Supplementary Information

Table S1. Kinetic constants for PEP³⁻ in WT-RMPK and in T113L-RMPK in the presence of saturating concentrations of the ADP-Mg complex and of the monovalent cation indicated in 100% water. Kinetic data (not shown) were fitted to the Michaelis-Menten equation ($v = Vmax \cdot S/K_m + S$) (Origin version 7.0). The standard deviations from two experiments are shown. k_{cat}/K_m values are expressed in log form.

N#+	WT			T113L			
IVI	K_m (μ M)	k_{cat} (s ⁻¹)	$\text{Log}(k_{cat}/K_m)$	$K_m (\mu M)$	k_{cat} (s ⁻¹)	$\log(k_{cat}/K_m)$	
K^+	47 ± 6.8	963 ± 39	7.31	108 ± 18	725 ± 49	6.83	
$\mathrm{NH_4}^+$	31 ± 3.6	645 ± 26	7.32	24 ± 3	610 ± 19	7.41	
Rb^+	25 ± 4.2	618 ± 26	7.40	62 ± 6.9	998 ± 34	7.21	
Cs^+	27 ± 1.7	171 ± 2.6	6.80	34 ± 7.3	193 ± 10	6.76	

Table S2. Kinetic constants for the ADP-Mg complex in WT-RMPK and in T113L-RMPK in the presence of saturating PEP^{3–} and the monovalent cation indicated in 100% water. Kinetic data (not shown) were fitted to the Michaelis-Menten equation ($v = Vmax \cdot S/K_m + S$) (Origin version 7.0). The standard deviations from two experiments are shown. k_{cat}/K_m values are expressed in log form.

М/+		WT		T113L			
IVI	K_m (μ M)	$k_{cat} (s^{-1})$	$\log(k_{cat}/K_m)$	K_m (μ M)	k_{cat} (s ⁻¹)	$\text{Log}(k_{cat}/K_m)$	
K^+	379 ± 28	1287 ± 44	6.53	1069 ± 264	831 ± 120	5.89	
$\mathrm{NH_4}^+$	221 ± 30	522 ± 27	6.37	264 ± 25	870 ± 31	6.52	
Rb^+	374 ± 76	892 ± 84	6.38	716 ± 139	686 ± 52	5.98	
Cs^+	305 ± 84	164 ± 19	5.73	1009 ± 260	260 ± 30	5.41	

Table S3. Kinetic constants for monovalent cations in WT-RMPK and in T113L-RMPK in the presence of saturating PEP³⁻ and ADP-Mg complex in 100% water. The data from Figure 2 were fitted to the Hill equation ($v = Vmax \cdot S^n/K_{0.5}^n + S^n$) and to the Michaelis-Menten equation ($v = Vmax \cdot S/K_m + S$) (Origin version 7.0). The standard deviations from two experiments are shown. $k_{cat}/^a K$ values are expressed in log form.

		W	/Τ		T113L				
\mathbf{M}^+	<i>K</i> _{0.5} (mM)	k_{cat} (s ⁻¹)	n	Log (<i>k_{cat}/K_{0.5}</i>)	<i>K</i> (mM) ^{<i>a</i>}	k_{cat} (s ⁻¹)	n	Log (<i>k_{cat}/^a K</i>)	
K^+	11.8 ± 0.9	1068 ± 40	1.4 ± 0.14	4.96	48 ± 10	1036 ± 108	-	4.33	
$\mathrm{NH_4}^+$	4.1 ± 0.4	698 ± 37	1.6 ± 0.25	5.23	7.84 ± 0.8	835 ± 39	1.37 ± 0.16	5.03	
Rb^+	12 ± 0.8	617 ± 21	1.46 ± 0.12	4.71	32 ± 3.7	1111 ± 57	-	4.54	
Cs^+	29.7 ± 8.4	240 ± 36	1.36 ± 0.28	3.91	26 ± 4.6	302 ± 32	-	4.06	

^{*a*} K represents the $K_{0.5}$ and K_m for the data fitted to the Hill and Michaelis-Menten equations, respectively.

Table S4. Kinetic constants for PEP³⁻ in WT-RMPK and in T113L-RMPK in the presence of saturating concentrations of the ADP-Mg complex and of the monovalent cation indicated in 20% DMSO. Kinetic data (not shown) were fitted to the Hill $(v = Vmax \cdot S^n/K_{0.5^n} + S^n)$ and Michaelis-Menten $(v = Vmax \cdot S/K_m + S)$ equations (Origin version 7.0). The standard deviations from two experiments are shown. $k_{cat}/^a K_{app}$ values are expressed in log form.

