

Wireless, Flexible, Ion-Selective Electrode System for Selective and Repeatable Detection of Sodium

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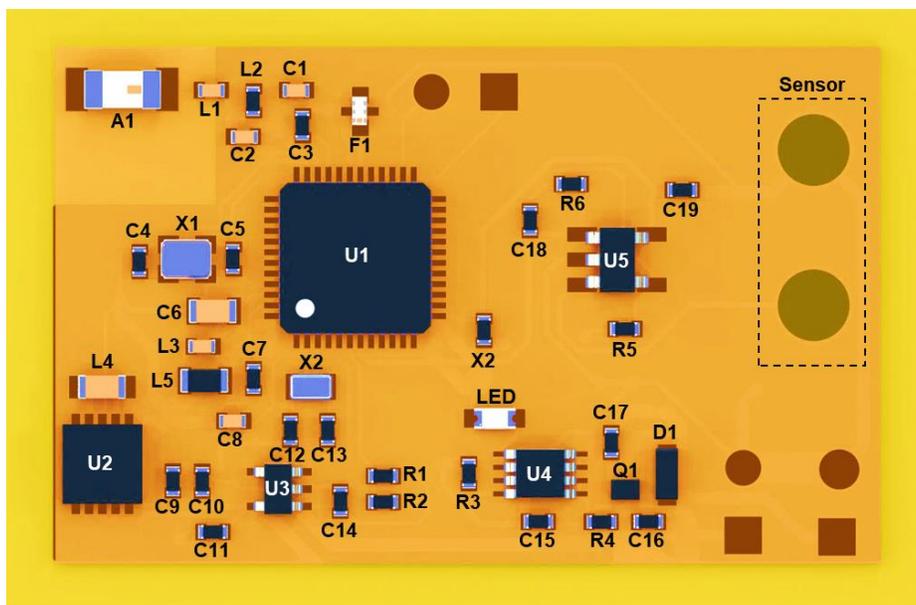


Figure S1. Illustration of wireless ion-selective sodium sensor circuit components with detail list of the surface mount chip components in Table S1. The sensor includes an ion-selective electrode (ISE) and an Ag/AgCl reference electrode (RE), which are directly formed onto Cu-pads of the circuit.

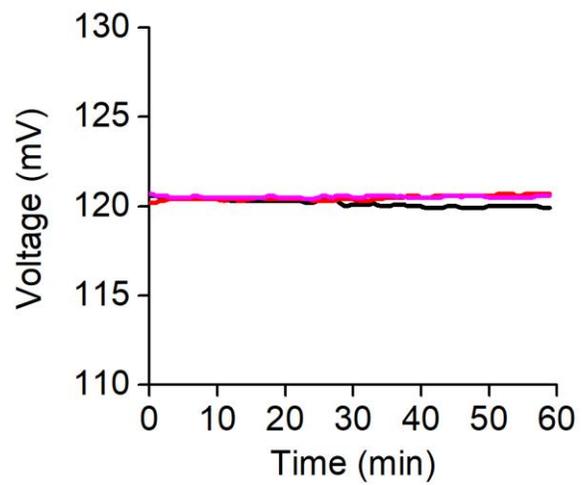


Figure S2. Voltage stability (0.53 mV/h; n = 3) of the all-solid-state film sodium ISE in 10^{-2} M sodium chloride.

