

# SUPPLEMENTARY MATERIAL

## Degradation of meropenem by heterogeneous photocatalysis using TiO<sub>2</sub>/fiberglass substrates

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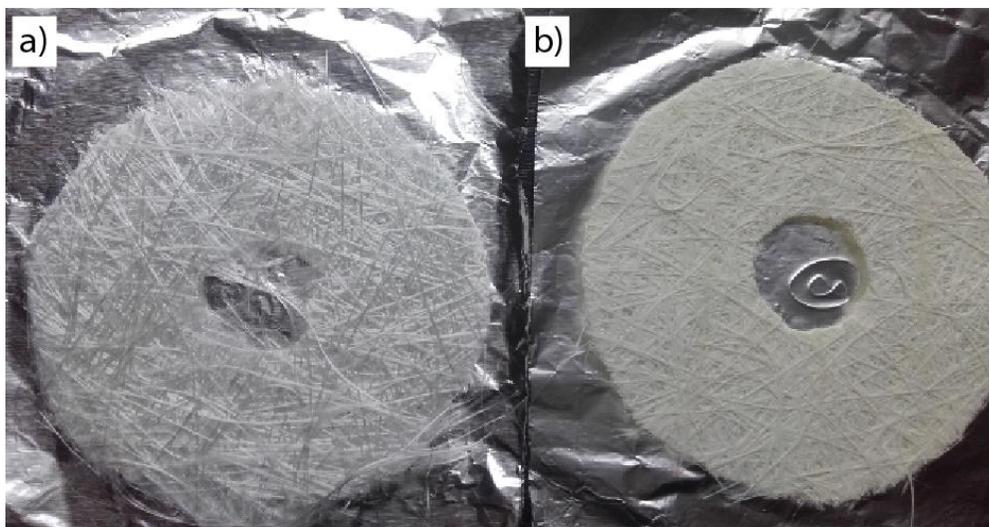
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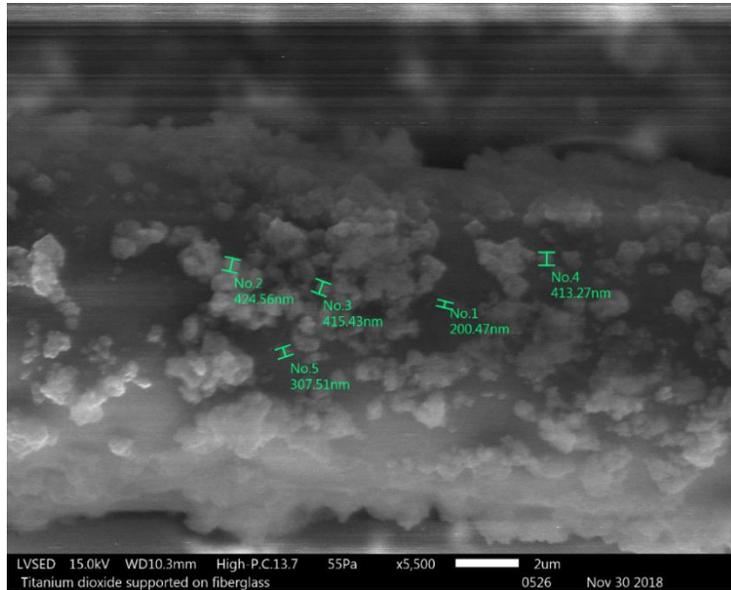
**Figure S1.** a) Fiberglass disc before immobilization b) TiO<sub>2</sub>/fiberglass substrate

#### **Average content of TiO<sub>2</sub> immobilized on the fiberglass substrates**

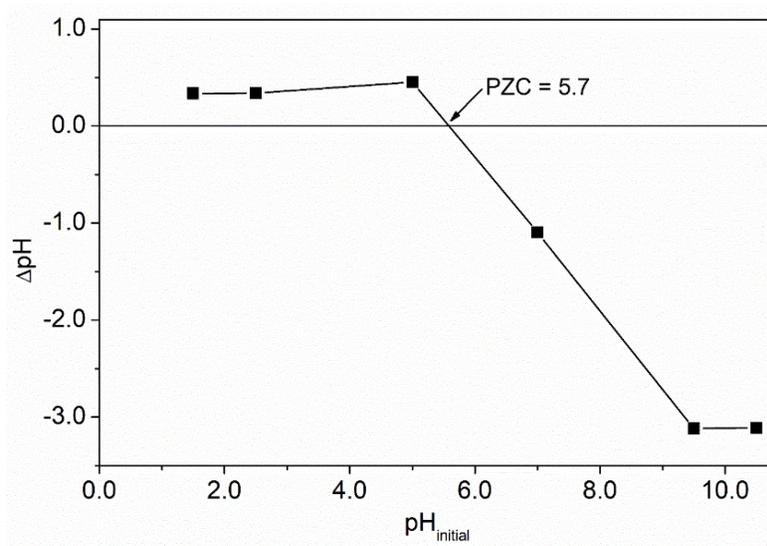
The average content of TiO<sub>2</sub> on each substrate was determined by a gravimetric analysis. For this, the initial weight of the cleaned fiberglass discs was measured. After the immobilization procedure, the plates were dried at 100 °C for 24 h, after this step, the weight of the substrates was measured again. For the calculation of the content of TiO<sub>2</sub>, it was assumed that TiO<sub>2</sub>, silicon and ethanol were homogeneously distributed in the suspension during the immobilization. As a result, the silicon/TiO<sub>2</sub> mass ratio right after the immobilizations was 3.0. Nevertheless, after the drying process, the silicon/TiO<sub>2</sub> mass ratio changed because of the loss of volatile compounds from the silicon mixture. For this reason, a correction was introduced considering a silicon<sub>(dry)</sub>/silicon<sub>(wet)</sub> mass ratio of 0.206 ± 0.02 determined in a separate gravimetric analysis. The area of the substrates (47.1 cm<sup>2</sup>) was calculated considering the internal and external diameters of the discs. Gravimetric data was measured for 115 substrates, and the average results are presented on Table S1.

