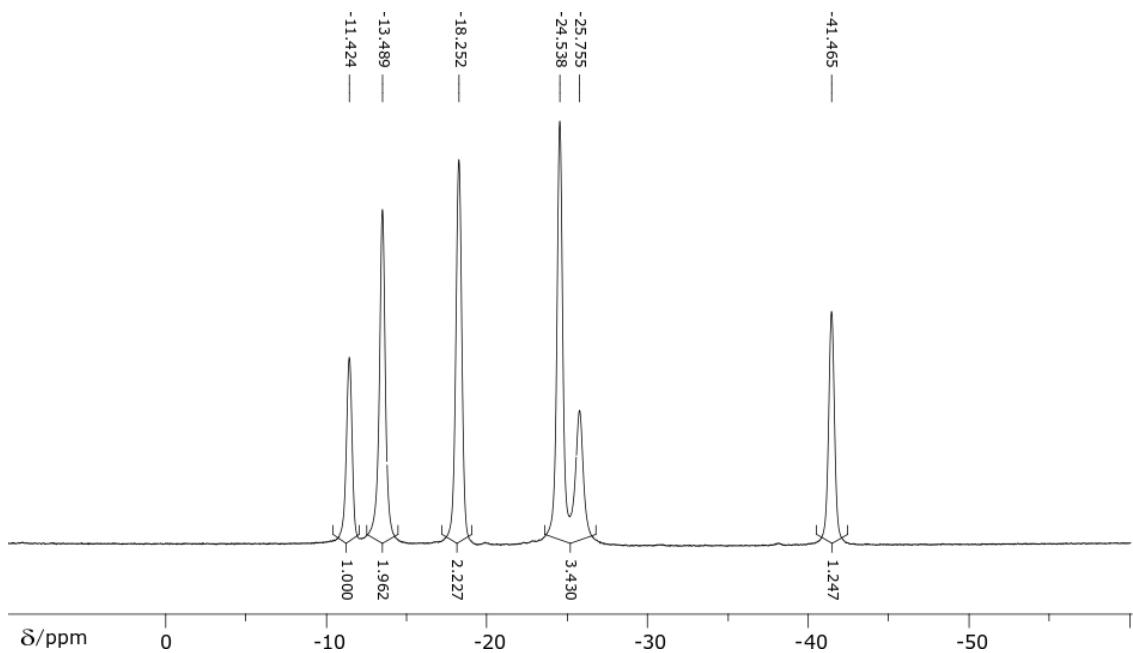


Fig S3a. ^{11}B NMR of compound 2

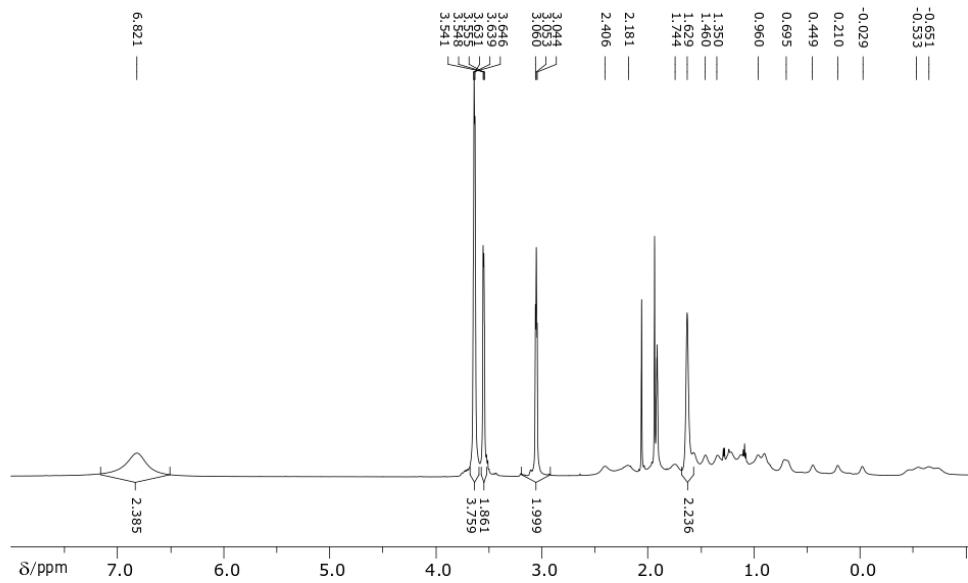


Fig S3b. ^1H NMR of compound 2

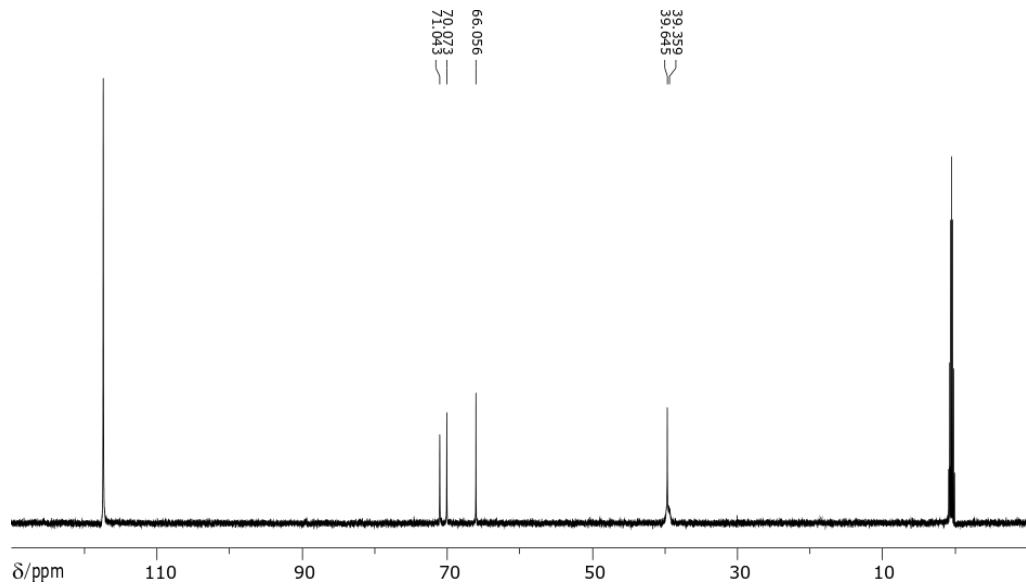


Fig S3c. ^{13}C NMR of compound 2

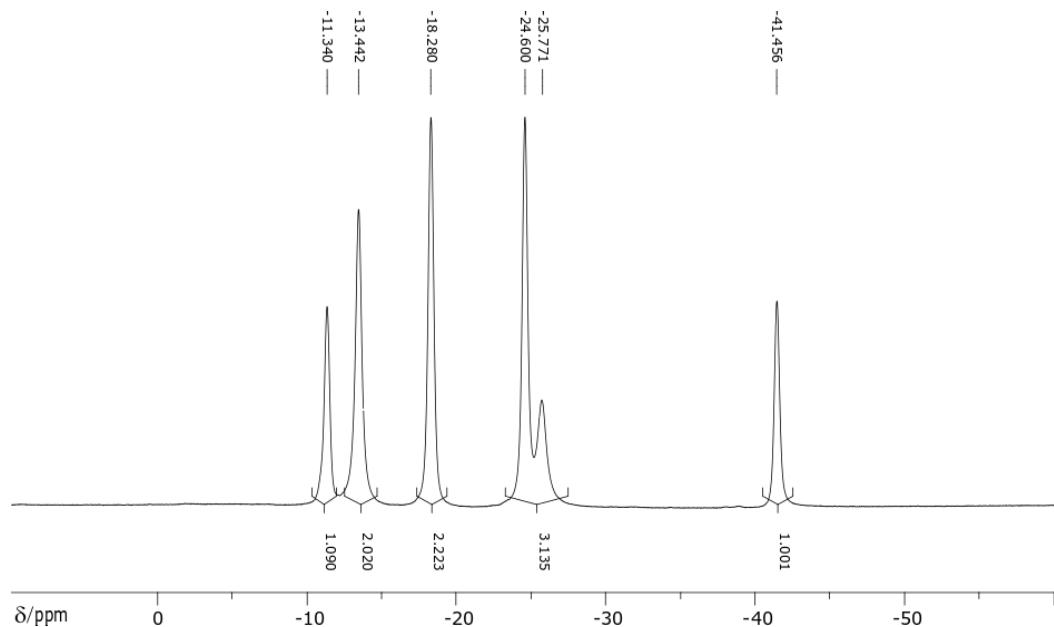


Fig S4a. ^{11}B NMR of compound 3

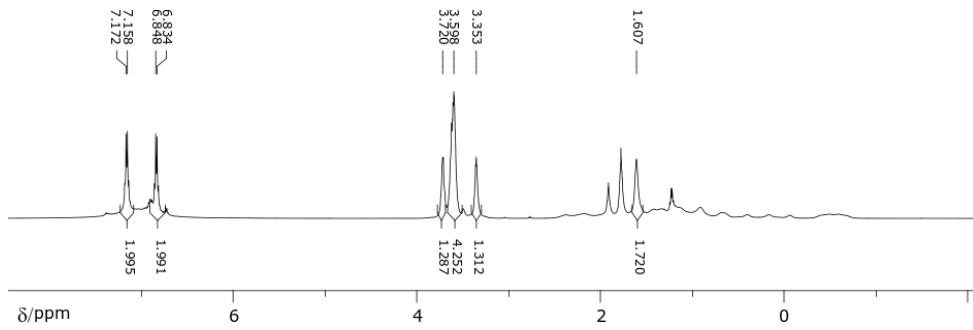


Fig S4b. ^1H NMR of compound 3

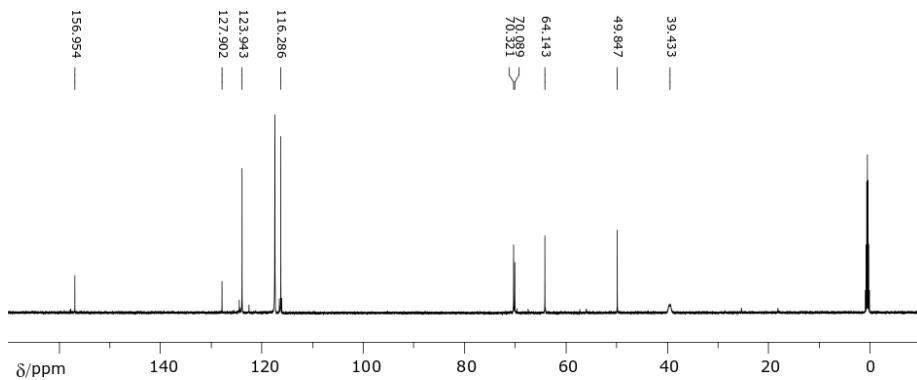


Fig S4c. ^{13}C NMR of compound 3

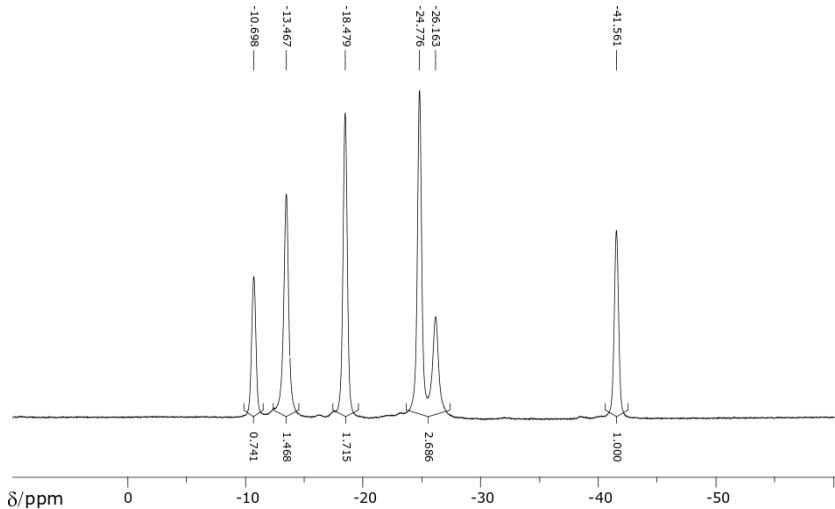


Fig S5a. ^{11}B NMR of compound $[\text{HOC}_6\text{H}_4\text{NH}_3]\text{4}$

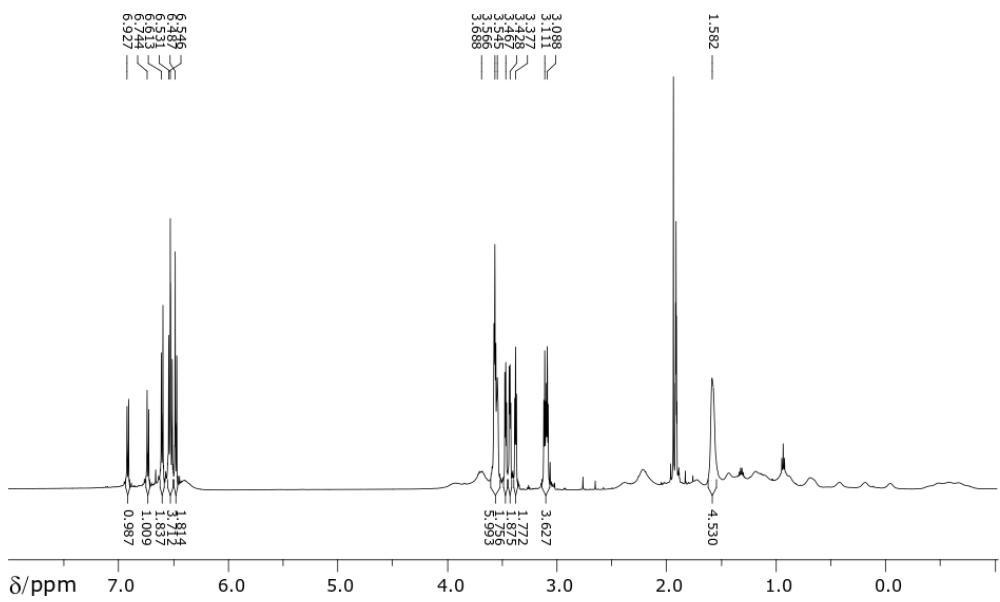


