

SUPPLEMENTARY MATERIALS

Photoelectrochemical Behavior of the Ternary Heterostructured Systems CdS/WO₃/TiO₂

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XRD data

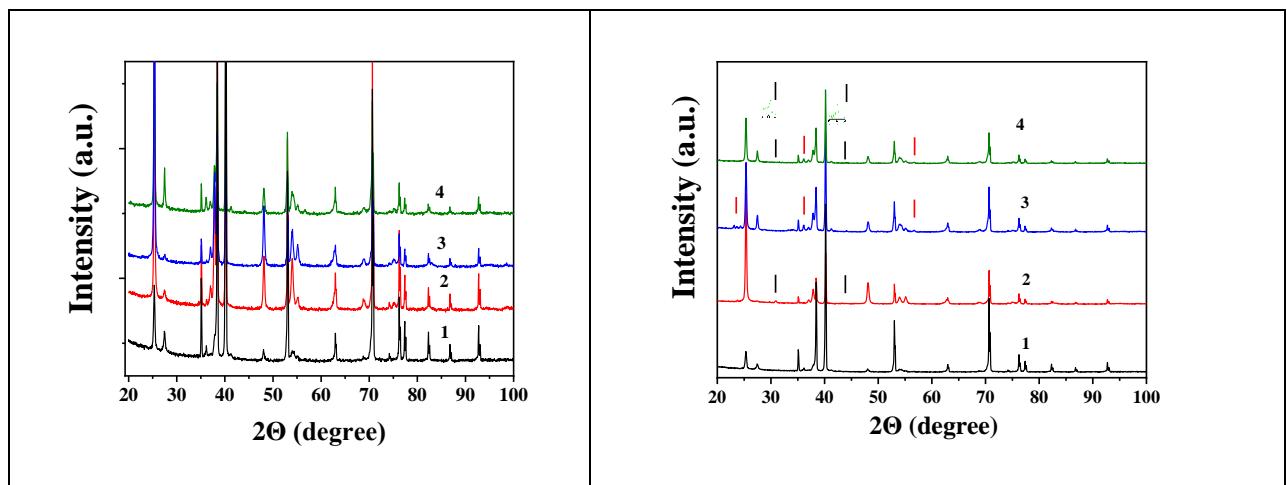
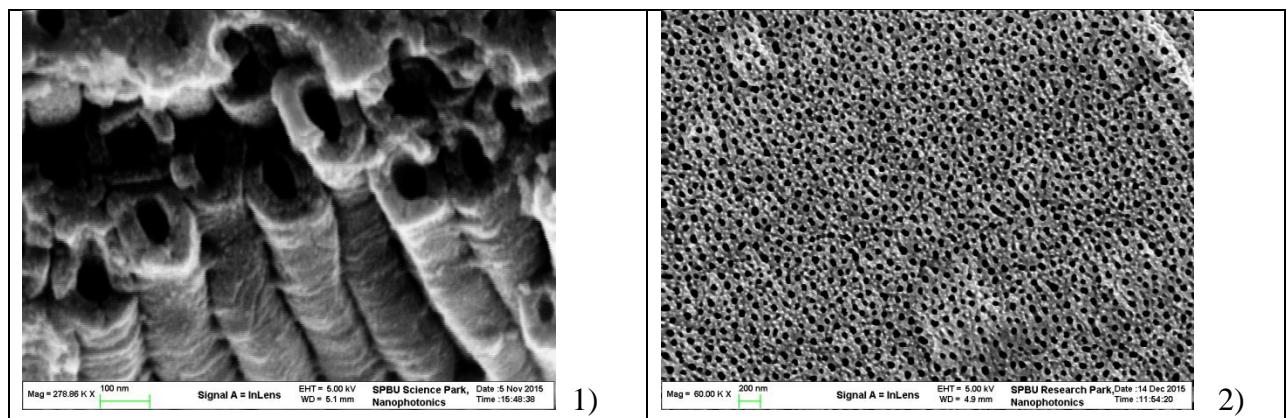


