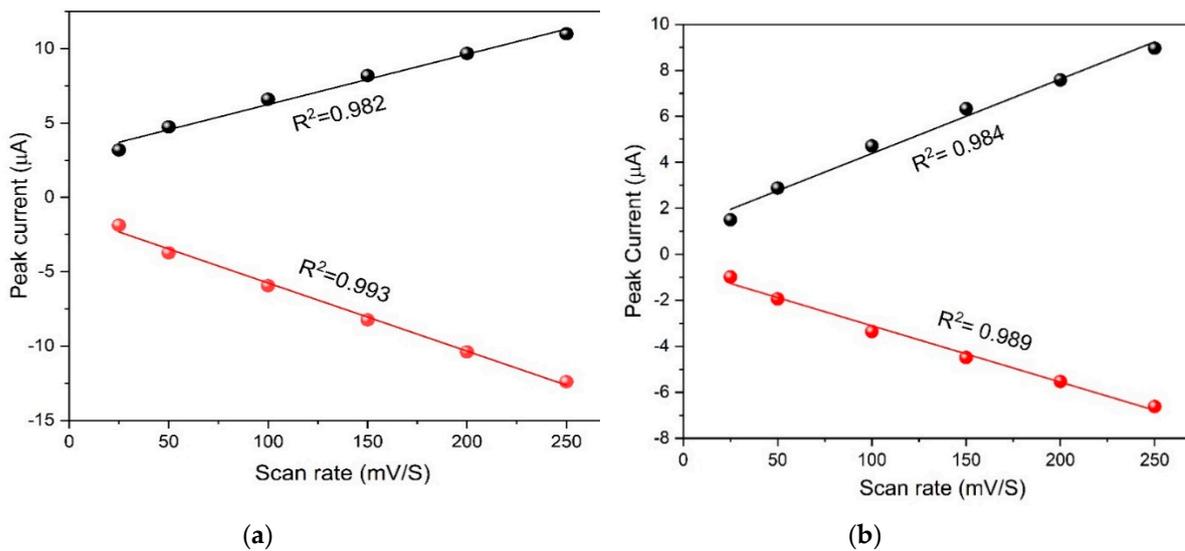
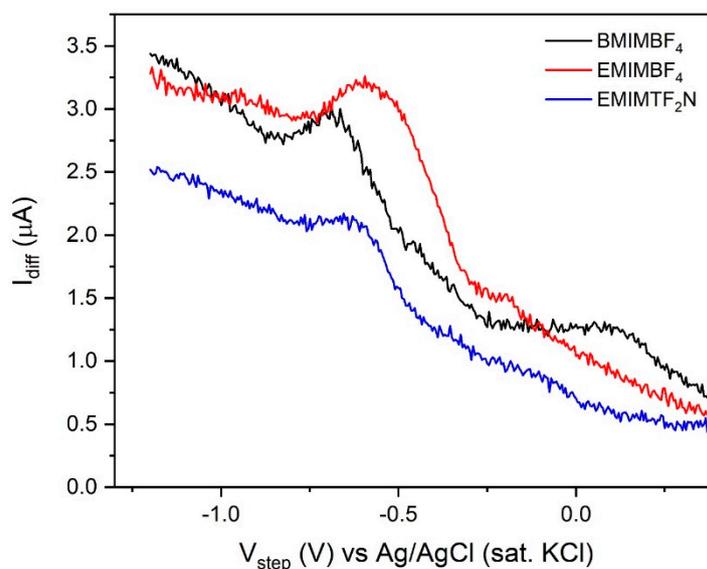