Fig S5b. ^1H NMR of compound $[\text{HOC}_6\text{H}_4\text{NH}_3]\text{4}$

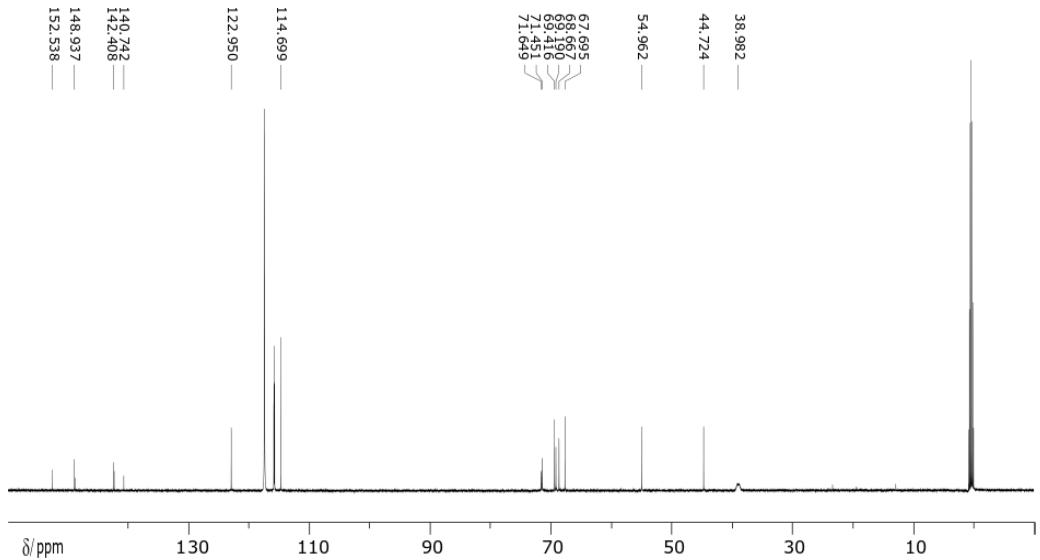


Fig S5c. ^{13}C NMR of compound $[\text{HOC}_6\text{H}_4\text{NH}_3]\text{4}$

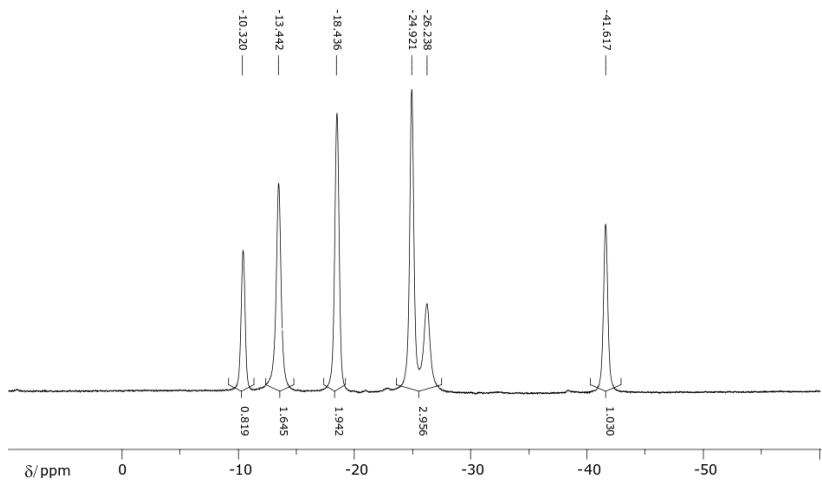


Fig S5d. ^{11}B NMR of compound K4

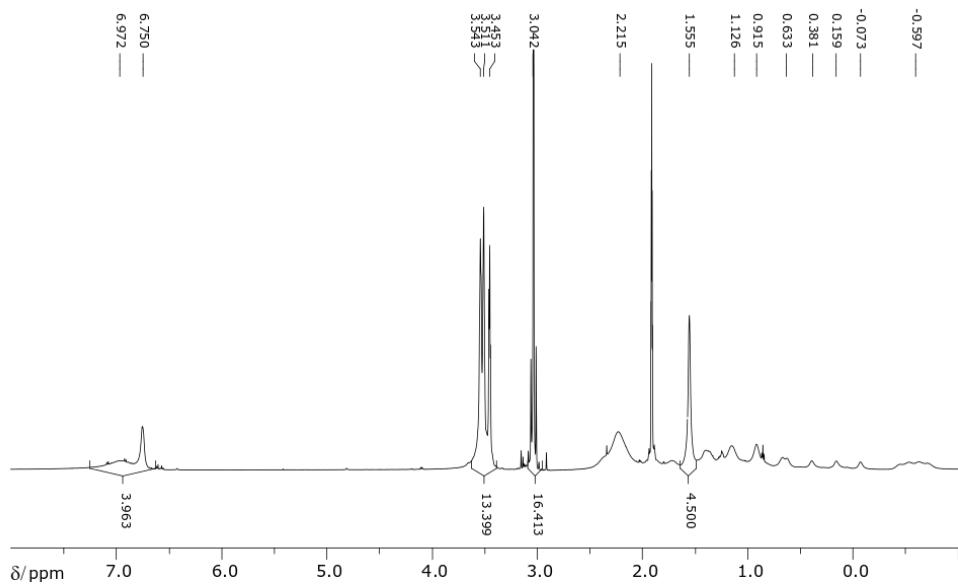


Fig S5e. ^1H NMR of compound K4

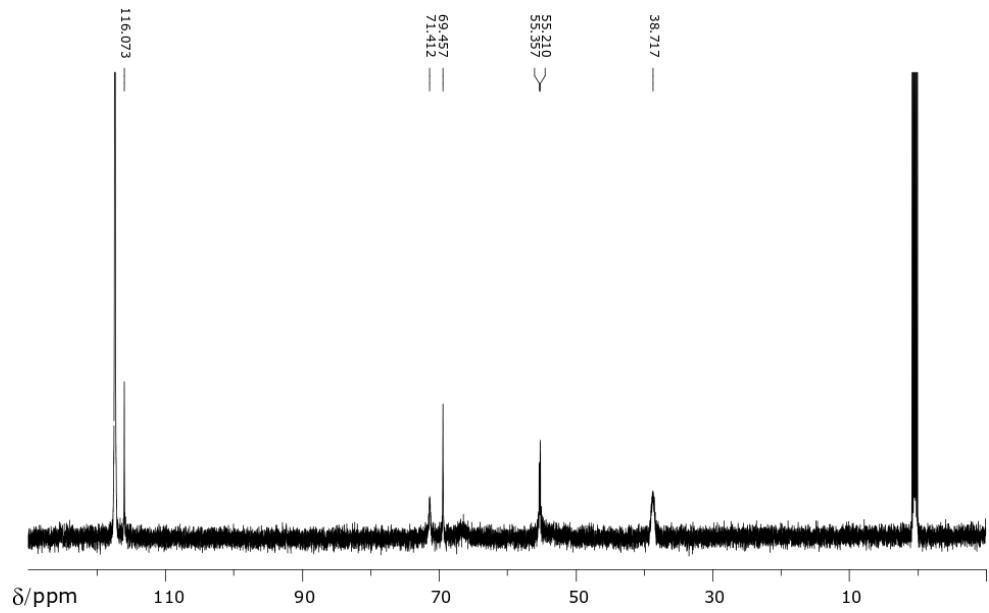


Fig S5f. ^{13}C NMR of compound K4

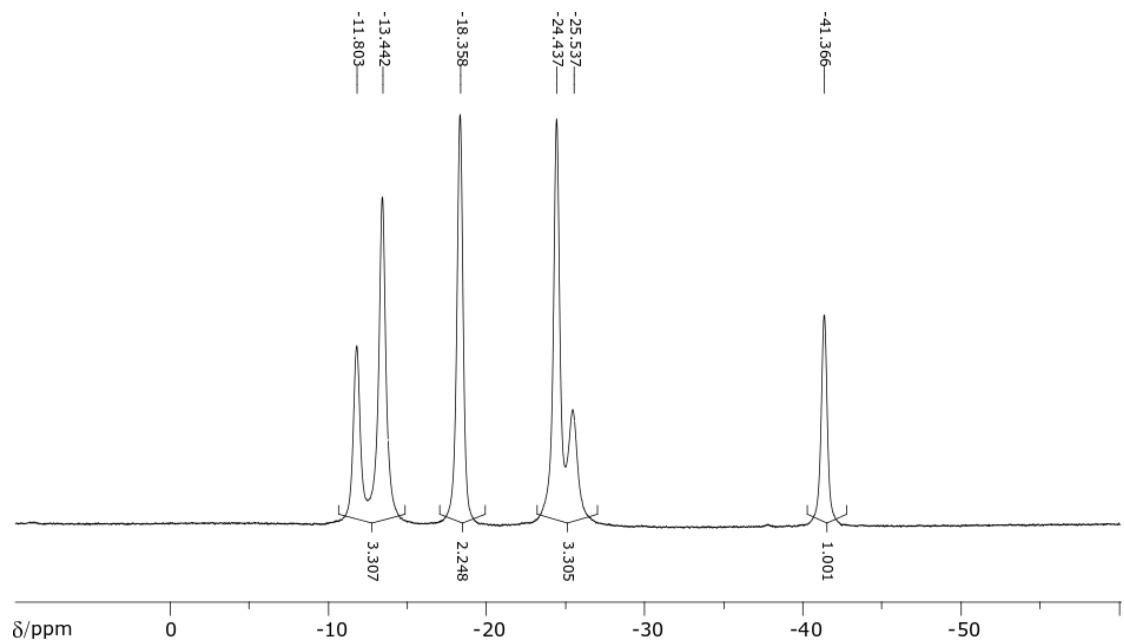


Fig S6a. ^{11}B NMR of compound 5 (zwitterion)

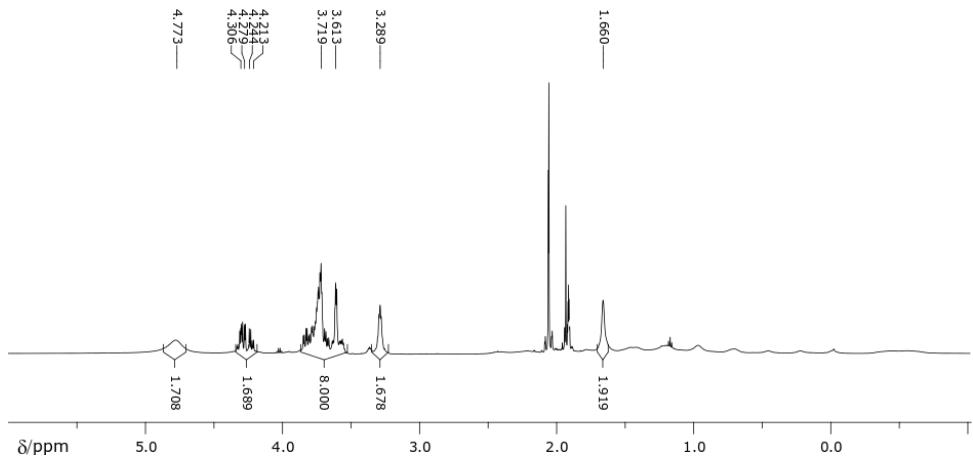


Fig S6b. ¹H NMR of compound 5 (zwitterion)

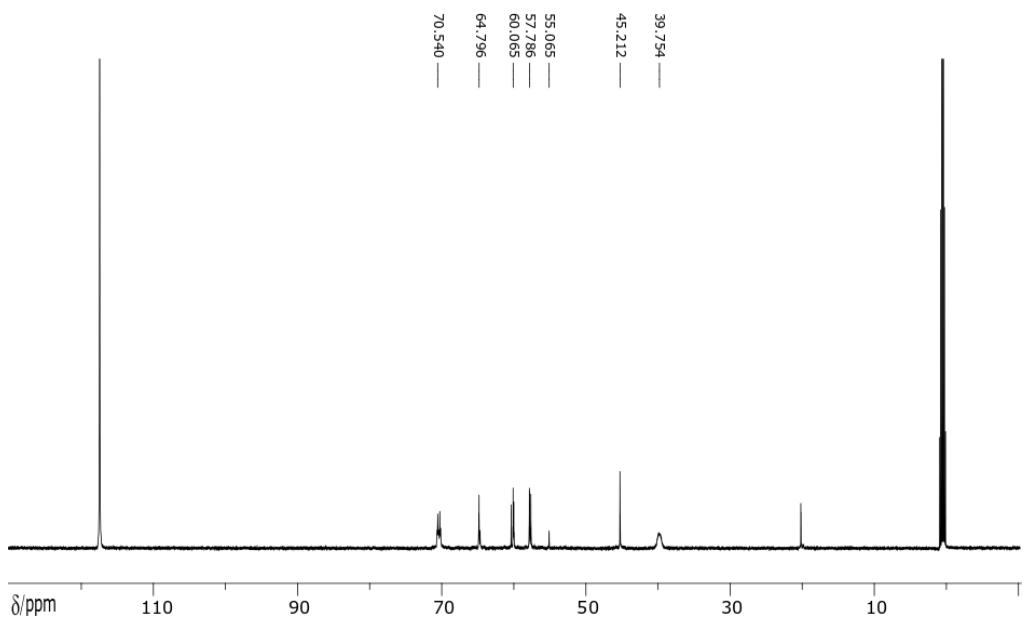


Fig S6c. ¹³C NMR of compound 5 (zwitterion)

