

Supplementary Materials

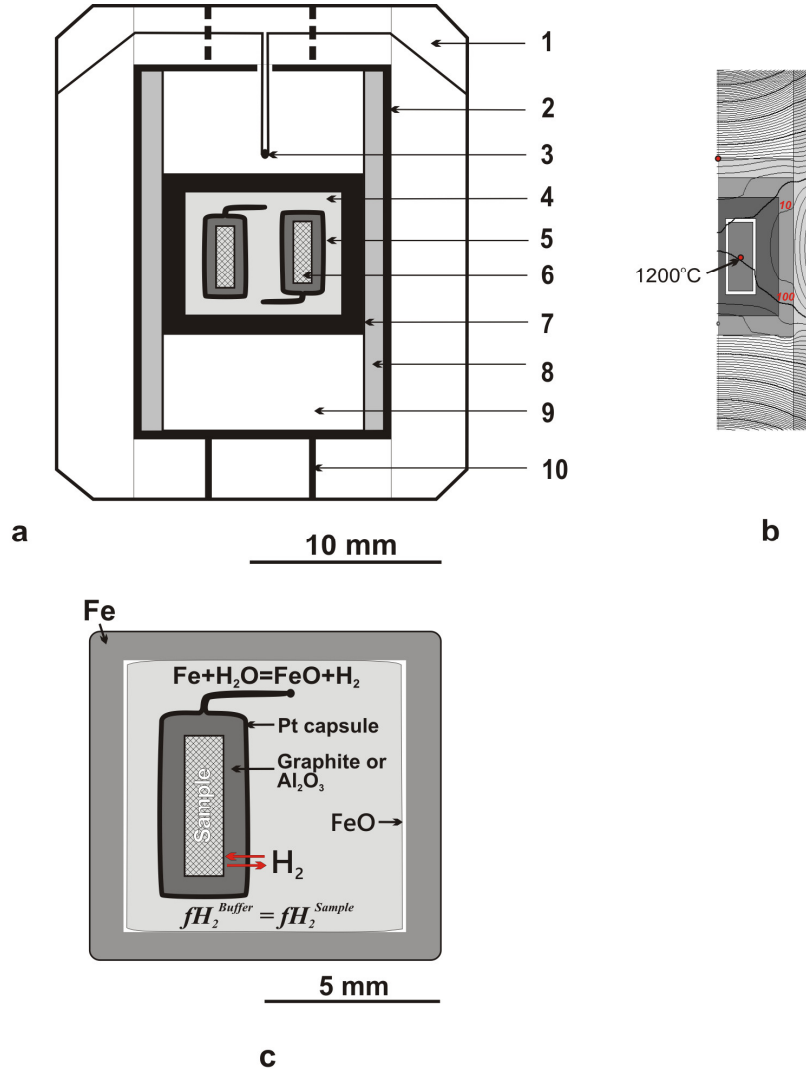


Figure S1. (a) The high-pressure cell; (b) temperature patterns in furnace assembly predicted using the software of Hernlund et al. [74] (simulations made for Pt capsule with graphite sample at 1200 °C, CsCl replaced by NaCl); (c) the double-capsule technique used to constrain the fH_2 conditions by the assemblage Fe + FeO + H₂O buffer (water in the outer capsule was released by talc decomposition into coesite, orthopyroxene under the applied P–T conditions; the main volume between outer and inner capsules was filled by high-melting CsCl). 1—ZrO₂ container; 2—cylindrical graphite heater; 3—PtRh₆/PtRh₃₀ thermocouple; 4—talc and CsCl; 5—Pt capsule; 6—sample; 7—Mo or Fe capsule; 8—MgO; 9—ZrO₂; 10—Mo leads.