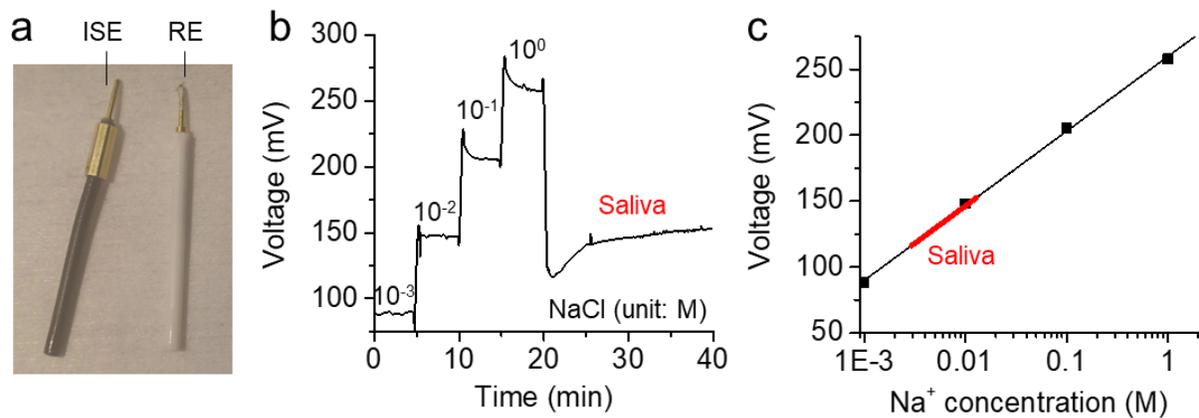


Figure S3. Saliva concentration measured with using a commercial all-solid-state ISE and RE. (a) Photograph of the commercial sodium ISE and RE. (b) Voltage response of the sensor in NaCl solutions and Saliva. (c) Saliva concentration shows 3-13 mM Na^+ (highlighted in red on the calibration line with a slope of 56 mV/decade).

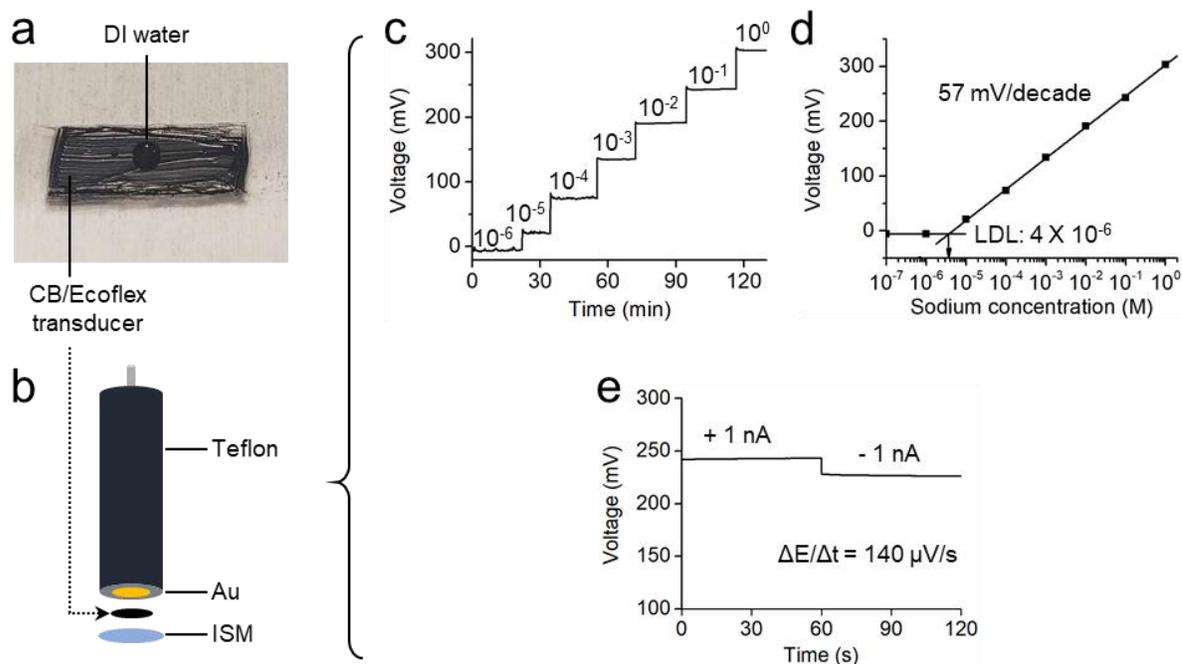


Figure S4. Sensing capabilities of a CB/Ecoflex-based ISE formed on a standard Teflon-body electrode. (a) Photograph showing a hydrophobicity of the CB/Ecoflex composite, ensuring a good contact with the PVC ISM. (b) Schematic diagram of the structure of CB/Ecoflex/ISM formed on an Au disc shrouded by Teflon. Sensing capabilities measured with using the ISE (b): (c) Real-time voltage response in 10^{-6} to 10^0 sodium chloride solutions, (d) calculated sensitivity and low detection limit (LDL), and (e) voltage stability measured by chronopotentiometry in 10^{-2} sodium chloride.

