

Supplementary Materials

Defective grey TiO₂ with minuscule anatase-rutile heterophase junctions for hydroxyl radicals formation in a visible-light triggered photocatalysis

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Table S1. Calculated g -values, line width Γ (peak-to-peak width of measured EPR spectra) for isotropic broadening, and intensities evaluated from the double-integrated EPR spectra. The uncertainties of the last digit are given in parenthesis.

Sample	g-value	Γ (mT)	Intensity (arb. u.)
PT	/	/	0
NT-3	2.0030(5)	0.31(2)	0.80(8)
NT-4	2.0025(5)	0.42(2)	2.2(2)
NT-5	2.0024(5)	0.32(2)	6.0(5)
NT-O-3	2.0025(5)	0.47(2)	0.85(9)
NT-O-4	2.0026(5)	0.36(2)	2.4(2)
NT-O-5	2.0022(5)	0.30(2)	8.6(9)

Table S2. Initial rates of 7-OHC formation obtained during the photocatalytic oxidation of aqueous solution of coumarin in the presence of investigated TiO₂ photocatalysts under visible-light illumination.

Sample	$10^3 \times (r_{7\text{-OHC}})_0$ ($\mu\text{mol}/(\text{g}_{\text{cat. min}})$)
PT	0.02
NT-3	0.08
NT-4	1.6
NT-5	5.9
NT-O-3	0.19
NT-O-4	2.6
NT-O-5	4.0

