

Supplementary Material

Highly Diastereoselective Chelation-controlled 1,3-*anti*-Allylation of (S)-3-(Methoxymethyl)hexanal Enabled by Hydrate of Scandium Triflate

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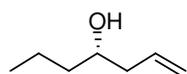
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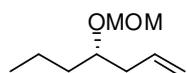
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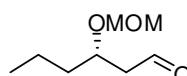
1. Synthesis of aldehyde 3.



(*S*)-Hept-1-en-4-ol (**SI-1**). Titanium(IV) isopropoxide (1.10 mL, 3.78 mmol) and CF_3COOH (28.0 μL , 0.37 mmol) were added in a sequence to a mixture of (*R*)-BINOL (2.16 g, 7.56 mmol) and 6.75 g of activated molecular sieves (4 Å) in anhydrous CH_2Cl_2 (155 mL). The reaction mixture was refluxed under inert atmosphere (argon) at vigorous stirring for 1 h. The obtained solution of titanium catalyst was cooled to -78°C (acetone-dry ice bath) and then a solution of butyraldehyde (4.86 mL, 54.0 mmol) in anhydrous CH_2Cl_2 (40 mL) was added at stirring. After 15 min, a solution of allyltributylstannane (36.0 g, 109 mmol) in anhydrous CH_2Cl_2 (70 mL) was added at the same temperature and the resulted mixture was kept in a refrigerator at -25°C for 120 h. The reaction mixture was treated at vigorous stirring with saturated aqueous solution of NaHCO_3 (300 mL). The organic layer was separated, the aqueous layer was extracted with CH_2Cl_2 (3 \times 100 mL), the combined organic extracts were washed with brine (100 mL) and dried over Na_2SO_4 . The solvent was removed under reduced pressure, and the title compound was isolated by column chromatography (SiO_2 , eluent PE:EtOAc, 20:1). Colorless oil (5.40 g, 88%). $R_f = 0.48$ (PE:EtOAc, 4:1). $[\alpha]_{\text{D}}^{20} = -12.2$ (c 1.18, CHCl_3). IR (neat): $\nu = 3362, 1642, 1465, 1123 \text{ cm}^{-1}$. ^1H NMR (400 MHz, CDCl_3): $\delta = 5.89\text{--}5.76$ (m, 1H), 5.17–5.07 (m, 2H), 3.69–3.61 (m, 1H), 2.33–2.25 (m, 1H), 2.19–2.07 (m, 1H), 1.63 (br.s, 1H), 1.52–1.28 (m, 4H), 0.93 (t, $J = 7.0 \text{ Hz}$, 3H). ^{13}C NMR (100.6 MHz, CDCl_3): $\delta = 134.90, 118.00, 70.38, 41.93, 38.94, 18.82, 14.04$. HRMS (ESI) calcd. for $\text{C}_7\text{H}_{15}\text{O}^+$ $[\text{M}+\text{H}]^+$ 115.1117, found m/z 115.1115. The enantiomeric excess ($ee > 99\%$) was determined by Mosher's method [1].



(*S*)-4-(Methoxymethoxy)hept-1-ene (**SI-2**). A solution of methoxymethyl chloride (2 M in toluene, 45.0 mL, 90.0 mmol, prepared as described in [2]) was added to a solution of (*S*)-hept-1-en-4-ol (**SI-1**) (5.13 g, 45.0 mmol), *N,N*-diisopropylethylamine (10.5 mL, 60.2 mmol) and catalytic amount of Bu_4NI in anhydrous CH_2Cl_2 (55 mL). The reaction mixture was stirred at room temperature for 4 h. Afterwards, the mixture was quenched at vigorous stirring with saturated aqueous solution of NaHCO_3 (100 mL) and additionally stirred for 1h. The organic layer was separated, the aqueous layer was extracted with CH_2Cl_2 (3 \times 30 mL), the combined organic extracts were washed with brine (50 mL) and dried over Na_2SO_4 . The solvent was removed under reduced pressure, and the title compound was isolated by column chromatography (SiO_2 , eluent PE:EtOAc, 50:1). Colorless oil (6.80 g, 96%). $R_f = 0.53$ (PE:EtOAc, 10:1). $[\alpha]_{\text{D}}^{20} = -33.1$ (c 2.08, CHCl_3). IR (neat): $\nu = 1641, 1466, 1378, 1213, 1145, 1101, 1043 \text{ cm}^{-1}$. ^1H NMR (400 MHz, CDCl_3): $\delta = 5.82$ (ddt, $J = 17.2, 10.2, 7.1 \text{ Hz}$, 1H), 5.13–5.01 (m, 2H), 4.68 (d, $J = 6.9 \text{ Hz}$, 1H), 4.64 (d, $J = 6.9 \text{ Hz}$, 1H), 3.65–3.57 (m, 1H), 3.38 (s, 3H), 2.28 (ddt, $J = 7.1, 5.8, 1.3 \text{ Hz}$, 2H), 1.54–1.26 (m, 4H), 0.91 (t, $J = 7.2 \text{ Hz}$, 3H). ^{13}C NMR (100.6 MHz, CDCl_3): $\delta = 134.86, 116.98, 95.36, 76.59, 55.47, 38.90, 36.39, 18.57, 14.12$. HRMS (ESI) calcd. for $\text{C}_9\text{H}_{18}\text{O}_2\text{Na}^+$ $[\text{M}+\text{Na}]^+$ 181.1199, found m/z 181.1204.



(*S*)-3-(Methoxymethoxy)hexanal (**3**). Ozonated oxygen was passed through a stirred solution of (*S*)-4-(methoxymethoxy)hept-1-ene (**SI-2**) (6.64 g, 42.0 mmol) in CH_2Cl_2 (150 mL) at -78°C until a persistent blue color appeared (ca. 2 h). Then, pure oxygen gas was passed through the solution to remove excess of ozone. Triphenylphosphine (22.0 g, 84.0 mmol) was added portion wise at -78°C and the reaction mixture was stirred until it warmed to room temperature (ca. 2 h). After treating with saturated

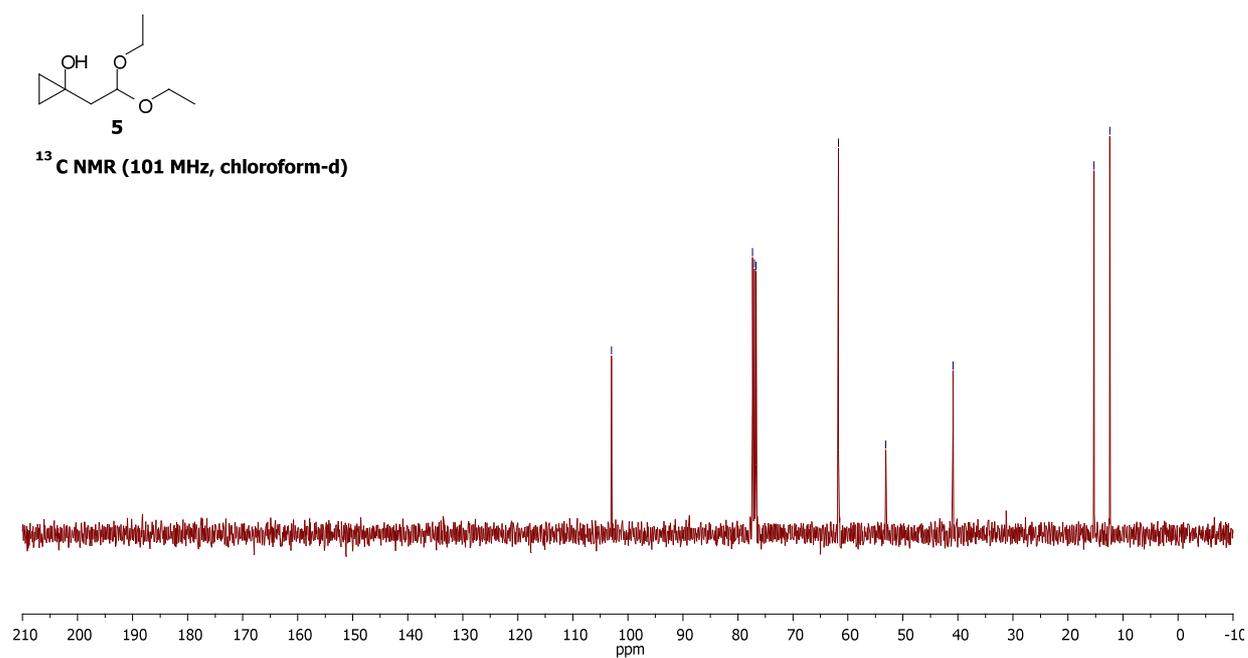
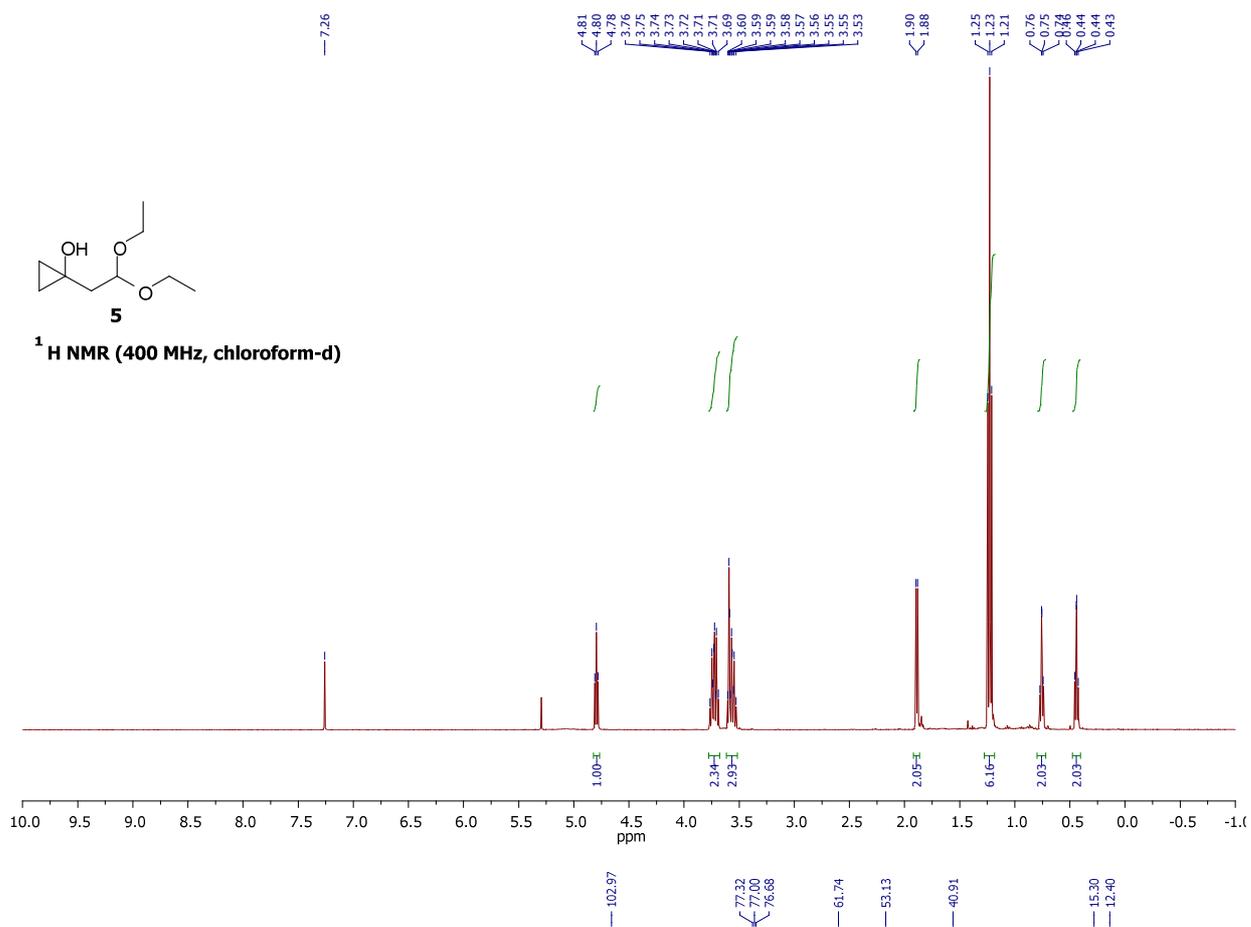
aqueous solution of NaHCO₃ (50 mL), the organic layer was separated, and the aqueous layer was extracted with CH₂Cl₂ (3 × 50 mL). The combined organic extracts were dried over Na₂SO₄. The solvent was removed under reduced pressure, and the title compound was isolated by column chromatography (SiO₂, PE:EtOAc, 20:1). Colorless oil (6.25 g, 93%). *R*_f = 0.47 (PE:EtOAc, 4:1). [α]_D²⁰ = +20.1 (c 1.48, CHCl₃). ¹H NMR (400 MHz, CDCl₃): δ = 9.79 (dd, *J* = 2.8, 1.8 Hz, 1H), 4.68 (d, *J* = 7.0 Hz, 1H), 4.64 (d, *J* = 7.0 Hz, 1H), 4.12–4.04 (m, 1H), 3.34 (s, 3H), 2.63 (ddd, *J* = 16.3, 7.0, 2.8 Hz, 1H), 2.55 (ddd, *J* = 16.3, 4.8, 1.8 Hz, 1H), 1.68–1.56 (m, 1H), 1.56–1.29 (m, 3H), 0.93 (t, *J* = 7.3 Hz, 3H). ¹³C NMR (100.6 MHz, CDCl₃): δ = 201.47, 95.79, 72.91, 55.59, 48.73, 37.10, 18.44, 13.97. HRMS (ESI) calcd. for C₈H₁₆O₃Na⁺ [M+Na]⁺ 183.0992, found *m/z* 183.0990.

References

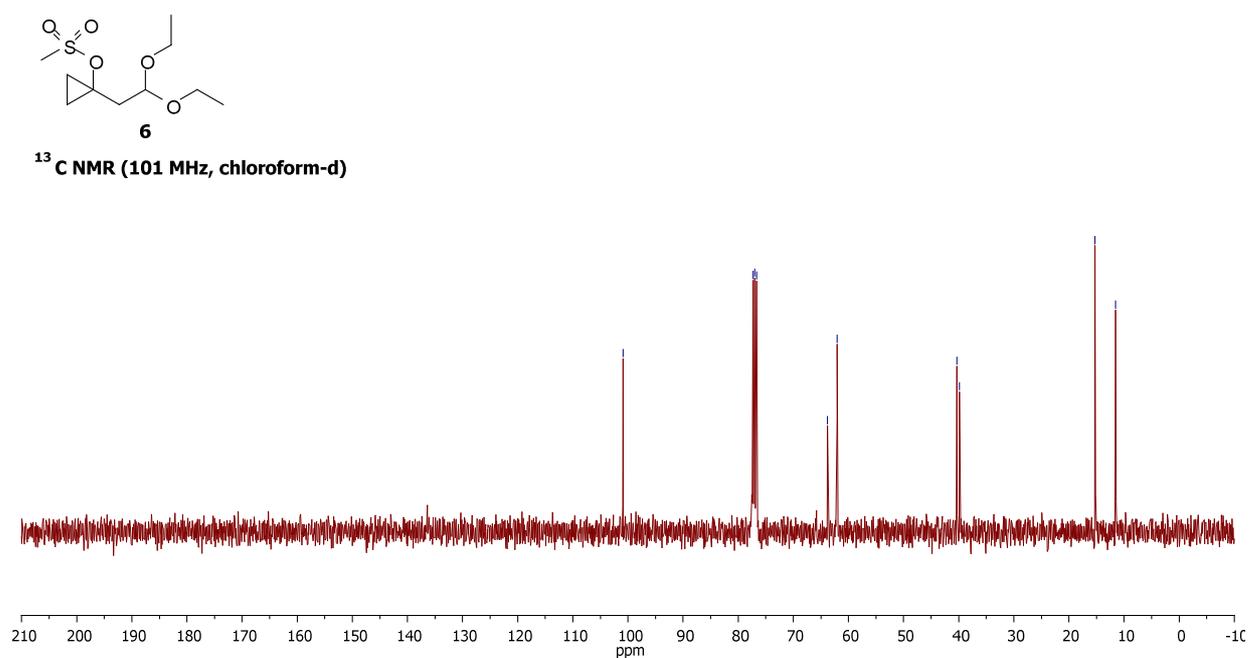
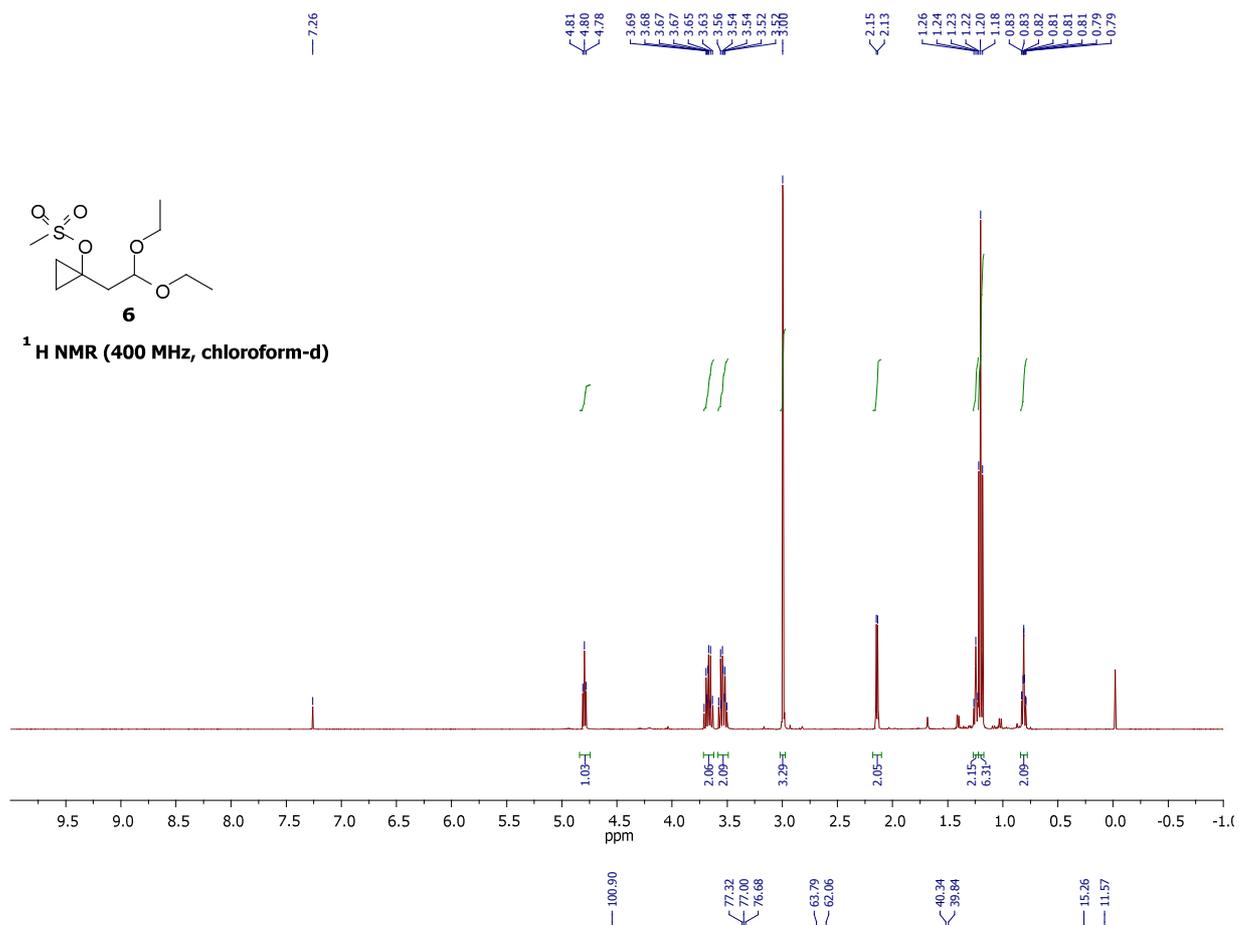
1. Dale, J.A.; Dull, D.L.; Mosher, H.S.; α-Methoxy-α-trifluoromethylphenylacetic acid, a versatile reagent for the determination of enantiomeric composition of alcohols and amines. *J. Org. Chem.* **1969**, *34*, 2543–2549.
2. Berliner, M.A.; Belecki, K.; Simple, rapid procedure for the synthesis of chloromethyl methyl ether and other chloro alkyl ethers. *J. Org. Chem.* **2005**, *70*, 9618–9621.