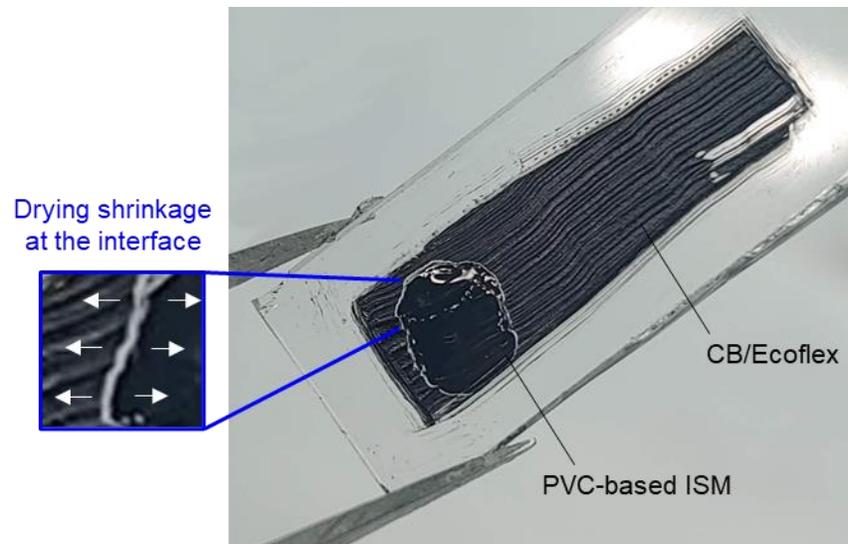


Figure S5. Photograph showing a drying shrinkage at the interface between a CB/Ecoflex transducer and ISM that contains THF, resulted in a delamination.

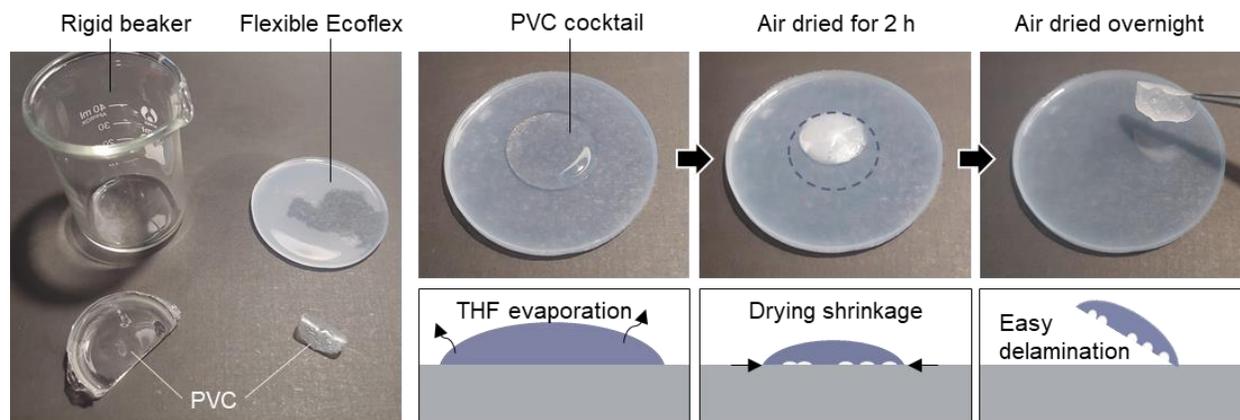


Figure S6. Photographs showing a drying shrinkage of PVC formed on the surface of Ecoflex compared to a rigid beaker. While the PVC cocktail in the beaker was dried stuck at the surface of the rigid glass, the film formed on the flexible substrate shows a significant shrinkage and pores.

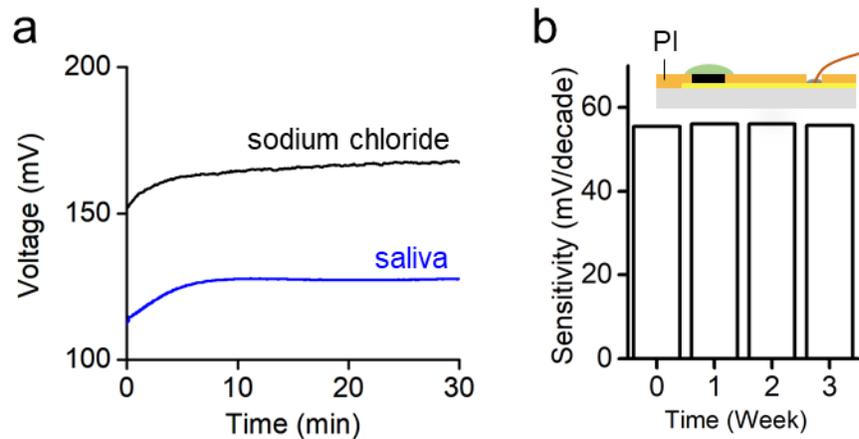


Figure S7. Short-term and long-term stable voltage reading of an all-solid-state film ISE enabled by efficient chemical insulation. (a) Voltage reading of the sensor in 10^{-1} sodium chloride solution and saliva. (b) Long-term stability of stability for three weeks.