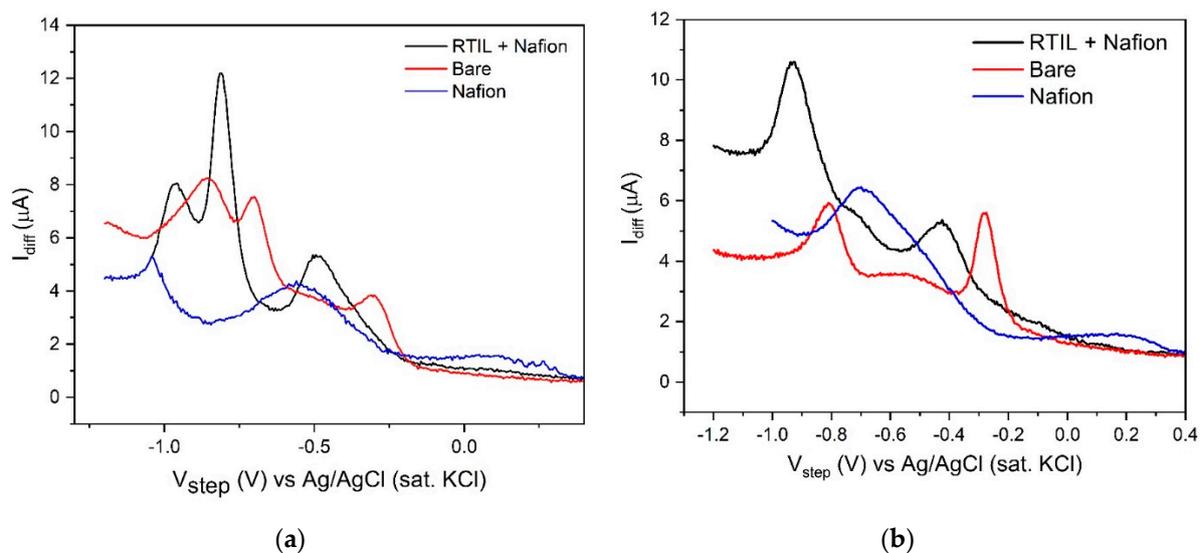
**Figure S1.** Cyclic voltammetry (CV) output of room-temperature ionic liquid (RTIL)-modified glassy carbon electrode (GCE) without analyte.



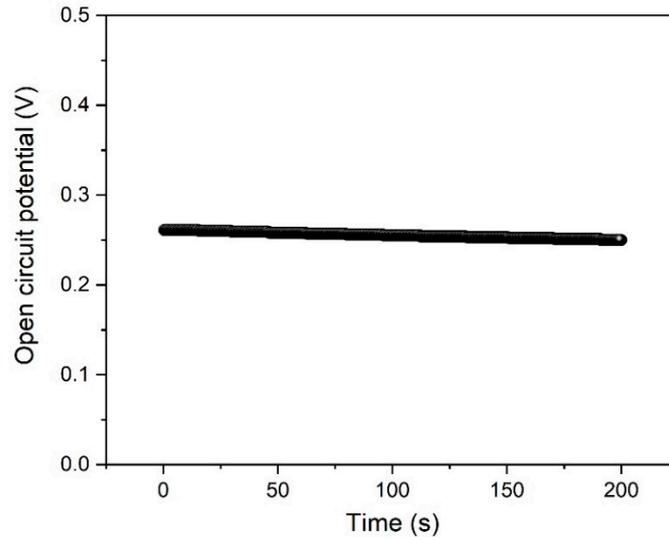
**Figure S2.** (a) The peak current vs. scan rate with  $R^2$  for 2,6-dinitrotoluene (2,6 DNT); (b) The peak current vs. scan rate with  $R^2$  for ethylnitrobenzene (ENB).