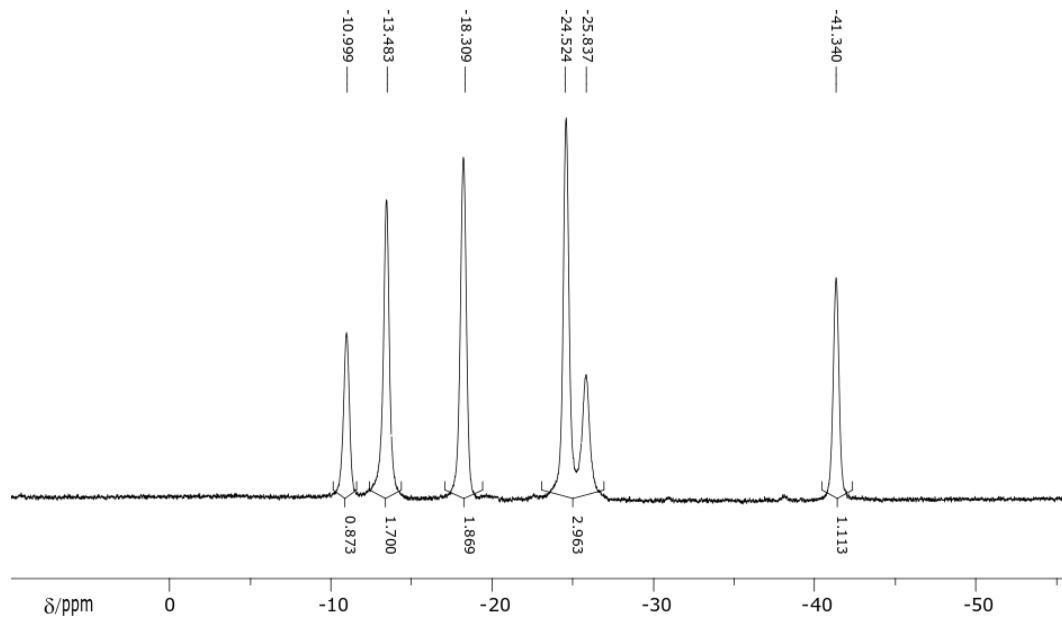


Fig S7a. ^{11}B NMR of compound $\mathbf{6}^-$

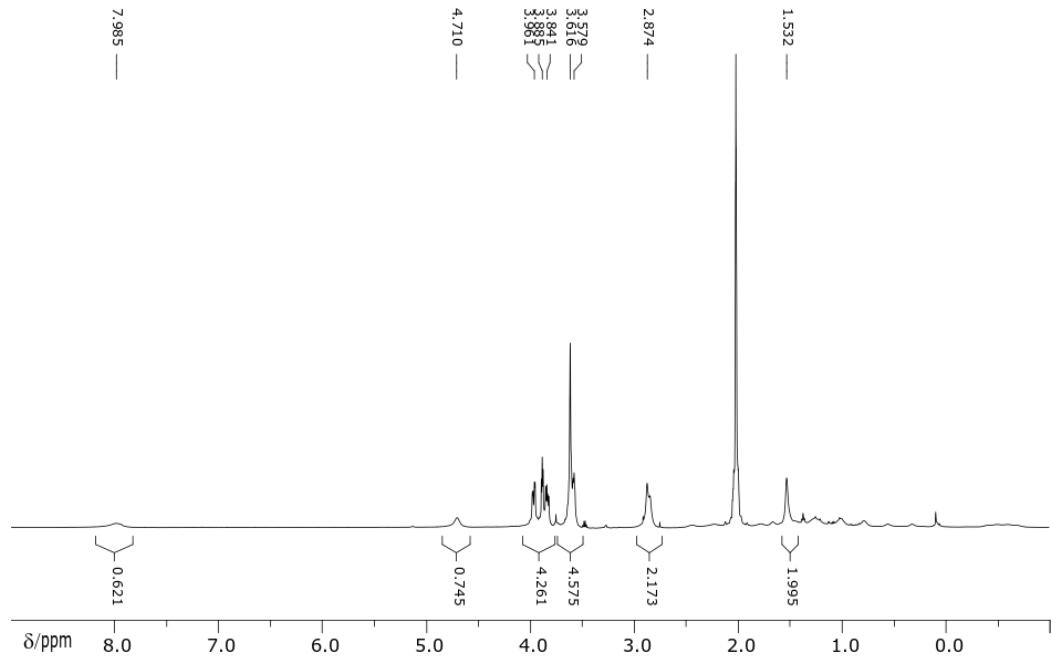


Fig S7b. ^1H NMR of compound $\mathbf{6}^-$

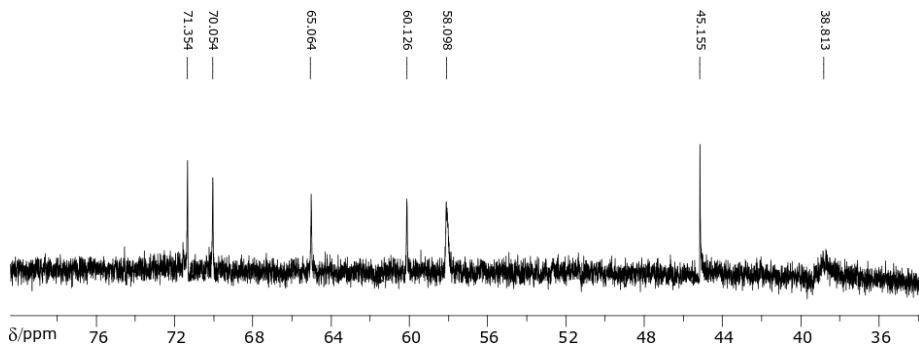


Fig S7c. ¹³C NMR of compound 6⁻

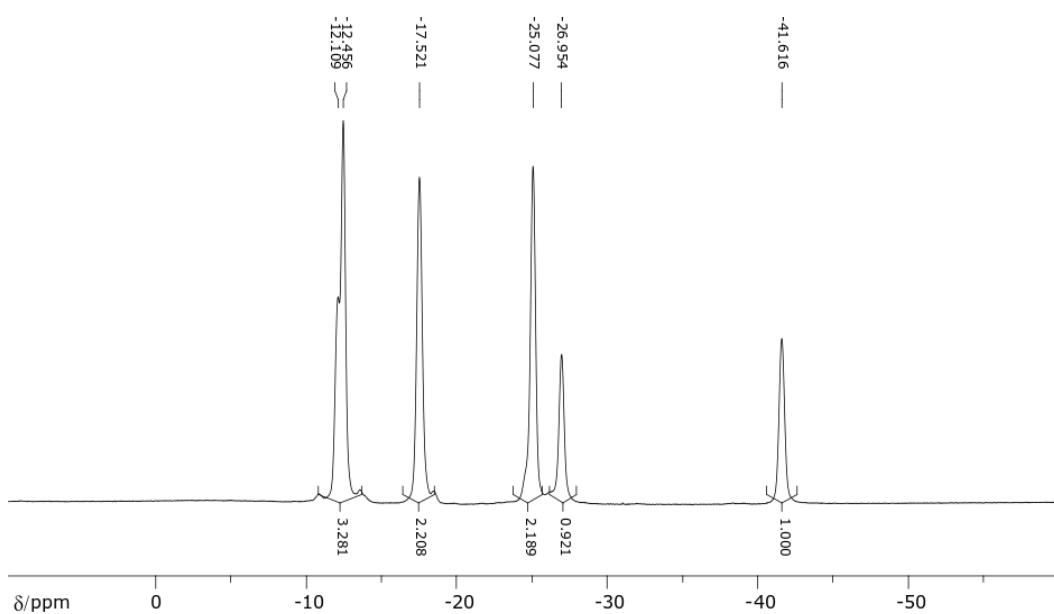


Fig S8a. ¹¹B NMR of compound 7⁻

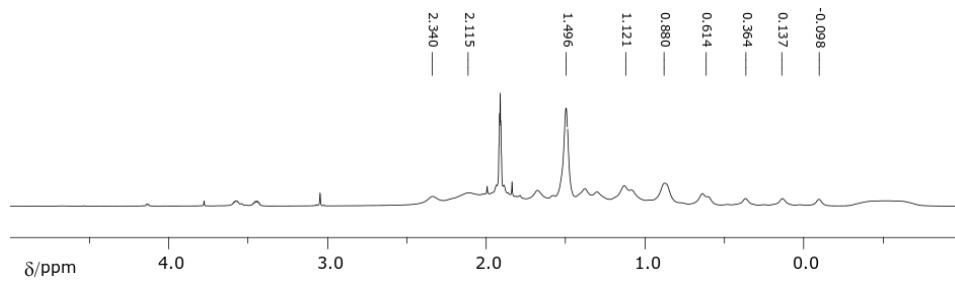


Fig S8b. ¹H NMR of compound 7

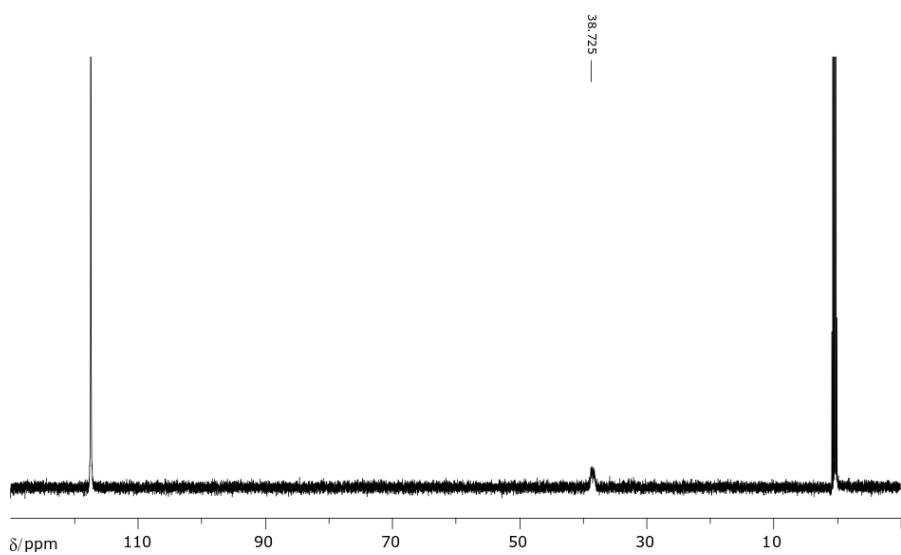


Fig S8c. ¹³C NMR of compound 7

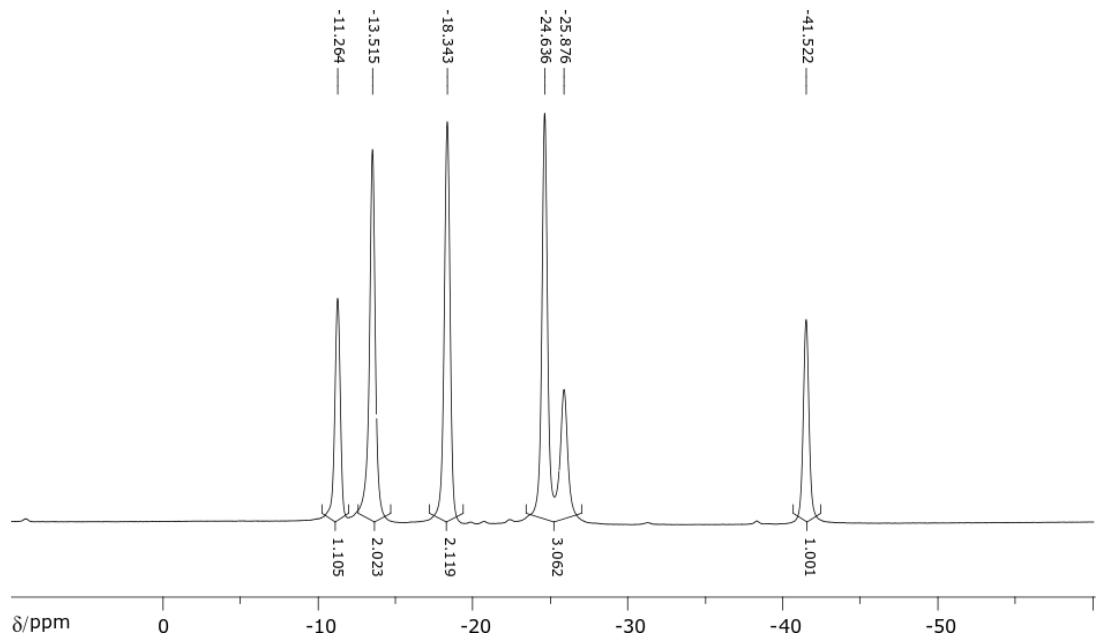


Fig S9a. ^{11}B NMR of compound $\mathbf{8}^-$

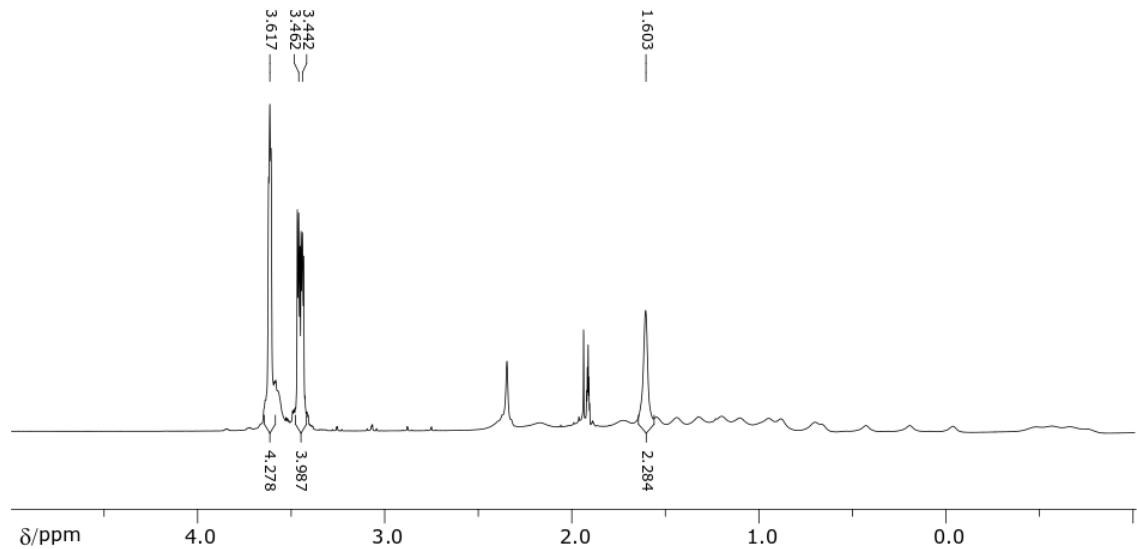