**Table S1.** Average content of TiO<sub>2</sub> immobilized on the fiberglass substrates

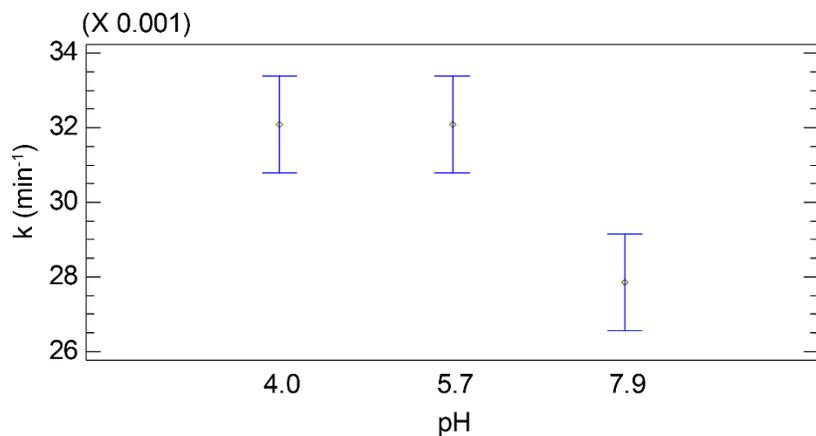
	<b>Initial weight (g)</b>	<b>Final weight (g)</b>	<b>TiO<sub>2</sub> and dry silicon immobilized (g)</b>	<b>*TiO<sub>2</sub> immobilized (mg/cm<sup>2</sup>)</b>
<b>Average**</b>	1.8857	2.0503	0.1645	2.1578
<b>Standard deviation**</b>	0.1573	0.1661	0.0253	0.3319



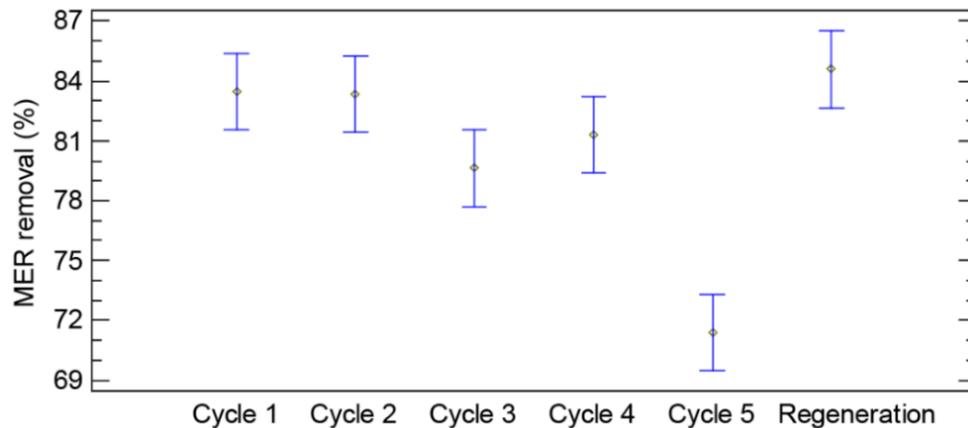
**Figure S2.** SEM images of the  $\text{TiO}_2$ /fiberglass substrate. The size of some agglomerates of  $\text{TiO}_2$  immobilized on fiberglass is reported



**Figure S3.** Point of zero charge of the  $\text{TiO}_2$ /fiberglass substrate



**Figure S4.** Statistical analysis of the effect of the pH value on the pseudo-first order rate constant of the photocatalytic degradation meropenem



**Figure S5.** Statistical analysis of the effect of the reuse of the TiO<sub>2</sub>/fiberglass substrate on the removal of MER after 60 min of reaction (pH = 5.7)