## Evaluation of *Matricaria aurea* Extracts as Effective Anticorrosive Agent for Mild Steel in 1.0 M HCl and Isolation of Their Active Ingredients

Merajuddin Khan<sup>1\*</sup>, Mahmood M. S. Abdullah<sup>1</sup>, Adeem Mahmood<sup>1</sup>, Abdullah M. Al-Mayouf<sup>2</sup>, Hamad Z. Alkhathlan<sup>1\*</sup>

<sup>1</sup>Department of Chemistry, College of Science, King Saud University, P.O.Box 2455, Riyadh -11451, Saudi Arabia.

<sup>2</sup> Electrochemical Sciences Research Chair, Department of Chemistry, College of Science, King Saud University, P.O.Box 2455, Riyadh - 11451, Saudi Arabia.

\*To whom correspondence should be addressed: *E-mail: mkhan3@ksu.edu.sa* (M. Khan), *khathlan@ksu.edu.sa* (H. Z. Alkhathlan), Tel: +966-1-4675910, Fax: +966-1-4675992.



**Fig. S1.** Tafel plots in absence and presence of 600 ppm of MeOH, mixture and water extracts of *M. aurea* in 1.0 M HCl.

**Table-S1:** Potentiodynamic polarization parameters obtained from Tafel plots for the corrosion of mild steel in 1.0 M HCl with 600 ppm of various extracts of *M. aurea* 

Inhibitors	E <sub>corr</sub> (mV)	<i>I<sub>corr</sub></i> (μAcm <sup>-2</sup> )	$\beta_a$ (mV/dec)	$\beta_c$ (mV/dec)	$R_p$	<i>IE</i> % Tafel	<i>IE</i> % LPR
Blank	-486.6	213.0	99.85	-110.73	54.5	-	-
МеОН	-481.58	20.0	82.29	-73.57	401.69	90.61	86.43
Mixture MeOH: H <sub>2</sub> O (85:15)	-474.32	26.0	80.31	-75.8	351.79	87.79	84.51
Water	-482.16	35.0	87.42	-74.12	271.26	83.57	79.91



**Fig. S2.** Nyquist plots in absence and presence of 600 ppm of MeOH, mixture and water extracts of *M. aurea* in 1.0 M HCl.

**Table-S2:** Electrochemical impedance parameters obtained from Nyquist plots for mild steel in 1.0 M HCl with 600 ppm of various extracts of *M. aurea* 

Inhibitors	$R_{ct}$ ( $\Omega$ cm <sup>2</sup> )	$C_{dl}$ (µF cm <sup>-2</sup> )	θ	IE %
Blank	57.1	533.0	-	-
MeOH	412.7	121.0	0.86	86.16
Mixture MeOH: H <sub>2</sub> O (85:15)	372.25	170.0	0.85	84.66
Water	302.0	200.0	0.81	81.09



**Fig. S3.** Tafel plots in absence and presence of *n*-BuOH and EtOAc extracts of *M. aurea* in 1.0 M HCl.

**Table-S3:** Potentiodynamic polarization parameters obtained from Tafel plots for the corrosion of mild steel in 1.0 M HCl with 600 ppm of *M.aurea* various extracts.

Extracts	E <sub>corr</sub>	I <sub>corr</sub>	$\beta_a$	$\beta_c$	$R_p$	E(%)	E(%)
	(mV)	$(\mu Acm^{-2})$	(mV/dec)	(mV/dec)	-	Tafel	LPR
1M HCl	-486.6	213.0	99.85	-110.73	54.5	-	-
<i>n</i> -BuOH extract	-468.62	14.7	80.61	-80.31	585.62	93.10	90.69
EtOAc extract	-478.65	27.8	82.45	-78.59	334.37	86.95	83.70



**Fig. S4.** Nyquist plots in absence and presence of *n*-BuOH and EtOAc extracts of *M.aurea* in 1.0 M HCl.

**Table-S4:** Electrochemical impedance parameters obtained from Nyquist plots for mild steel in 1.0 M HCl with 600 ppm of *M.aurea* various extracts.

