

Insights into the structural dynamics of Pt/CeO₂ single-site catalysts during CO oxidation

Paolo Dolcet ¹, Florian Maurer ¹, Maria Casapu ¹, and Jan-Dierk Grunwaldt ^{1,2,*}

¹ Institute for Chemical Technology and Polymer Chemistry (ITCP), Karlsruhe Institute of Technology (KIT), Engesserstraße 20, 76131 Karlsruhe, Germany; paolo.dolcet@kit.edu (P.D.), florian.maurer@kit.edu (F.M.), maria.casapu@kit.edu (M.C.)

² Institute of Catalysis Research and Technology (IKFT), Karlsruhe Institute of Technology (KIT), Hermann-von-Helmholtz-Platz 1, 76344 Eggenstein-Leopoldshafen, Germany; grunwaldt@kit.edu (J.-D.G.)

* Correspondence: grunwaldt@kit.edu;

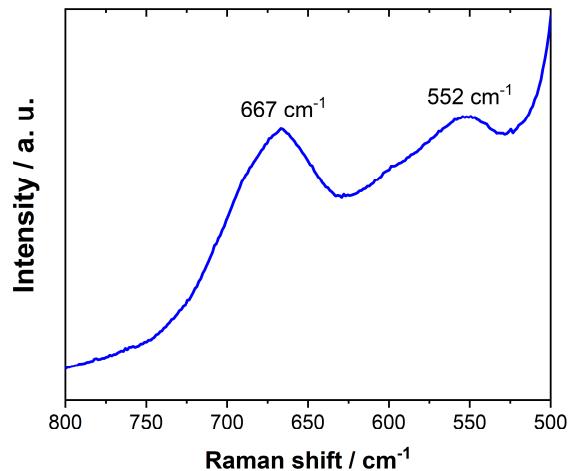


Figure S1. Raman spectra of as-prepared Pt single site catalyst

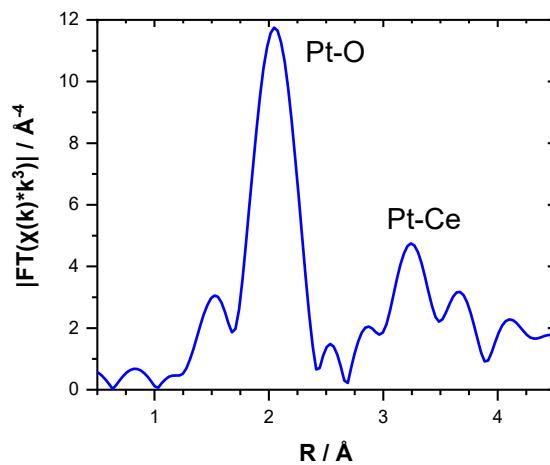


Figure S2. Fourier transformed k^2 -weighted EXAFS spectrum of the as prepared Pt/CeO₂ catalyst

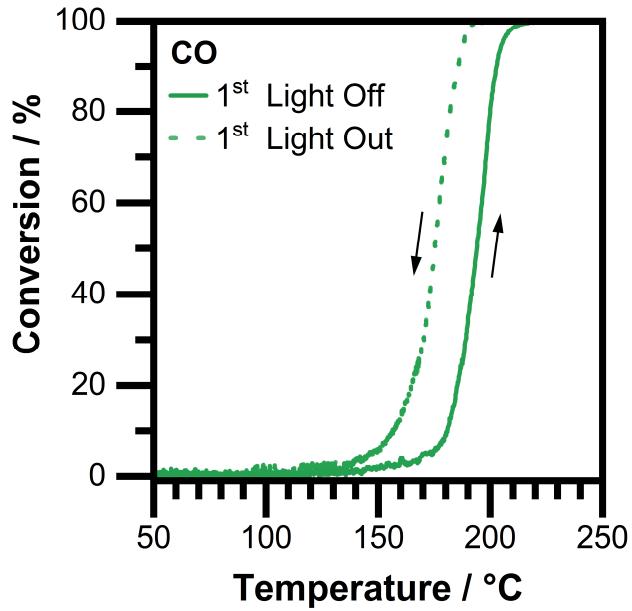


Figure S3. CO conversion curves for consecutive light-offs (1000 ppm CO and 8% O₂)