Figure S1. XRD patterns of planar samples: 1 - TiO₂, 2 – CdS/TiO₂, 3 - WO₃/TiO₂, 4 - WO₃/CdS/TiO₂ (left); nanotubes: 1 - TiO₂, 2 – CdS/TiO₂, 3 - WO₃/TiO₂, 4 - WO₃/CdS/TiO₂, Black lines show the replicas of CdS, red lines – WO₃

Electron microscopy



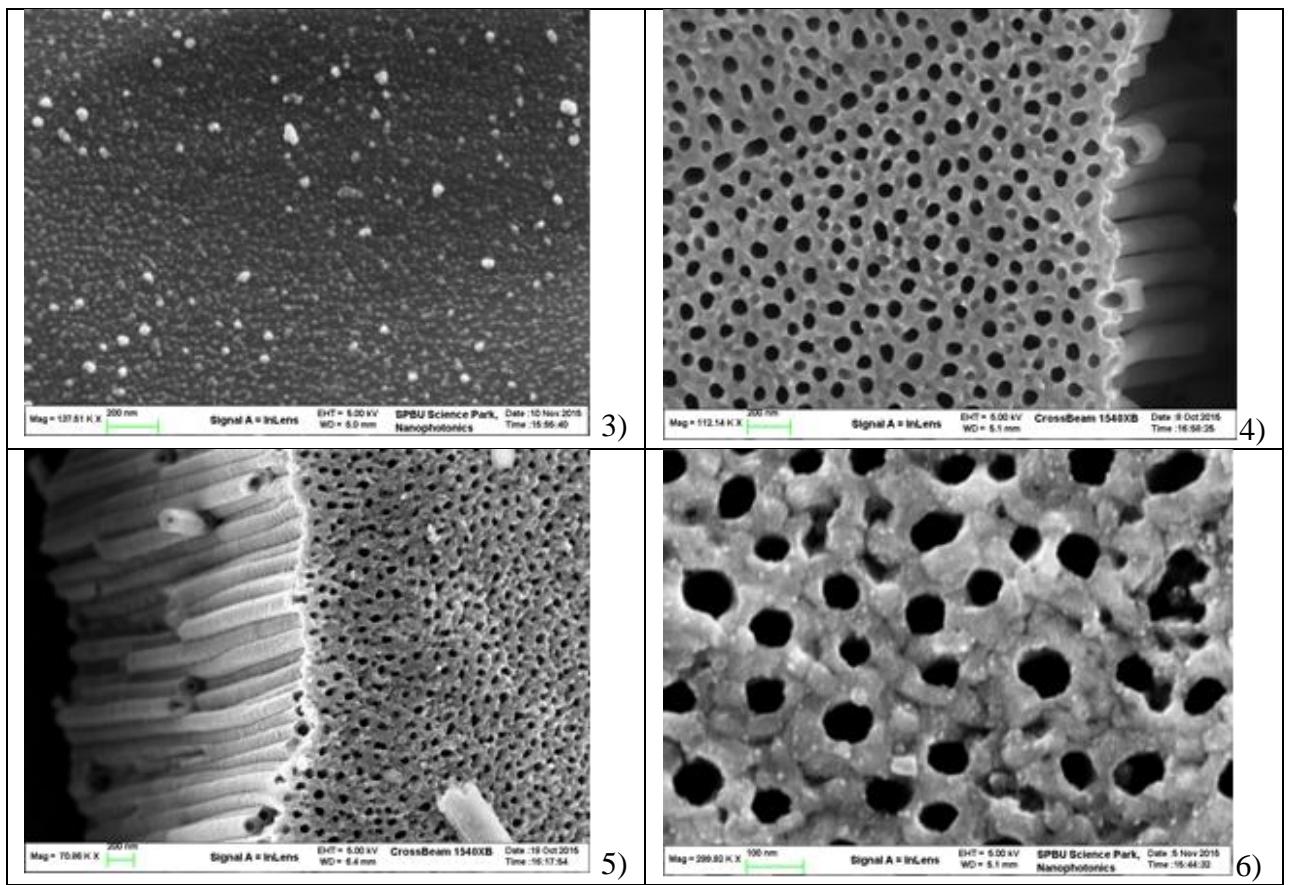