2. Copies of ^1H and ^{13}C NMR spectra

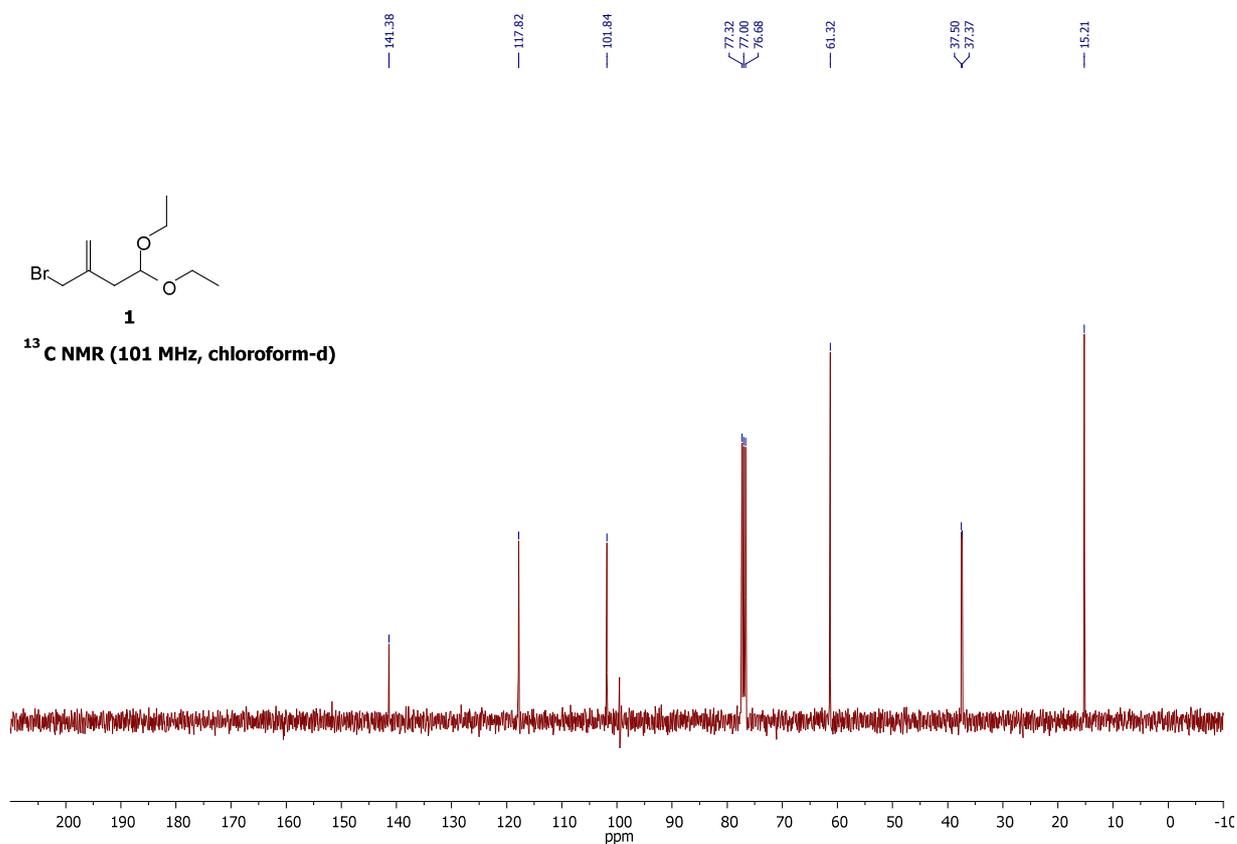
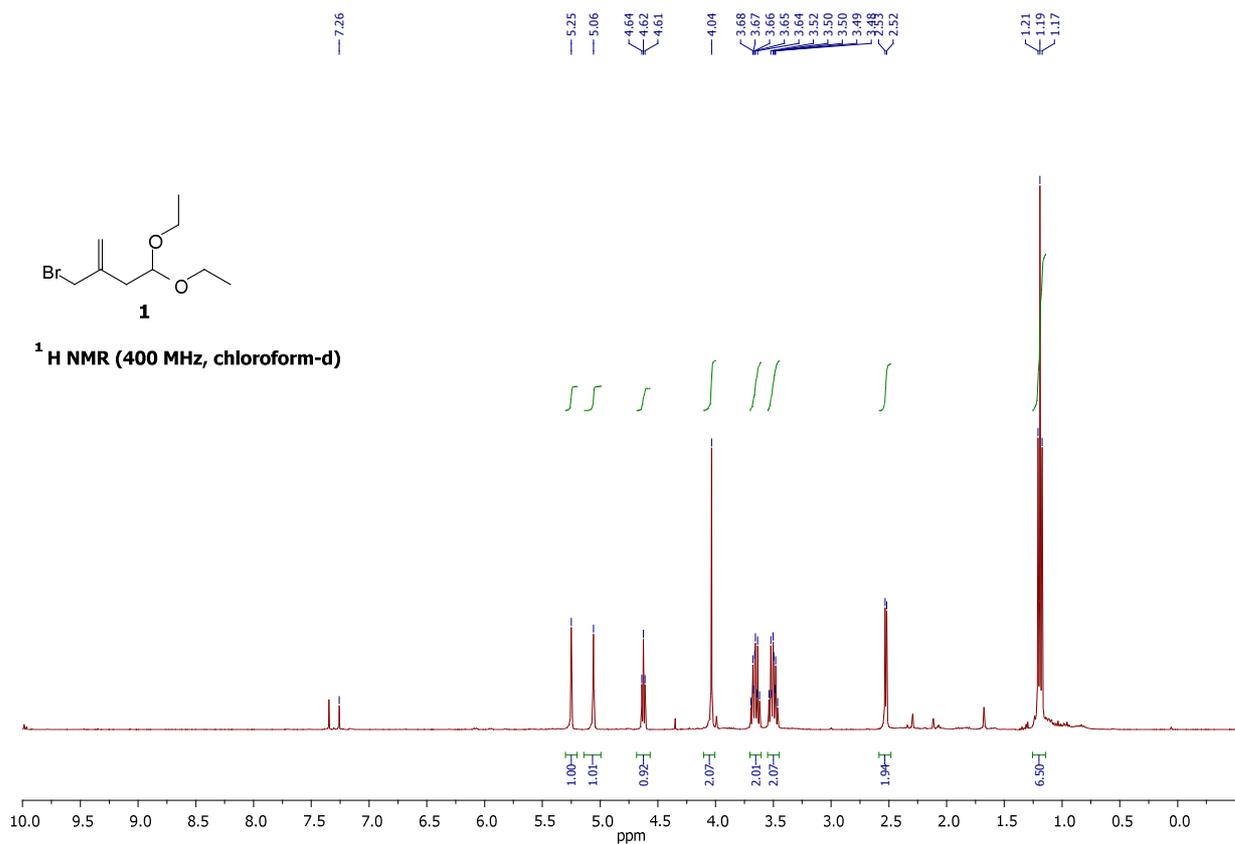
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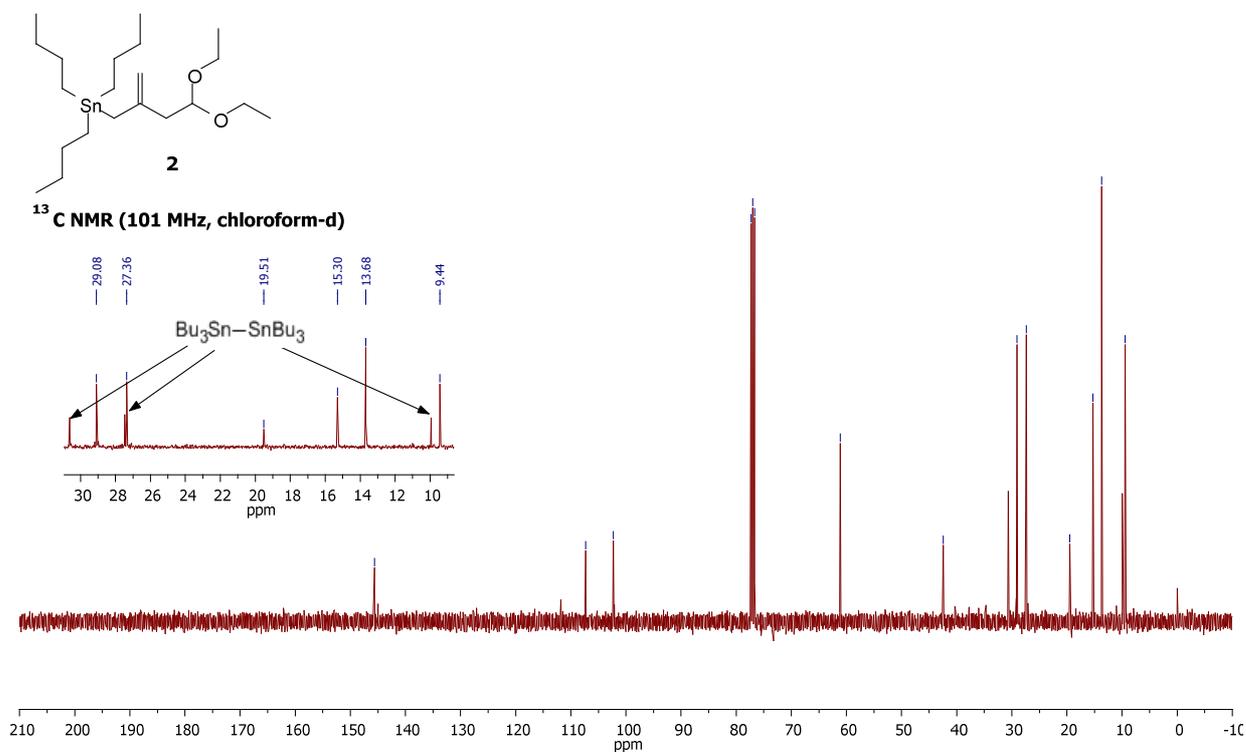
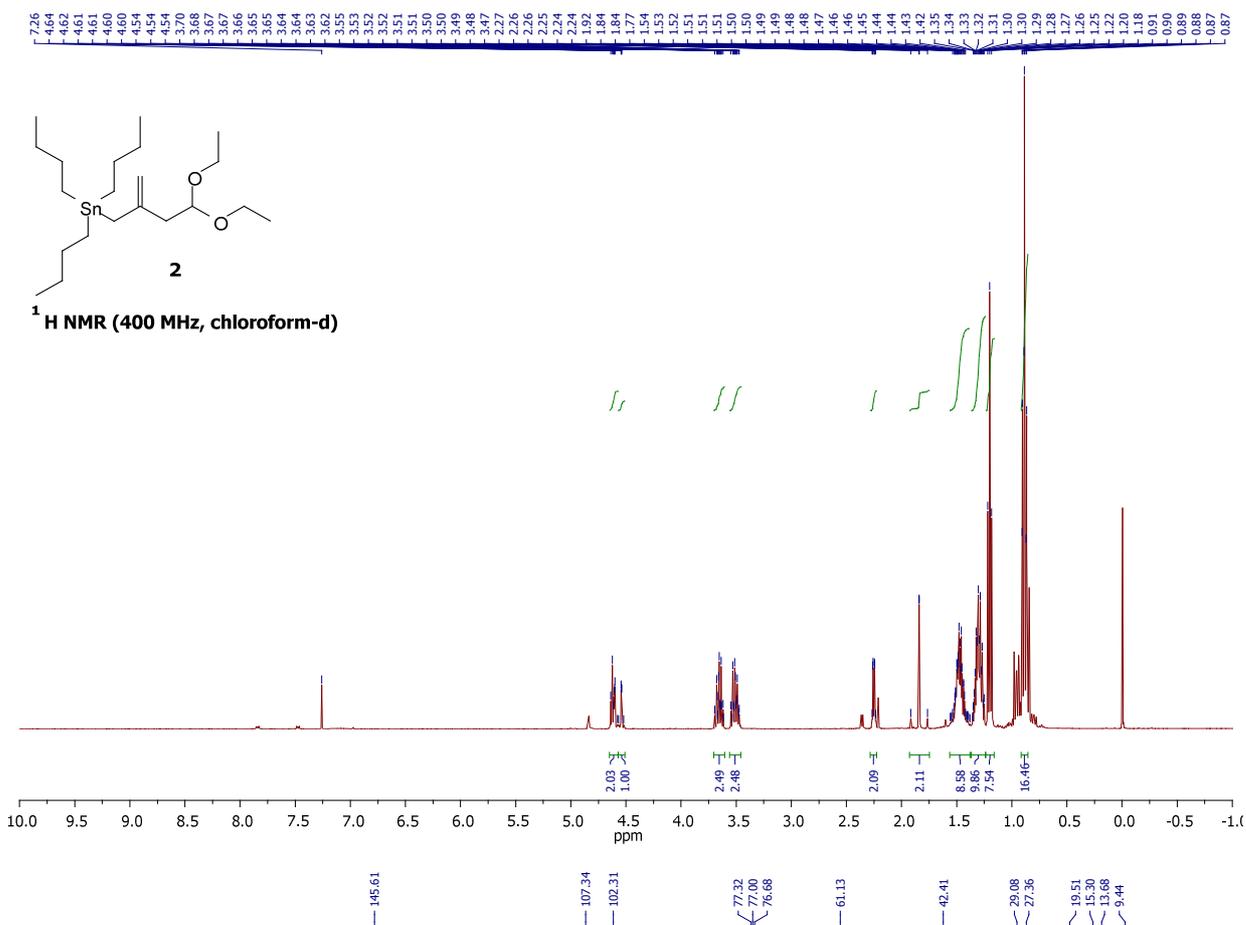
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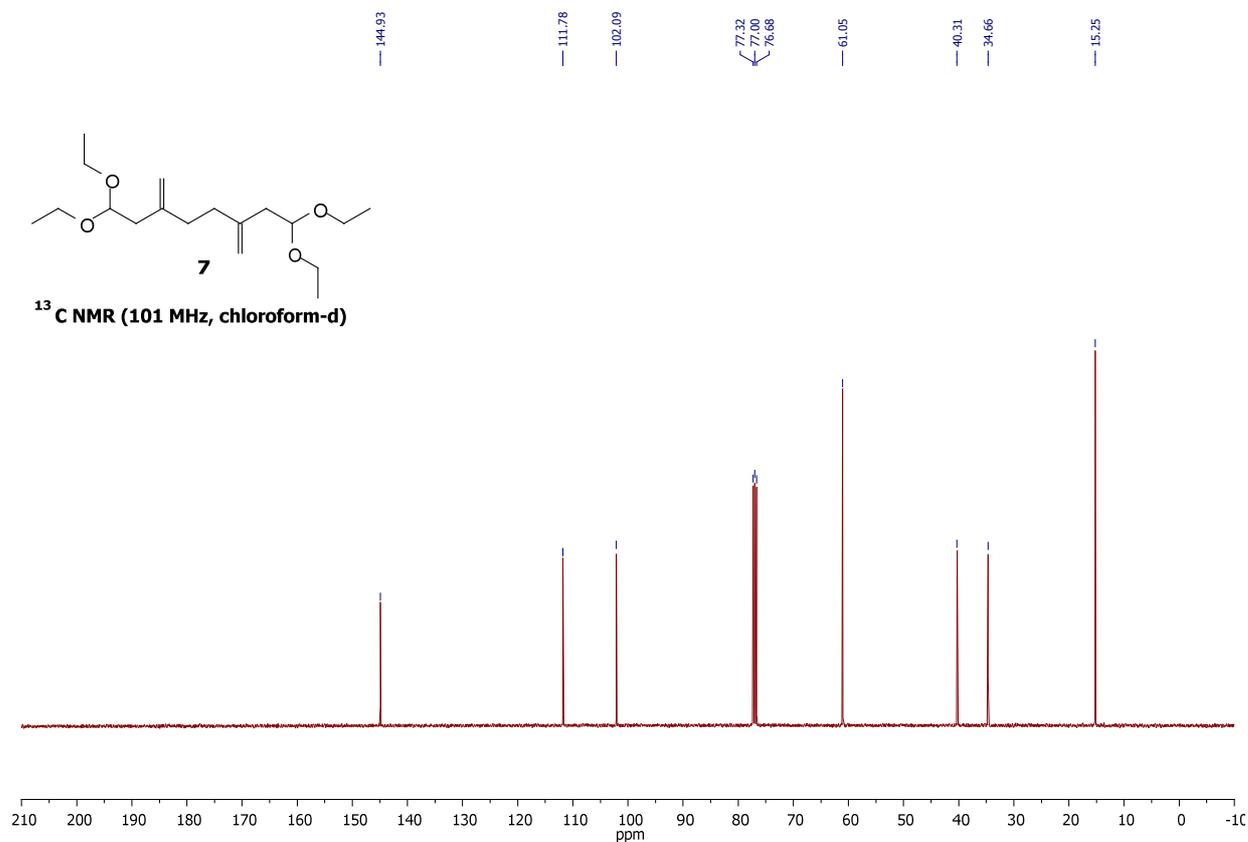
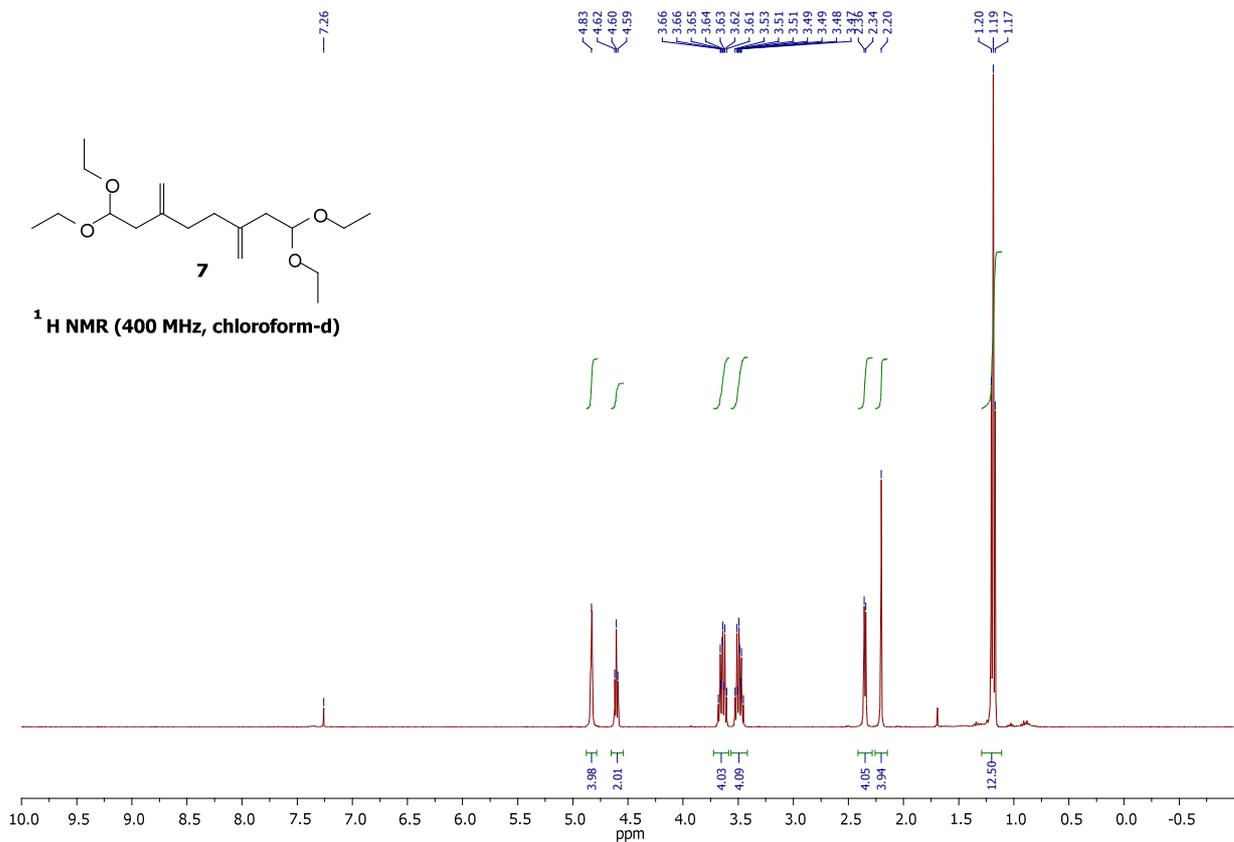
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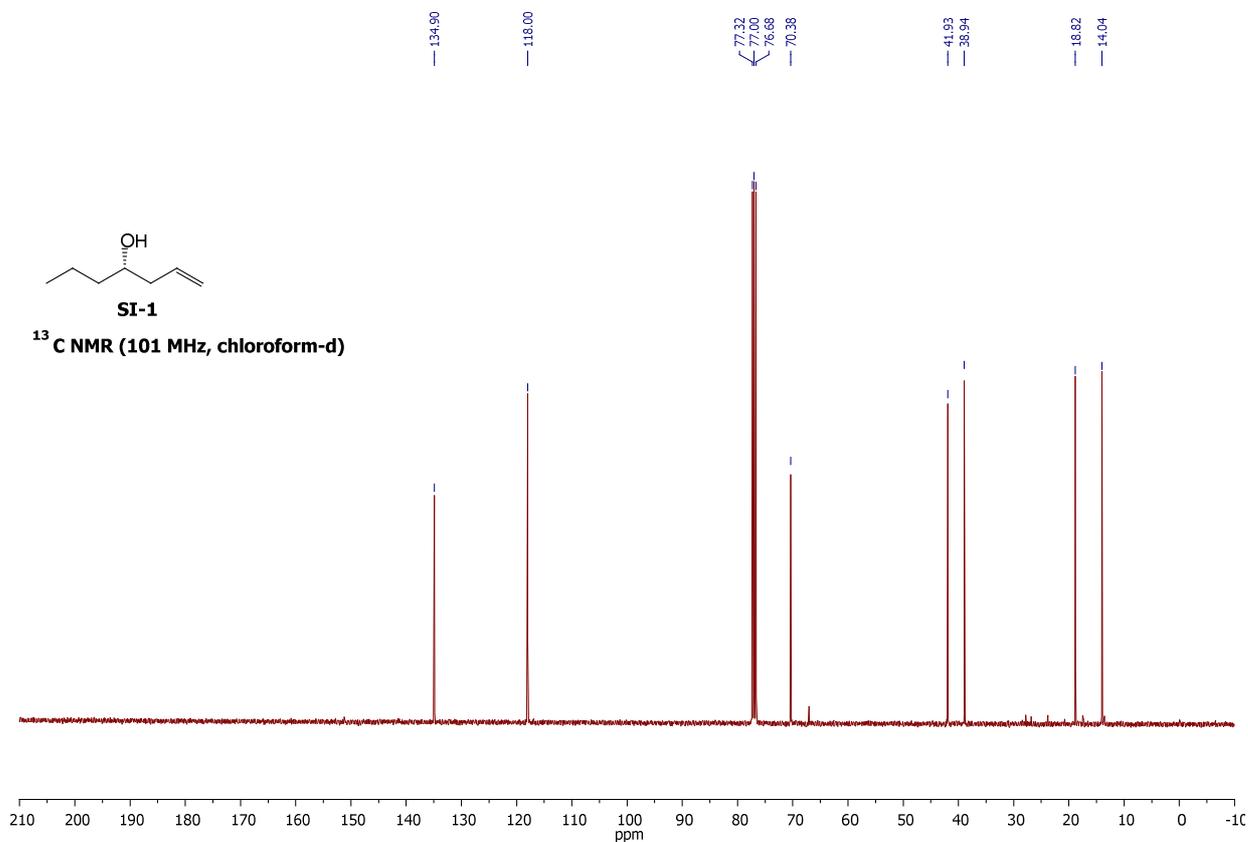
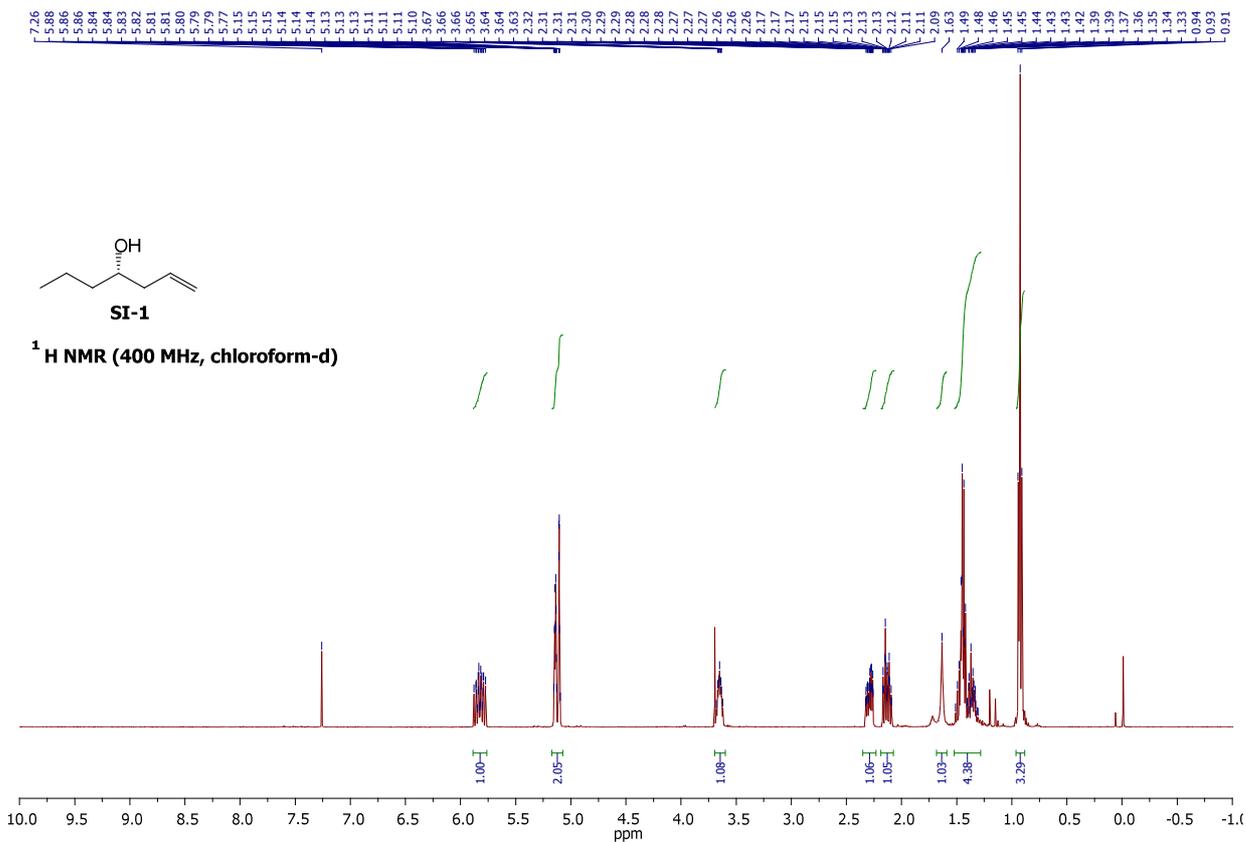
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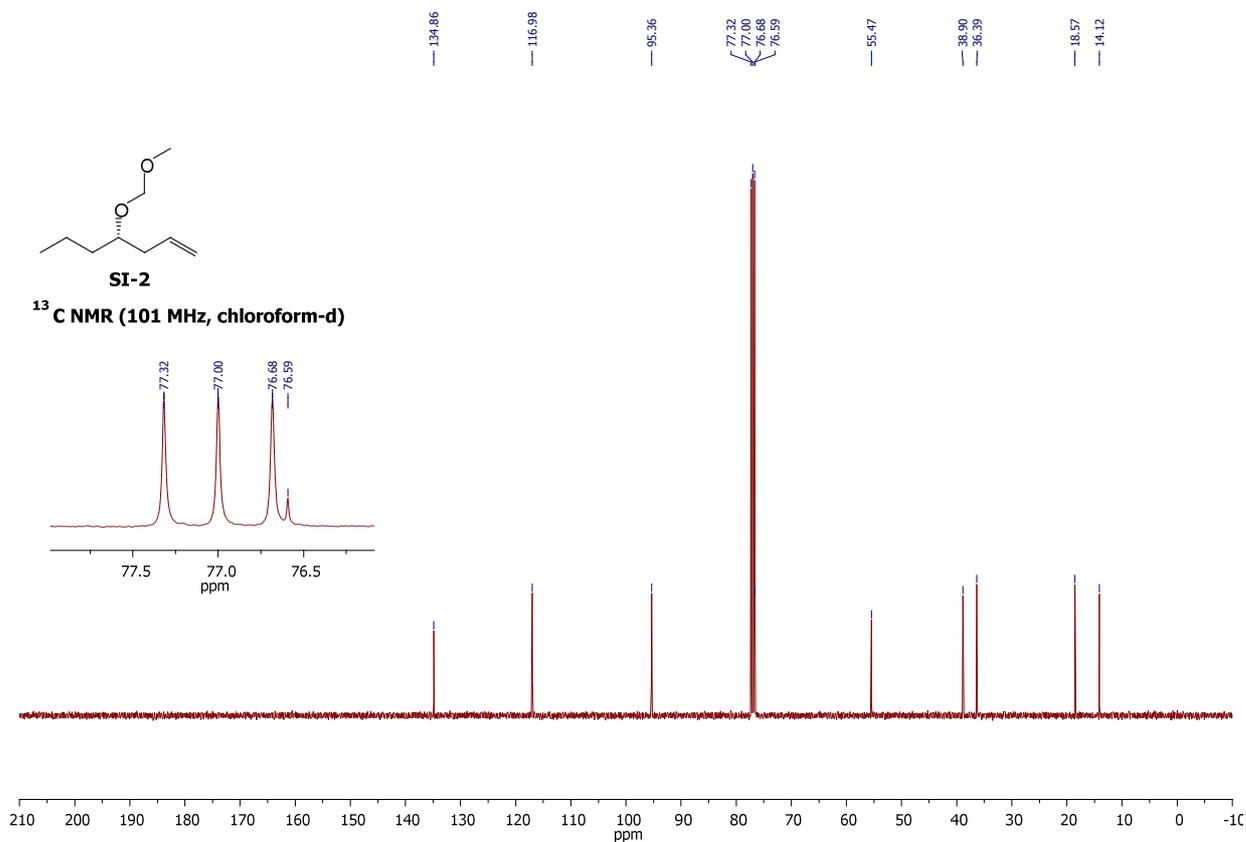