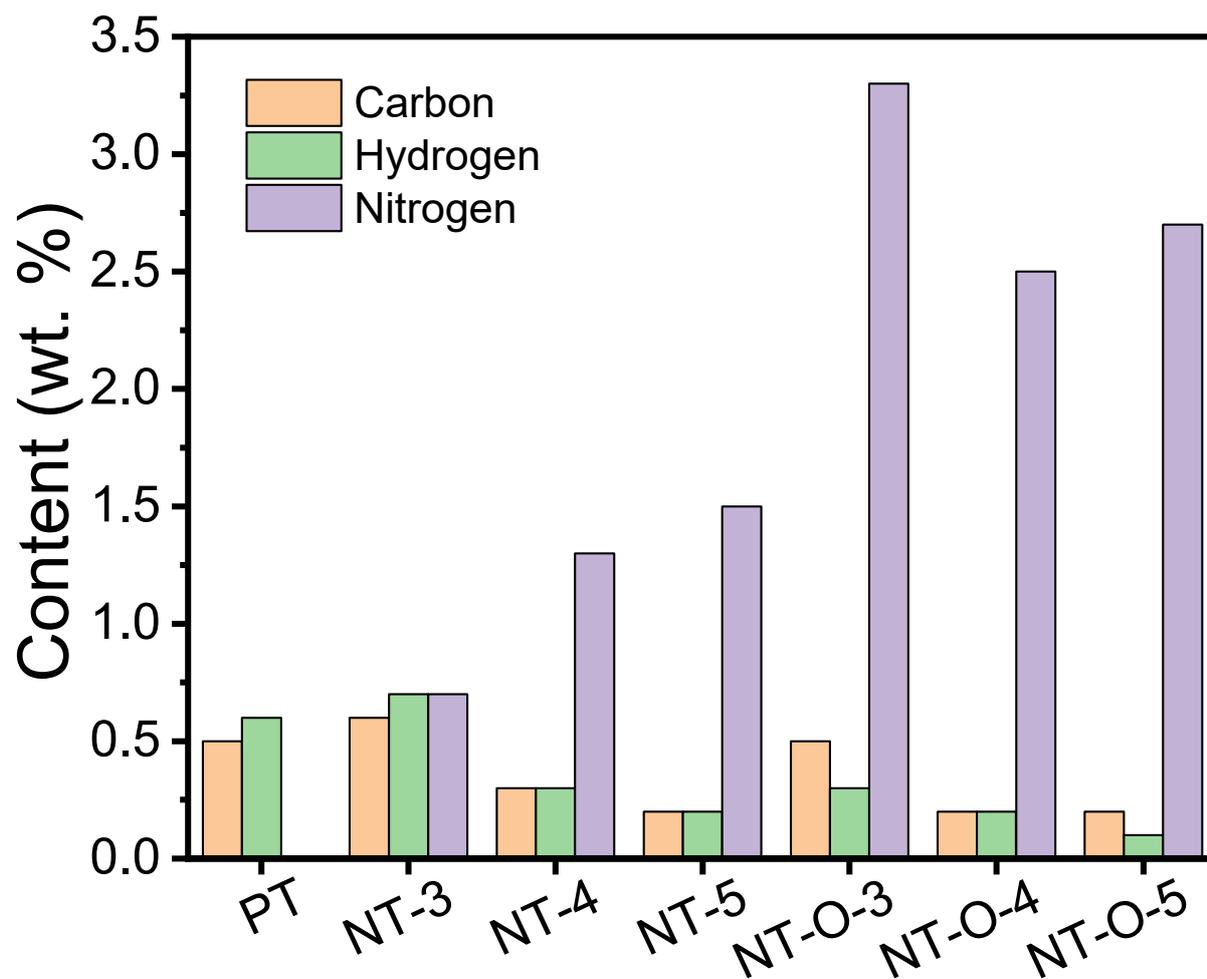


Figure S1. Results of CHN elemental analysis of fresh TiO₂ samples.