Fig S9b. ^1H NMR of compound $\mathbf{8}^-$

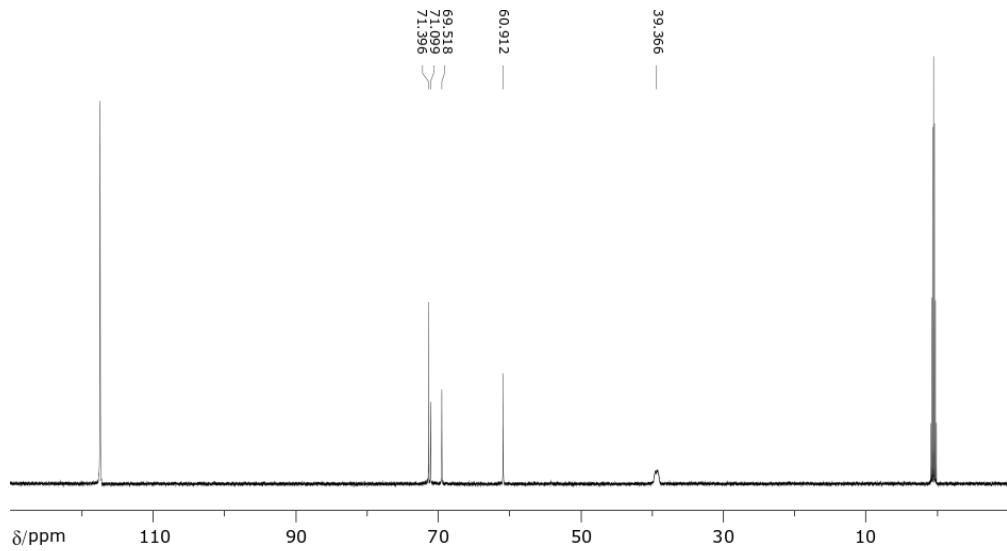


Fig S9c. ^{13}C NMR of compound $\mathbf{8}^-$

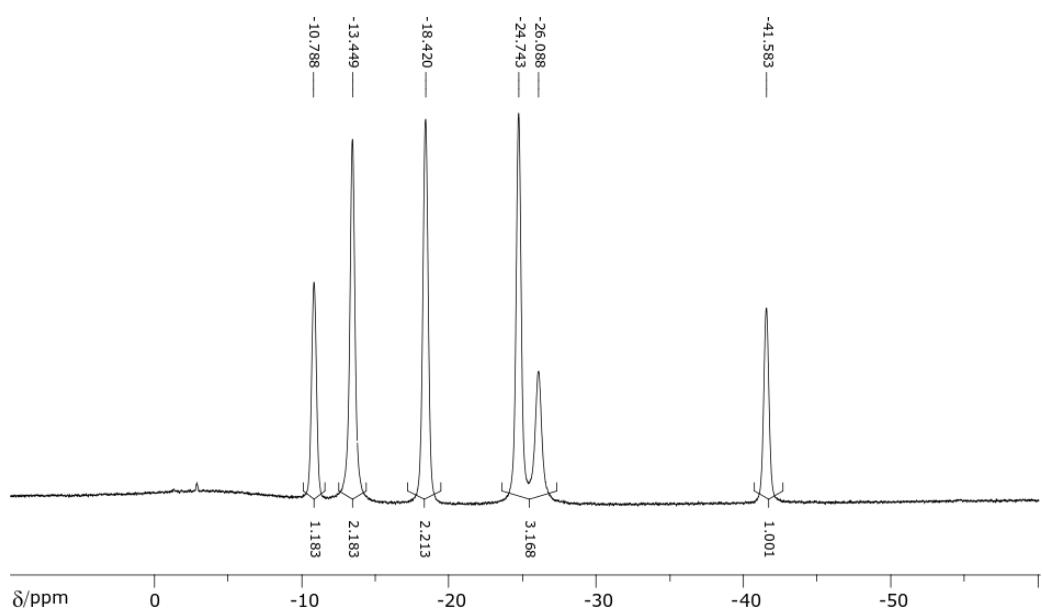


Fig S10a. ^{11}B NMR of compound $\mathbf{9}^-$

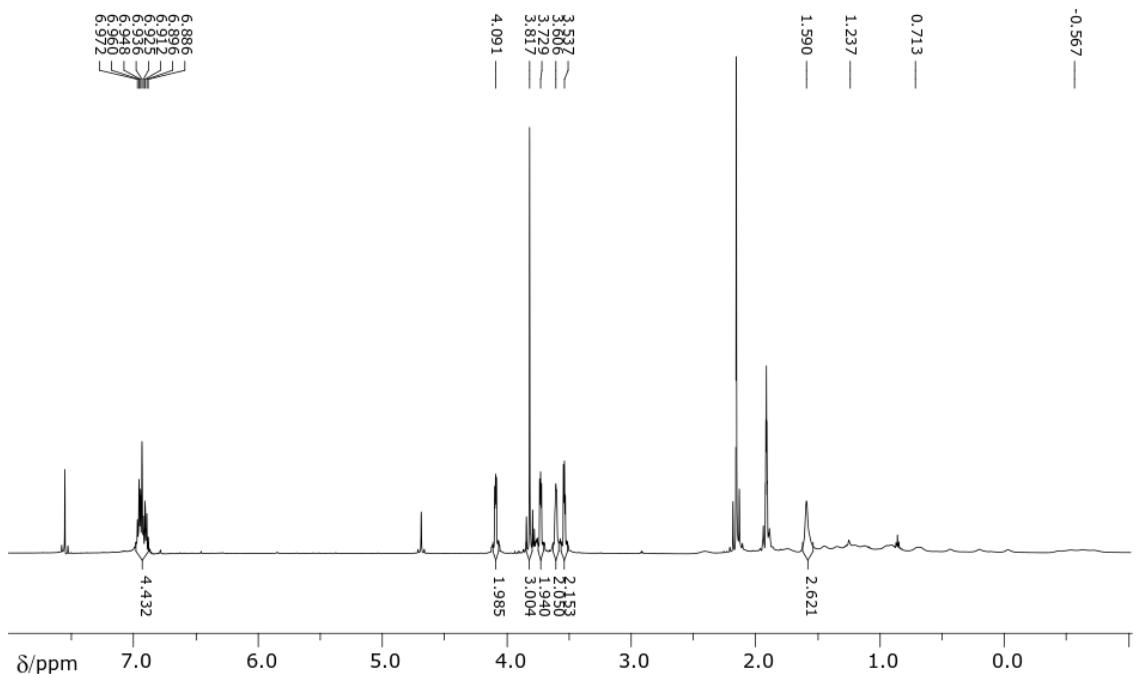


Fig S10b. ^1H NMR of compound $\mathbf{9}^-$

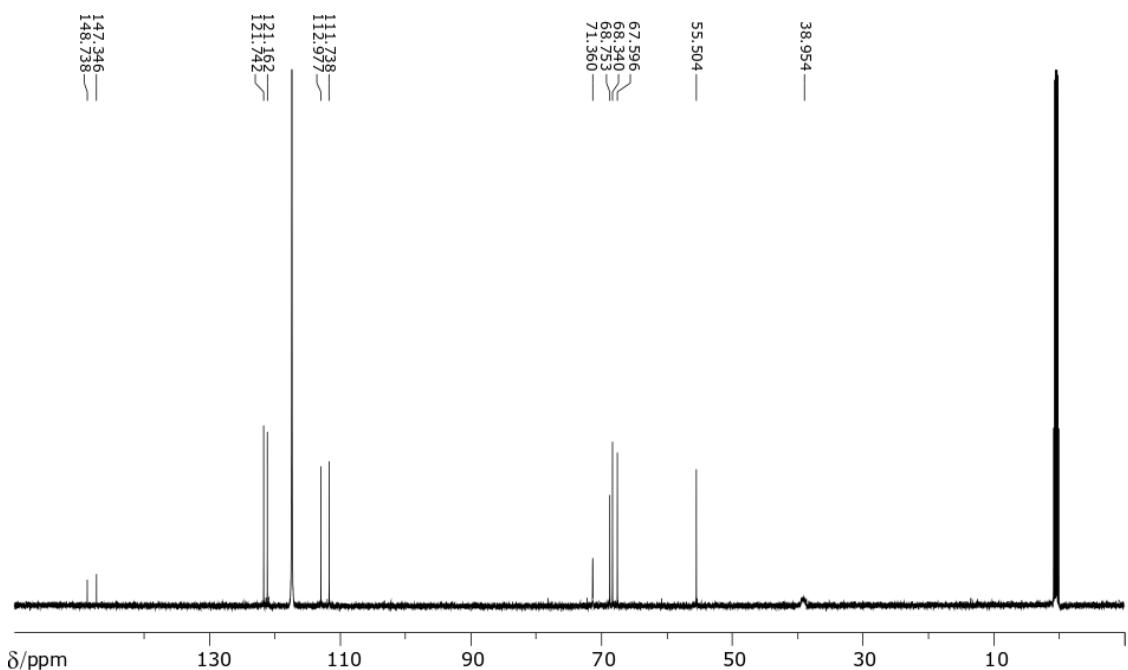


Fig S10c. ^{13}C NMR of compound $\mathbf{9}^-$

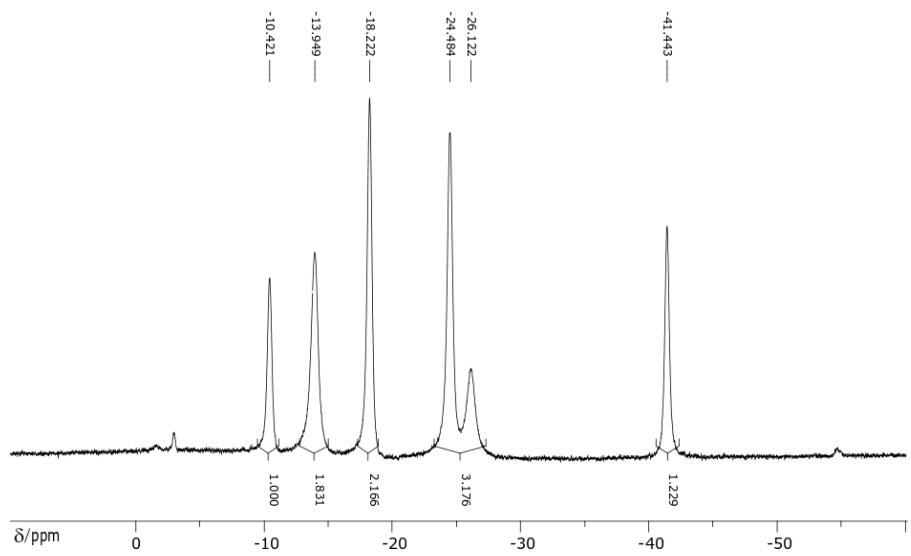


Fig S11a. ^{11}B NMR of compound $\mathbf{10}^{2-}$

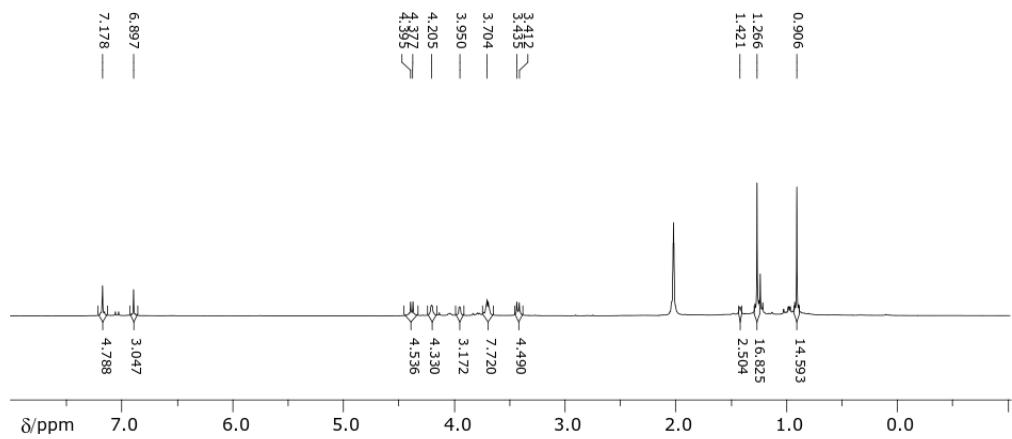


Fig S11b. ^1H NMR of compound $\mathbf{10}^{2-}$

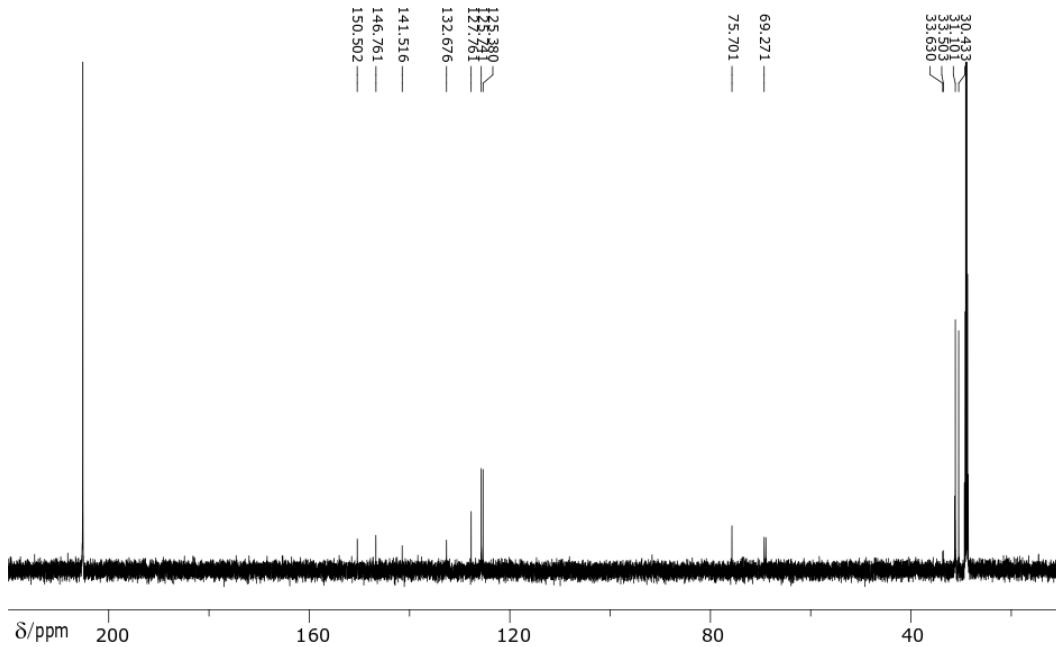