Extract	$R_{ct}$ ( $\Omega$ cm <sup>2</sup> )	$C_{dl}$ (µF cm <sup>-2</sup> )	θ	E (%)
1M HCl	57.10	533.0	-	-
<i>n</i> -BuOH extract	580.6	98.0	0.91	90.17
EtOAc extract	364.0	158.0	0.84	84.31



Fig. S5. Tafel plots in absence and presence of various fractions of *n*-BuOH extracts of *M. aurea* in 1.0 M HCl.

**Table-S5:** Potentiodynamic polarization parameters obtained from Tafel plots for the corrosion of mild steel in 1.0 M HCl with 600 ppm of various fractions of *M. aurea n*-BuOH extarcts.

Fractions	E <sub>corr</sub>	I <sub>corr</sub>	$\beta_a$	$\beta_c$	$R_p$	IE %	IE %
	(mV)	$(\mu A cm^{-2})$	(mV/dec)	(mV/dec)	F	Tafel	LPR
Blank	-486.6	213.0	99.85	-110.73	54.5	-	-
11-13	-484.77	41.0	74.97	-67.91	251.99	80.75	78.37
15-19	-481.75	31.0	70.48	-68.19	274.41	85.45	80.14
28-34	-511.52	19.24	68.33	-68.32	395.29	90.97	86.21
41-81	-477.19	30.0	71.99	-69.26	294.31	85.92	81.48



**Fig. S6.** Nyquist plots in absence and presence of various fractions of *n*-BuOH extracts of *M. aurea* in 1.0 M HCl.

**Table-S6:** Electrochemical impedance parameters obtained from Nyquist plots for mild steel in 1.0 M HCl with 600 ppm of various fractions of *M. aurea n*-BuOH extarcts.

Fractions	$R_{ct}$ ( $\Omega$ cm <sup>2</sup> )	$C_{dl}$ (µF cm <sup>-2</sup> )	θ	IE %
Blank	57.1	533.0	-	-
11-13	265.1	200	0.78	78.46
15-19	284.15	185	0.80	79.90
28-34	436	171.5	0.87	86.90
41-81	320.3	200.0	0.82	82.17



Fig. S7a. <sup>1</sup>H NMR spectrum of apigenin-7-O- $\beta$ -D-glucoside (MAB) in DMSO- $d_6$ .



Fig. S7b. Expanded <sup>1</sup>H NMR spectrum of apigenin-7-O- $\beta$ -D-glucoside (MAB) in DMSO- $d_6$ .



Fig. S8a. <sup>13</sup>C NMR spectrum of apigenin-7-O- $\beta$ -D-glucoside (MAB) in DMSO- $d_6$ .



Fig. S8b. Expanded <sup>13</sup>C NMR spectrum of apigenin-7-O-β-D-glucoside (MAB) in DMSO-d<sub>6</sub>.



Fig. S9. DEPT spectrum of apigenin-7-O- $\beta$ -D-glucoside (MAB) in DMSO- $d_6$ .



Fig. S10. COSY spectrum of apigenin-7-O- $\beta$ -D-glucoside (MAB) in DMSO- $d_6$ .



Fig. S11a. HMQC spectrum of apigenin-7-O-β-D-glucoside (MAB) in DMSO-d<sub>6</sub>.



Fig. S11b. Expanded HMQC spectrum of apigenin-7-O-β-D-glucoside (MAB) in DMSO-d<sub>6</sub>.



Fig. S12a. HMBC spectrum of apigenin-7-O-β-D-glucoside (MAB) in DMSO-d<sub>6</sub>.



Fig. S12b. Expanded HMBC spectrum of apigenin-7-O-β-D-glucoside (MAB) in DMSO-d<sub>6</sub>.



Fig. S12c. Expanded HMBC spectrum of apigenin-7-O-β-D-glucoside (MAB) in DMSO-d<sub>6</sub>.