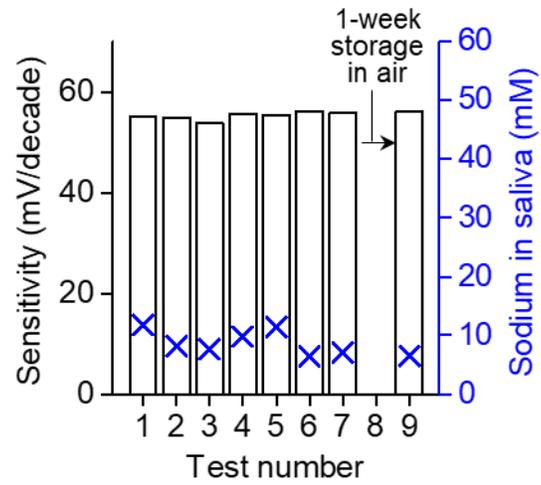


Figure S8. Sensitivity and applicability of sodium detection of our film ISE during 1-week storage in air. The 7 times measurement was conducted continuously.

Table S1. List of surface mount components used in wireless ion-selective sodium sensor circuit.

Component	Description	Value	Part number
U1	Bluetooth PSoC	N/A	NRF52832-QFAA-R
U2	3.3 voltage regulator	N/A	TPS63001
U3	Current limit active-low load switch	N/A	TPS22941
U4	Battery recharge	N/A	MCP73831
U5	Single channel Op-amp IC	N/A	AD8603AUJZREEL7
F1	2.45 GHz low pass filter	N/A	2450FM07A0029
A1	2.45 GHz RF chip antenna	N/A	2450AT18A100
D1	Schottky Diode	N/A	641-1285-1-ND
Q1	P-MOSFET	N/A	DMP21D5UFB4-7BDICT-ND
X1	32 MHz crystal	N/A	ECS-320-8-37CKM
X2	32.768 kHz crystal	N/A	ECS-327-9-12-TR
L1	0402 inductor	3.9 nH	N/A
L2	0402 inductor	2.7 nH	N/A
L3	0402 inductor	15 nH	N/A
L4	0603 inductor	2.2 μ H	N/A
L5	0603 inductor	10 μ H	N/A
C1	0402 ceramic capacitor	1.0 pF	N/A
C2, C8	0402 ceramic capacitor	100 nF	N/A
C3	0402 ceramic capacitor	100 pF	N/A
C4, C5, C12, C13	0402 ceramic capacitor	12 pF	N/A
C6	0603 ceramic capacitor	1.0 μ F	N/A
C7, C15, C17	0402 ceramic capacitor	4.7 μ F	N/A
C9	0402 ceramic capacitor	22 μ F	N/A
C10, C11, C14, C18	0402 ceramic capacitor	10 μ F	N/A
C16	0402 ceramic capacitor	1.0 μ F	N/A
C19	0402 ceramic capacitor	10 nF	N/A
R1, R2	0402 resistor	1 M Ω	N/A
R3	0402 resistor	2 k Ω	N/A
R4	0402 resistor	100 k Ω	N/A
R5, R6	0402 resistor	10 k Ω	N/A
Battery	Lithium-ion polymer		

Table S2. Comparison of sensing capabilities based on carbon-polymer composite transducers.

Year ^{Reference}	Transducer	Sensitivity mV/decade)	Range (M)	Ion
This work ²⁰²⁰	Carbon black/Ecoflex	57.0	10 ⁻⁶ to 1	Na ⁺
[1] ²⁰⁰⁹	CNT/block polymer	58.6	10 ⁻⁷ to 10 ⁻¹	Na ⁺
[2] ²⁰¹⁰	CNT/block polymer/ISM	58.2	10 ⁻⁶ to 10 ⁻¹	Na ⁺
[3] ²⁰¹⁹	CNT/copolymer	56.3	10 ⁻⁷ to 10 ⁻¹	K ⁺
[4] ²⁰¹⁷	CNT/porphyrinoids	60.5	10 ⁻⁷ to 10 ⁻¹	K ⁺
[5] ²⁰¹⁵	Graphene/AgTFPB	56.5	10 ⁻⁵ to 10 ⁻²	K ⁺

References

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