Fig S11c. ¹³C NMR of compound **10**²⁻

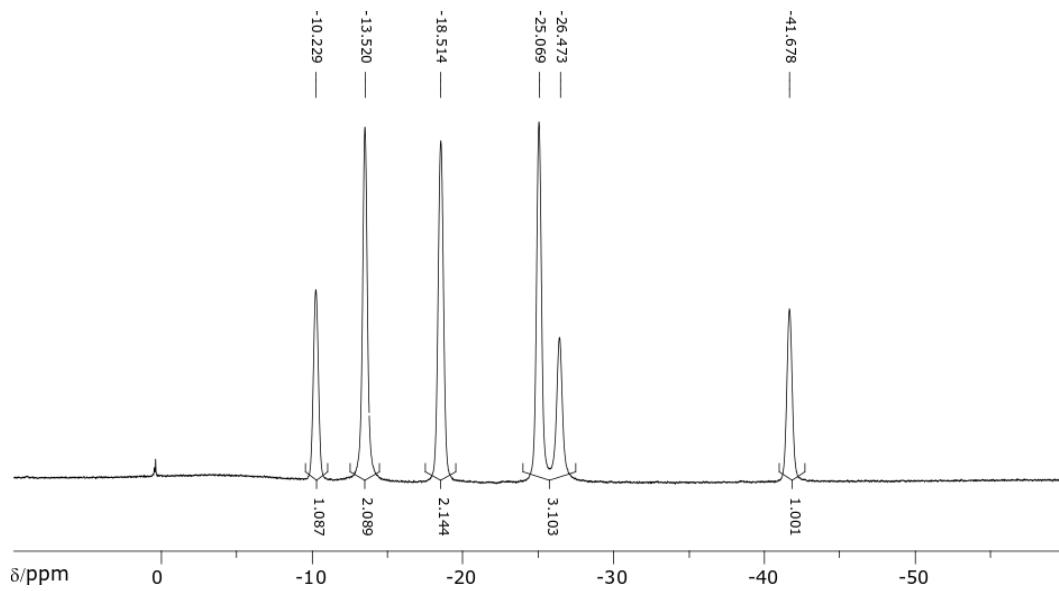


Fig S12a. ¹¹B NMR of compound **11**⁻

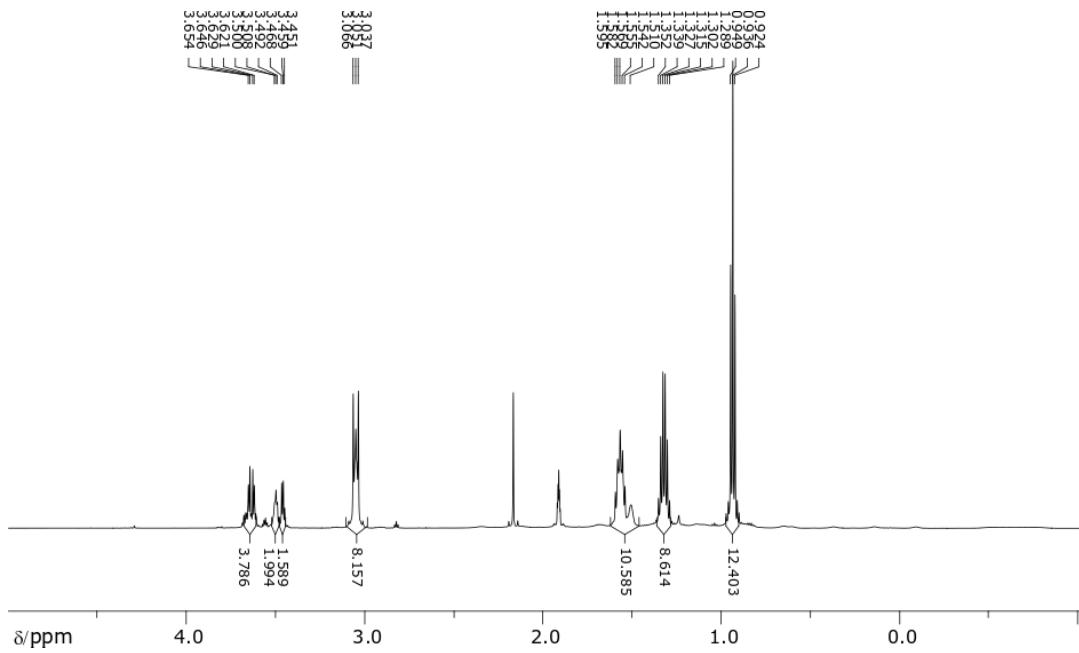


Fig S12b. ¹H NMR of compound **11**⁻

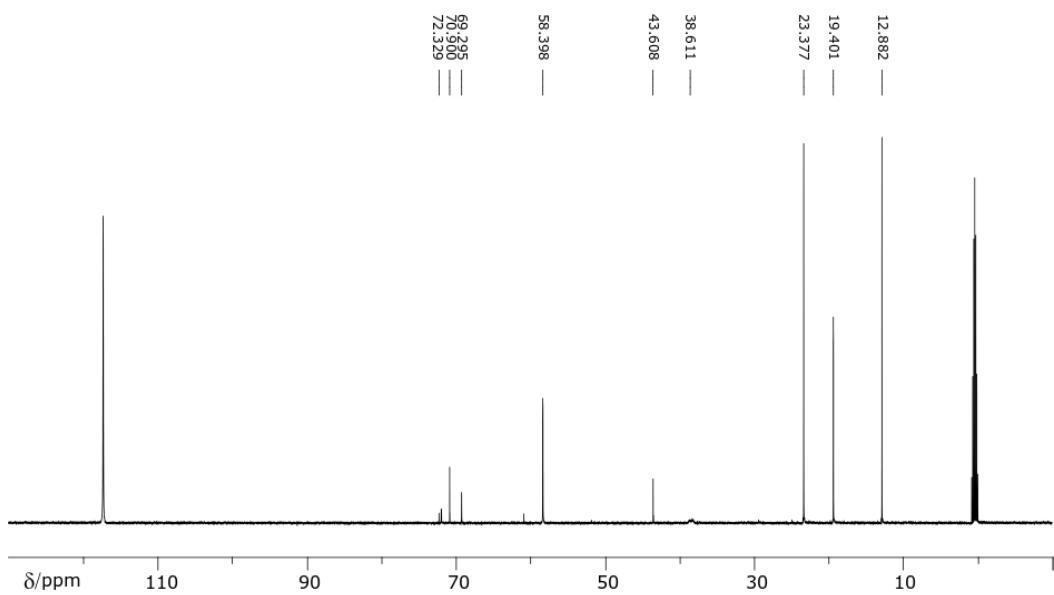


Fig S12c. ¹³C NMR of compound **11**⁻

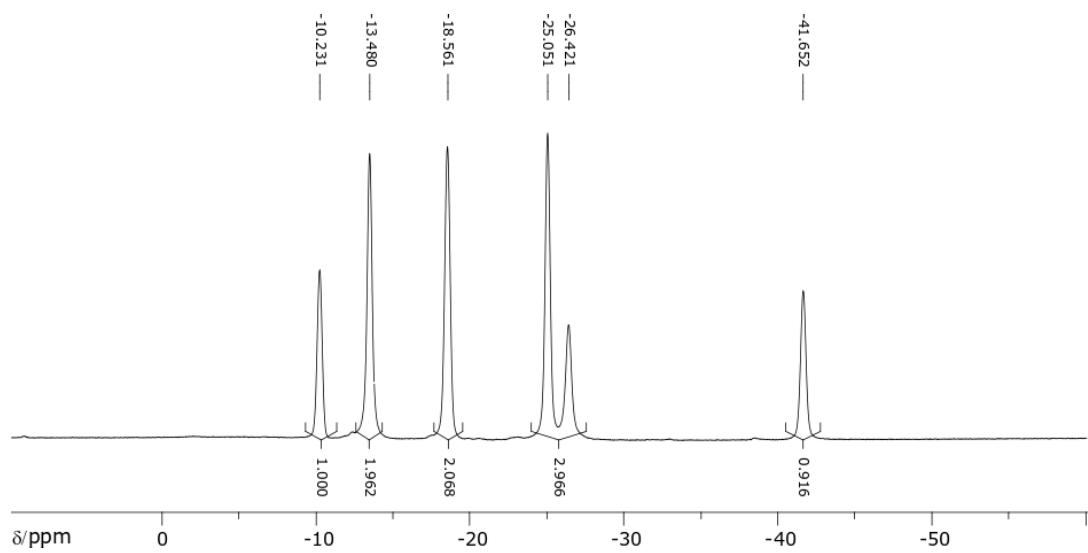


Fig S13a. ^{11}B NMR of compound $\mathbf{12}^-$

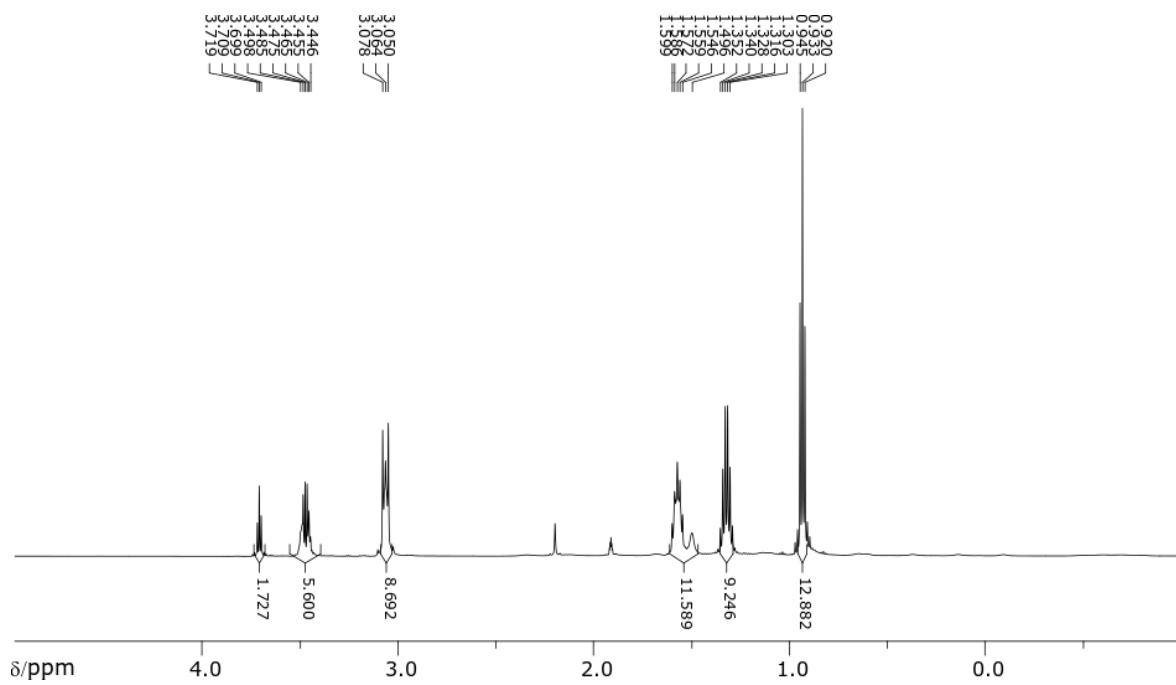


Fig S13b. ^1H NMR of compound $\mathbf{12}^-$

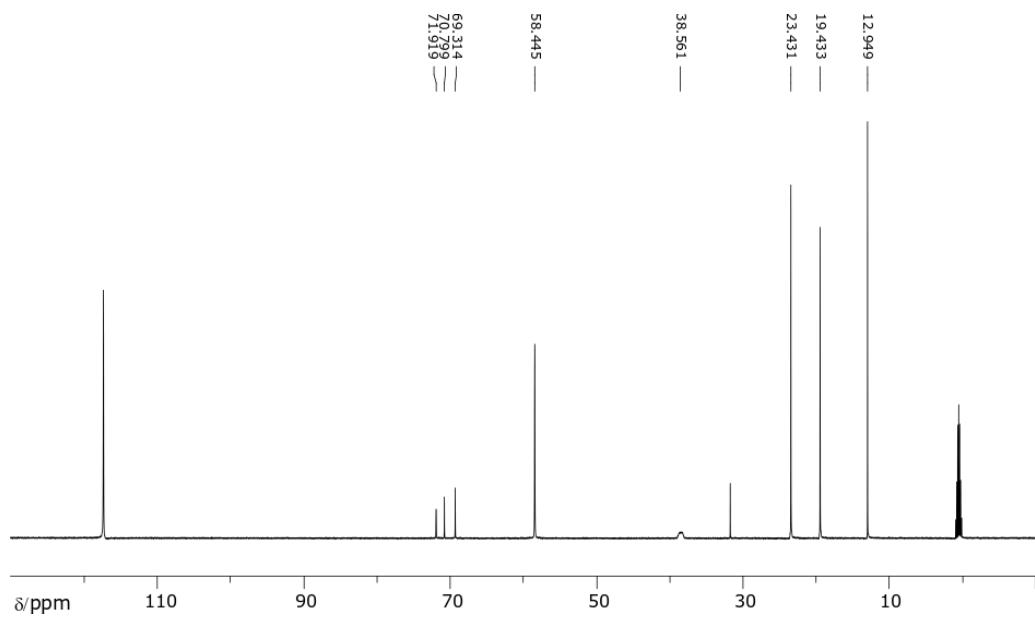


Fig S13c. ^{13}C NMR of compound $\mathbf{12}^-$

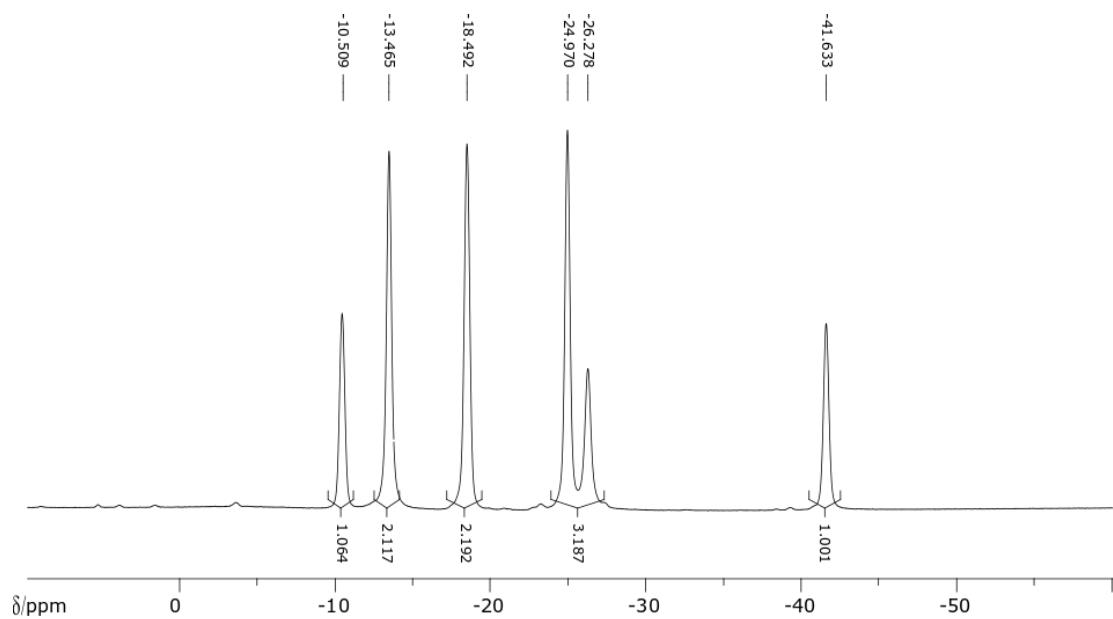