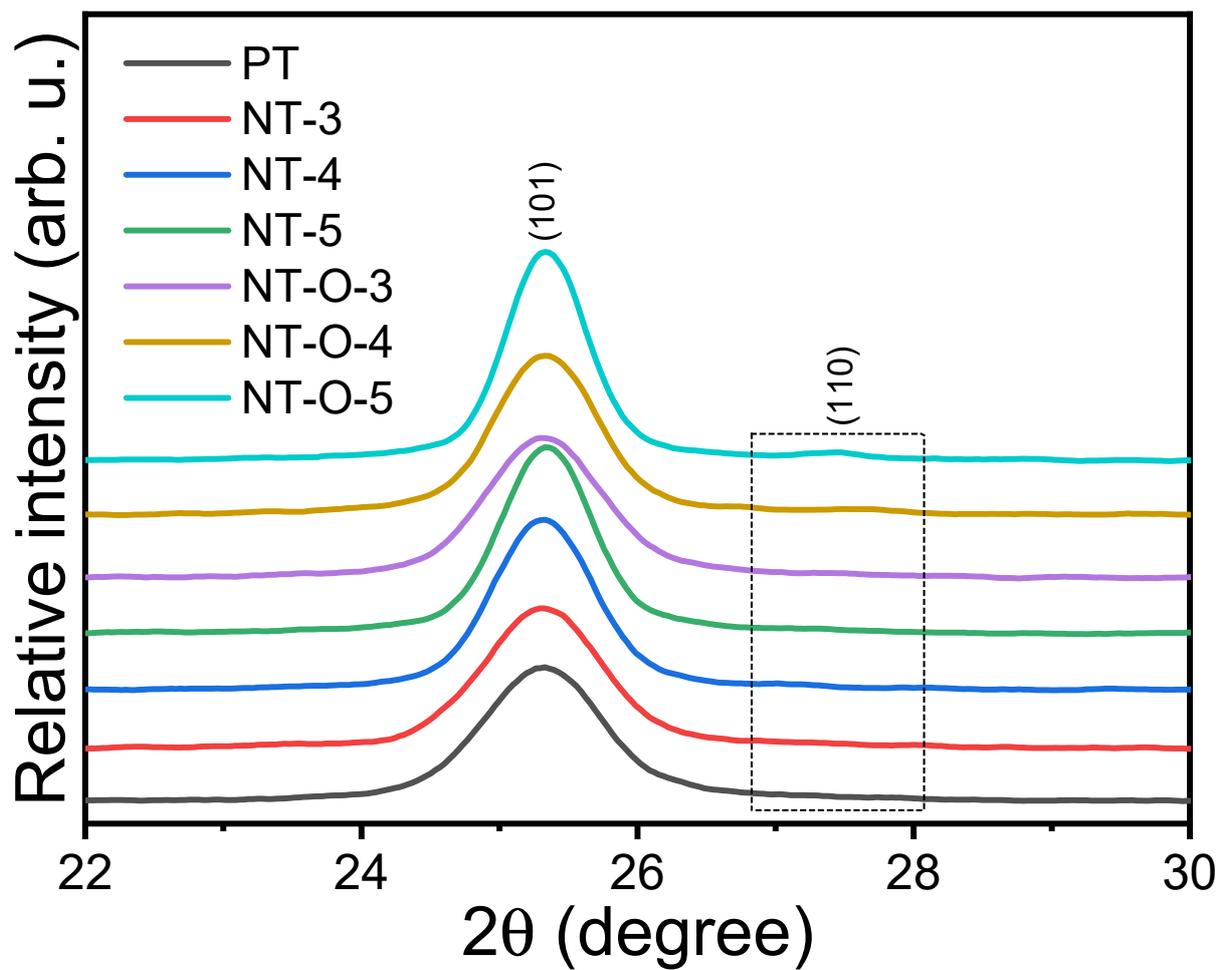


Figure S2. Zoomed XRD diffractograms of solids examined in the present study in the range of 22-30°.