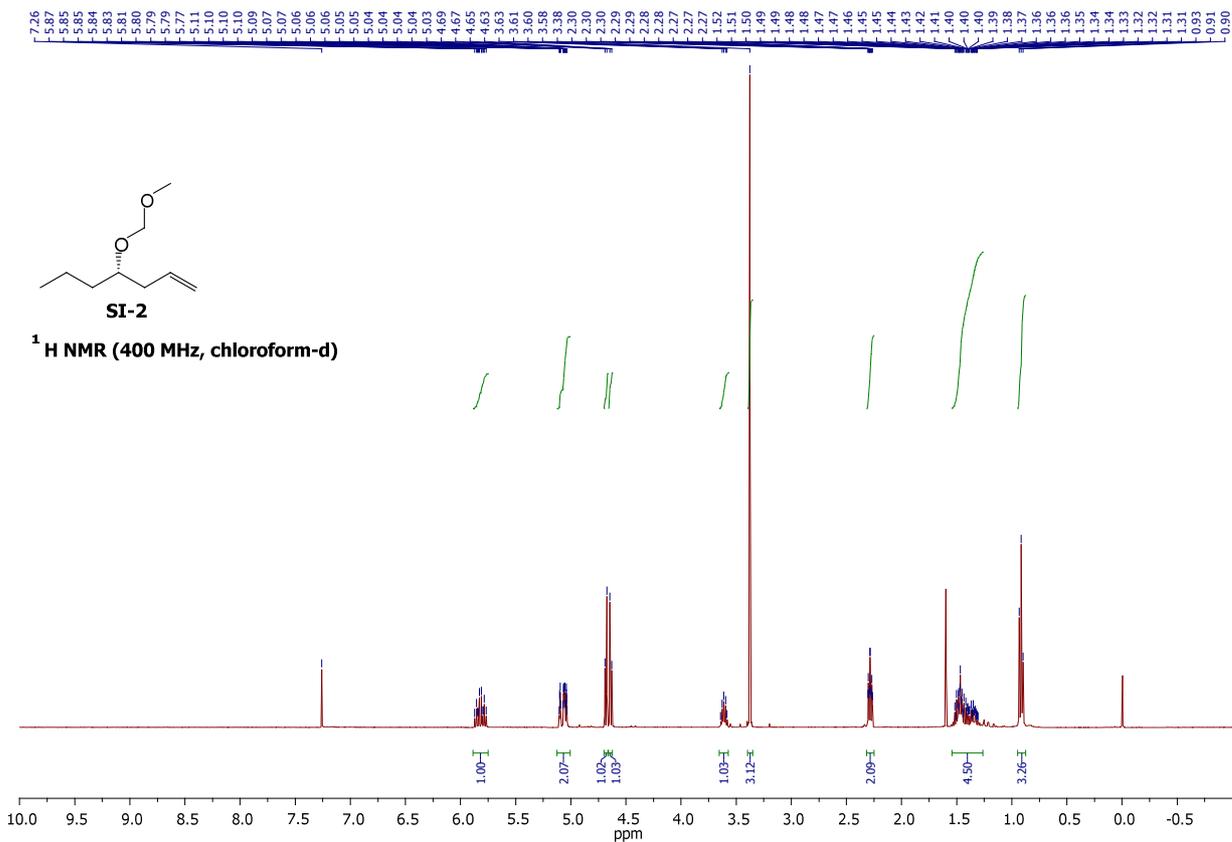
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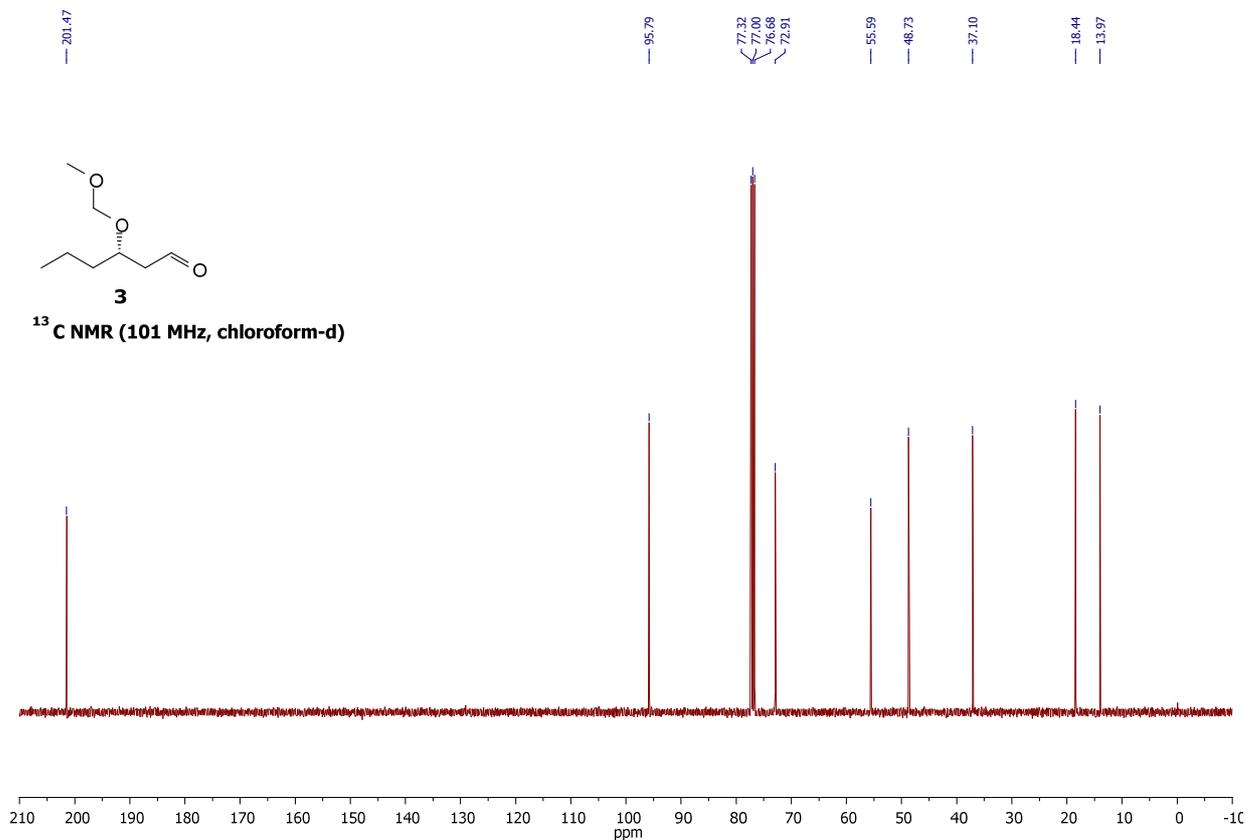
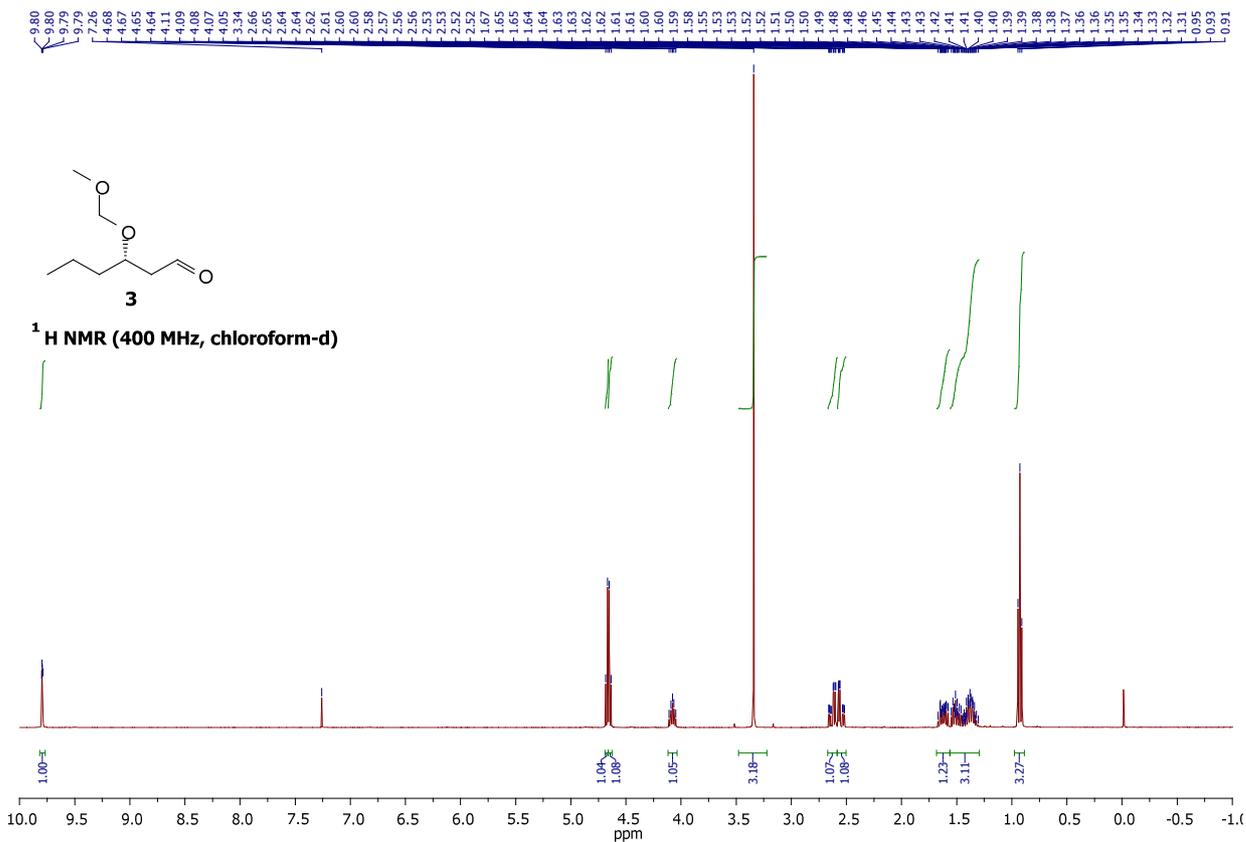
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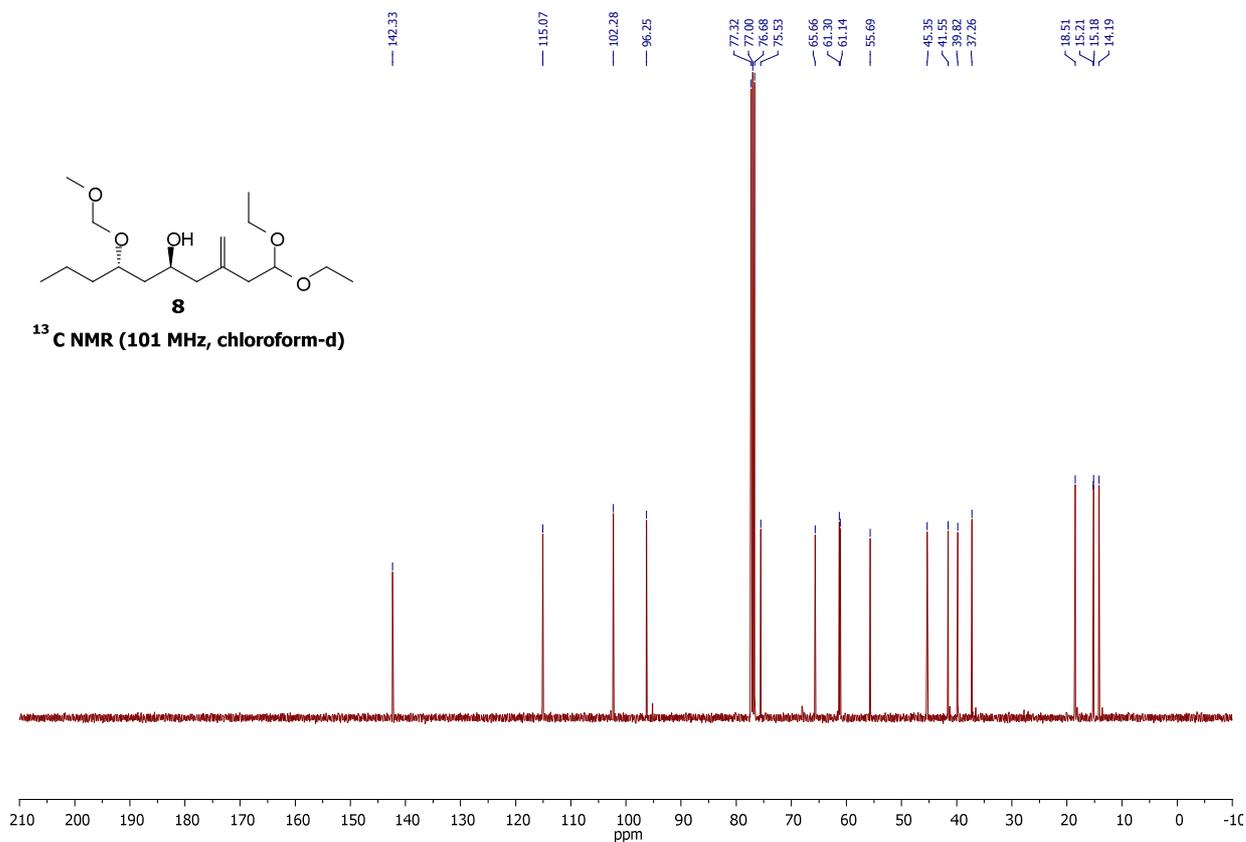
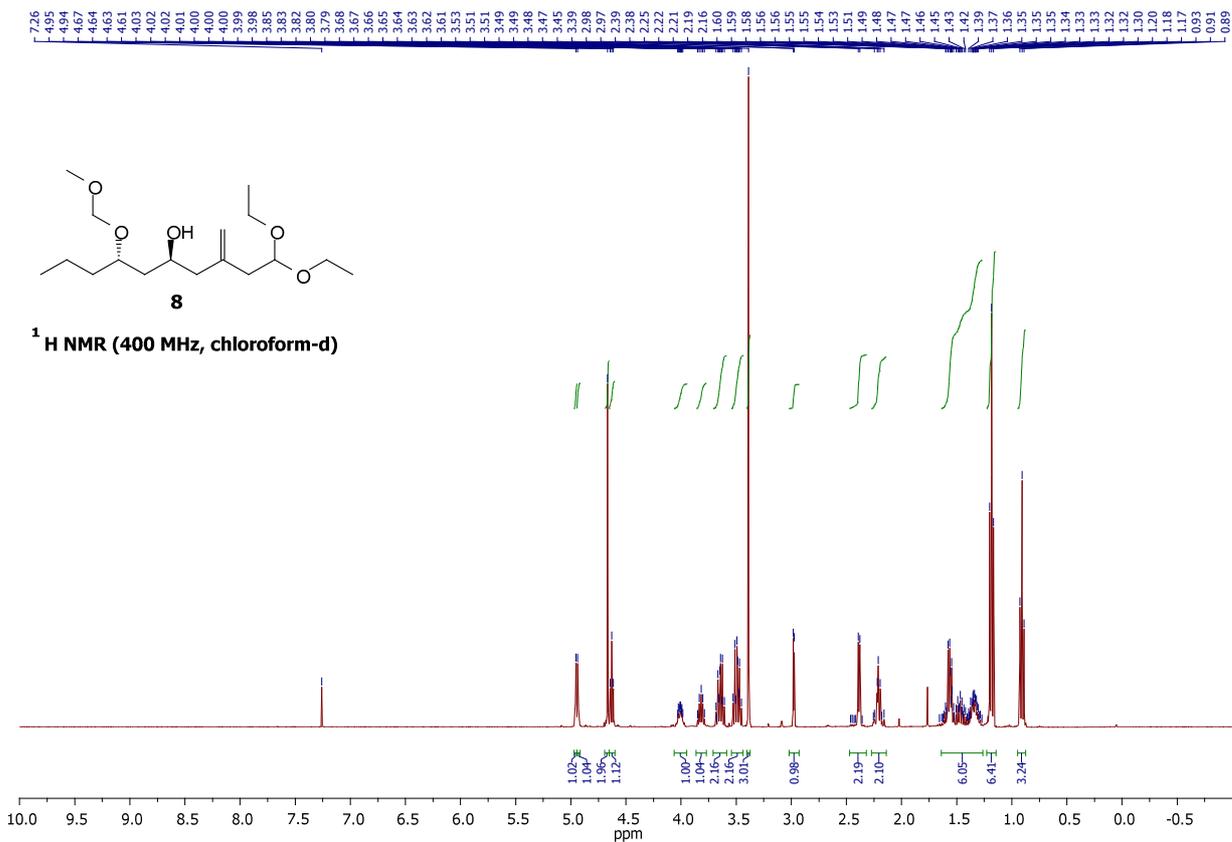
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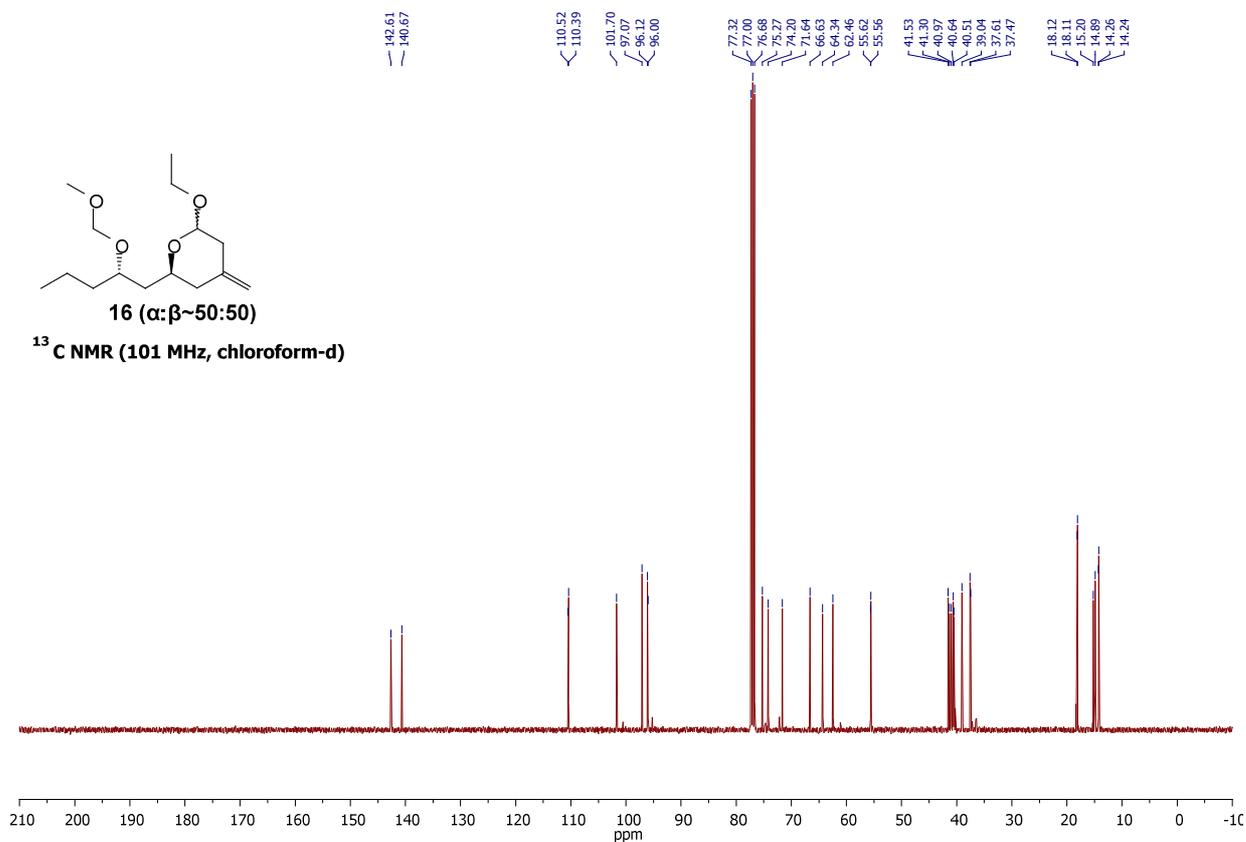
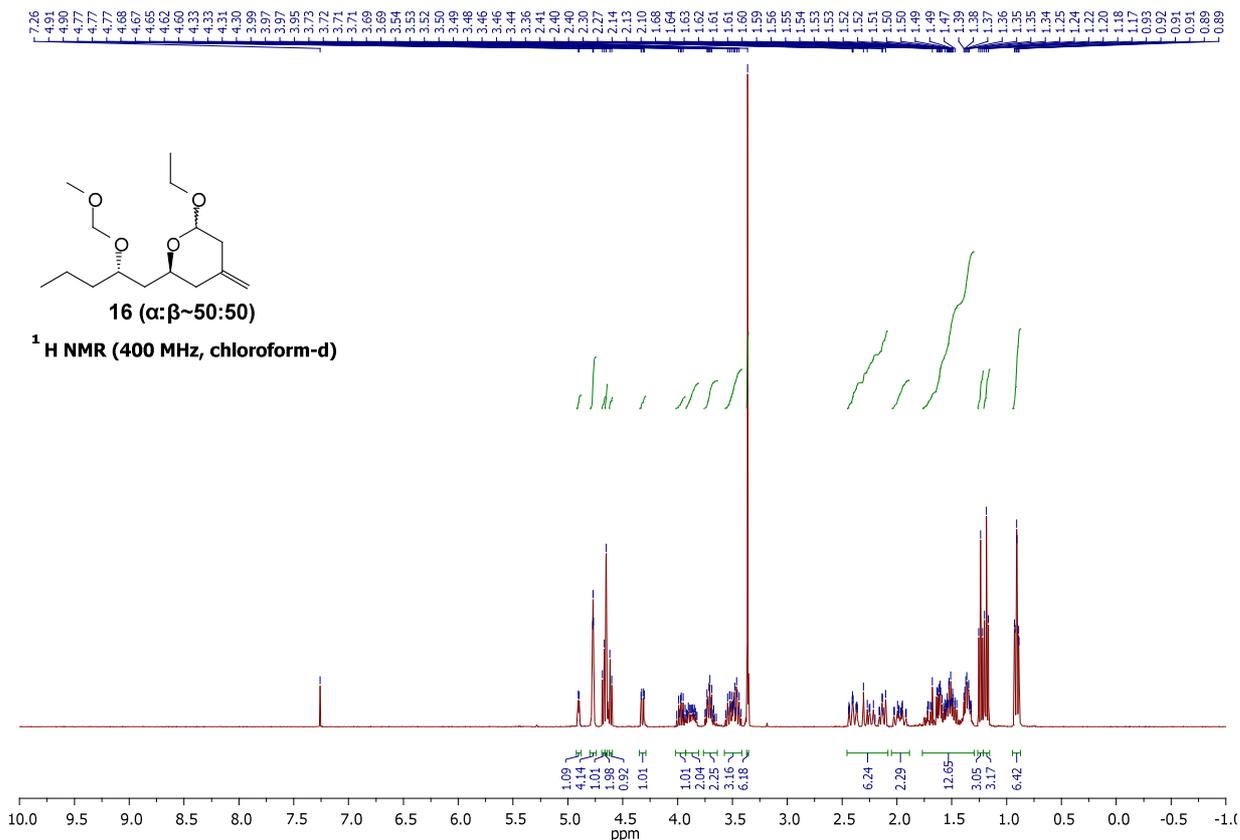
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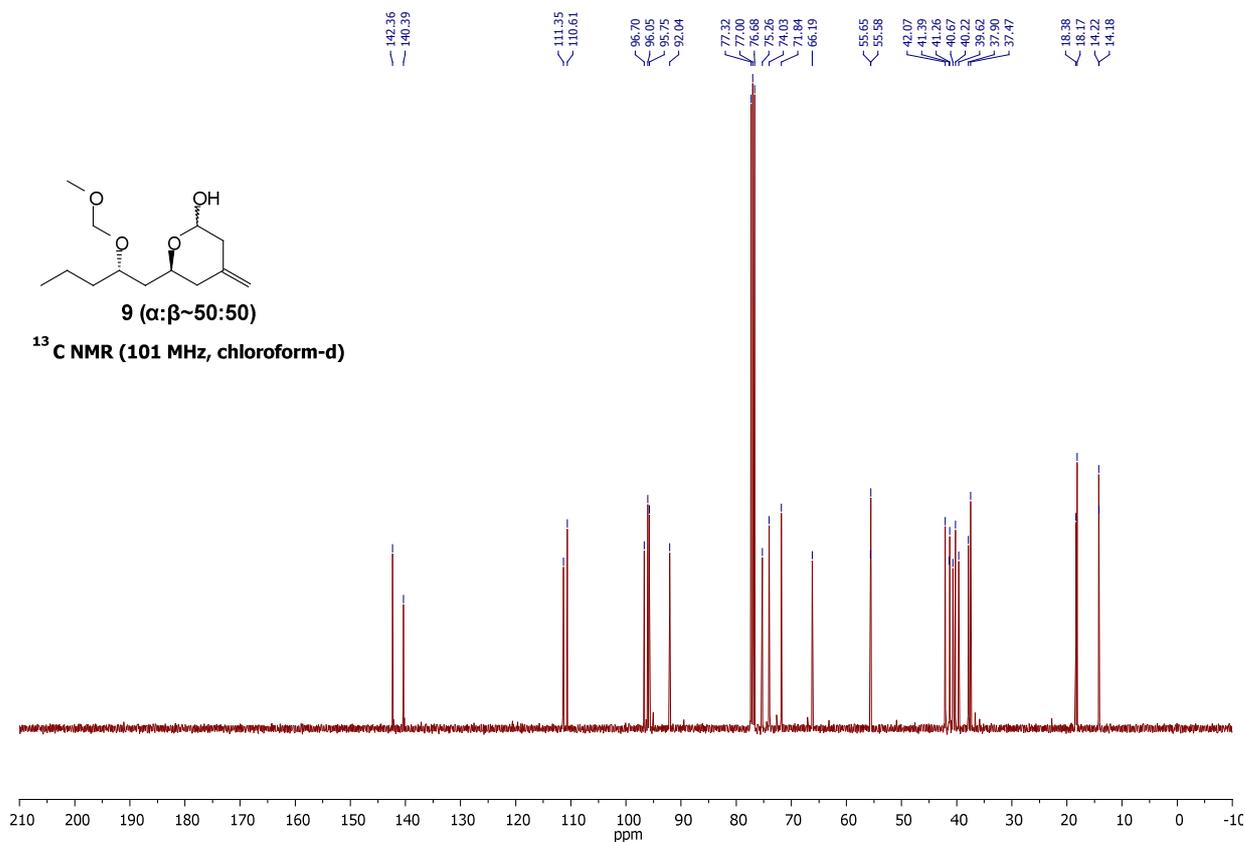
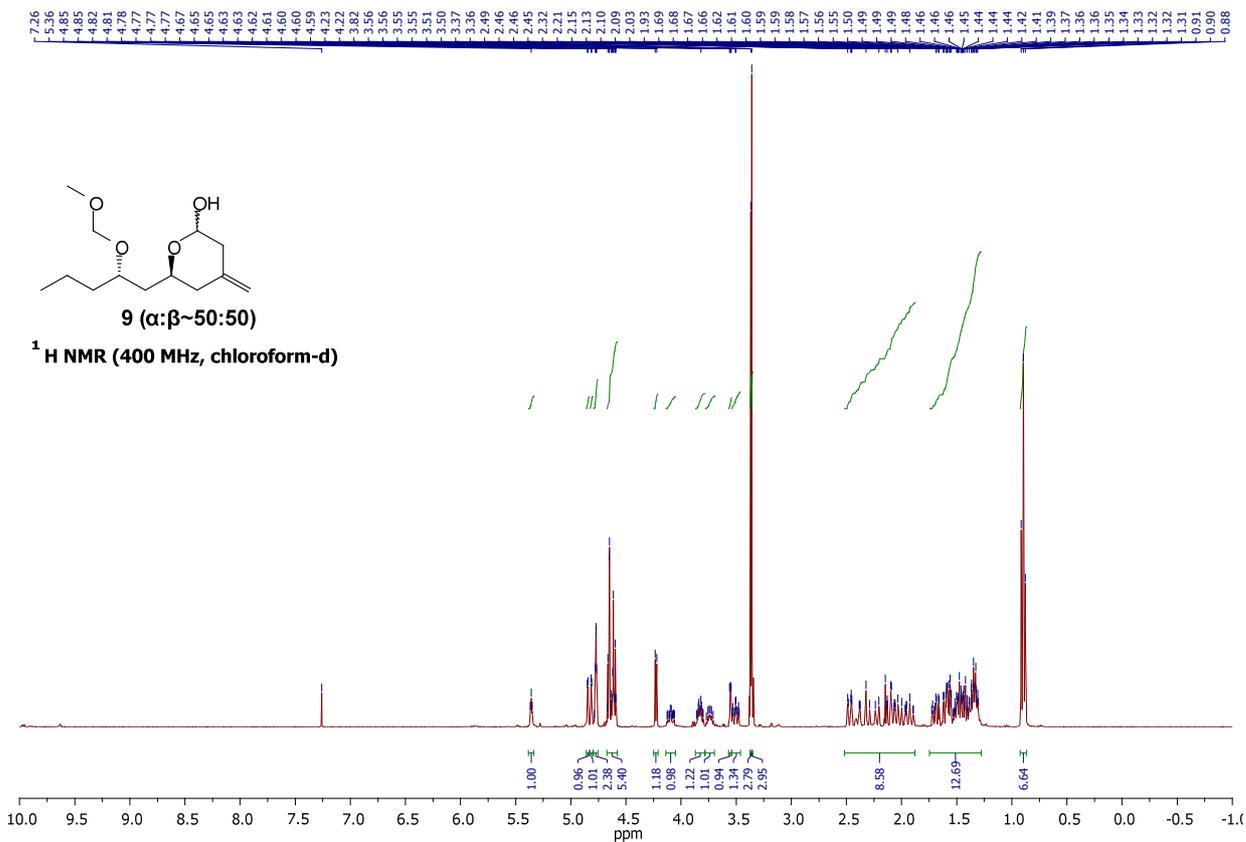