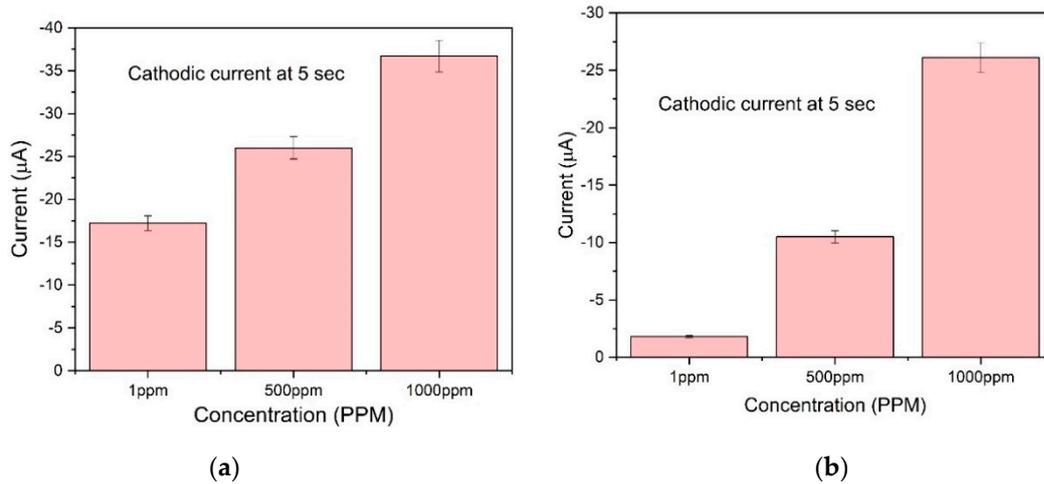
**Figure S3.** Square wave voltammetry (SQWV) was performed using different RTILs in the absence of the target analyte to obtain the baseline peak current. SQWV parameter: frequency 25 Hz, amplitude of 25 mV and step size of 5 mV.



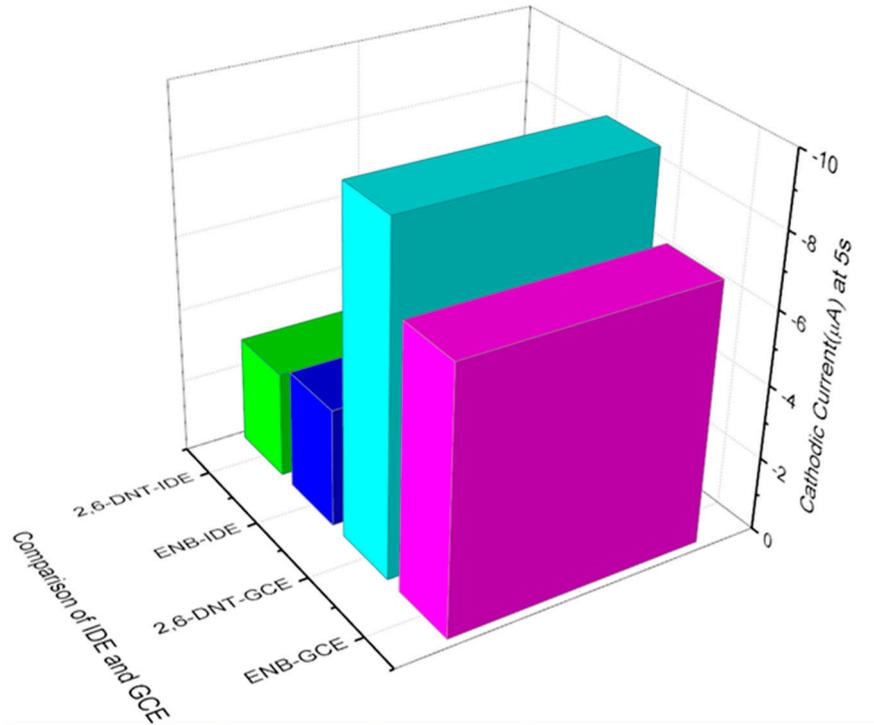
**Figure S4.** (a) SQWV output of RTIL ([EMIM][BF<sub>4</sub>)]-nafion modified GCE, bare electrode and nafion-GCE for detection of 100 ppm 2,6 DNT; (b) SQWV output of RTIL ([EMIM][BF<sub>4</sub>)]-nafion modified GCE, bare electrode and nafion-GCE for detection of 100 ppm ENB.



**Figure S5.** Open circuit potential measurement to show the stability of the fabricated sensor. The sensor is stable for over 200 s with the OCP in the lower millivolt regime.



**Figure S6.** (a) Calibration dose response plotted for 2,6 DNT concentrations of 1, 500 and 1000 ppm in terms of steady state current at 5 sec; (b) Calibration dose response plotted for ENB concentrations of 1, 500 and 1000 ppm in terms of steady state current at 5 sec.



**Figure S7.** Comparison of the current obtained at 5 s for interdigitated electrode (IDE) vs. GCE for both the target analytes.



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