



Supplementary materials

Magnetic carbon nanocomposites from Fe₃O₄ reduction and their application as Cr (VI) adsorbents

Laura Cervera-Gabaldá ^{1,2} and Cristina Gómez-Polo ^{1,2,*}

¹ Departamento de Ciencias, Universidad Pública de Navarra, Campus de Arrosadia, 31006 Pamplona, Spain

² Institute for Advanced Materials and Mathematics (INAMAT2), Universidad Pública de Navarra, Campus de Arrosadia, 31006, Pamplona, Spain

* Correspondence: gpolo@unav.es; Tel.: +34-948169576

Supplementary Figures

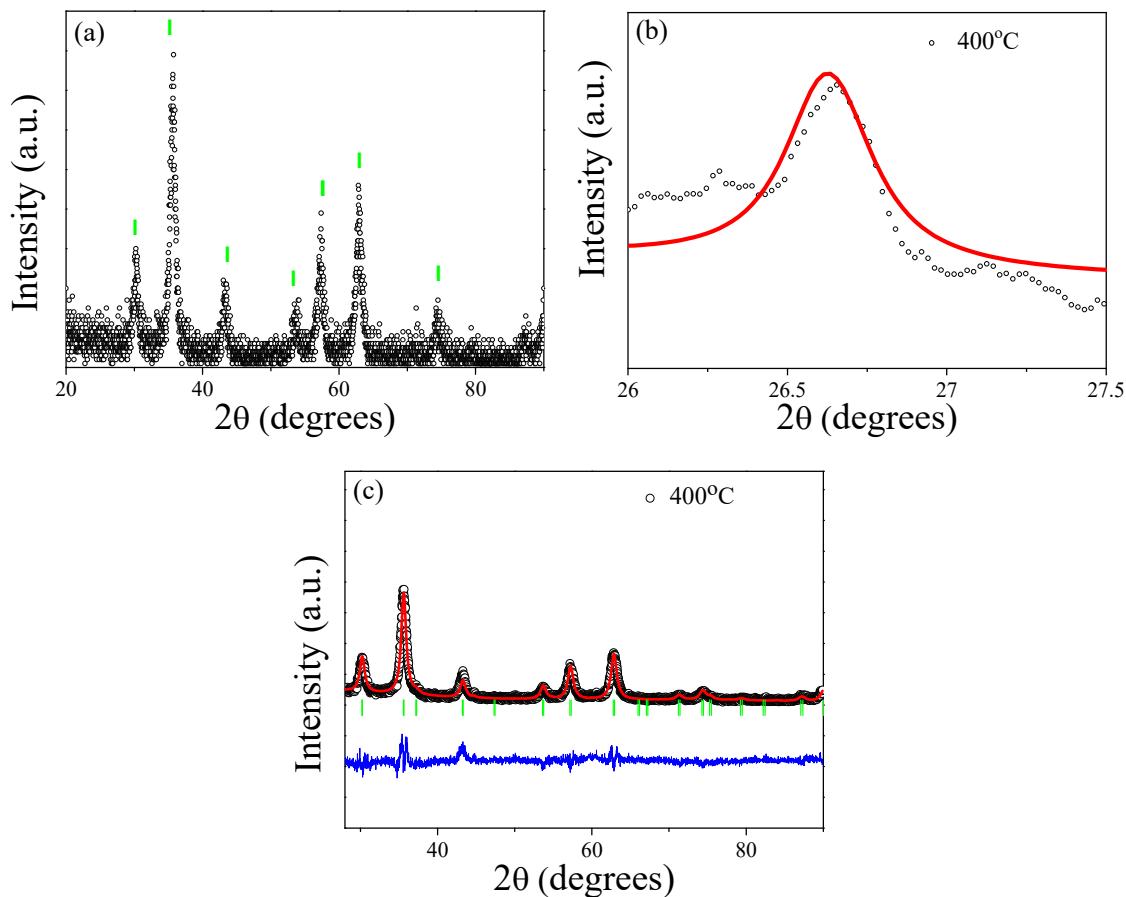


Figure S1. XRD patterns for the (a) Fe₃O₄ initial MNPs and (b,c) MNPs + fructose sample at $T_{ann} = 400^{\circ}\text{C}$: (○) Experimental, (—) calculated (Rietveld refinement) intensities and (—) difference between both intensities. The Bragg reflections are shown for (|) Fe₃O₄.