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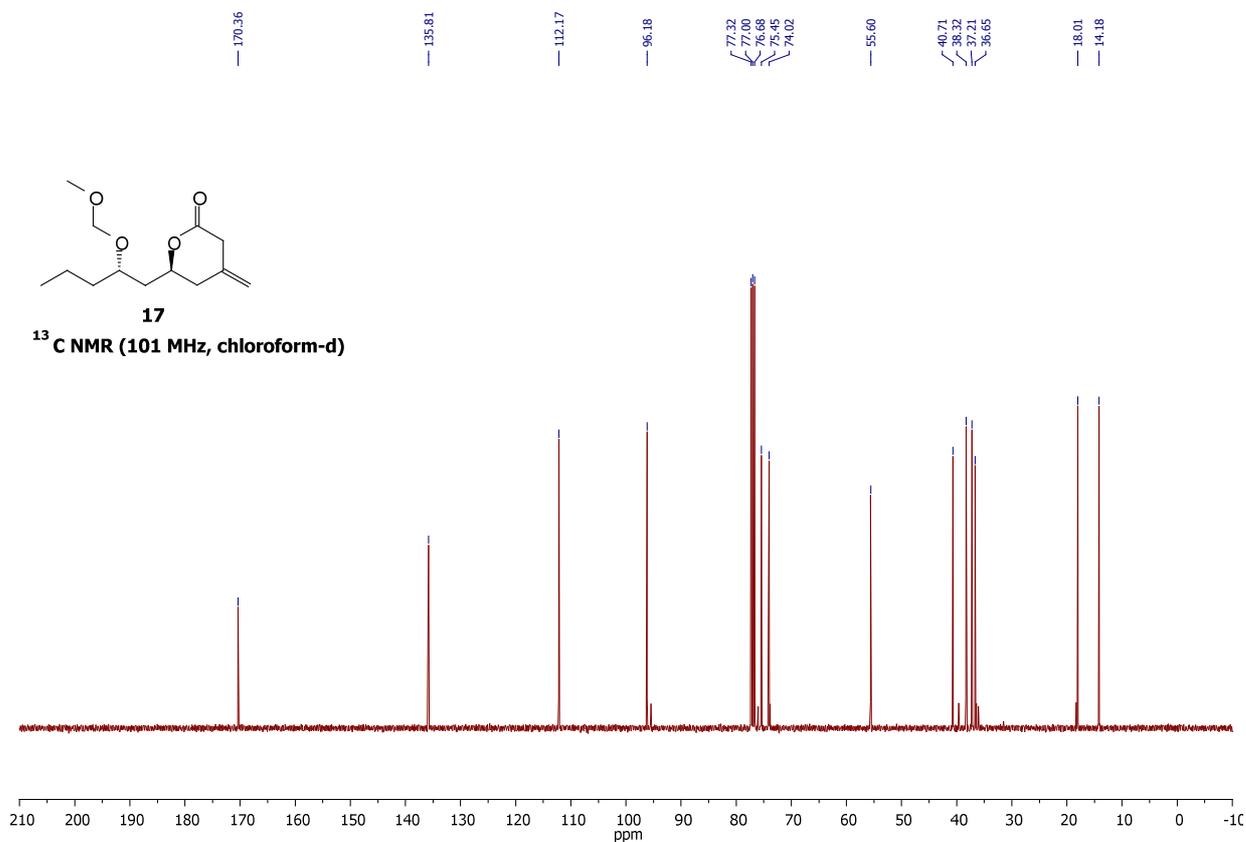
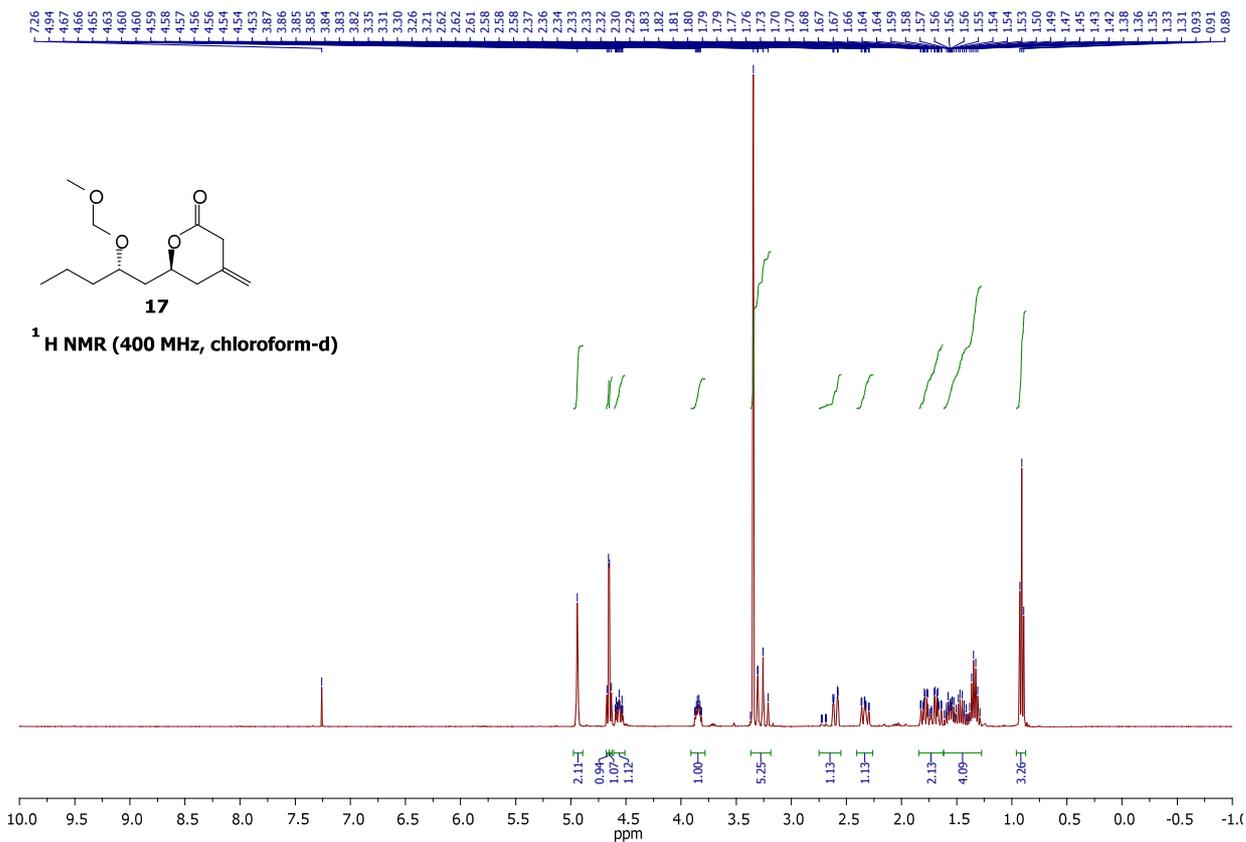
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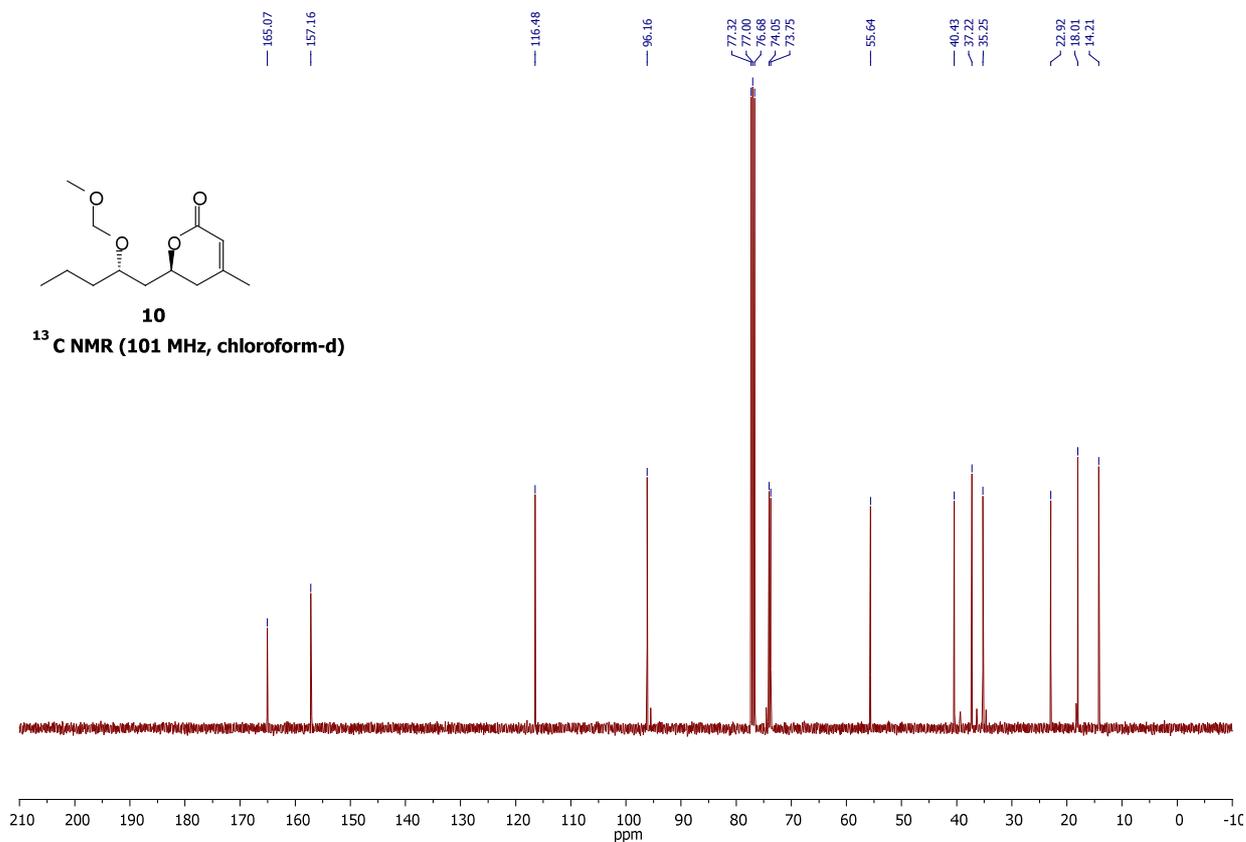
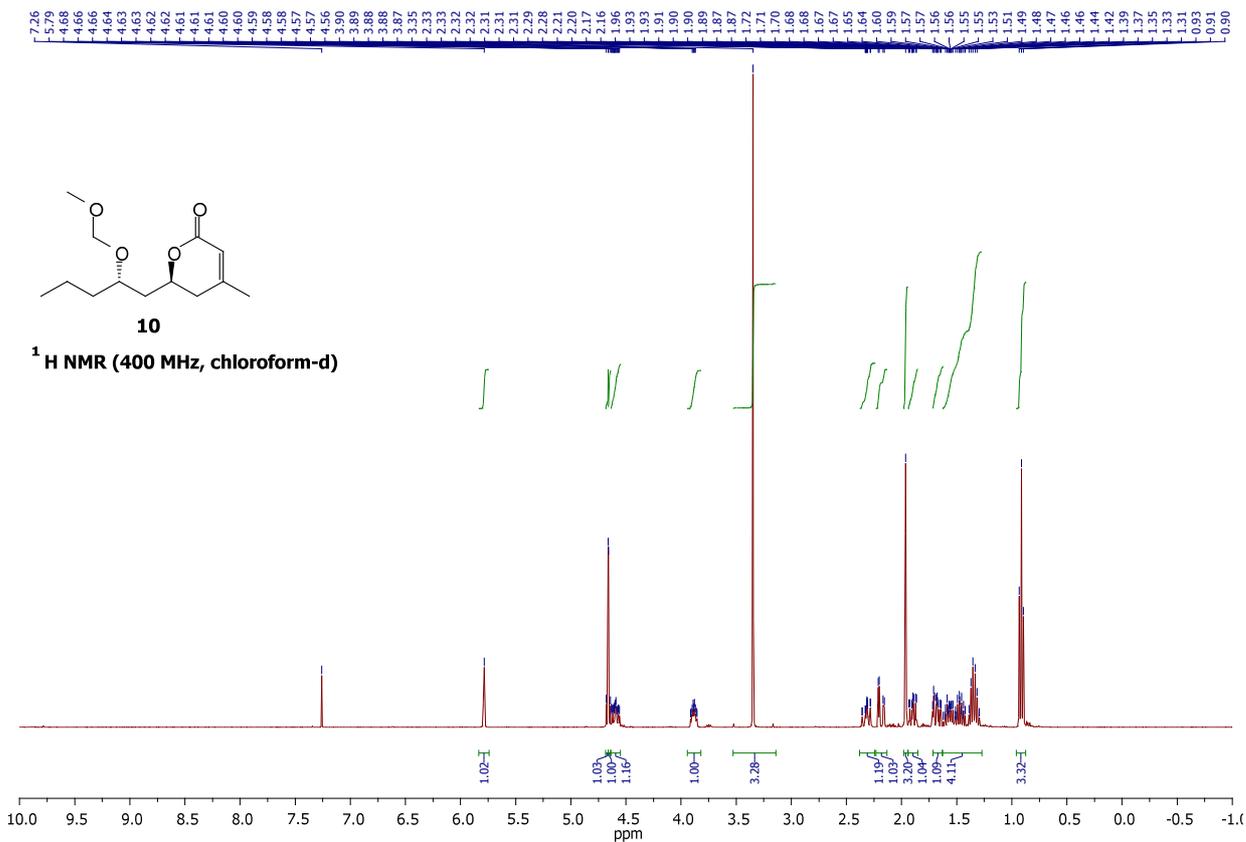
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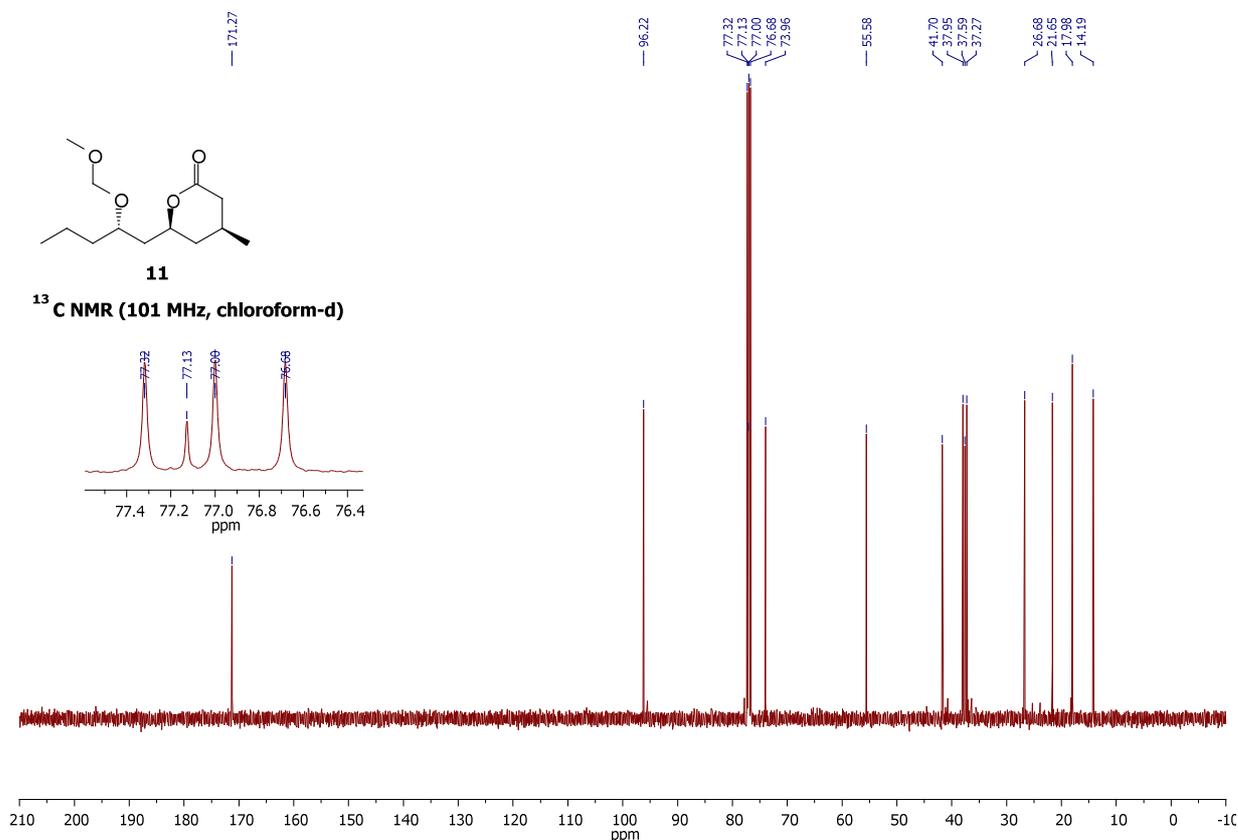
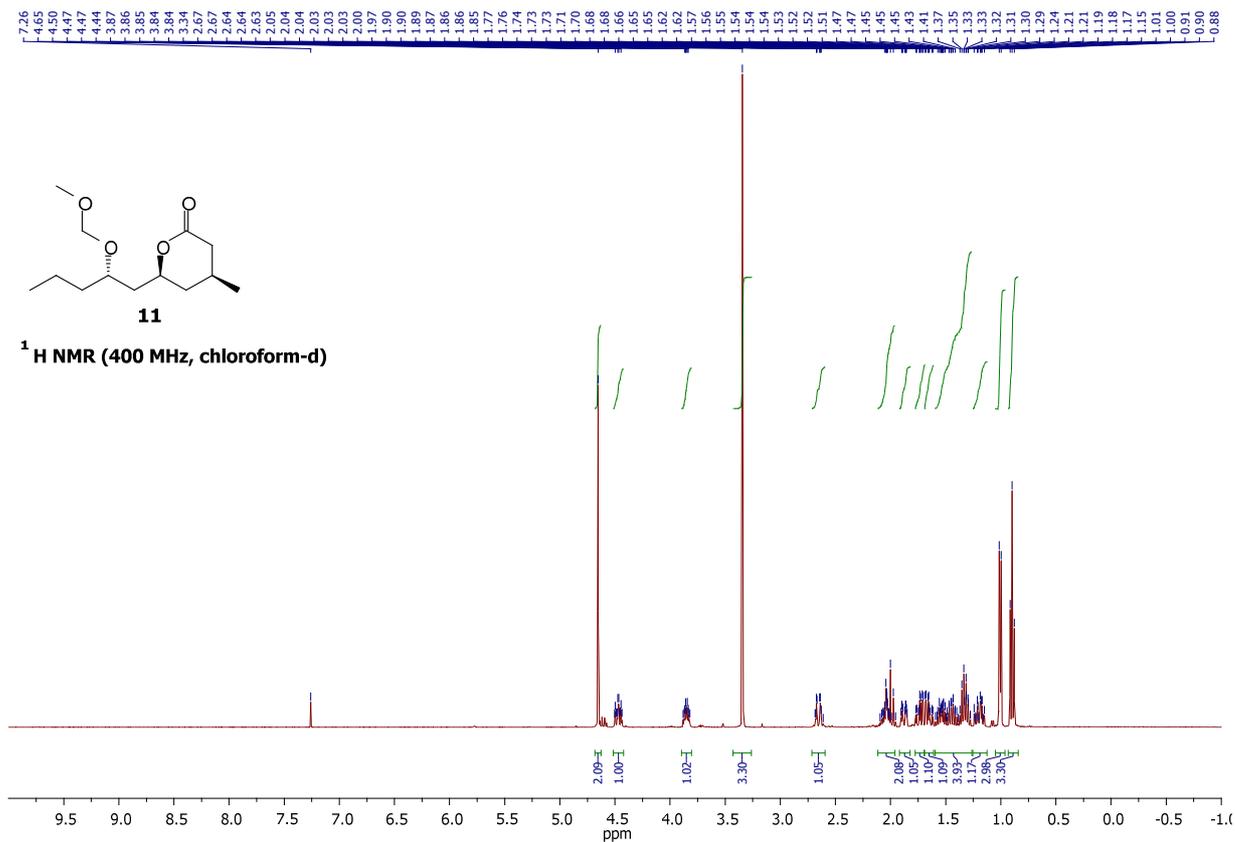
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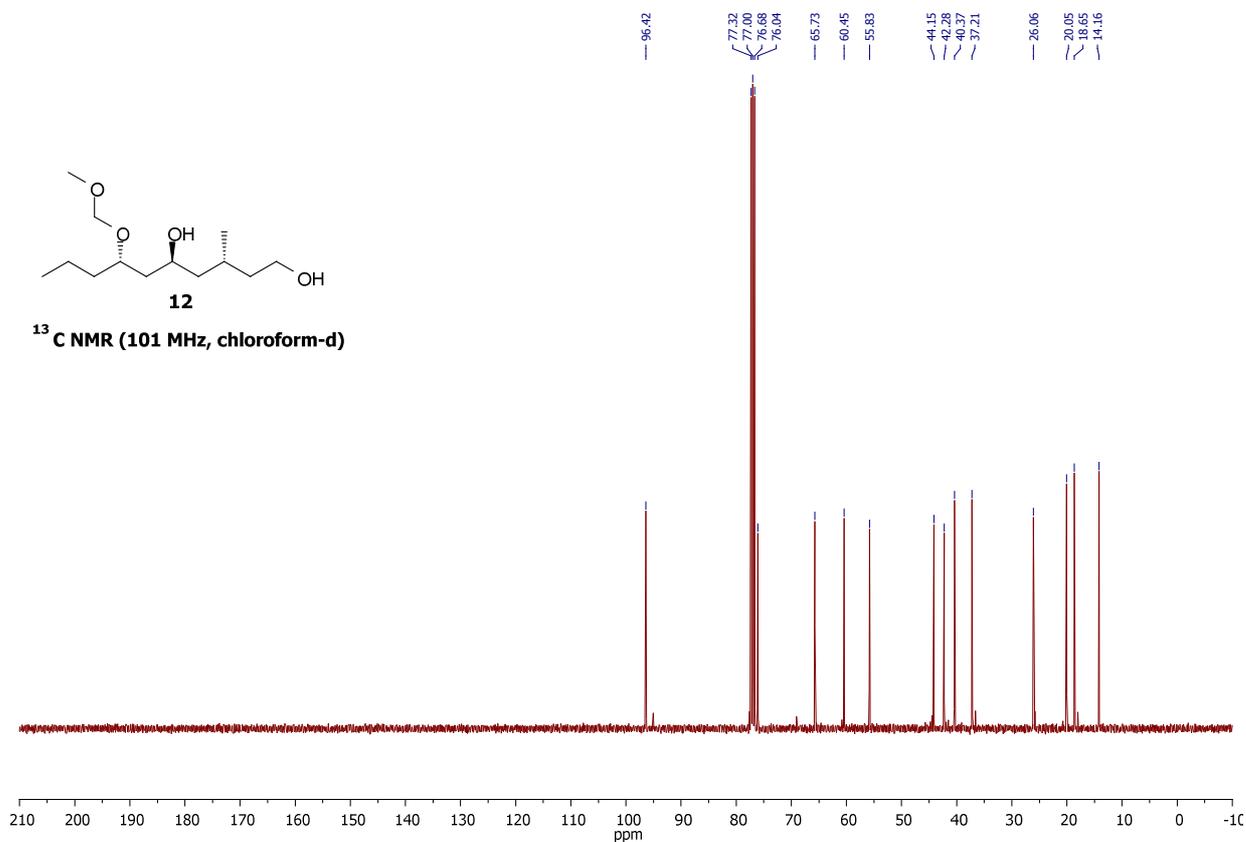
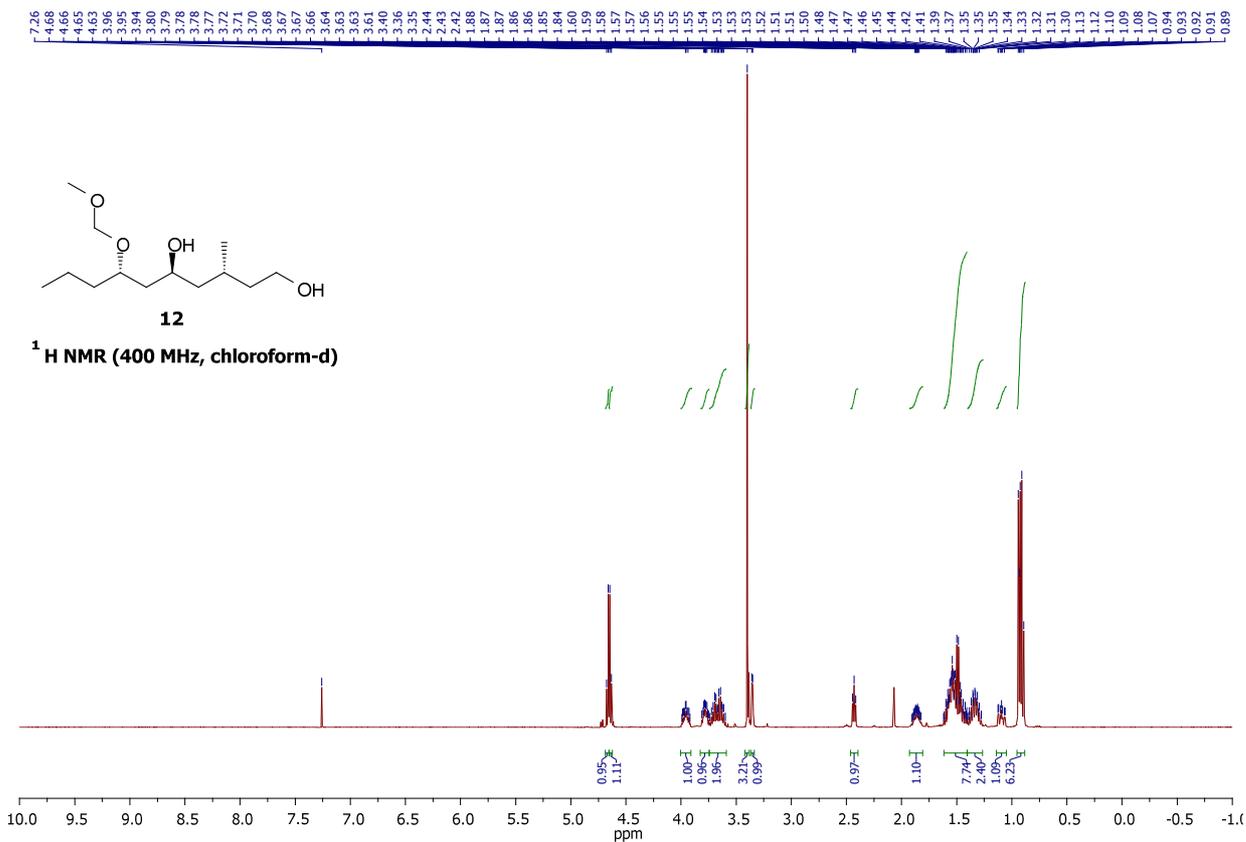
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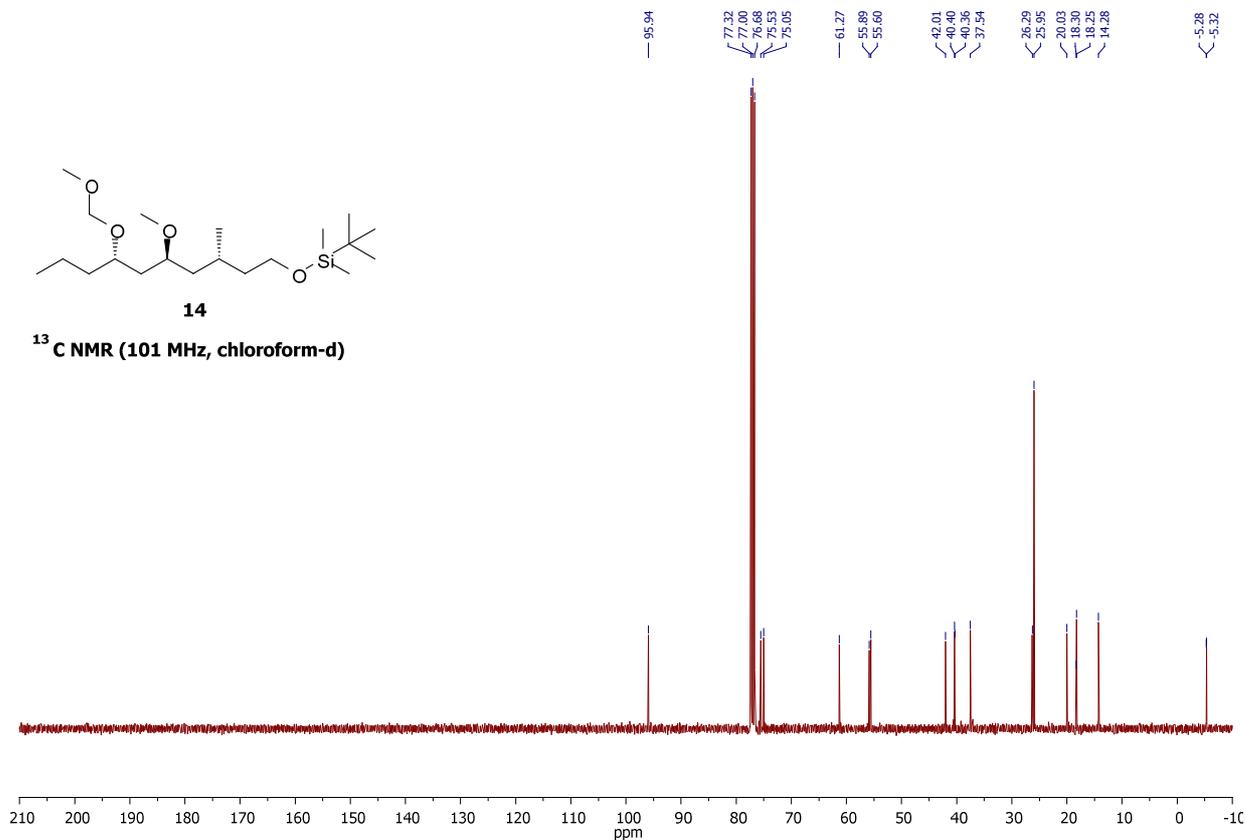