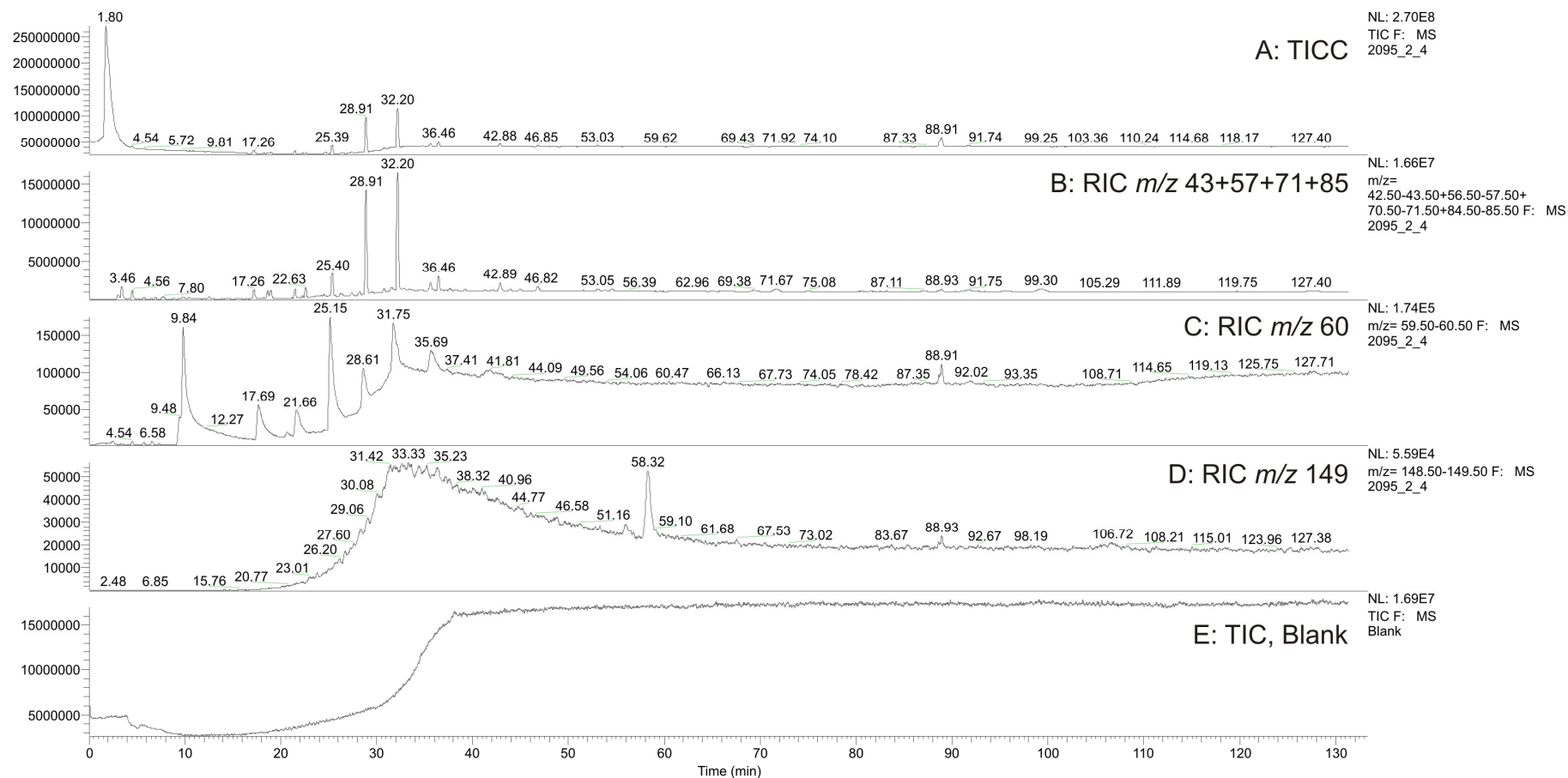


Figure S2. Results of GC-MS analysis of quenched fluid extracted by mechanical shock destruction from run 2095_2_4. (a) Total ion current chromatogram (TICC) and reconstructed ion chromatograms (RIC) for m/z 43+57+71+85 (b), m/z 60 (c), m/z 149 (d), and TIC of a pre-run system blank (e).