Fig S14a. ^{11}B NMR of compound $\mathbf{13}^-$

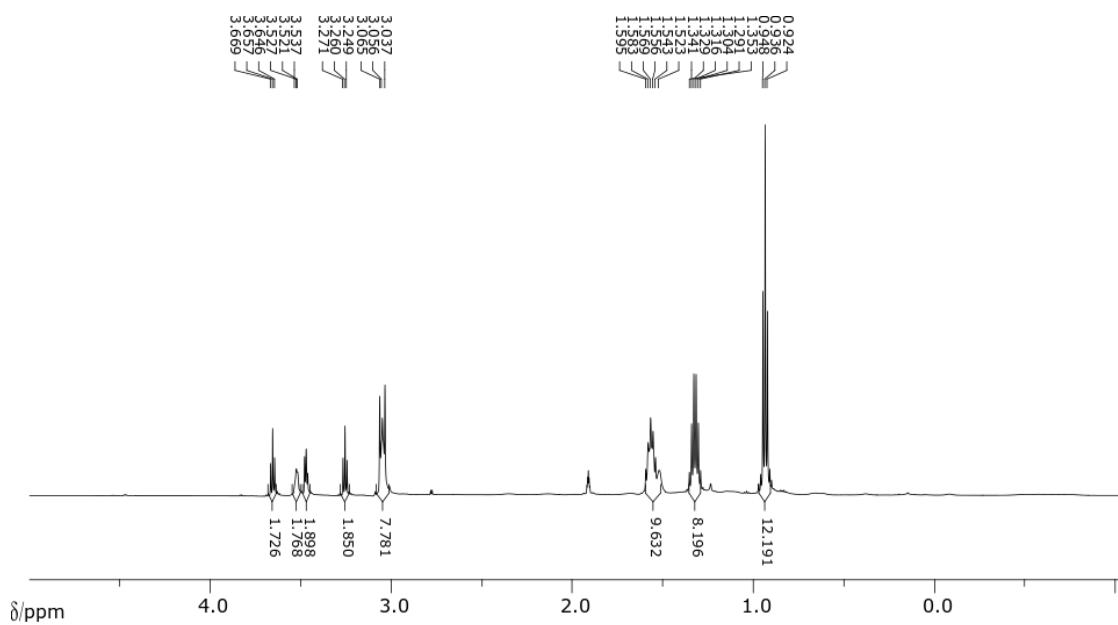


Fig S14b. ¹H NMR of compound **13⁻**

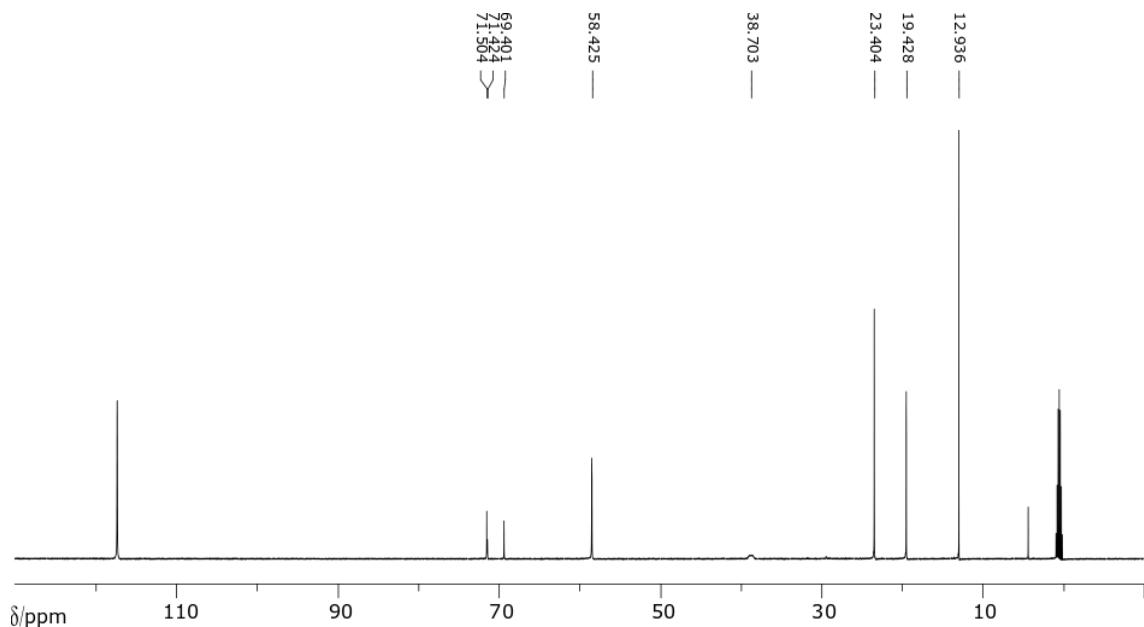


Fig S14c. ¹³C NMR of compound **13⁻**

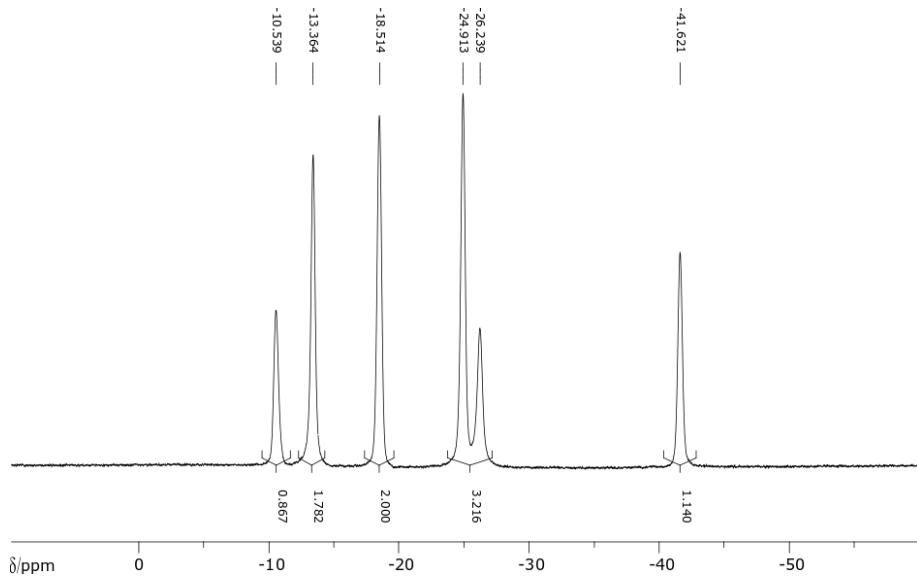


Fig S15a. ^{11}B NMR of compound $\mathbf{14}^-$

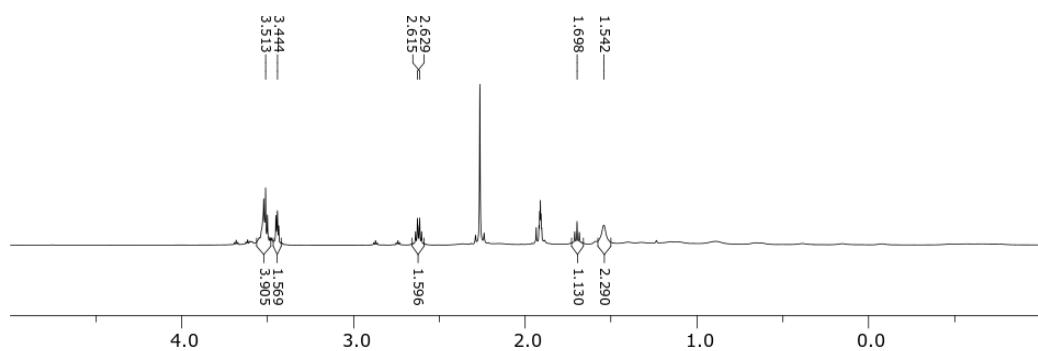


Fig S15b. ^1H NMR of compound $\mathbf{14}^-$

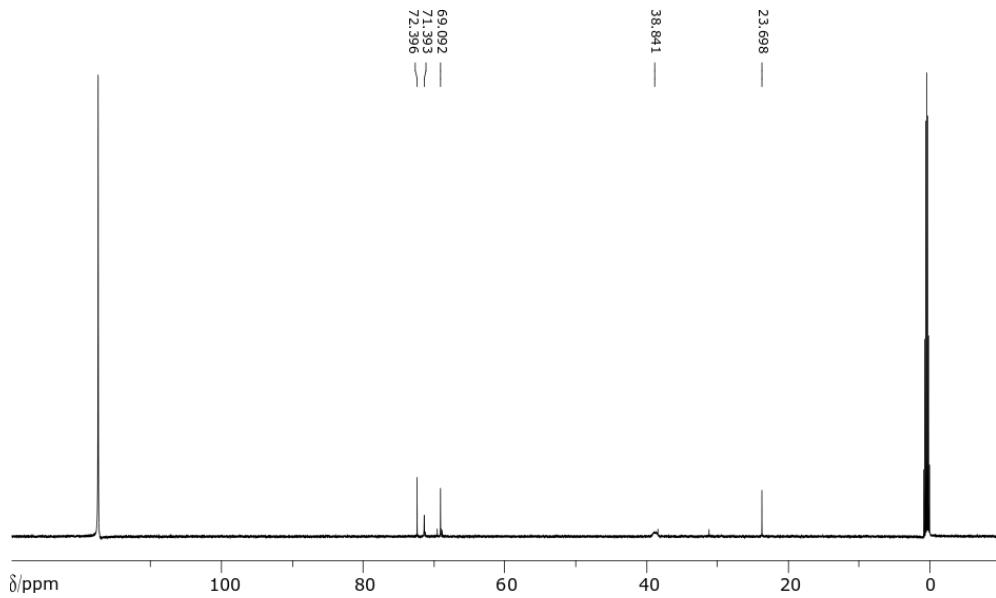


Fig S15c. ¹³C NMR of compound **14⁻**

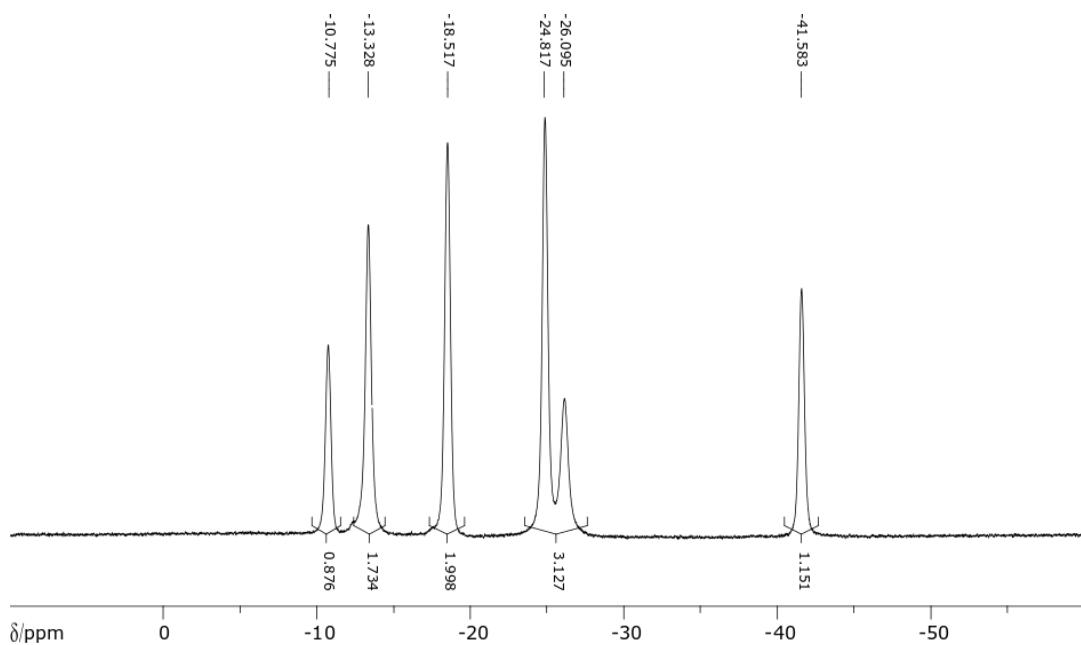


Fig S16a. ¹¹B NMR of compound **15²⁻**

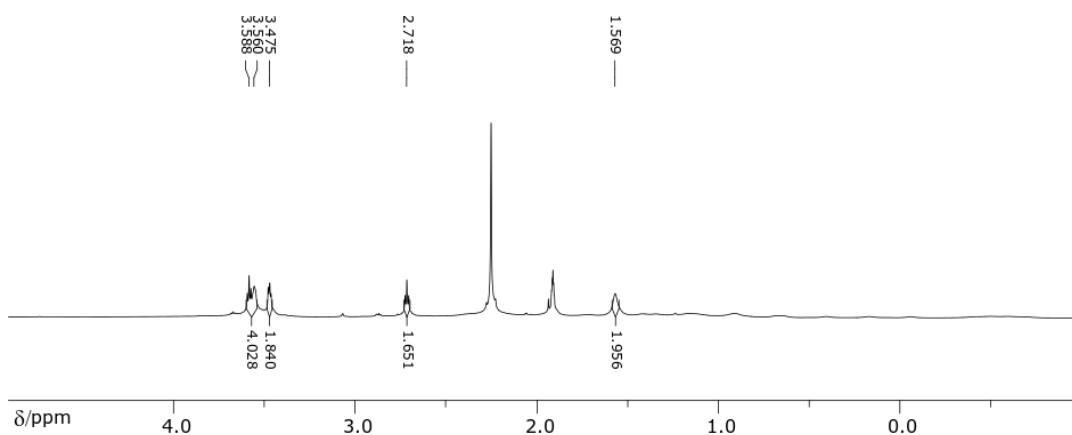