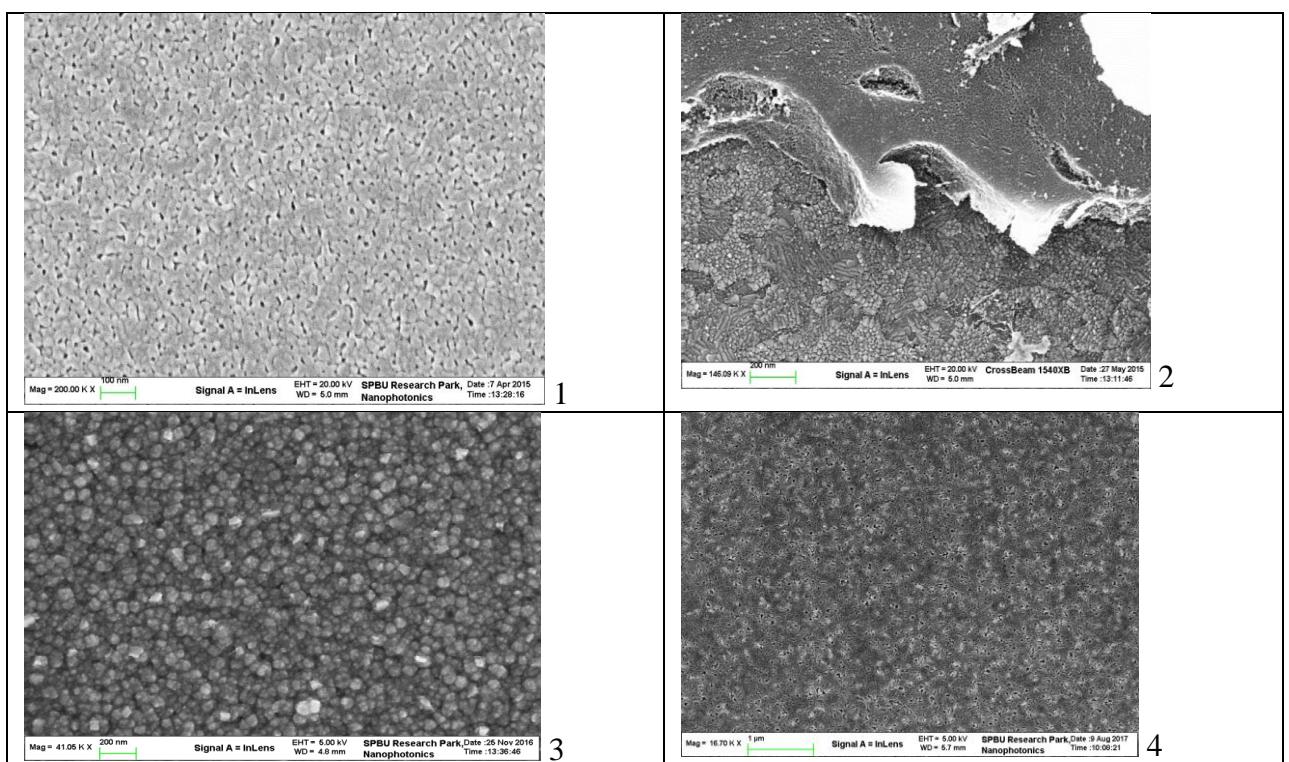


Figure S2. SEM images of the titania nanotube (TNT) based systems: 1, 2 – TNTs (side and top view), 3 – tungsten metal particles deposited on Ti foil (before anodizing), 4 – TNTs/WO₃, 5 – CdS/TNTs, 6 – CdS/TNTs/WO₃.