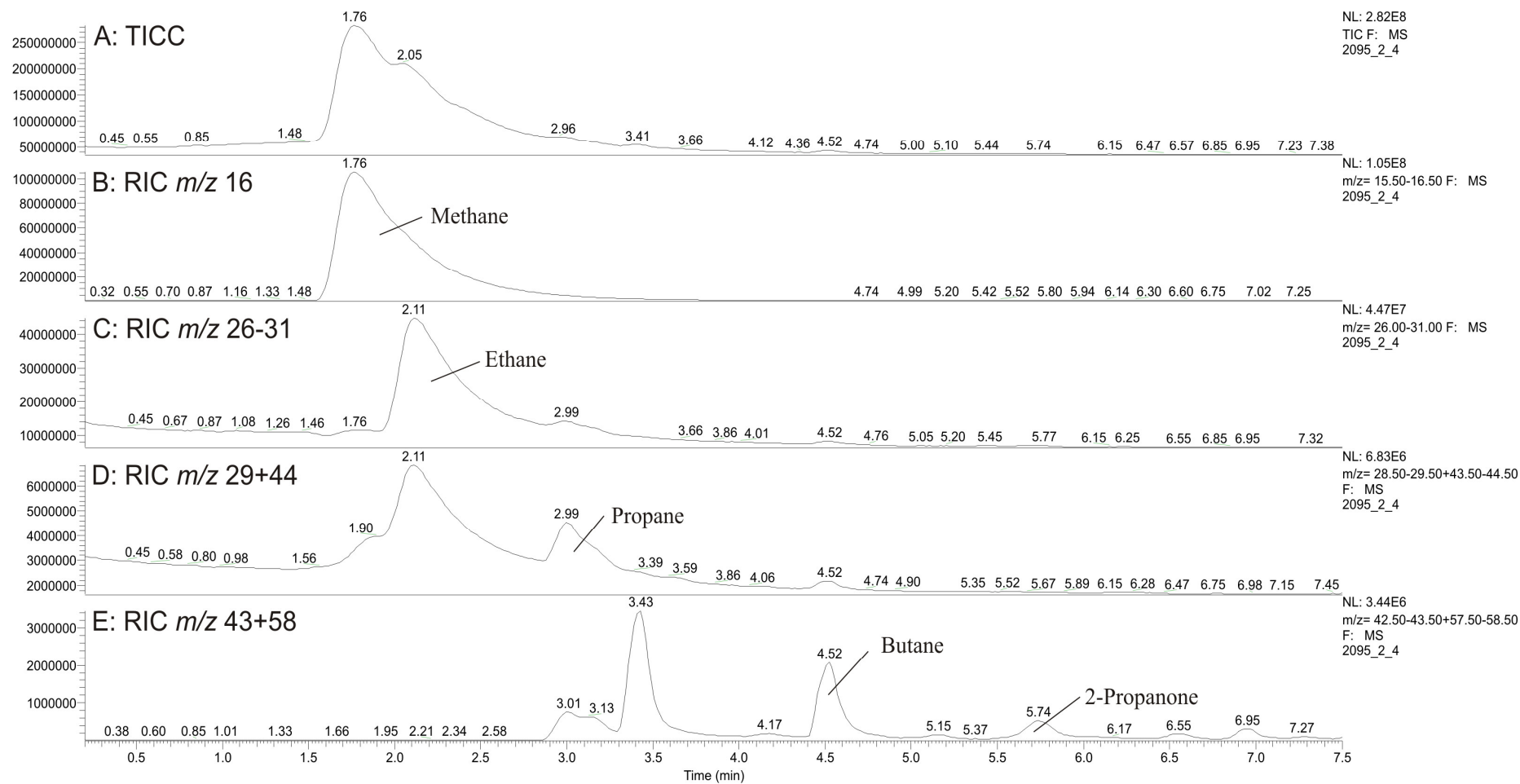


Figure S3. Results of GC-MS analysis of quenched fluid extracted by mechanical shock destruction from run 2095_2_4. (a) Fragments of a total ion current chromatogram (TICC) and reconstructed ion chromatograms (RIC) for m/z 16 (b), m/z 26-31 (c), m/z 29+44 (d), and m/z 43+58 (e).