Fig S16b. ¹H NMR of compound **15**²⁻

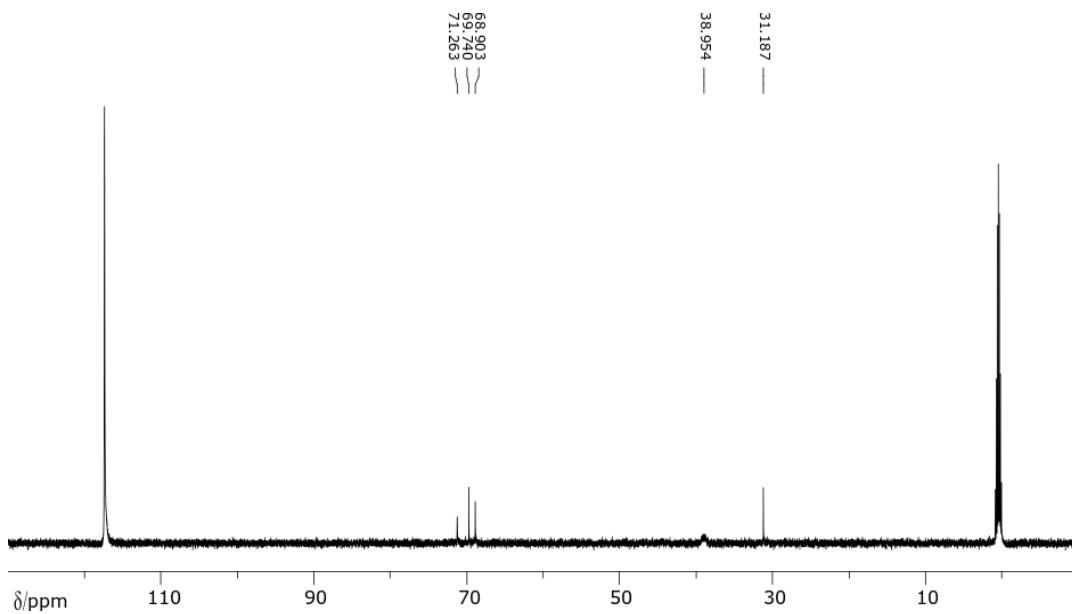


Fig S16c. ¹³C NMR of compound **15**²⁻

III. Mass Spectra

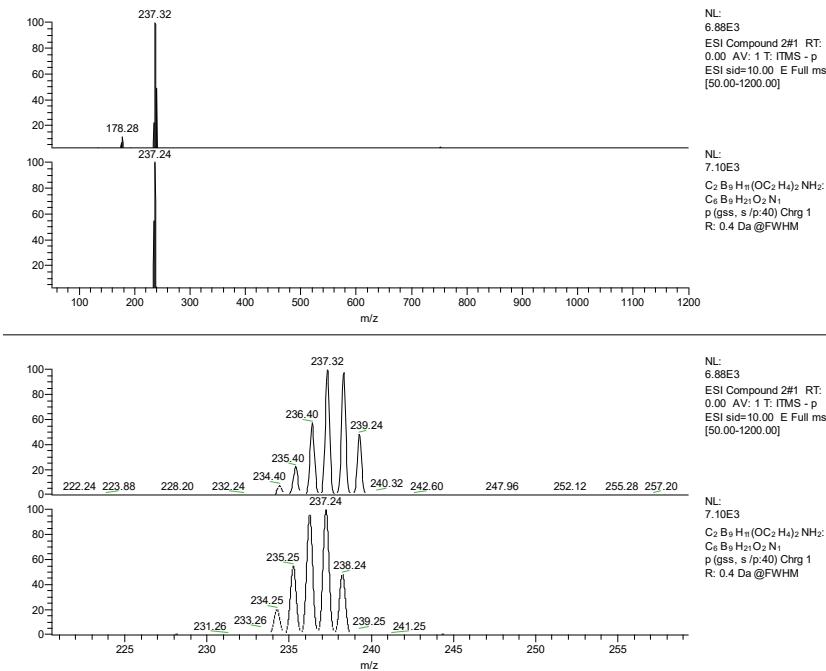


Fig S17. MS spectra of compound 2

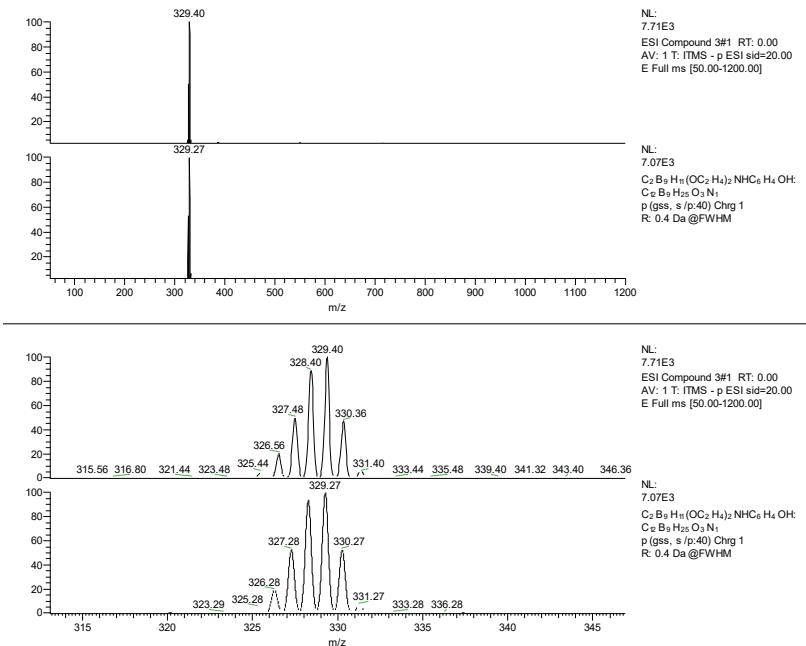


Fig S18. MS spectra of compound 3

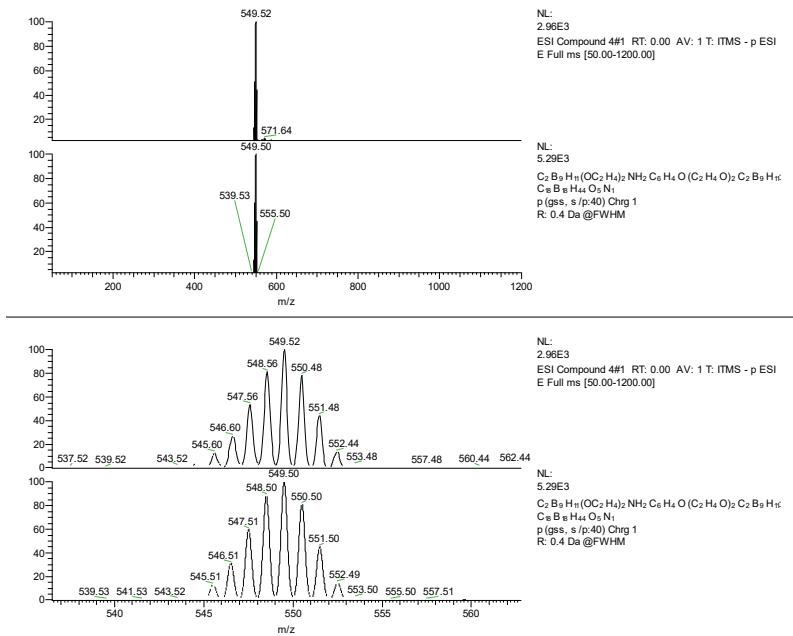


Fig S19. MS spectra of compound 4⁻

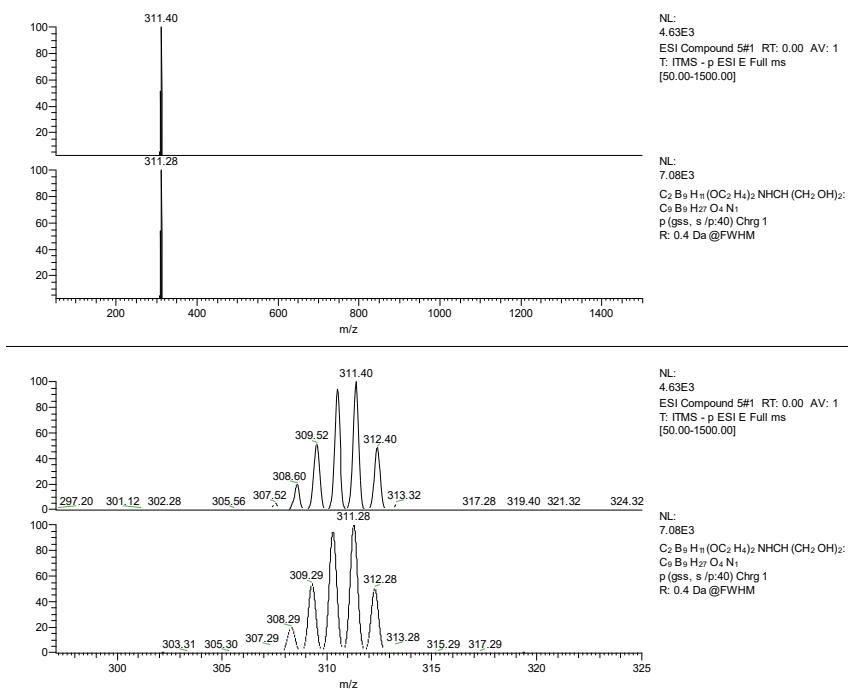


Fig S20. MS spectra of compound 5

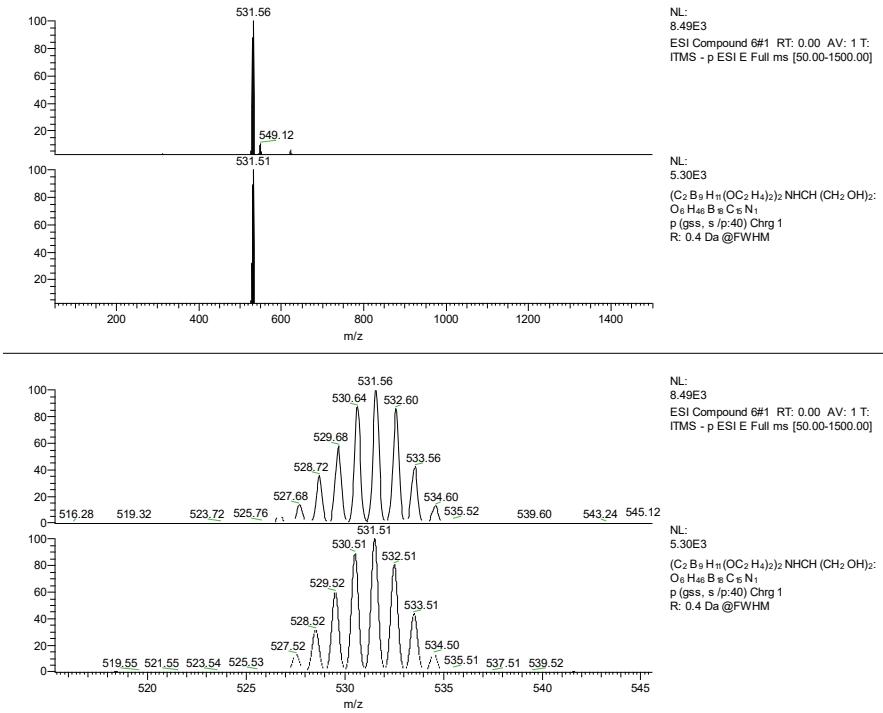


Fig S21. MS spectra of compound 6^-

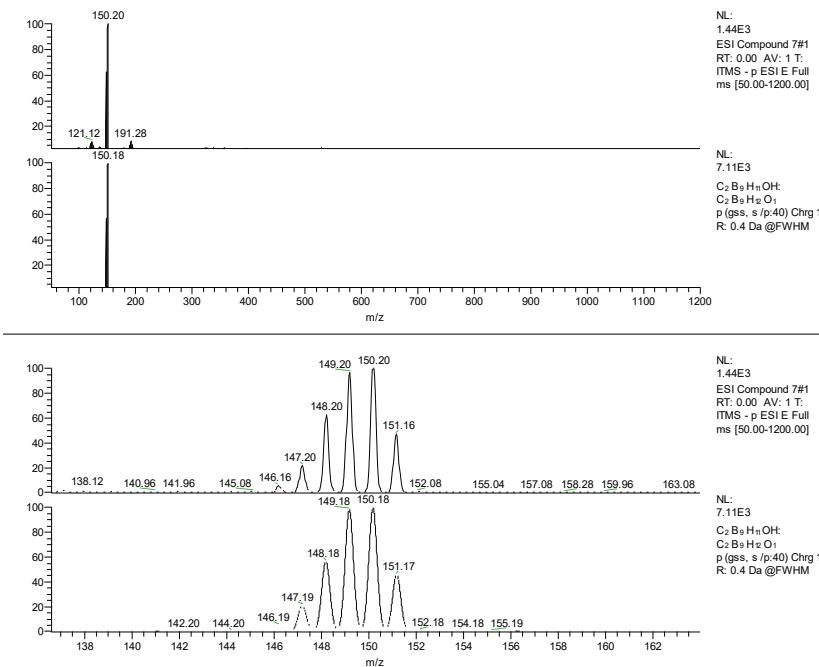


Fig S22. MS spectra of compound 7^-

