Supplementary Material

Cu^{III} and Co^{III} based MOFs: { $[La_2Cu_3(\mu-H_2O)(ODA)_6(H_2O)_3]\cdot 3H_2O\}_n$ and { $[La_2Co_3(ODA)_6(H_2O)_6]\cdot 12H_2O\}_n$. The relevance of physicochemical properties in the catalytic aerobic oxidation of cyclohexene.

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Table S1: Thermal dependence of aerobic cyclohexene oxidation catalyzed by LaCoODA						
т (°С)	Conversion (%)	2-Cyclohexen-1-ol (%)	2-Cyclohexen-1-one (%)			
30	22	28	72			
50	51	25	75			
85	87	26	74			
Reaction conditions : 50 mmol of cyclohexene and 0.01 mmol of LaCo(ODA), 1 bar of continuous oxygen flow. The mixture is stirred for 24 hours at 75°C.						

Table S2. Conversions for the non-activated and thermally activated catalytic system.Selectivities for the activated catalyst are also presented.

Time	Non-activated LaCoODA conversión (%)	Activated LaCoODA conversion (%)	Activated LaCoODA alcohol selectivity (%)	Activated LaCoODA ketone selectivity (%)		
3h	21	24	30	70		
6h	50	53	26	74		
12h	73	76	27	73		
24h	81	85	27	73		

Reaction conditions: 50 mmol of cyclohexene and 0.01 mmol of LaCo(ODA), 1 bar of continuous oxygen flow. The mixture is stirred at 75°C.



Figure S1. IR spectra of LaCuODA (a) and LaCoODA (b).







Figure S2. Linear (a) and logarithmic (b) adsorption isotherms for LaCoODA







Figure S3 Linear (a) and logarithmic (b) adsorption isotherms for LaCuODA