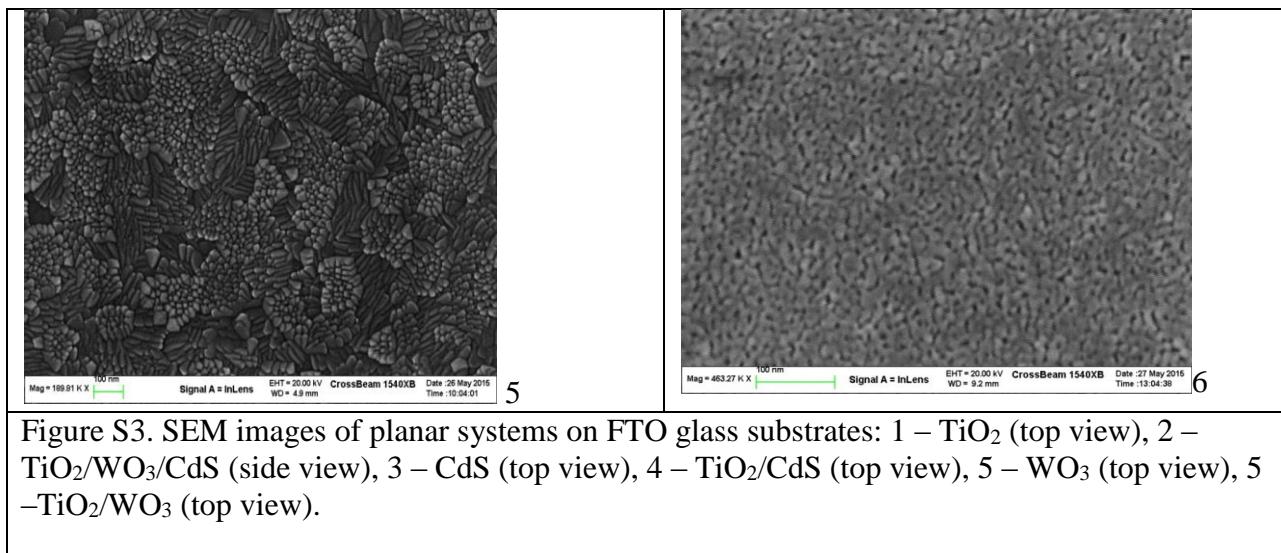


Figure S3. SEM images of planar systems on FTO glass substrates: 1 – TiO_2 (top view), 2 – $\text{TiO}_2/\text{WO}_3/\text{CdS}$ (side view), 3 – CdS (top view), 4 – TiO_2/CdS (top view), 5 – WO_3 (top view), 5 – TiO_2/WO_3 (top view).

XPS data

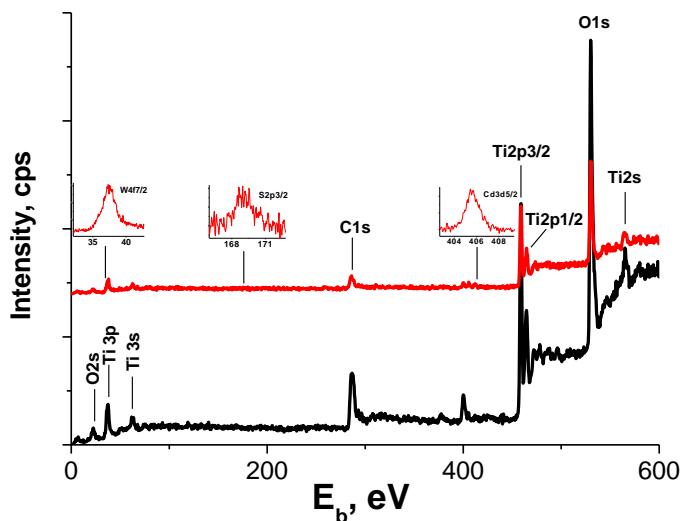


Figure S4. XPS spectra: TNT (black), CdS/WO_3 TNT (red).