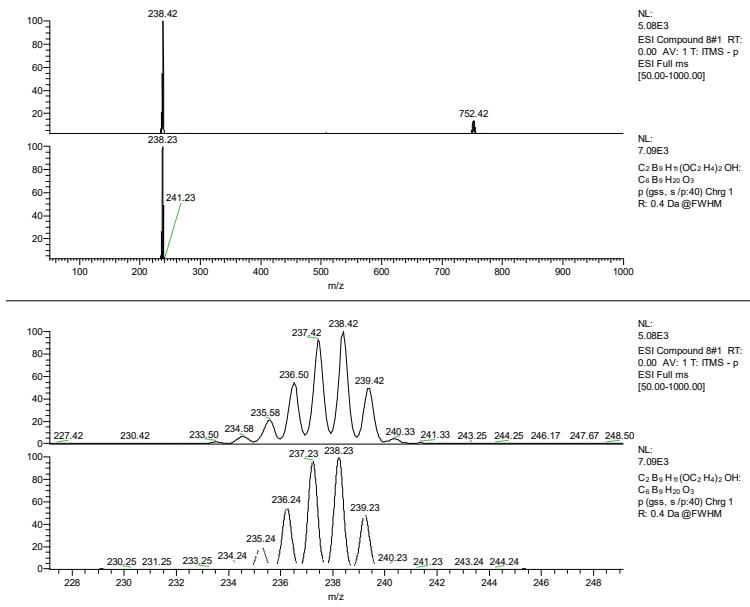


Fig S23. MS spectra of compound 8⁻

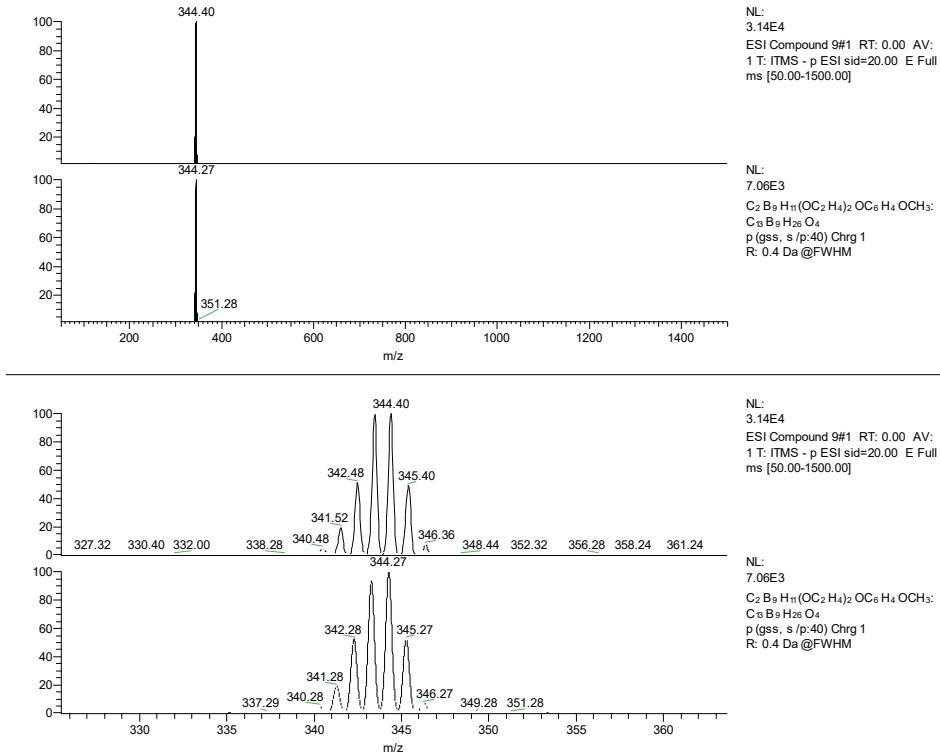


Fig S24. MS spectra of compound 9⁻

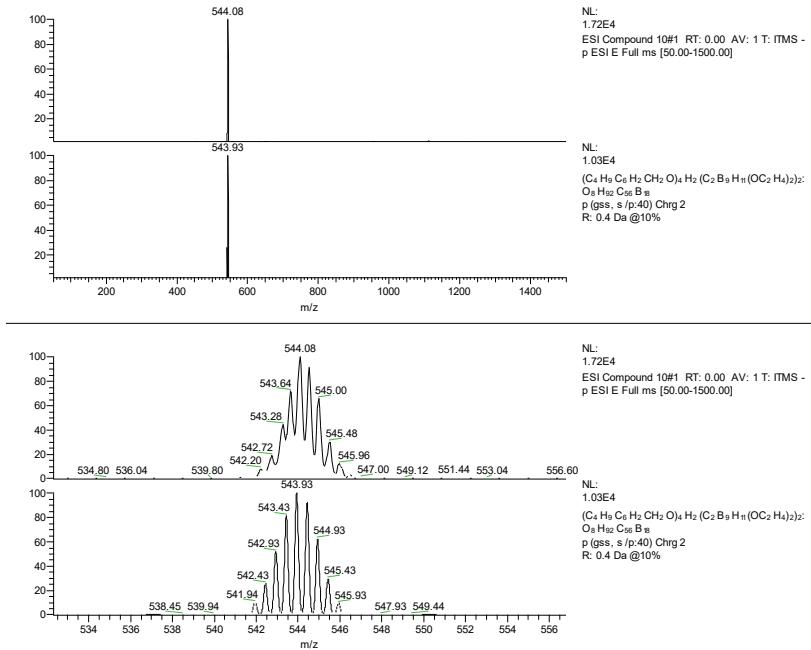


Fig S25. MS spectra of compound 10²⁻

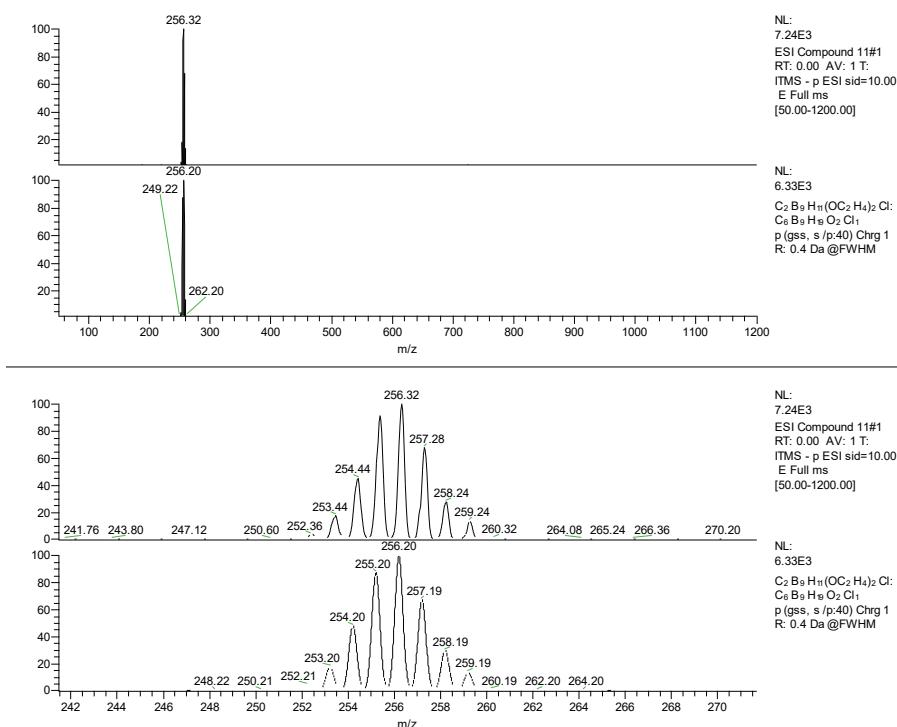


Fig S26. MS spectra of compound 11⁻

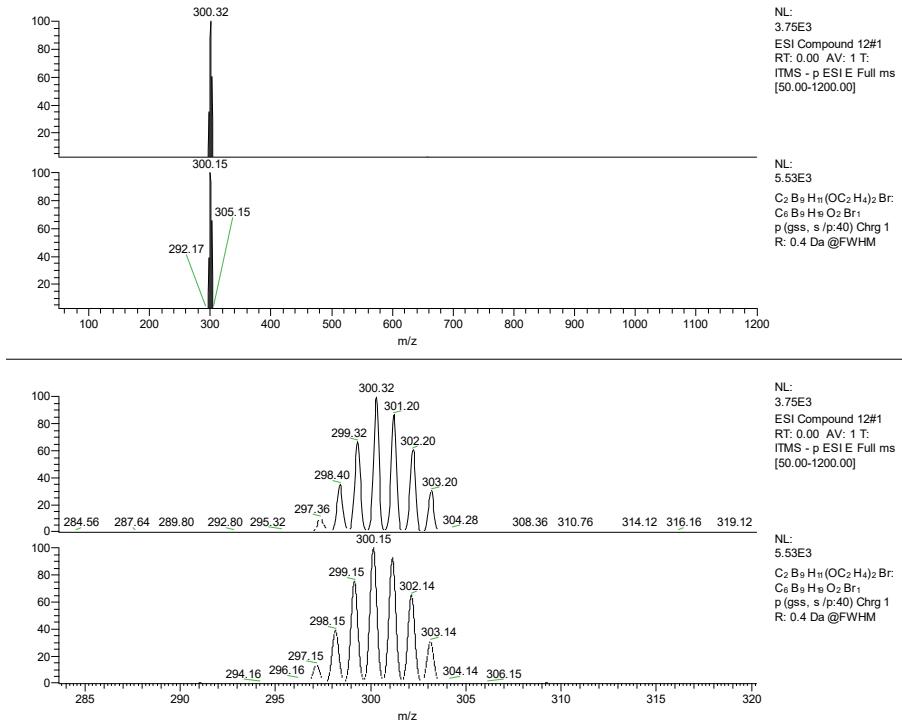


Fig S27. MS spectra of compound 12^-

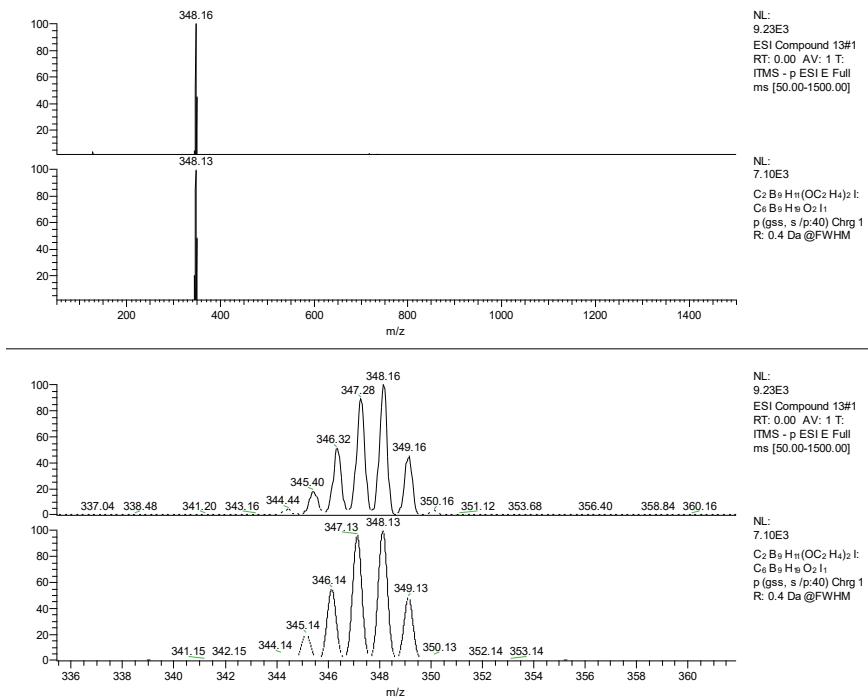


Fig S28. MS spectra of compound 13^-

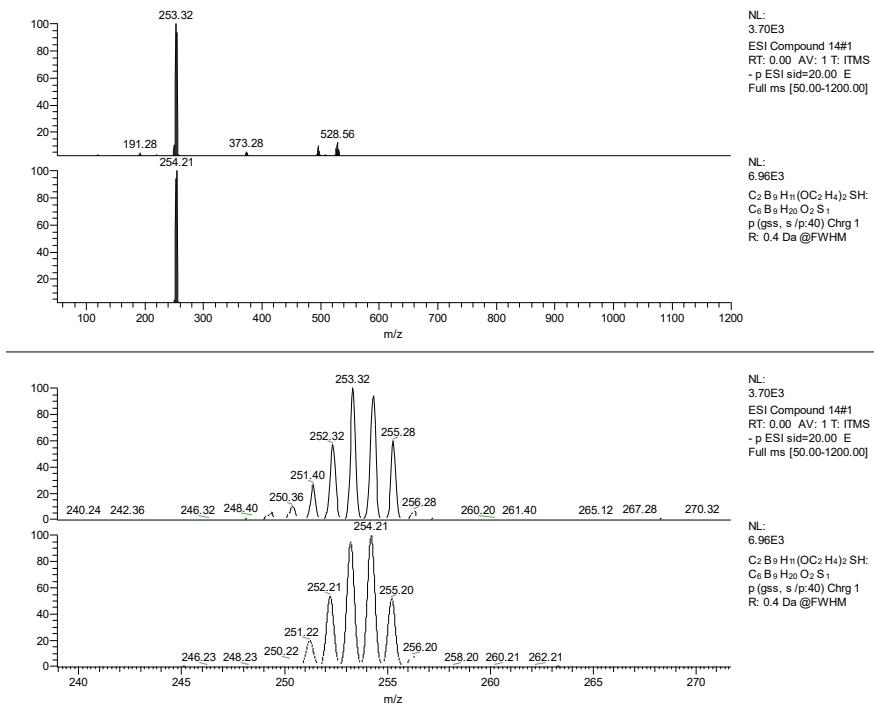


Fig S29. MS spectra of compound 14⁺

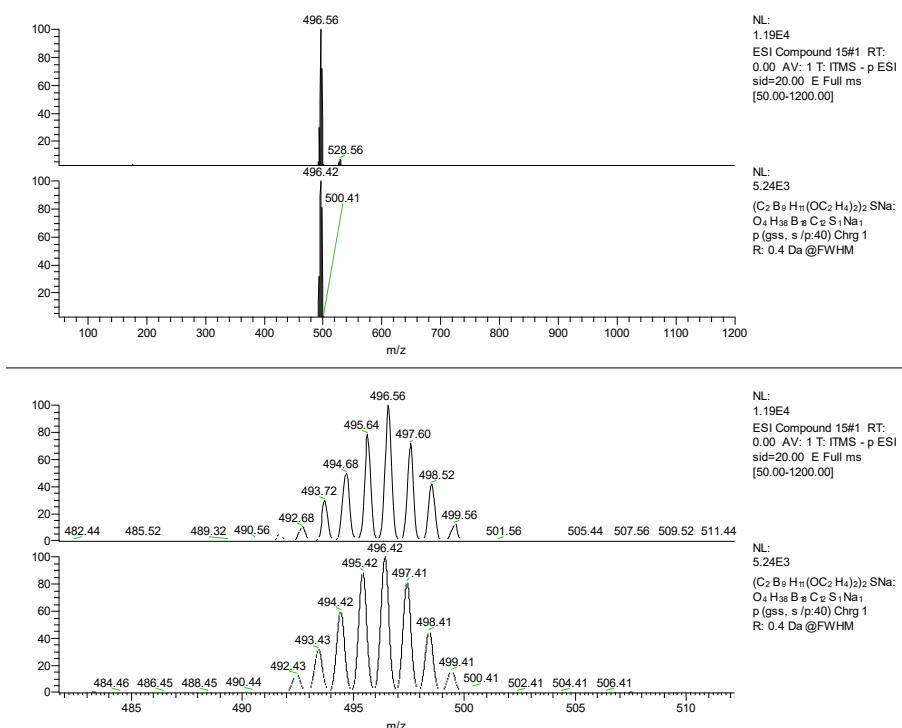


Fig S30. MS spectra of compound 15²⁻