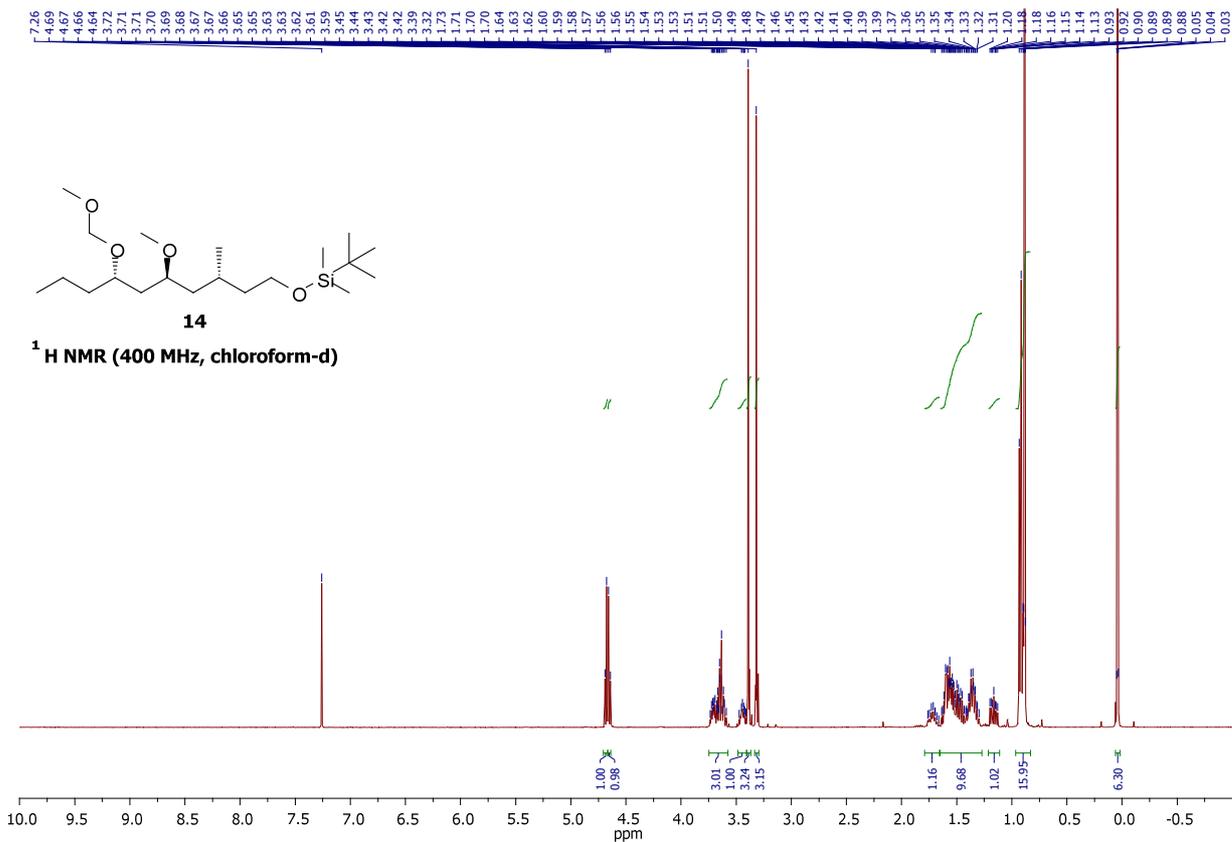
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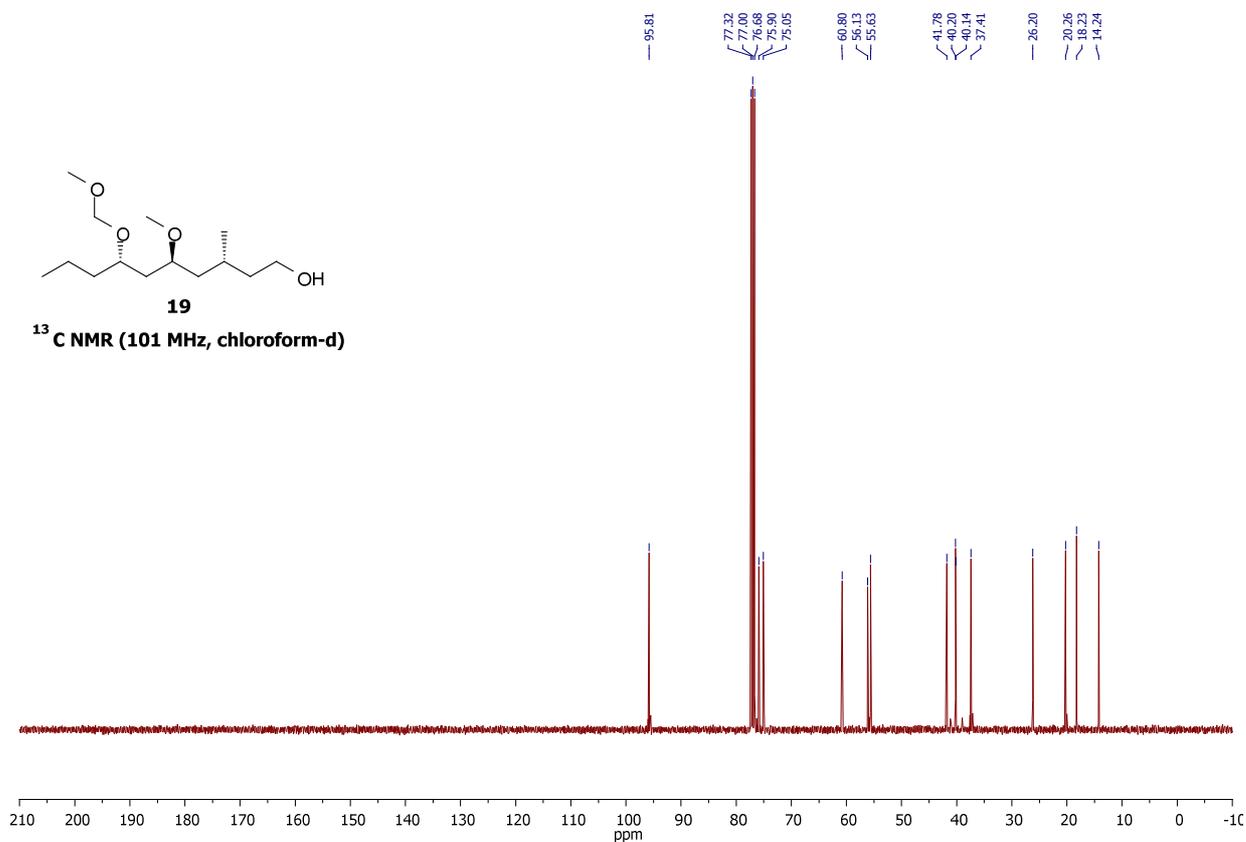
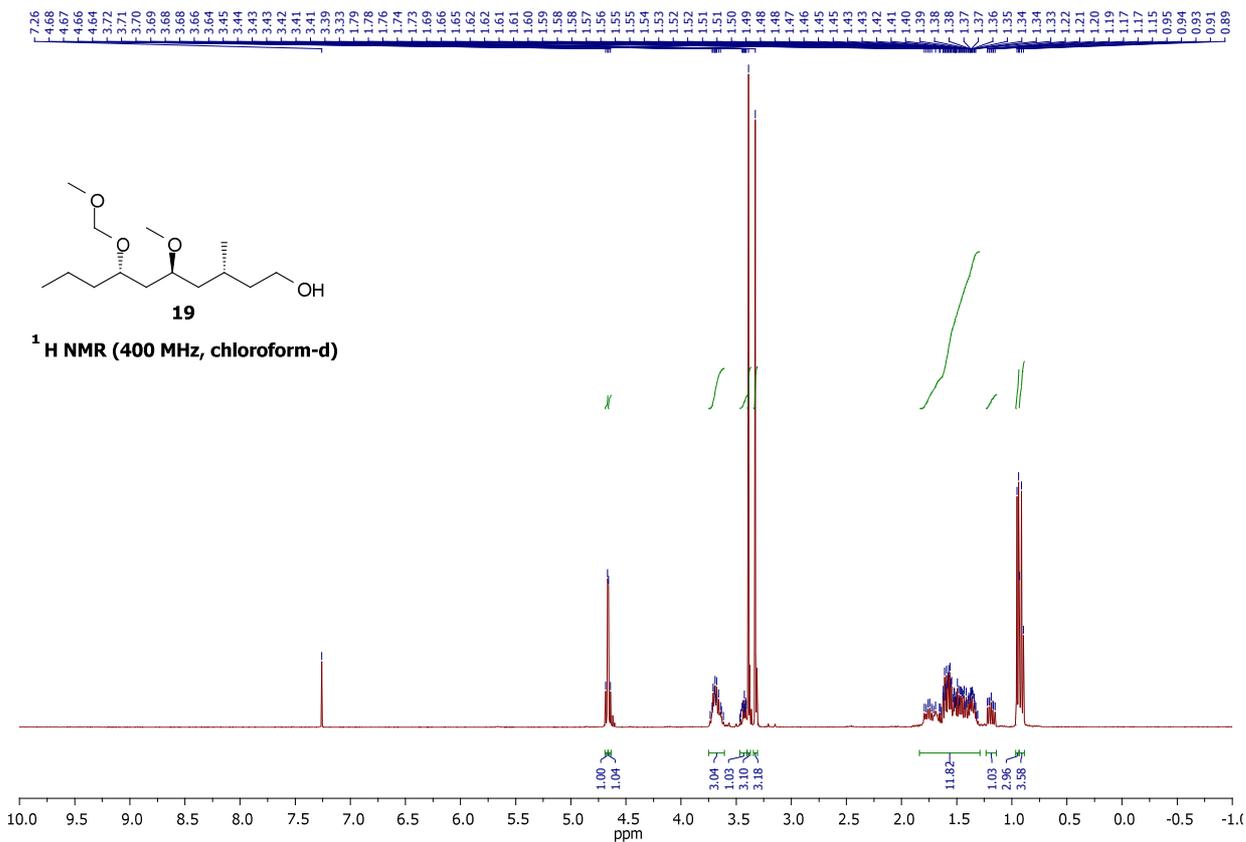
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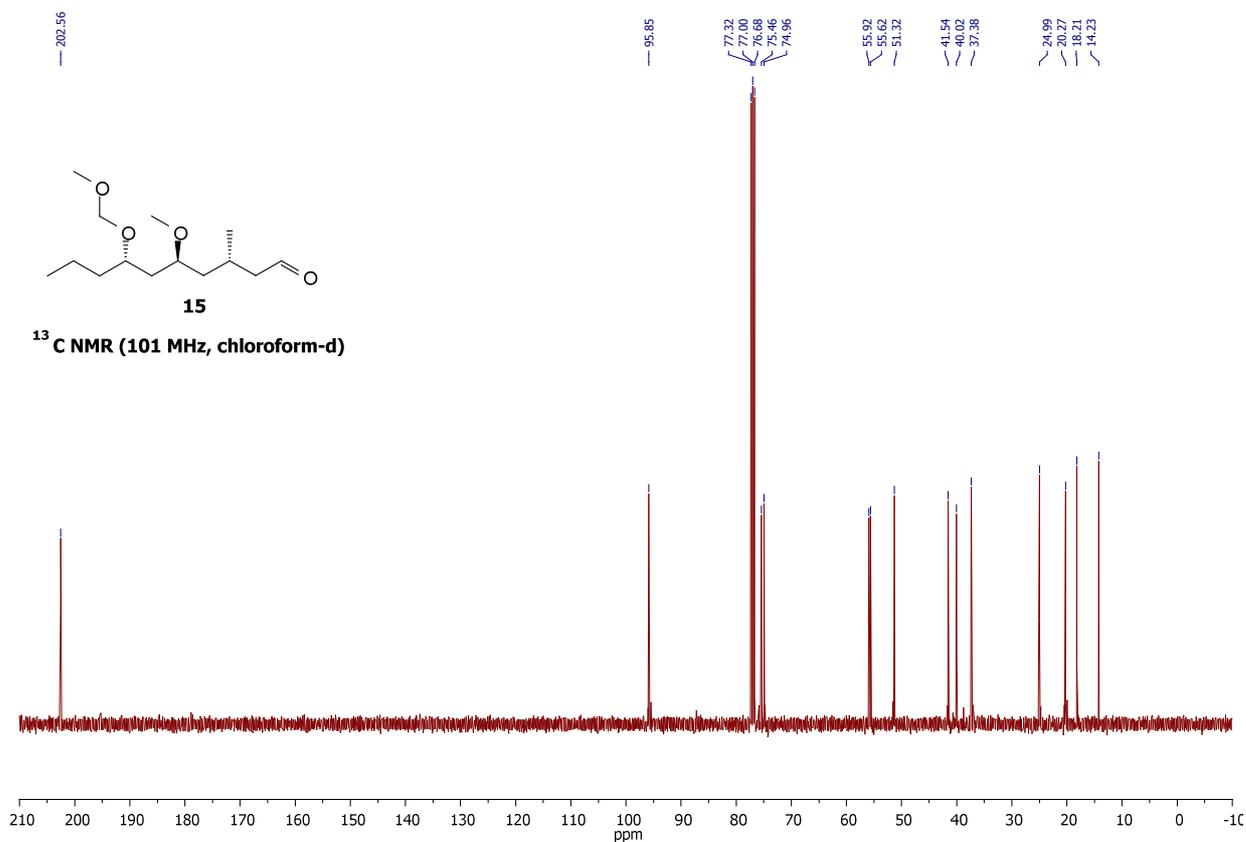
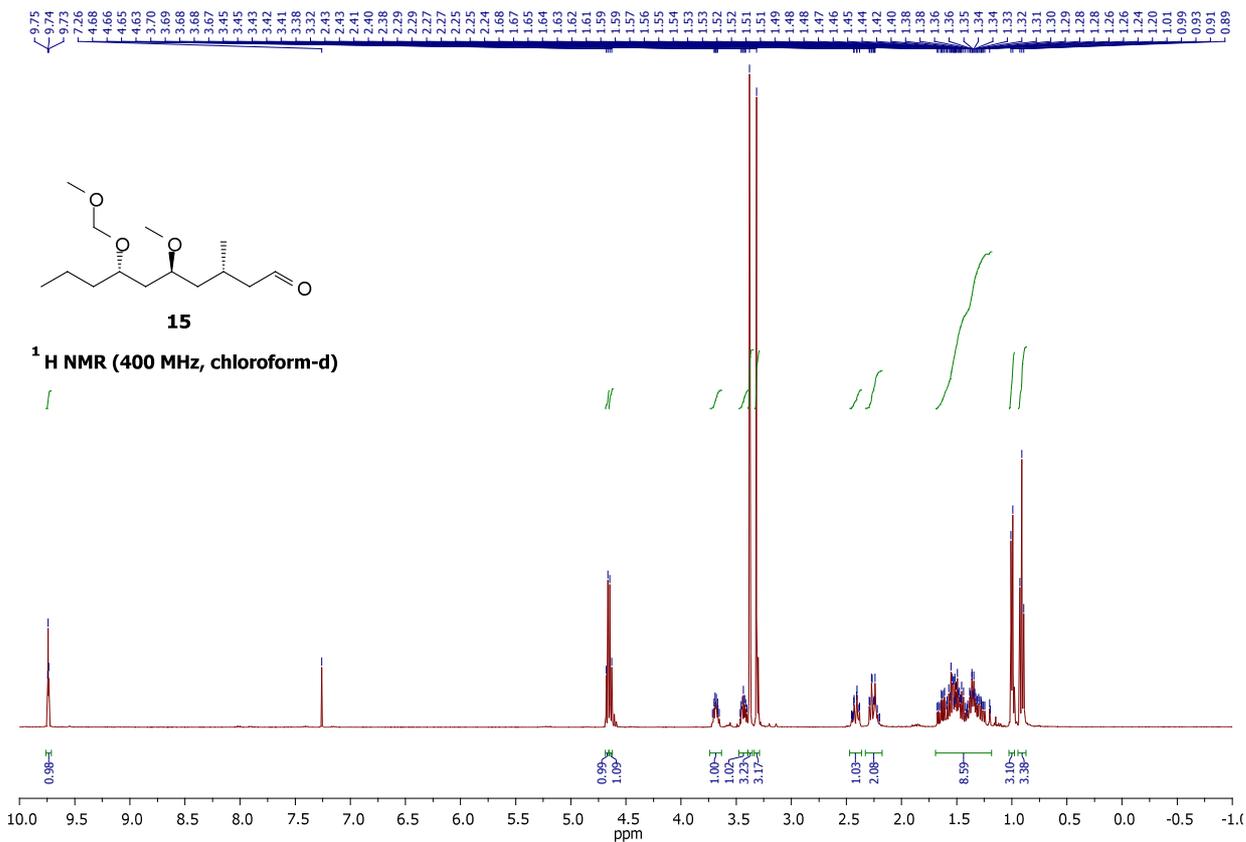
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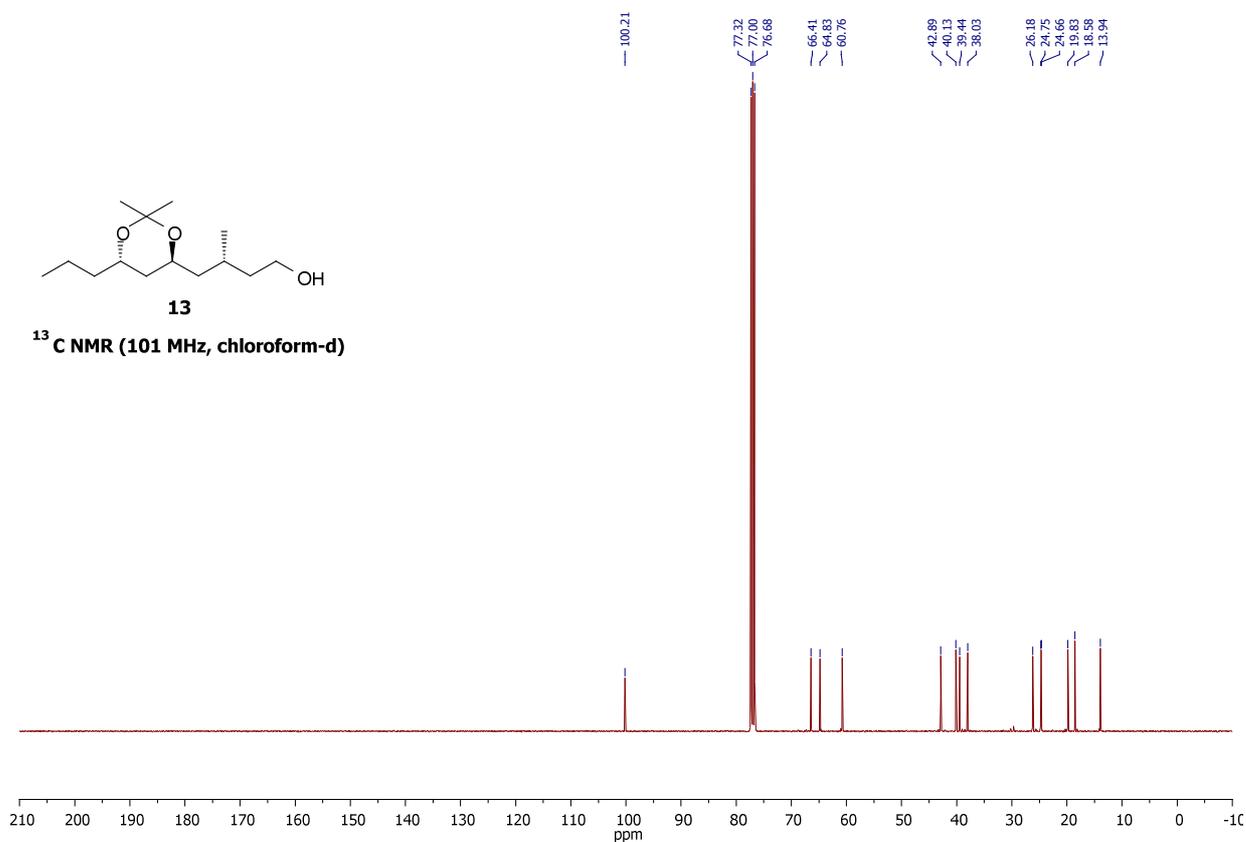
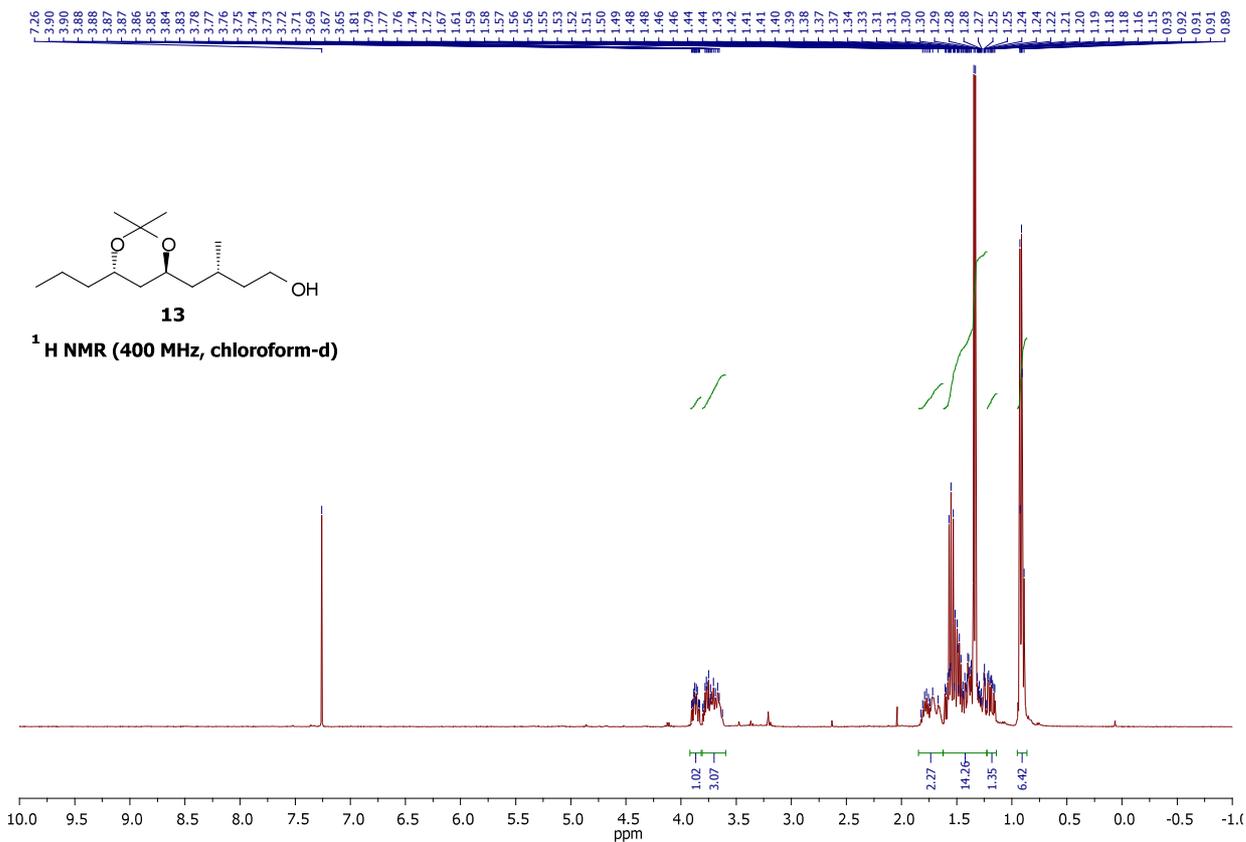
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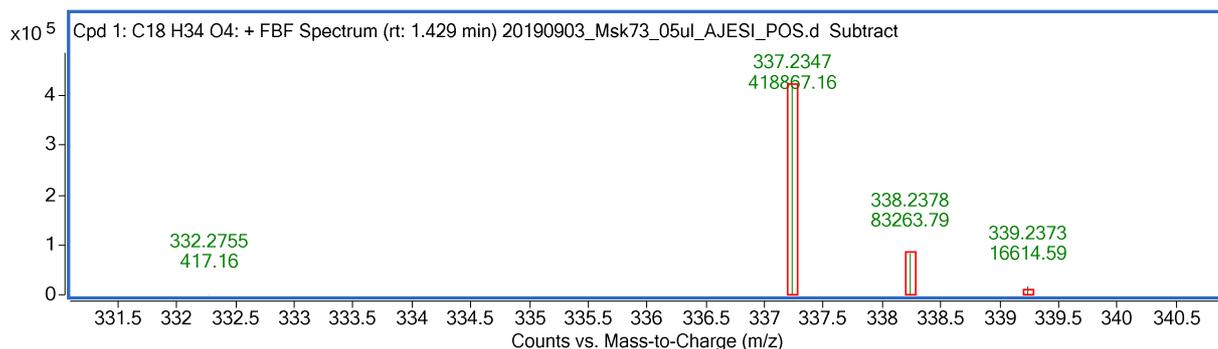
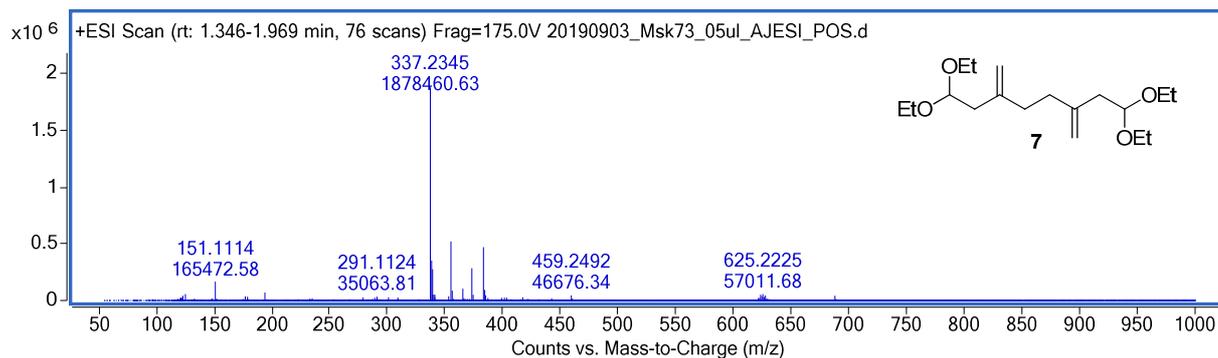