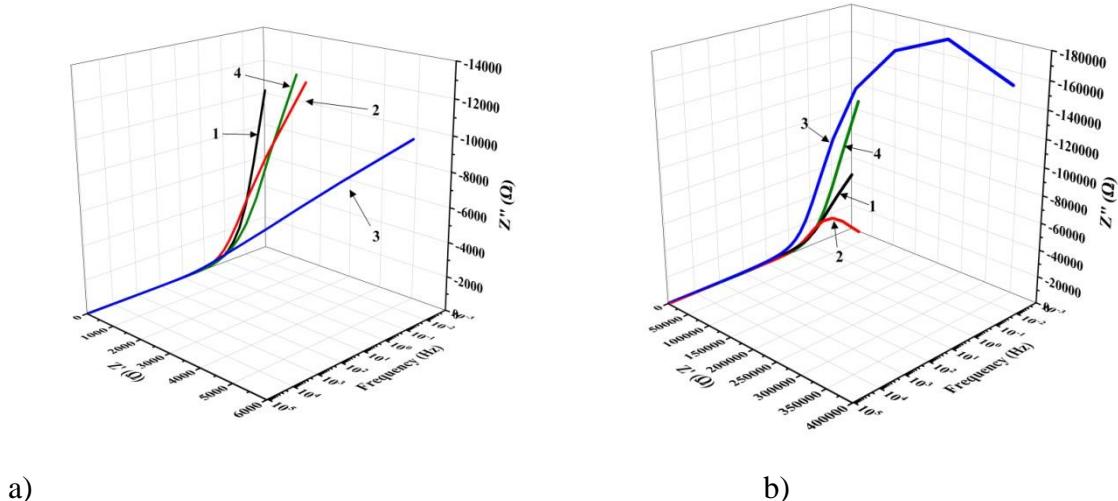
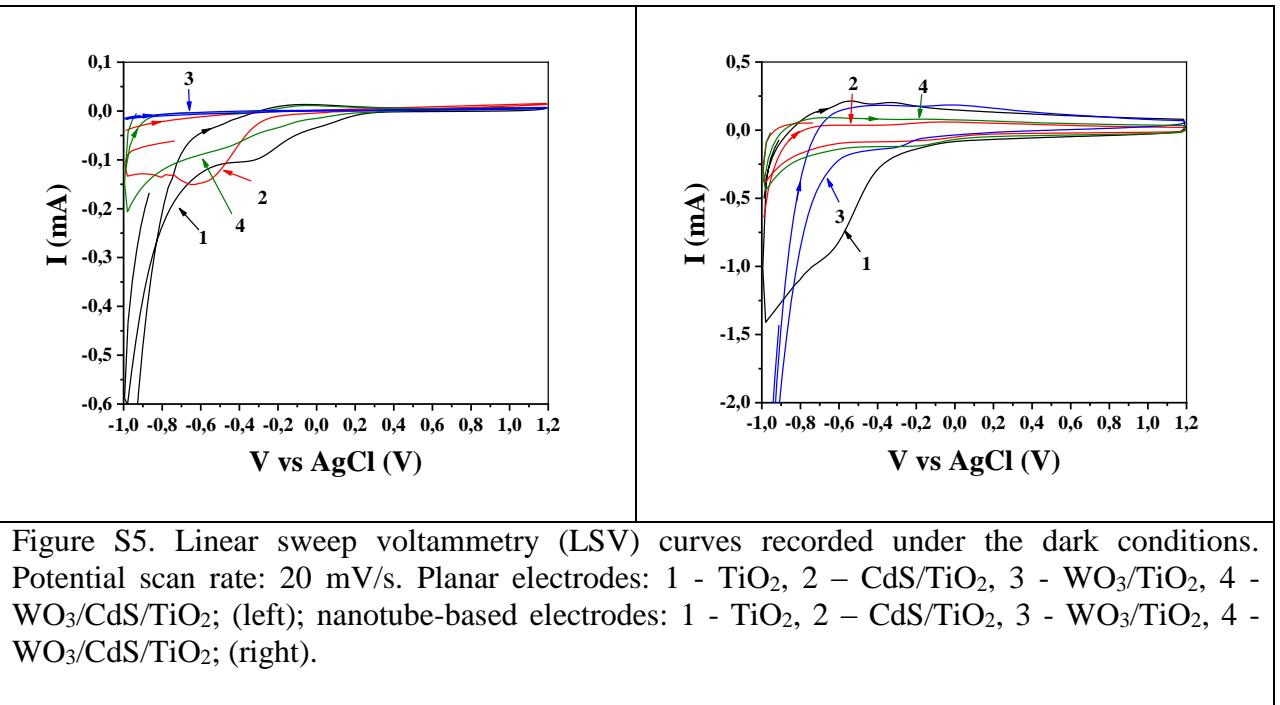


