## Supplementary Materials

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Assignment

C-OH/C-OR

Carbonate

C-C/C-H

C=O

BE(eV)

285.0

286.5

287.9

289.8

1

2

3

4



В

		BE(eV)	Assignment	Peak Area %
	1	285.0	C-C/C-H	28.3
	2	286.3	C-OH/C-OR	33.5
	3	287.4	C=0	15.8
	4	288.8	C(O)OH/C(O)OR	8.0
	5	289.8	Carbonate	14.4

*Figure S1*. C1s spectra for the catalyst La(III)@KIT-6 before (A) and after (B) three recycling experiments.

Peak Area %

41.3

21.8

7.5

29.4



Figure S2. Survey spectrum of recovered La(III)@KIT-6 after three recycling experiments.



Figure S3. N2 adsorption-desorption isotherm for KIT-6



Figure S4. N<sub>2</sub> adsorption–desorption isotherm for La(III)O-KIT-6 catalyst  ${\bf 2}$ 



Figure S5. N2 adsorption-desorption isotherm for La(III)O-KIT-6 after III cycle (2')



Figure S6. SEM image 100 µm of KIT-6.



Figure S7. SEM image 20 µm of KIT-6.



Figure S8. SEM image 100  $\mu m$  of La(III)OKIT-6.



Figure S9. SEM image 100  $\mu$ m of La(III)OKIT-6 after three recycling experiments.



Figure S10. SEM image 20 µm of La(III)OKIT-6 after three recycling experiments.



*Figure S11.* HPLC chromatograms at 0 h (a), 8 h (b), 16 h (c) and 24 h (d) reaction time. C18 Column Supelcosil (15 cm  $\times$  4.6 mm id), a solvent program starting with H<sub>2</sub>O to Methanol for 40 minutes, and a flow rate of 1.0 mL/min, IT-TOF detector.



*Figure S12.* GC/MS analysis of the 1<sup>st</sup> catalytic run with La(III)OKIT-6.



*Figure S13.* GC/MS analysis of 2<sup>nd</sup> catalytic run with La(III)OKIT-6.



*Figure S14*. GC/MS analysis of 3<sup>rd</sup> catalytic run with La(III)OKIT-6.



Figure S15. HRMS of linear and cyclic hybrid oligomers.



*Figure S16.* <sup>1</sup>H-NMR (300 MHz, DMSO-d6) spectrum of reaction mixture at 24 hours.



*Figure S17.* <sup>13</sup>C-NMR (75 MHz, DMSO-d6) spectrum of reaction mixture at 24 hours.