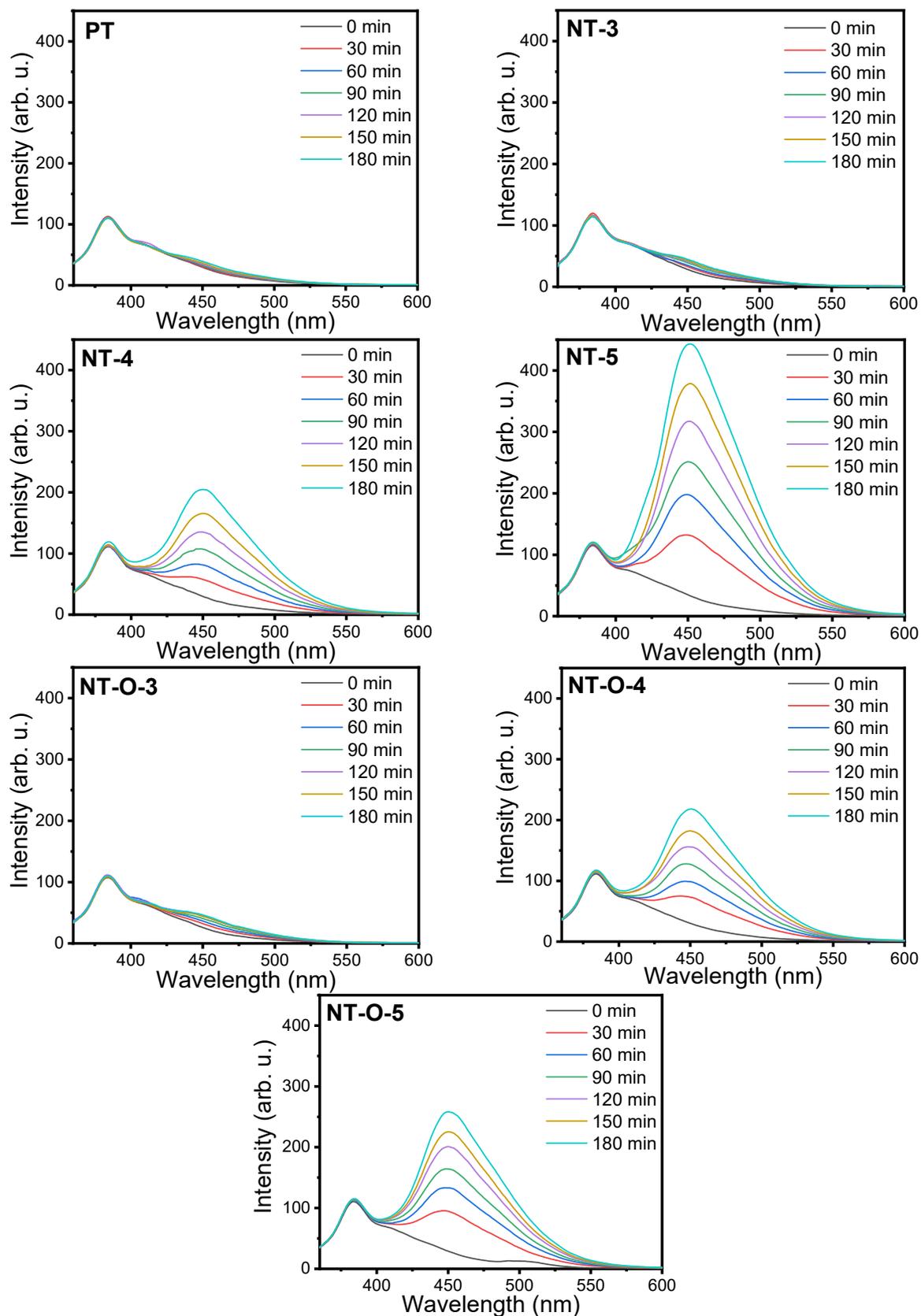


Figure S3. Photoluminescence spectra obtained during the photocatalytic oxidation of aqueous solution of coumarin conducted in the presence of investigated TiO₂ samples.

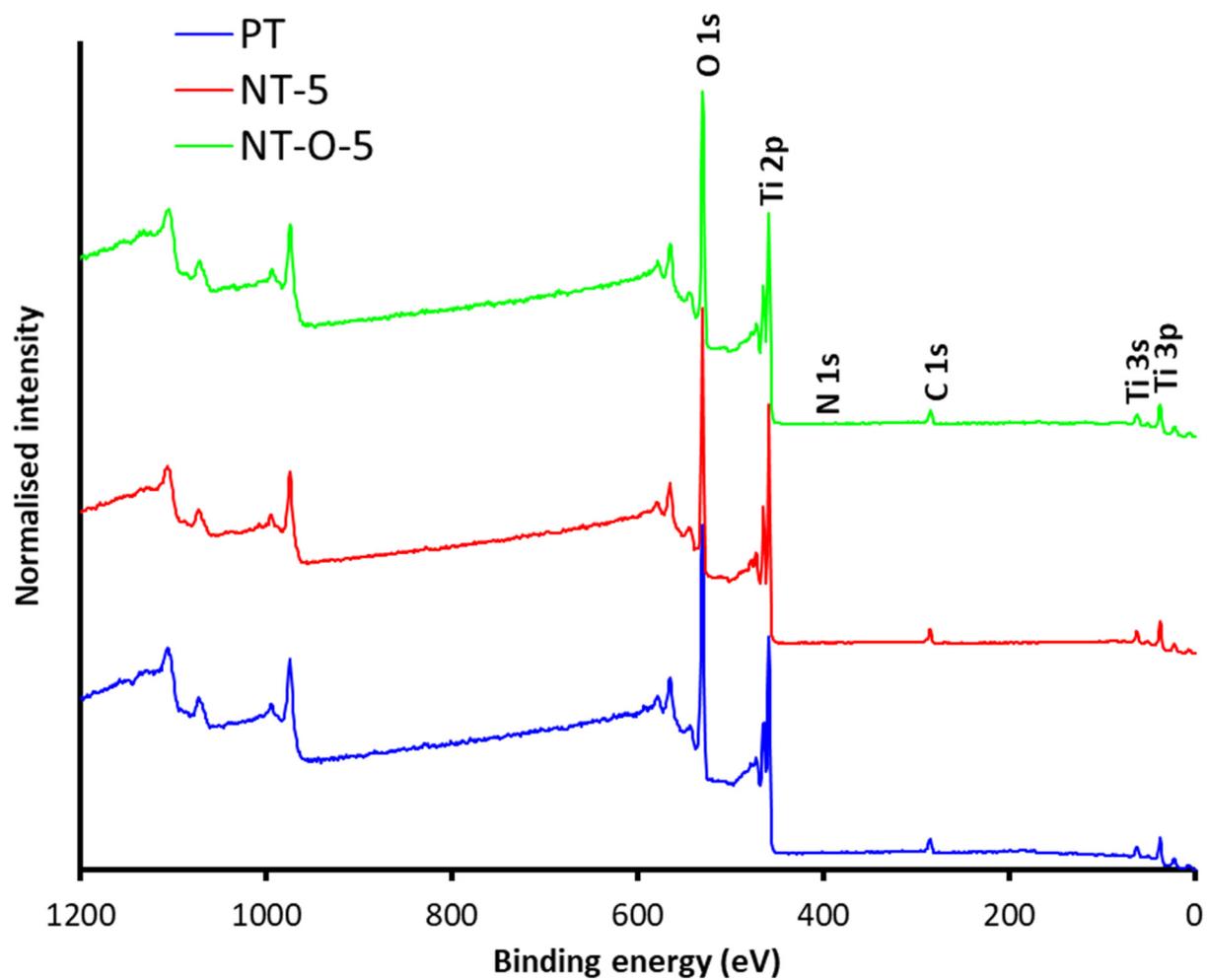


Figure S4. XPS survey spectra measured for PT, NT-5 and NT-O-5 samples.