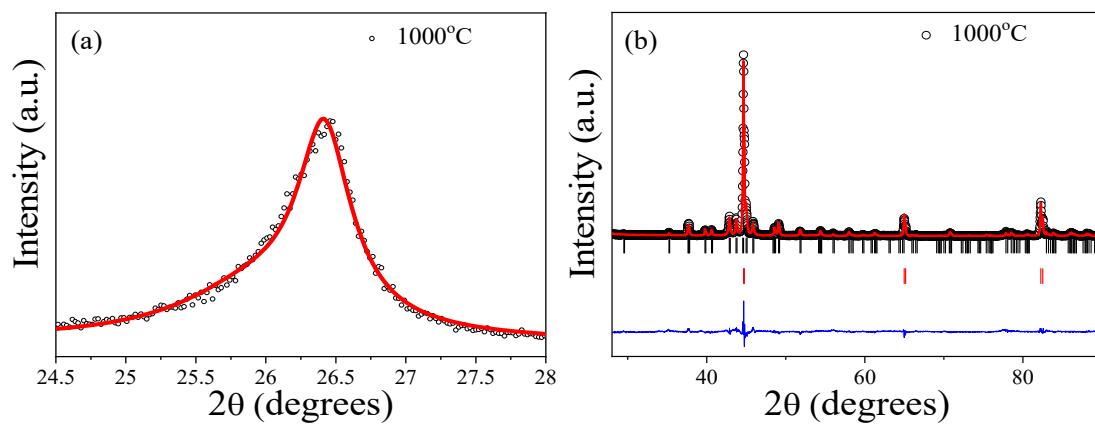


Figure S2. XRD patterns for the MNPs + fructose sample annealed at $T_{ann} = 1000^{\circ}\text{C}$. (○) Experimental, (—) calculated (Rietveld refinement) intensities and (—) difference between both intensities. The Bragg reflections are shown for (|) Fe_3C and (|) $\alpha\text{-Fe}$.

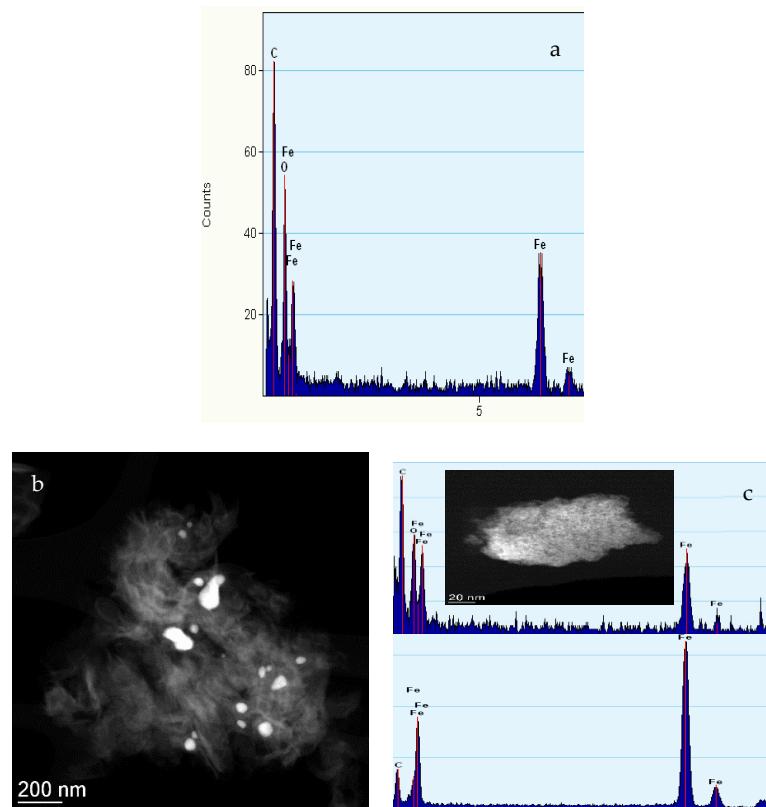


Figure S3. (a,c) EDX analysis for the MNPs + fructose sample annealed at $T_{ann} = 400$ and 600°C , respectively. (b) STEM image of the MNPs + fructose sample annealed at $T_{ann} = 600^{\circ}\text{C}$.

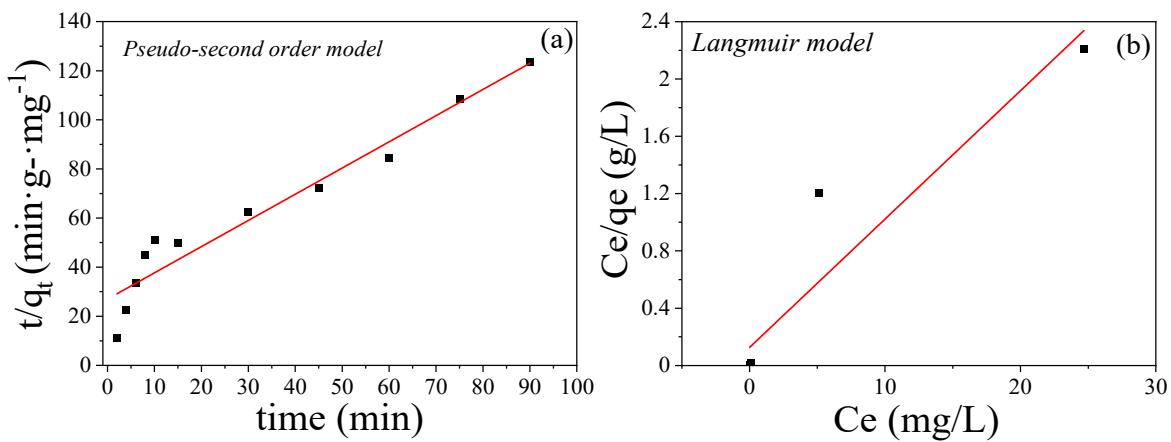


Figure S4. (a) Adsorption kinetics of Cr(VI) and (b) adsorption isotherms in the presence of the initial Fe₃O₄ MNPs.