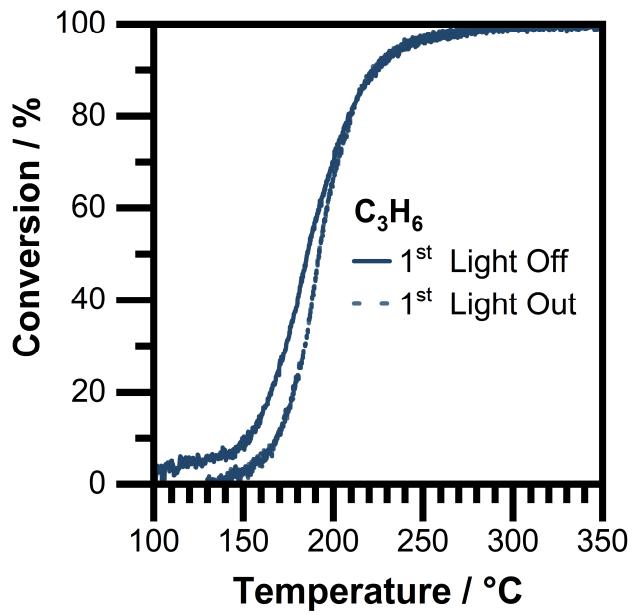


Figure S4. C₃H₆ conversion curves for consecutive light-offs (150 ppm C₃H₆ and 8% O₂)

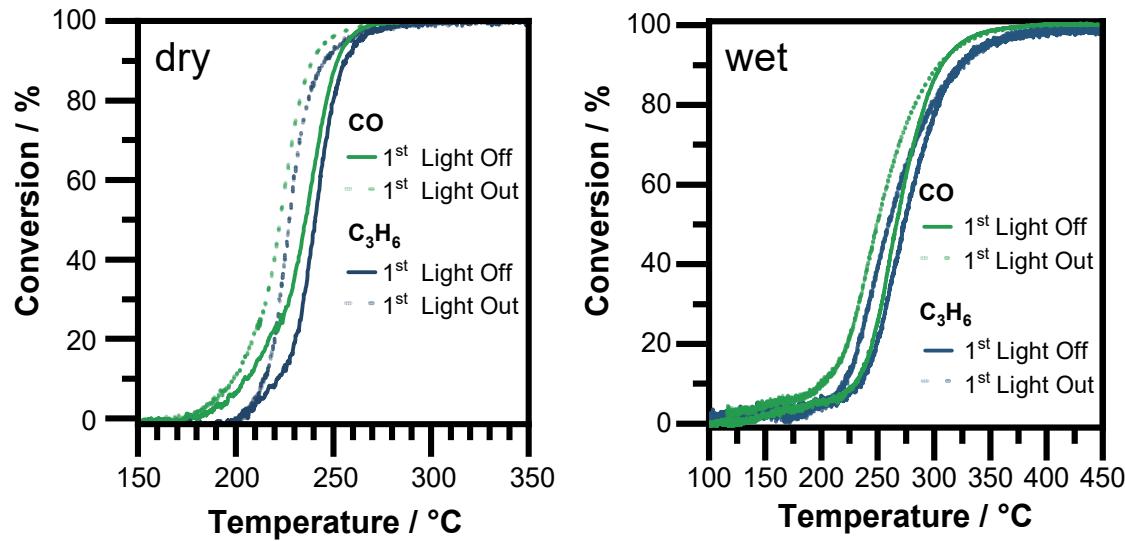


Figure S5. CO (green lines) and (blue lines) C₃H₆ conversion curves light-offs in dry (left, 1000ppm/150 ppm C₃H₆ and 8% O₂) and wet (right, 1000ppm/150 ppm C₃H₆, 6.6% H₂O and 8% O₂) conditions