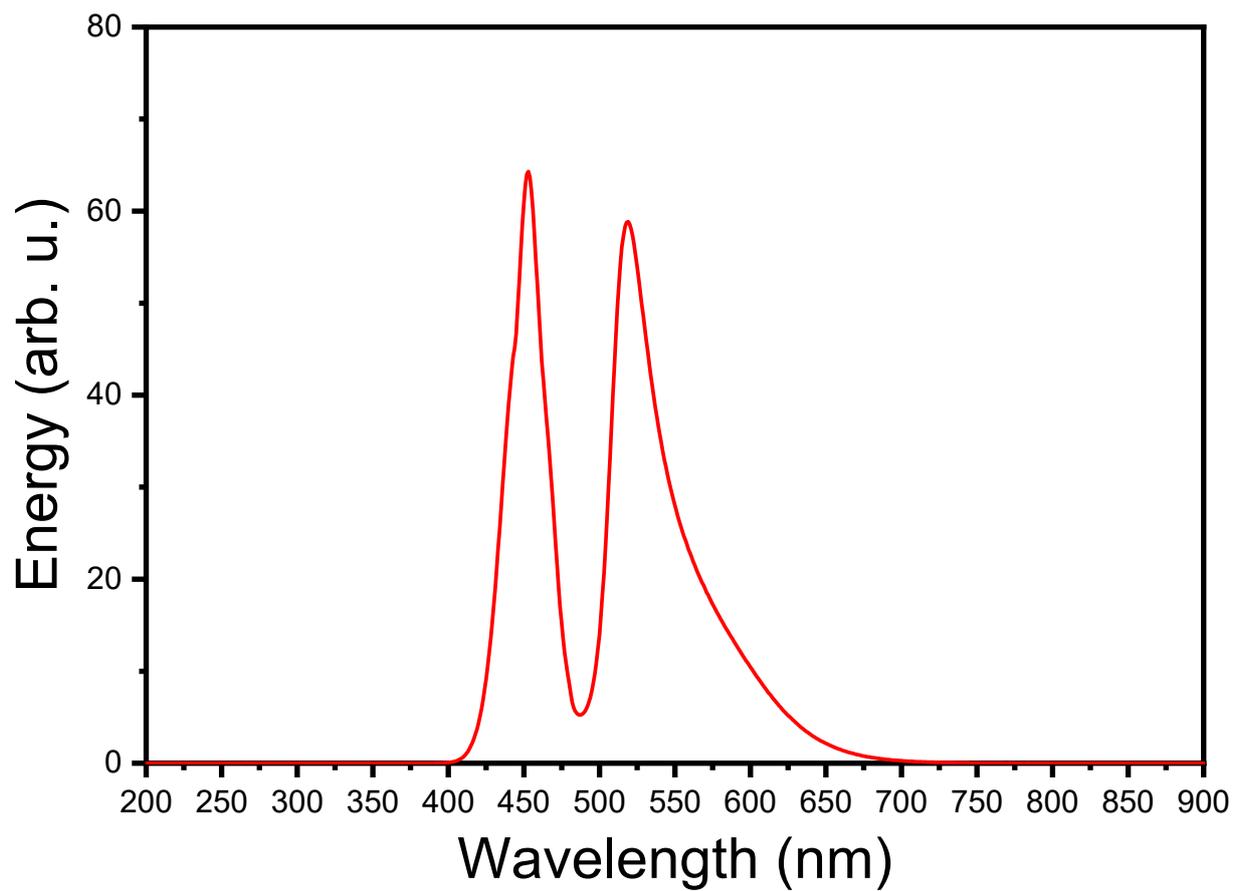


Figure S5. Energy spectrum of LED lamp (KL 1600 LED, Schott, Mainz, Germany) used in the present study.