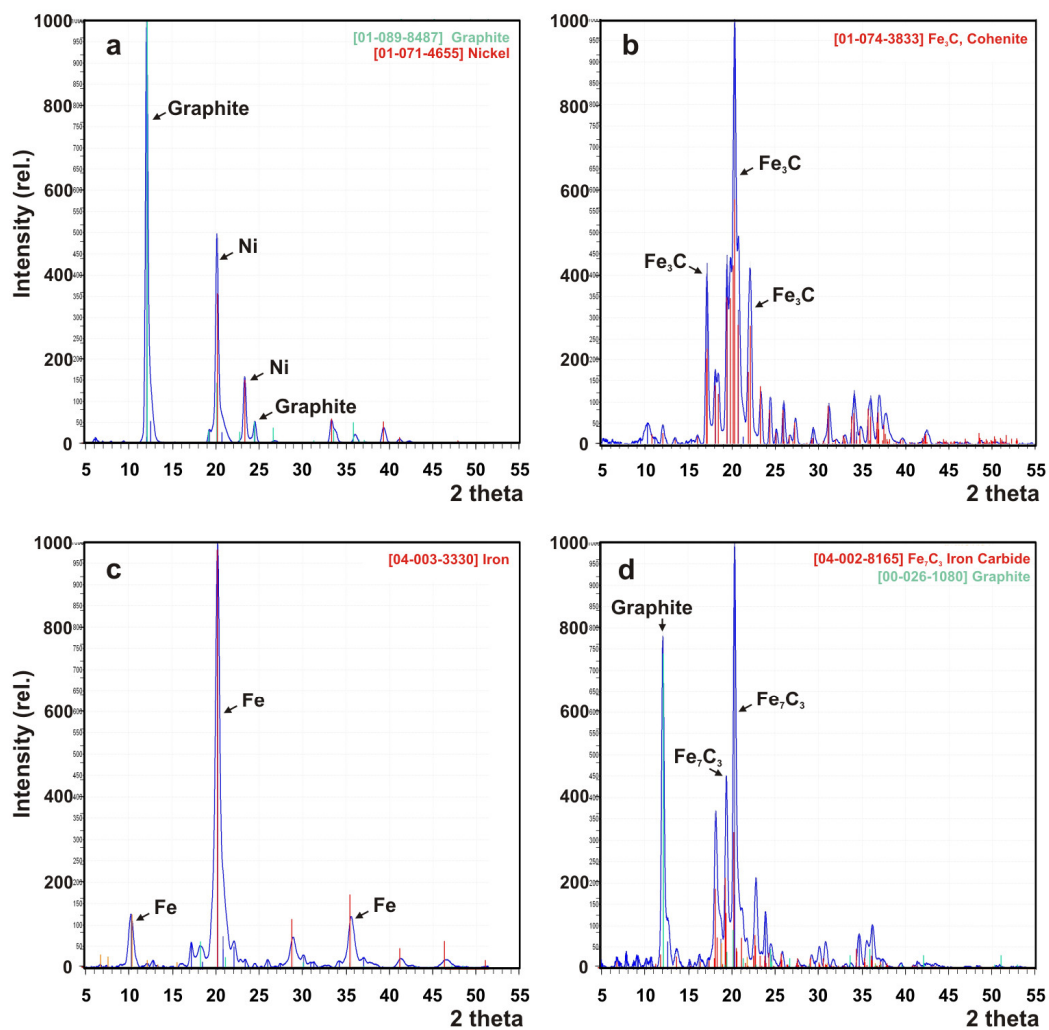


Figure S4. Phases identified by X-ray powder diffraction in the Gandolfi mode. The database of PDF-4 Minerals (The Powder Diffraction File PDF-4 +, 2006) was used for phase analysis. a: graphite and nickel, run 1001_4_3; b: iron carbide (Fe₃C), run 1001_4_4; c: iron, run 2095_2_2; d: iron carbide (Fe₇C₃) and carbon, run 2100_2_4.