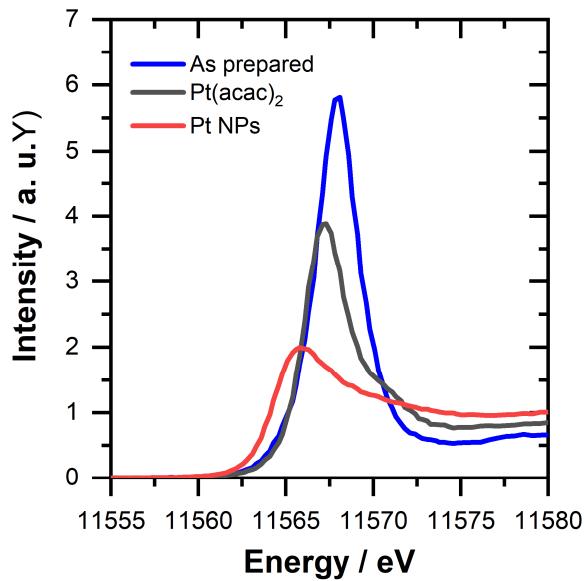


Figure S6. Pt-L_{III} edge HERFD-XANES spectrum of as prepared single sites catalyst compared to metallic Pt nanoparticles and Pt(II)(acac)₂ reference spectra

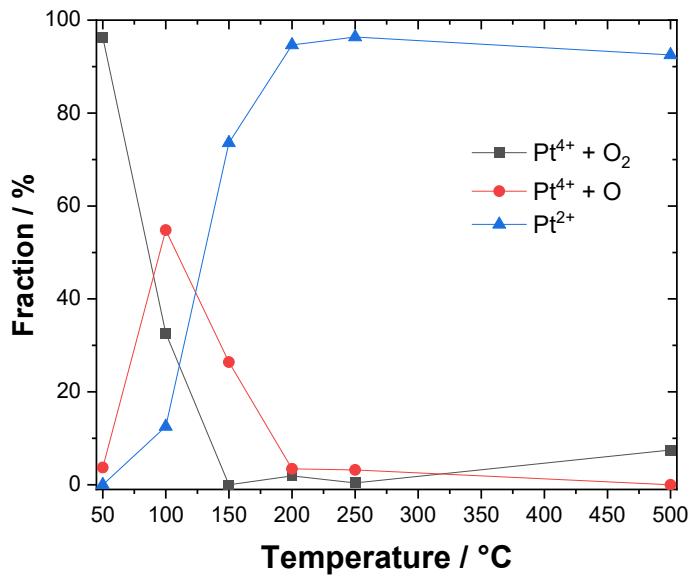


Figure S7. Pt species identified by MCR-ALS during temperature programmed oxidation in 10 % O₂/He

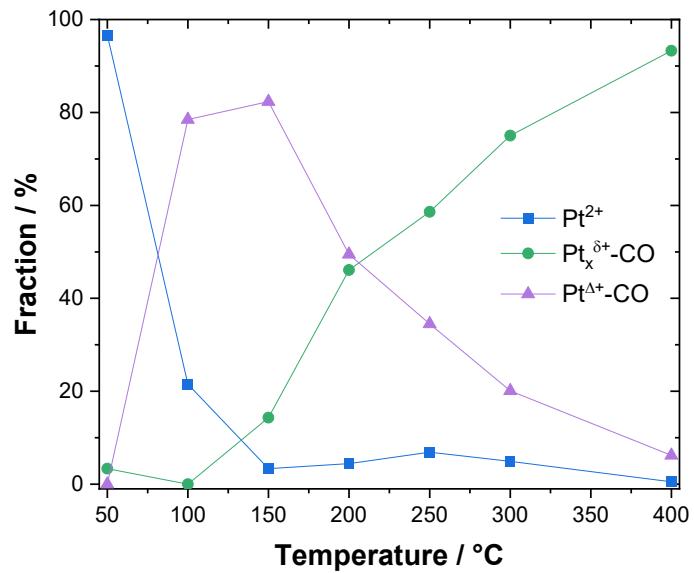


Figure S8. Pt species identified by MCR-ALS during temperature programmed reduction in 1000 ppm CO/He