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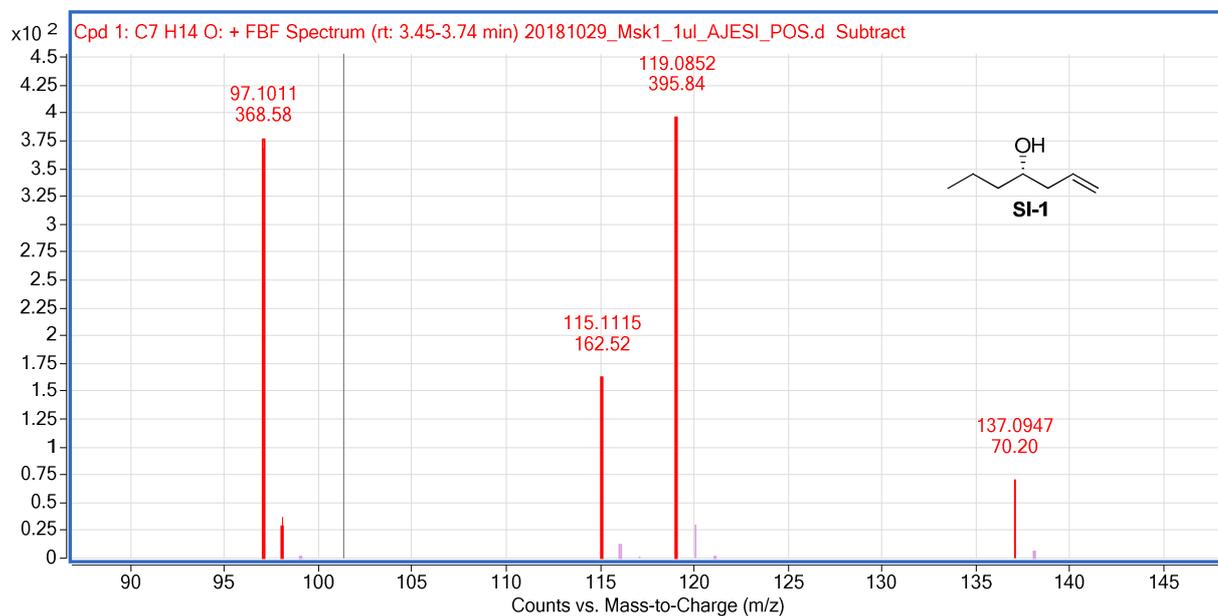