Figure S6. 3D EIS plots recorded in the dark. a) nanotube-based electrodes: 1 - TiO_2 , 2 - CdS/TiO_2 , 3 - WO_3/TiO_2 , 4 - $\text{WO}_3/\text{CdS}/\text{TiO}_2$; b) planar electrodes: 1 - TiO_2 , 2 - CdS/TiO_2 , 3 - WO_3/TiO_2 , 4 - $\text{CdS}/\text{WO}_3/\text{TiO}_2$.

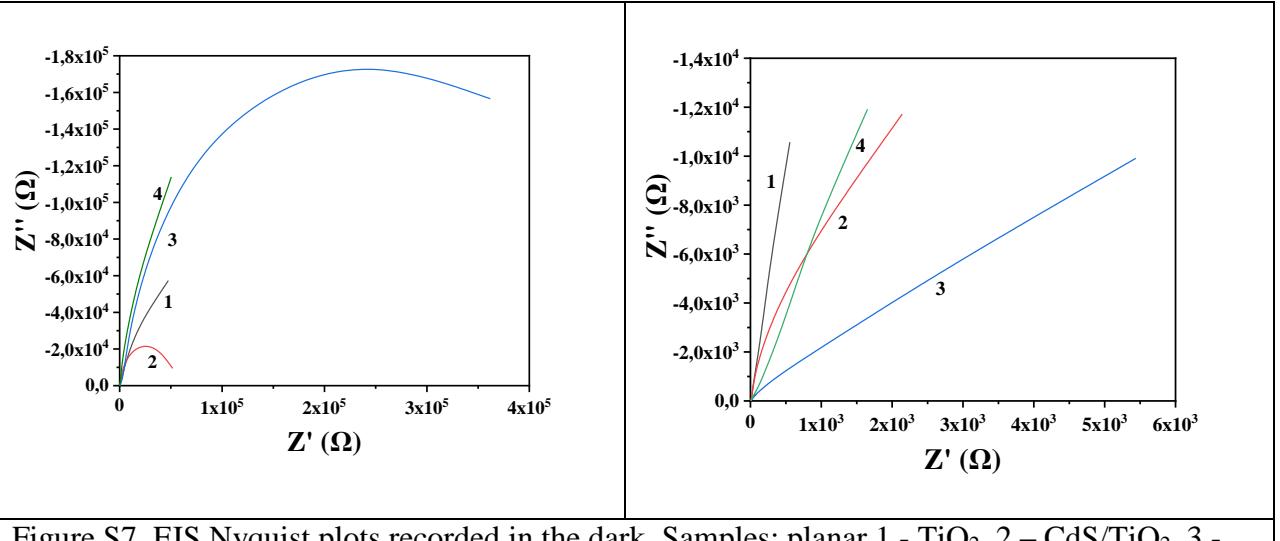


Figure S7. EIS Nyquist plots recorded in the dark. Samples: planar 1 - TiO₂, 2 – CdS/TiO₂, 3 - WO₃/TiO₂, 4 - WO₃/CdS/TiO₂ (left); nanotube-based 1 - TiO₂, 2 – CdS/TiO₂, 3 - WO₃/TiO₂, 4 - WO₃/CdS/TiO₂ (right).