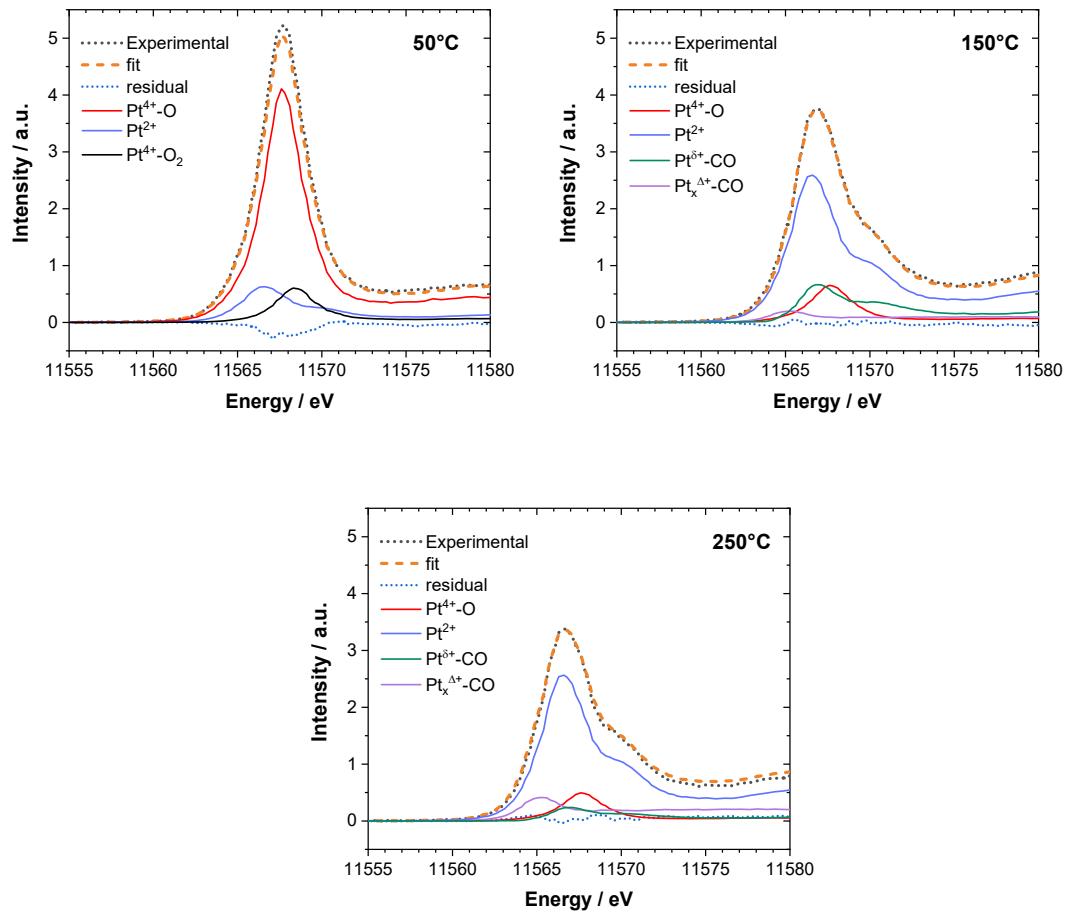


Figure S9. Pt species identified by MCR-ALS during temperature programmed reduction in 1000 ppm CO/He