

Supporting Information: Study on the Mechanism of a Side Coupling Reaction during the Living Anionic Copolymerization of Styrene and 1-(Ethoxydimethylsilyphenyl)-1-phenylethylene (DPE-SiOEt)

Pibo Liu, Hongwei Ma, Heyu Shen, Li Han, Shuang Chang, Long Zang, Yiyu Bian, Yu Bai and Yang Li

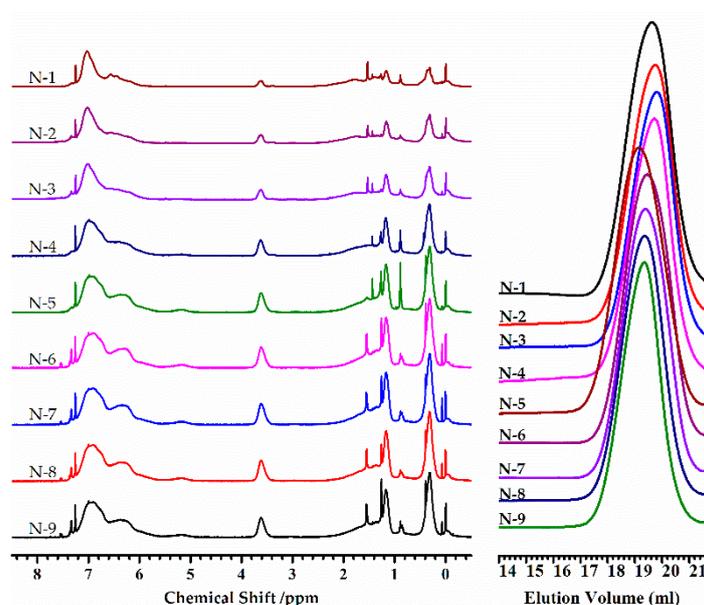


Figure S1. The SEC and ^1H NMR spectra of NT-1~NT-9.

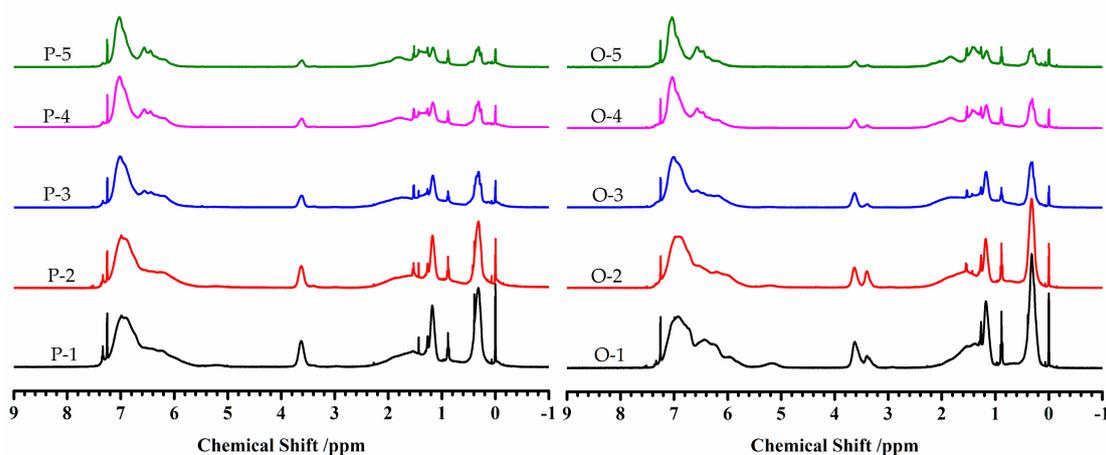


Figure S2. The ^1H NMR spectra of O-1~O-5 and P-1~P-5.

Equations:

$$\frac{2N_D}{5N_S+9N_D} = \frac{\text{Area}(-\text{OCH}_2-)}{\text{Area}(\text{aromatic region})} \quad (\text{S1})$$

$$M_S N_S + M_D N_D = \overline{M}_n \quad (\text{S2})$$

where N_D refers to the average number of DPE-SiOEt units incorporated into the copolymer. M_D , M_S and \overline{M}_n are the molecular weights of DPE-SiOEt, styrene and the copolymer, respectively.

$$\ln \frac{[M_D]}{[M_D]_0} + \frac{1}{r_{St}-1} \ln \left\{ \frac{[M_S]_0}{[M_D]_0} (r_{St} - 1) + 1 \right\} = 0 \quad (S3)$$

$$\lg \frac{r_{St}^0}{r_{St}} = \rho\sigma \quad (S4)$$

where r_{St}^0 is the reactivity ratio of the copolymerization for the styrene and DPE ($r_{St}^0=0.45$) in benzene[1]. $\rho=+1.8$ [2,3] is substituted into Equation (S4).

1. Yamagishi, A.; Szwarc, M. Kinetics of styrene addition in benzene solution to living lithium polymers terminated by 1,1-diphenylethylene units. The effect of mixed dimerization of monomeric polymers. *Macromolecules* **1978**, *11*, 504-506.
2. Quirk, R.P.; Lee, B. Experimental criteria for living polymerization. *Polym. Int.* **1992**, *27*, 359-367.
3. Wu, L.L.; Wang, Y.S.; Wang, Y.R.; Shen, K.H.; Li, Y. In-chain multi-functionalized polystyrene by living anionic copolymerization with 1,1-bis(4-dimethylaminophenyl)ethylene: Synthesis and effect on the dispersity of carbon black in polymer-based composites. *Polymer* **2013**, *54*, 2958-2965.