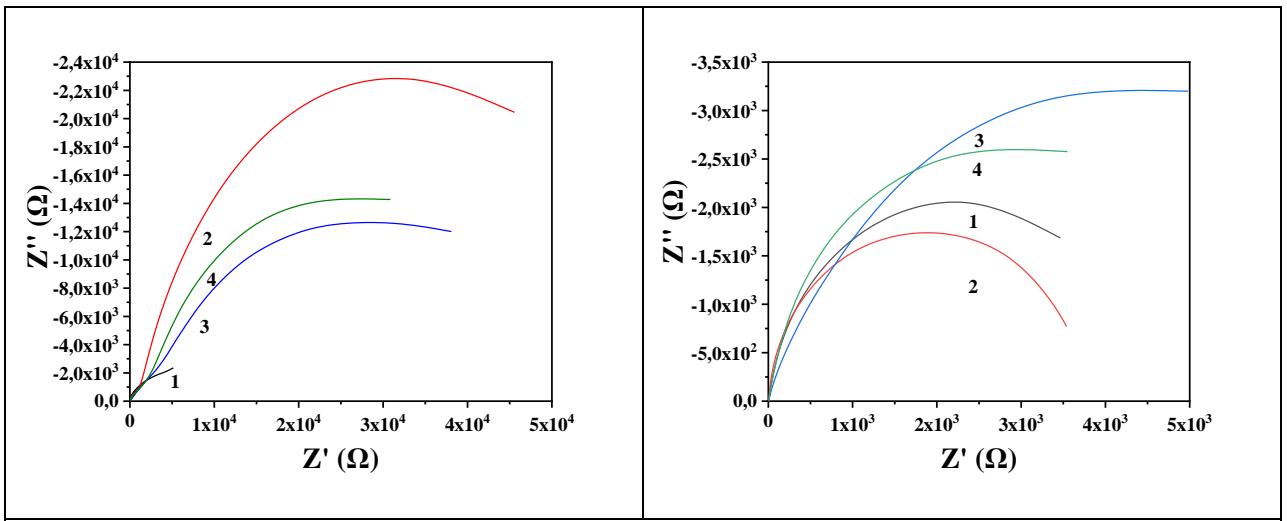


Figure S8. EIS Nyquist plots. Data recorded under illumination. Samples: planar 1 - TiO₂, 2 – CdS/TiO₂, 3 - WO₃/TiO₂, 4 - WO₃/CdS/TiO₂ (left); nanotube-based 1 - TiO₂, 2 – CdS/TiO₂, 3 - WO₃/TiO₂, 4 - WO₃/CdS/TiO₂ (right).

Table S1. Parameters of equivalent circuit scheme for nanotube-based systems in the dark.

Sample	R1, Ohm	R2, Ohm	P1	n1
TiO ₂	~0	700000 ± 100000	$0,0027 \pm 0,0003$	$0,97 \pm 0,01$
CdS/TiO ₂	~0	80000 ± 5000	$0,00135 \pm 0,00006$	$0,97 \pm 0,01$
WO ₃ /TiO ₂	~0	80000 ± 5000	$0,000714 \pm 0,000004$	$0,75 \pm 0,02$
CdS/WO ₃ /TiO ₂	~0	100000 ± 300000	$0,00199 \pm 0,000008$	$0,91 \pm 0,01$

Table S2.

dark	R1	R2	P1	n1
TiO ₂	30±80	176000±7000	0,000118±0,000001	0,79±0,01
CdSTiO ₂	200±100	49100±800	0,000103±0,00002	0,95±0,01
WO ₃ TiO ₂	700±400	399000±8000	0,0000268±0000003	0,89±0,02
CdSWO ₃ TiO ₂	70±30	380000±20000	0,000173±0,000007	0,90±0,02