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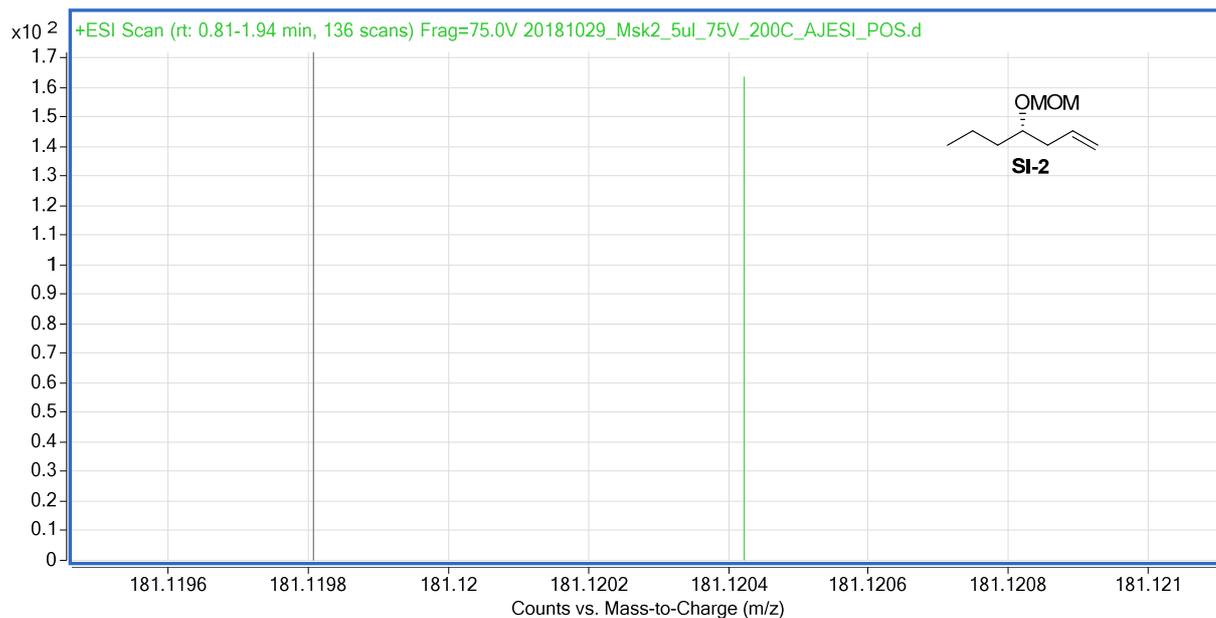
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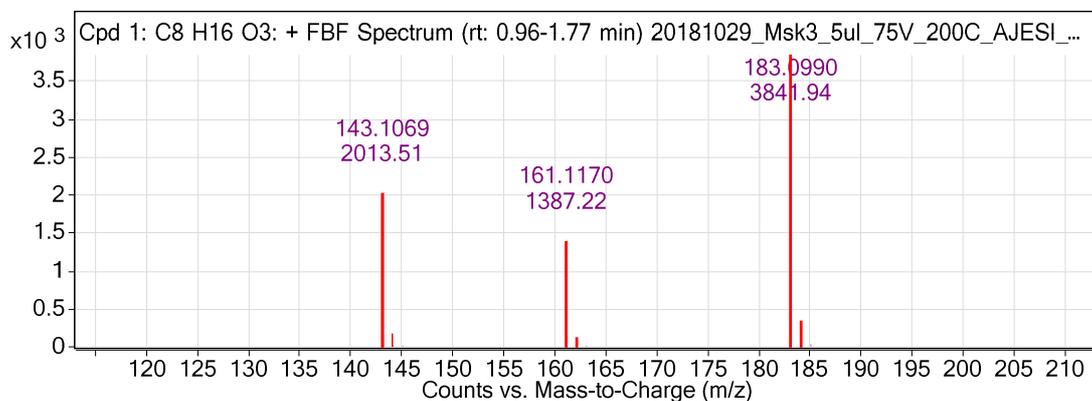
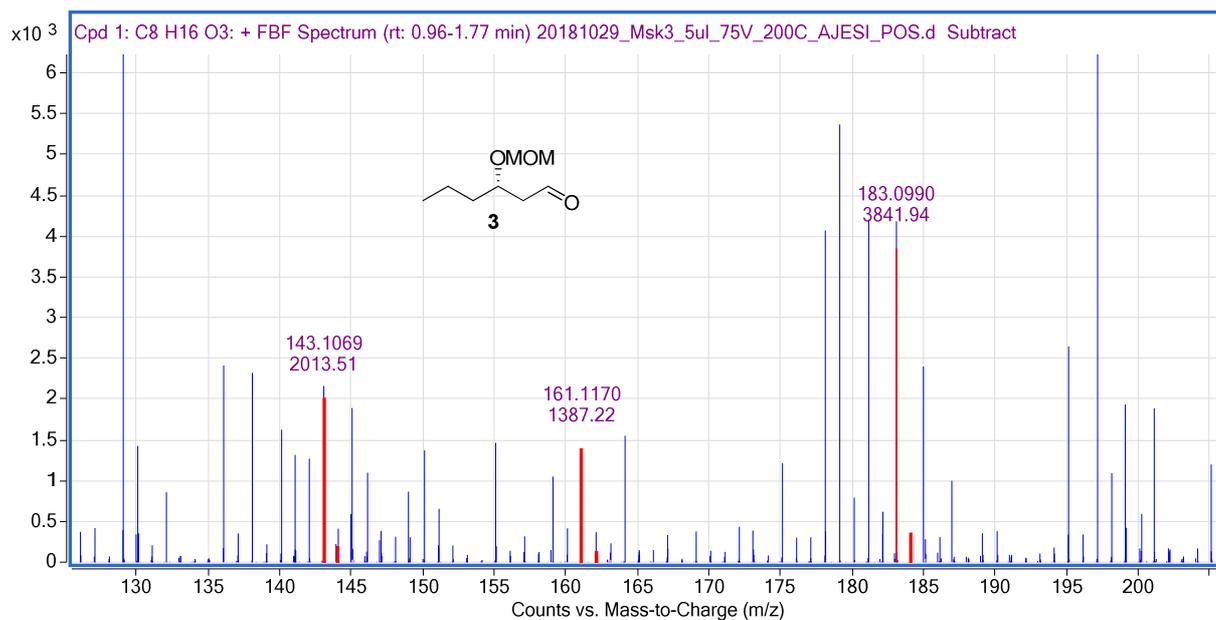
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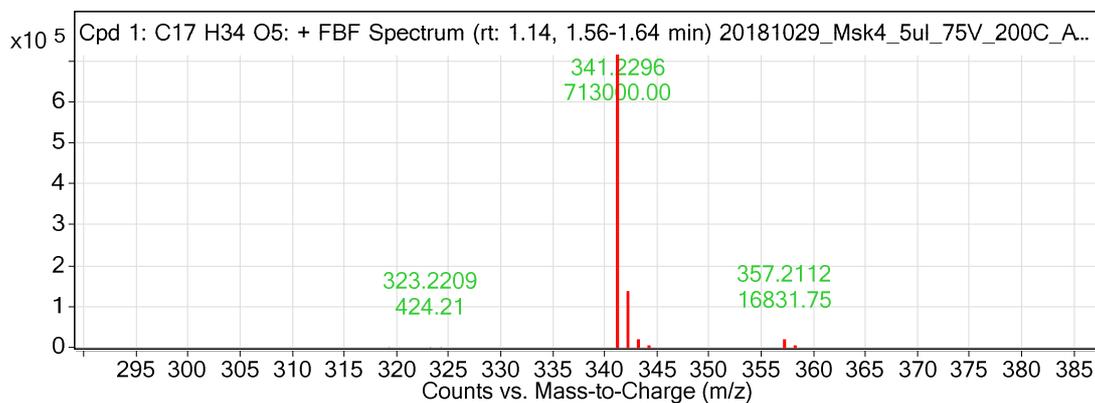
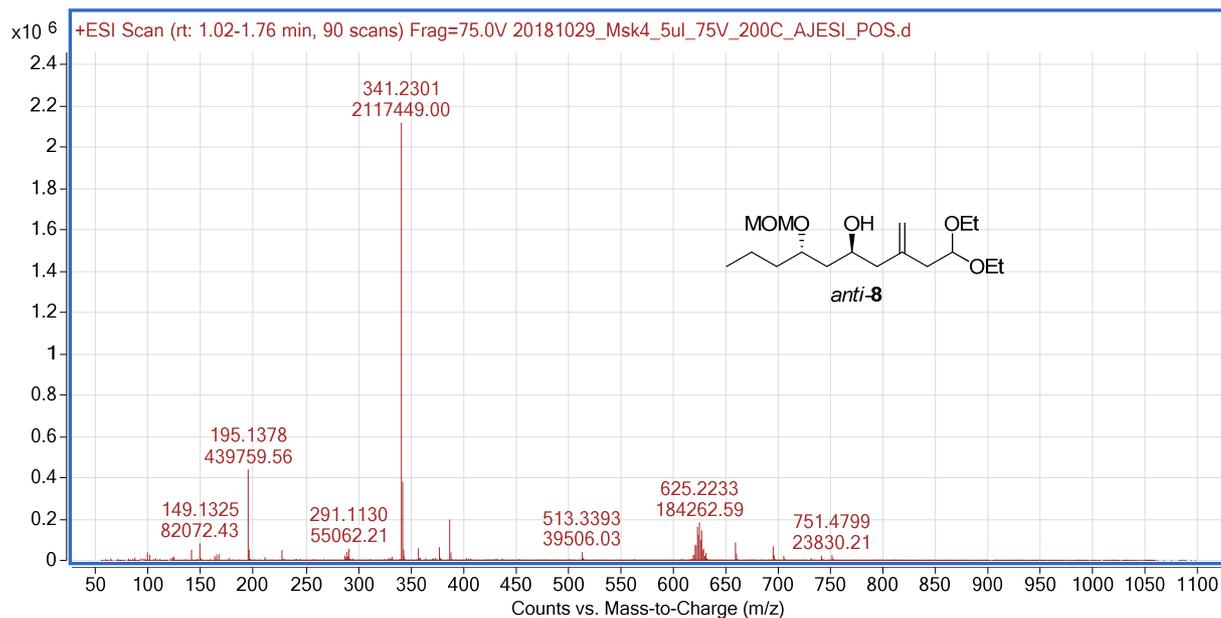
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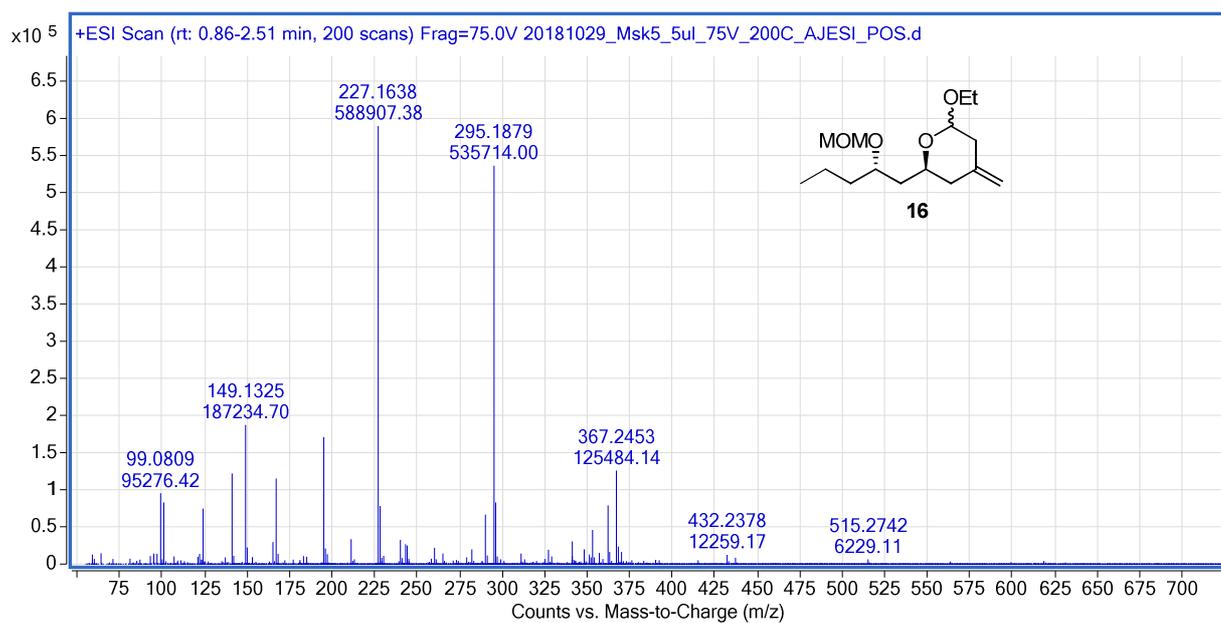
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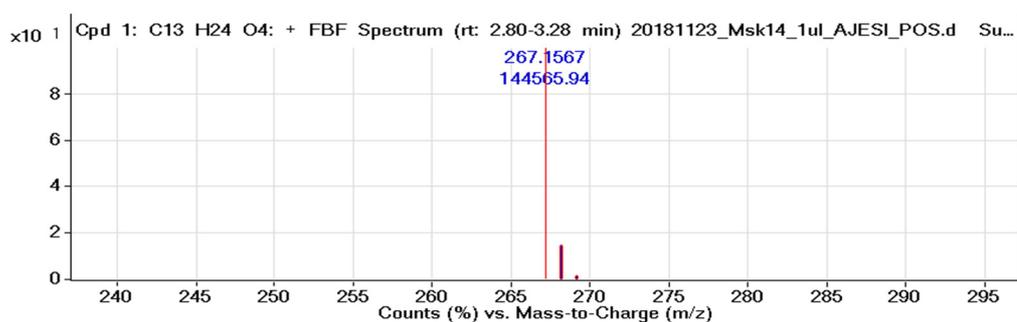
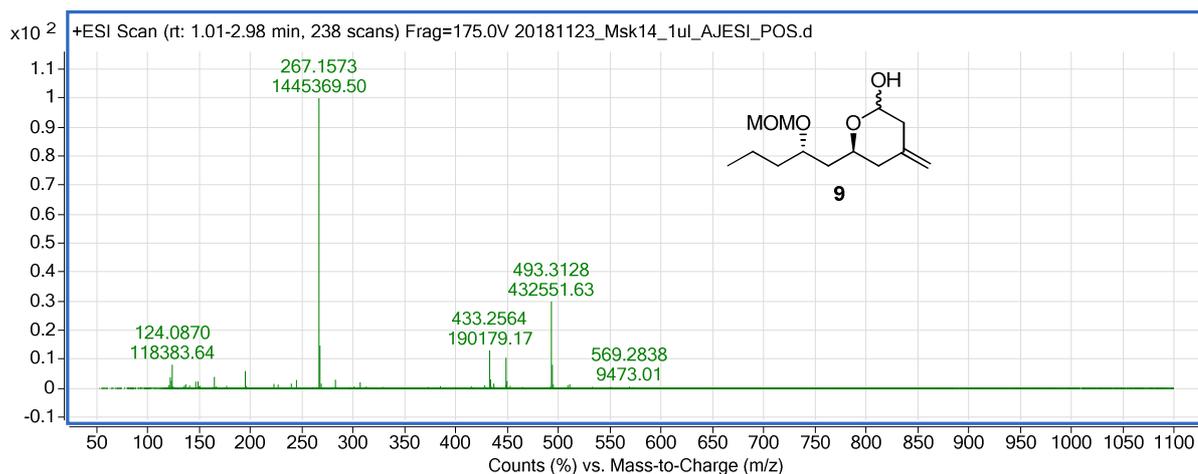
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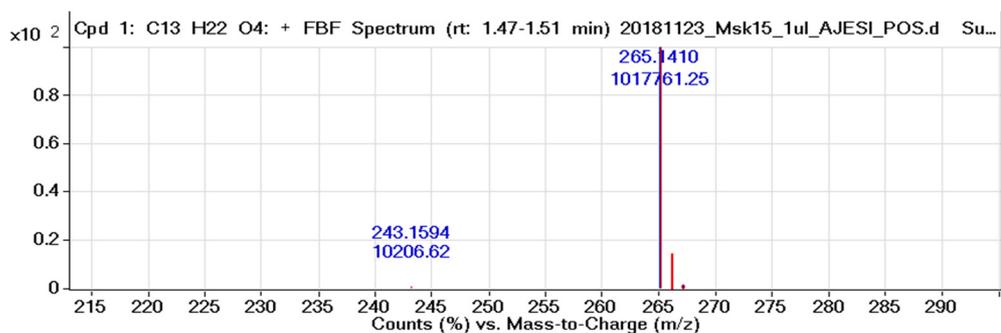
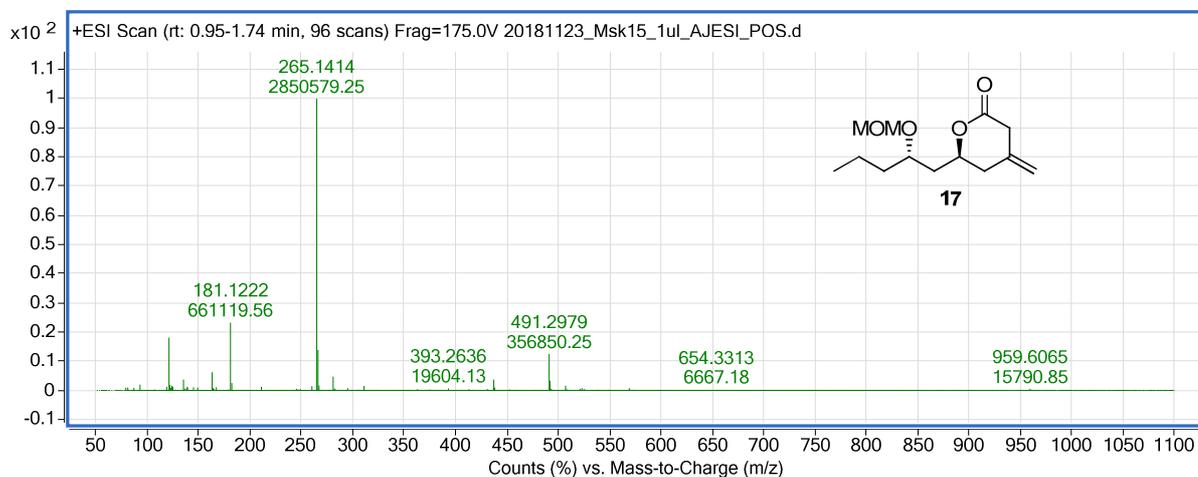
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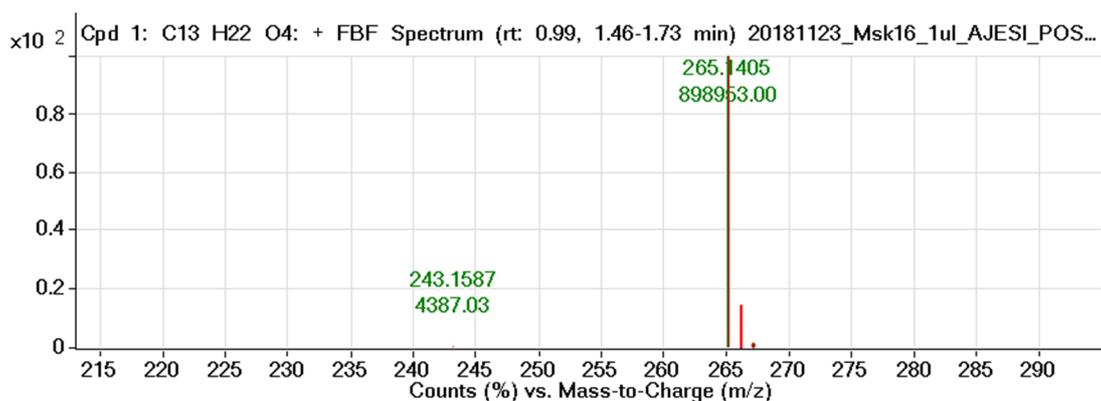
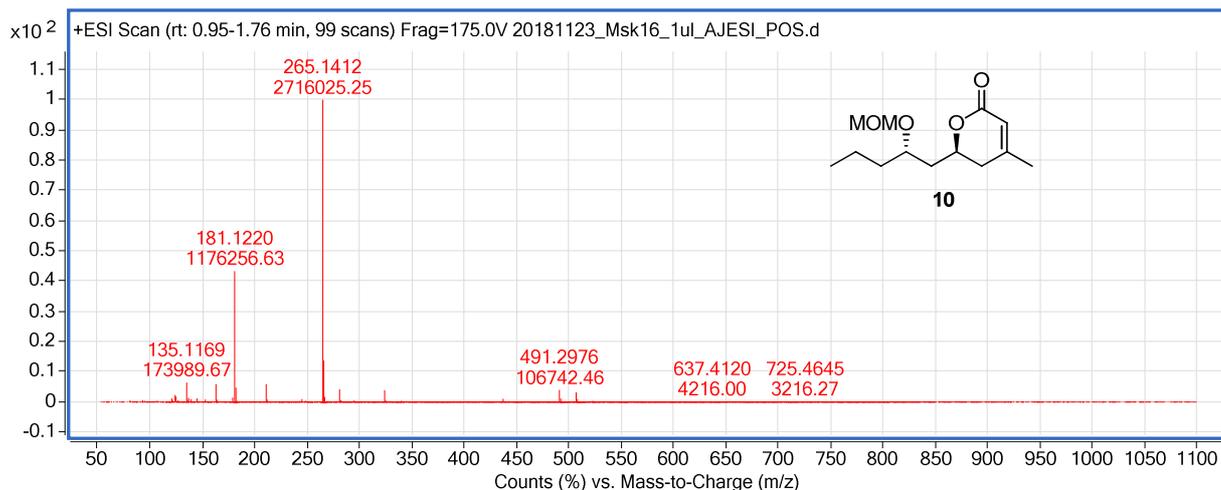
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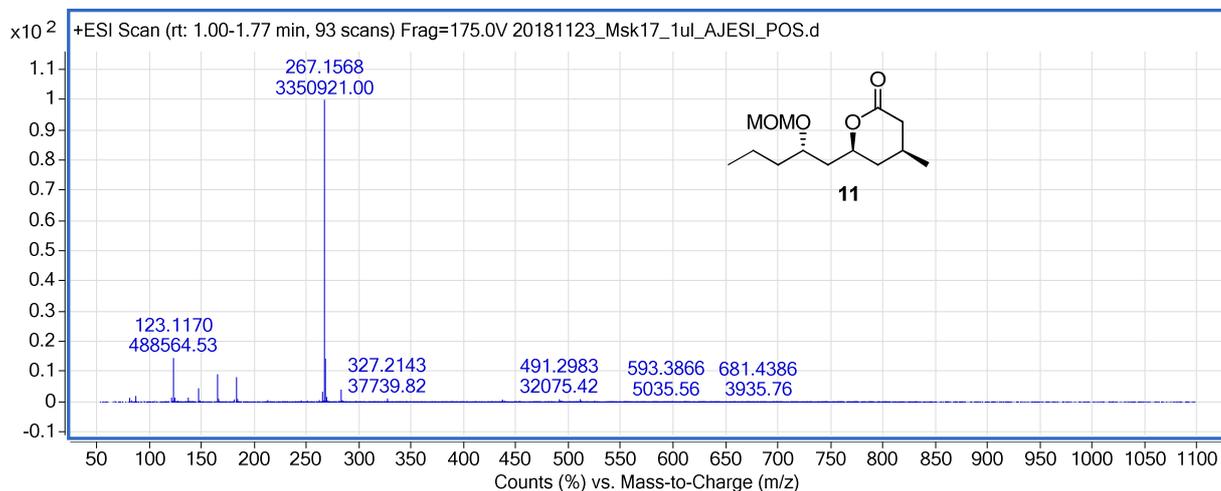
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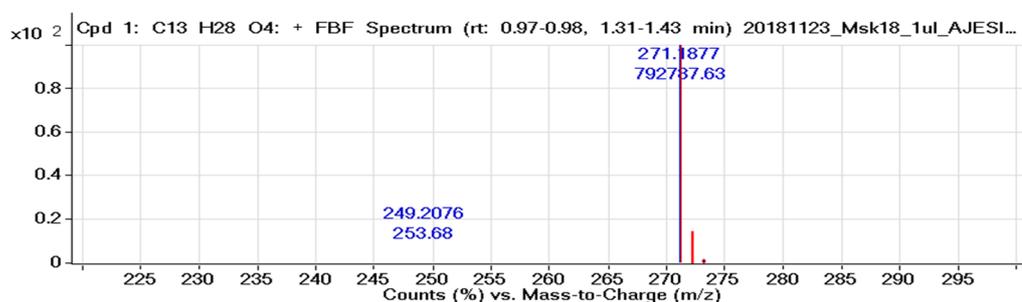
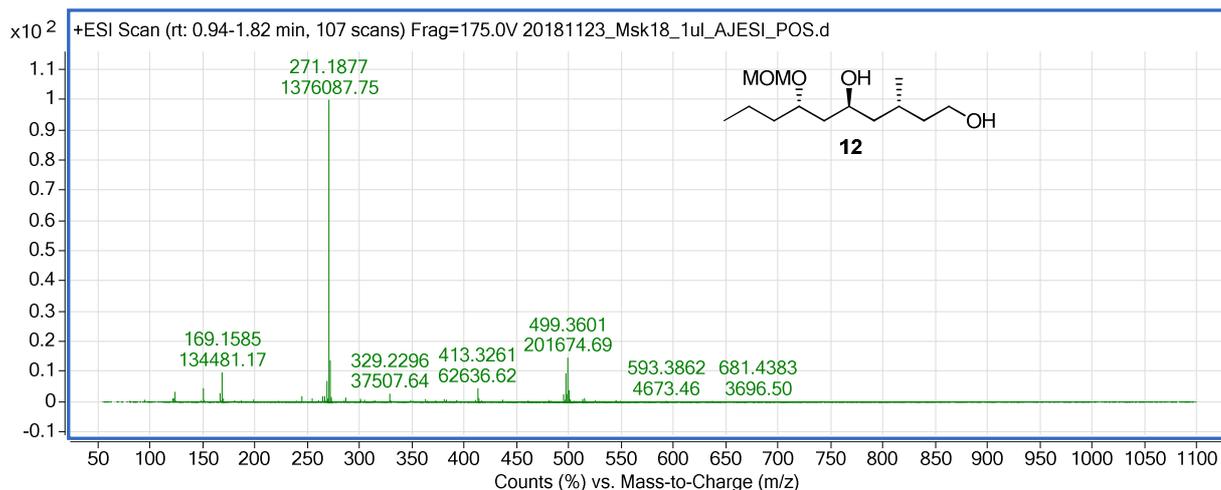
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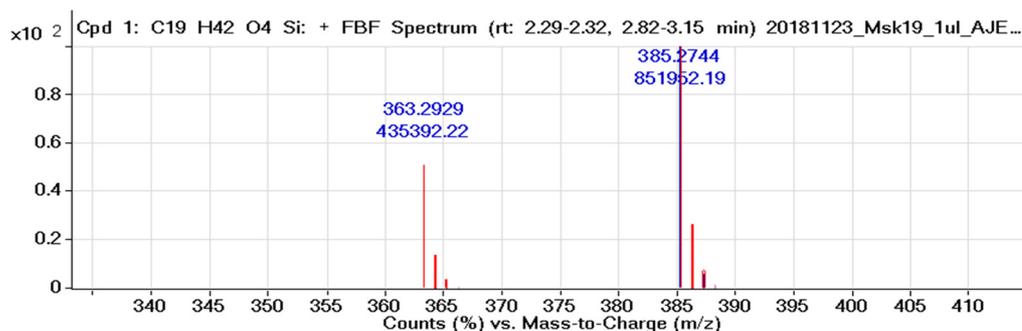
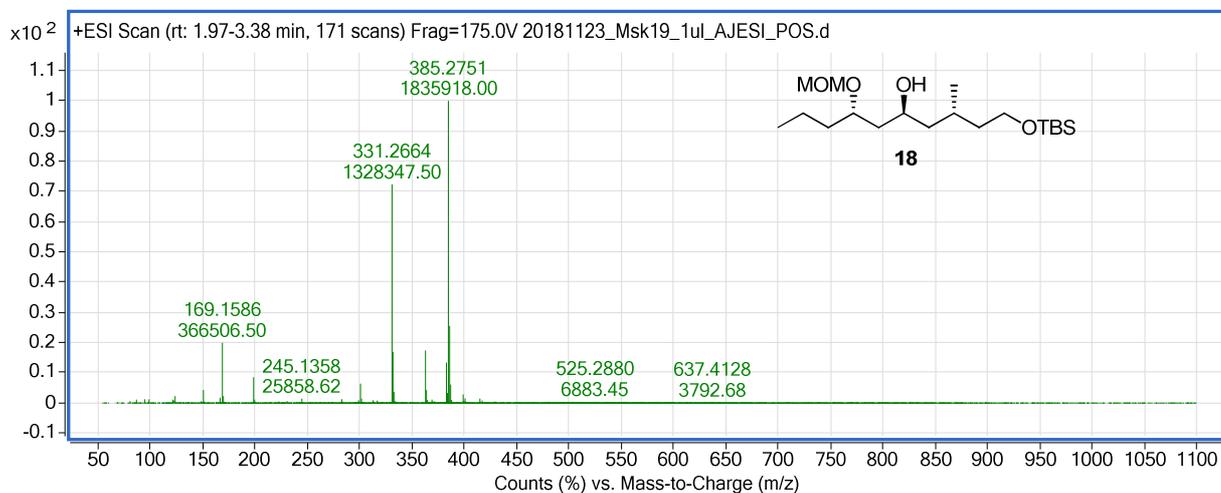
HRMS of compound 11