M ⁺		W	/T			T113L		
	K_{app} (μ M) ^a	k_{cat} (s ⁻¹)	п	Log	<i>К_{т(арр)}</i> (ш М)	k_{cat} (s ⁻¹)	Log	
V^+	24 ± 1.1	959 ± 17	1.62 ± 0.12	(<i>R_{cat}/ <i>R_{app}</i>)</i>	(μN)	714 ± 46	$(\kappa_{cat}, \kappa_{m(app)})$	
ĸ	24 ± 1.1	636 ± 17	1.02 ± 0.12	7.30	44 ± 1.8	$/14 \pm 40$	1.22	
$\mathrm{NH_{4}^{+}}$	11.2 ± 1.5	439 ± 23	1.61 ± 0.34	7.59	28 ± 3.8	776 ± 30	7.44	
Rb^+	13.8 ± 0.9	548 ± 12	1.87 ± 0.24	7.6	30 ± 3.4	616 ± 17	7.31	
Cs^+	2.9 ± 0.6	162 ± 4	-	7.75	5.2 ± 0.98	204 ± 5	7.59	

^{*a*} K_{app} represents the $K_{0.5(app)}$ and $K_{m(app)}$ for the data fitted to the Hill and Michaelis-Menten equations, respectively.

Table S5. Kinetic constants for the ADP-Mg complex in WT-RMPK and in T113L-RMPK in the presence of saturating PEP³⁻ and the monovalent cation indicated in 20% DMSO. Kinetic data (not shown) were fitted to the Michaelis-Menten equation $(v = Vmax \cdot S/K_{m(app)} + S)$ (Origin version 7.0). The standard deviations from two experiments are shown. $k_{cat}/K_{m(app)}$ values are expressed in log form.

M ⁺ -		WT		T113L			
	$K_{m(app)}$ (μ M)	k_{cat} (s ⁻¹)	$\text{Log}(k_{cat}/K_{m(app)})$	$K_{m(app)}$ (μ M)	k_{cat} (s ⁻¹)	$\text{Log}(k_{cat}/K_{m(app)})$	
K^+	99.7 ± 13.2	993 ± 39	7.00	252 ± 30	1131 ± 60	6.65	
$\mathrm{NH_4}^+$	71 ± 15	472 ± 23	6.82	124 ± 19	579 ± 26	6.67	
Rb^+	83 ± 11	674 ± 28	6.91	156 ± 29	621 ± 39	6.6	
Cs^+	44 ± 14	174 ± 10	6.6	73 ± 13	232 ± 10	6.5	

Table S6. Kinetic constants for monovalent cations in WT-RMPK and in T113L-RMPK in the presence of saturating PEP³⁻ and ADP-Mg complex in 20% DMSO. The data from Figure 3 were fitted to the Michaelis-Menten ($v = Vmax \cdot S/K_m + S$) and to the Hill ($v = Vmax \cdot S^n/K_{0.5}^n + S^n$) equations (Origin version 7.0). The standard deviations from two experiments are shown. $k_{cat}/^a K_{app}$ values are expressed in log form.

		W	Τ		T113L				
\mathbf{M}^+	Kapp (mM) ^a	k_{cat} (s ⁻¹)	n	Log	K 0.5(app)	$k_{cat}(s^{-1})$	п	Log	
				(kcat/ ^a Kapp)	(mM)			$(k_{cat}/K_{0.5(app)})$	
\mathbf{K}^+	1.5 ± 0.18	889 ± 36	-	5.76	1.8 ± 0.4	587 ± 78	1.15 ± 0.10	5.50	
$\mathrm{NH_4^+}$	0.6 ± 0.09	432 ± 33	1.3 ± 0.19	5.84	0.92 ± 0.1	425 ± 33	1.81 ± 0.27	5.66	
Rb^+	1.4 ± 0.23	640 ± 50	1.18 ± 0.16	5.67	1.8 ± 0.22	548 ± 35	1.3 ± 0.1	5.49	
Cs^+	3.8 ± 0.6	198 ± 16	-	4.72	7.3 ± 2	216 ± 38	1.32 ± 0.19	4.47	

^a K_{app} represents the $K_{0.5(app)}$ and $K_{m(app)}$ for the data fitted to the Hill and Michaelis-Menten equations, respectively.

Figure S1. Stick model of the water-protein and water-water interactions in the vicinity of O^{γ} of T113 of pyruvate kinase from rabbit muscle. The dashed lines show hydrogen bond distances. Water molecules (red spheres) are illustrated. This figure was constructed from the coordinates of PDB 2G50 [10] using the program, PYMOL v0.99.