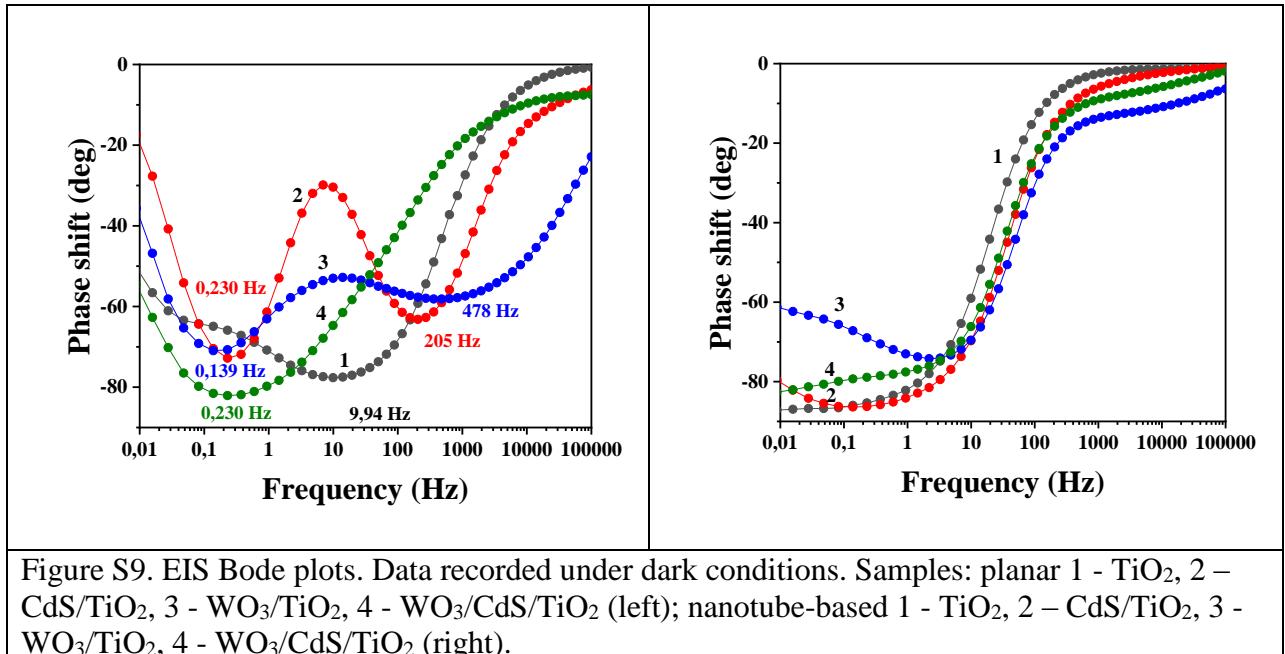


Figure S9. EIS Bode plots. Data recorded under dark conditions. Samples: planar 1 - TiO₂, 2 – CdS/TiO₂, 3 - WO₃/TiO₂, 4 - WO₃/CdS/TiO₂ (left); nanotube-based 1 - TiO₂, 2 – CdS/TiO₂, 3 - WO₃/TiO₂, 4 - WO₃/CdS/TiO₂ (right).

Table S3. The time constants τ calculated from Bode diagrams for nanotube heterostructures in the dark.

sample	frequency, Hz	τ , sec
TiO ₂	0,00889	17,9
CdS/TiO ₂	0,139	1,145
WO ₃ /TiO ₂	2,19	0,0727
CdS/WO ₃ /TiO ₂	0,00889	17,9

Table S4. The time constants τ calculated from Bode diagrams for planar heterostructures in the dark.

sample	frequency, Hz	t, sec
TiO ₂	9,94	0,0160
CdS/TiO ₂	0,23	0,692
CdS/TiO ₂	205	0,000776
WO ₃ /TiO ₂	0,139	1,145
WO ₃ /TiO ₂	478	0,000333
CdS/WO ₃ /TiO ₂	0,23	0,692