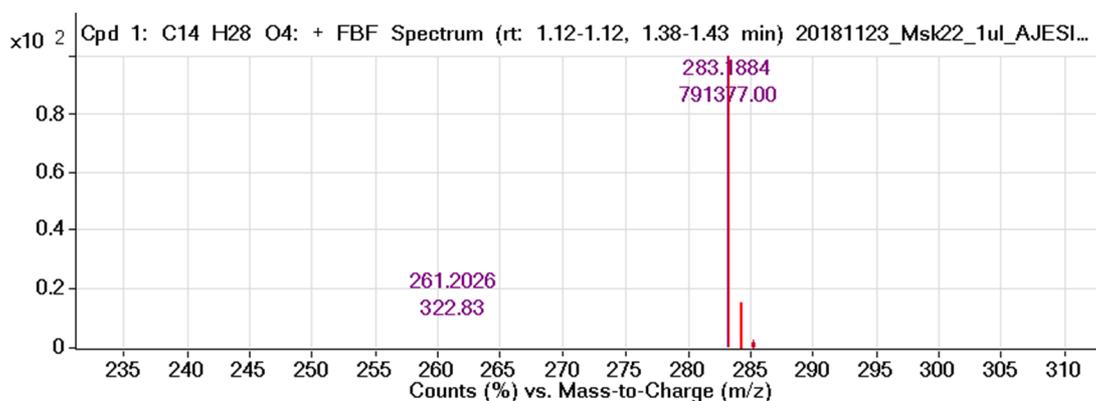
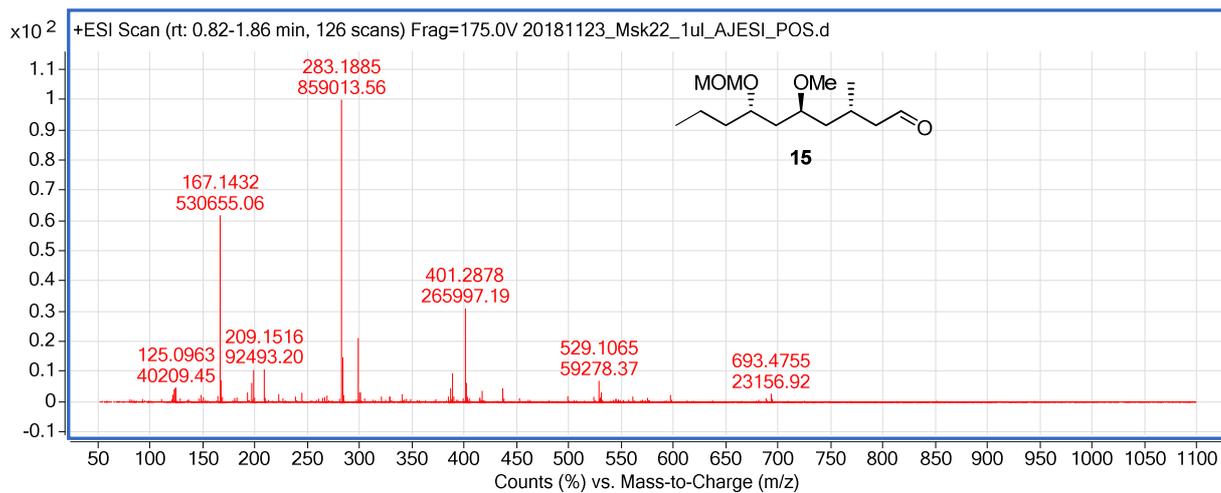
HRMS of compound 12



HRMS of compound 18



HRMS of compound 15



HRMS of compound 13

