

Supporting Information

System Theoretical Study on the Effect of Variable Nonmetallic Doping on Improving Catalytic Activity of 2D-Ti₃C₂O₂ for Hydrogen Evolution Reaction

Ye Su [†], Minhui Song [†], Xiaoxu Wang ^{*}, Jihang Jiang, Xiaolong Si, Tianhang Zhao and Ping Qian ^{*}

Beijing Advanced Innovation Center for Materials Genome Engineering, School of Mathematics and Physics, ,
University of Science and Technology Beijing, Beijing 100083, China; suyechina@163.com (Y.S.);
mhuisong@163.com (M.S.); xuying13520506589@163.com (J.J.); sixiaolong2018@outlook.com (X.S.);
ztx13520805687@163.com (T.Z.)

^{*} Correspondence: wxx18811349251@163.com (X.W.); pingqian@ustb.edu.cn (P.Q.)

[†] Ye Su and Minhui Song are co-first authors.

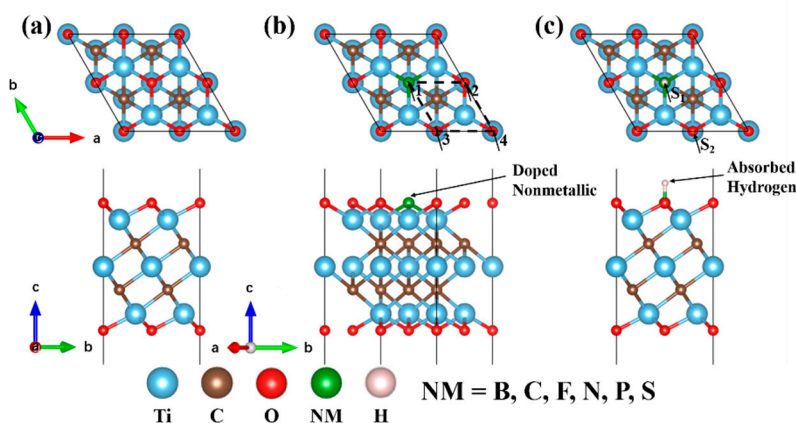
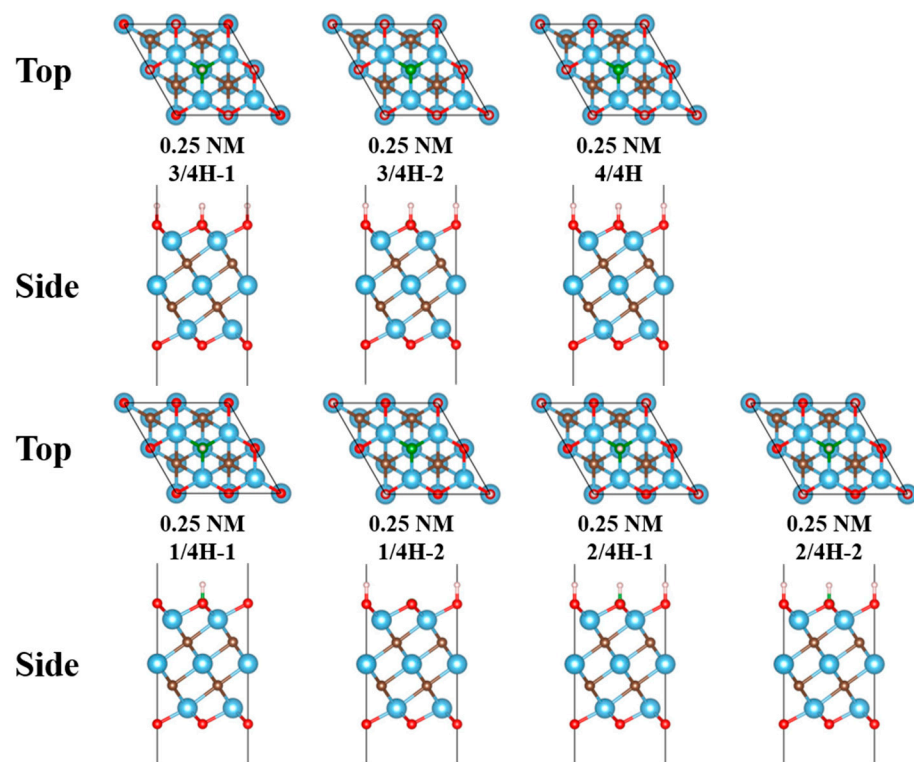
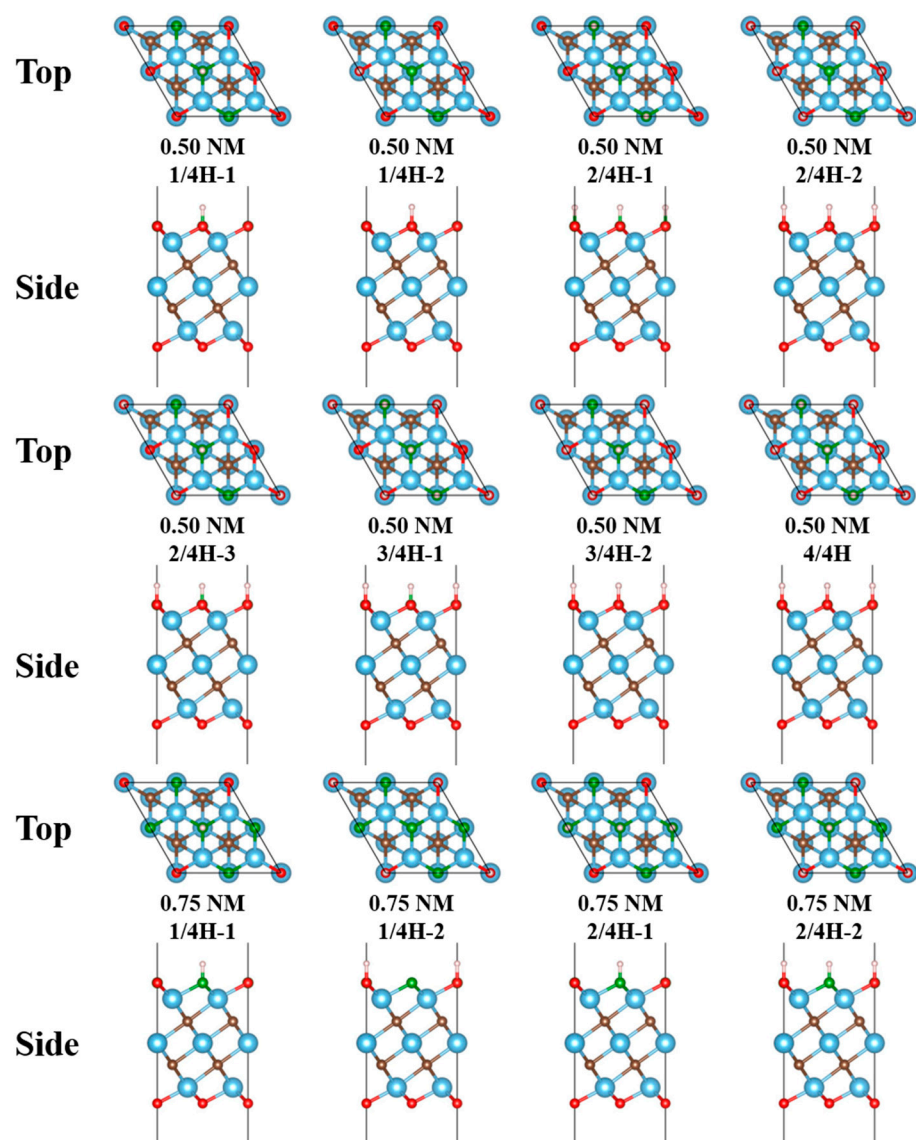


Figure S1. 0.25-1.00 concentrations nonmetallic element doping configurations.





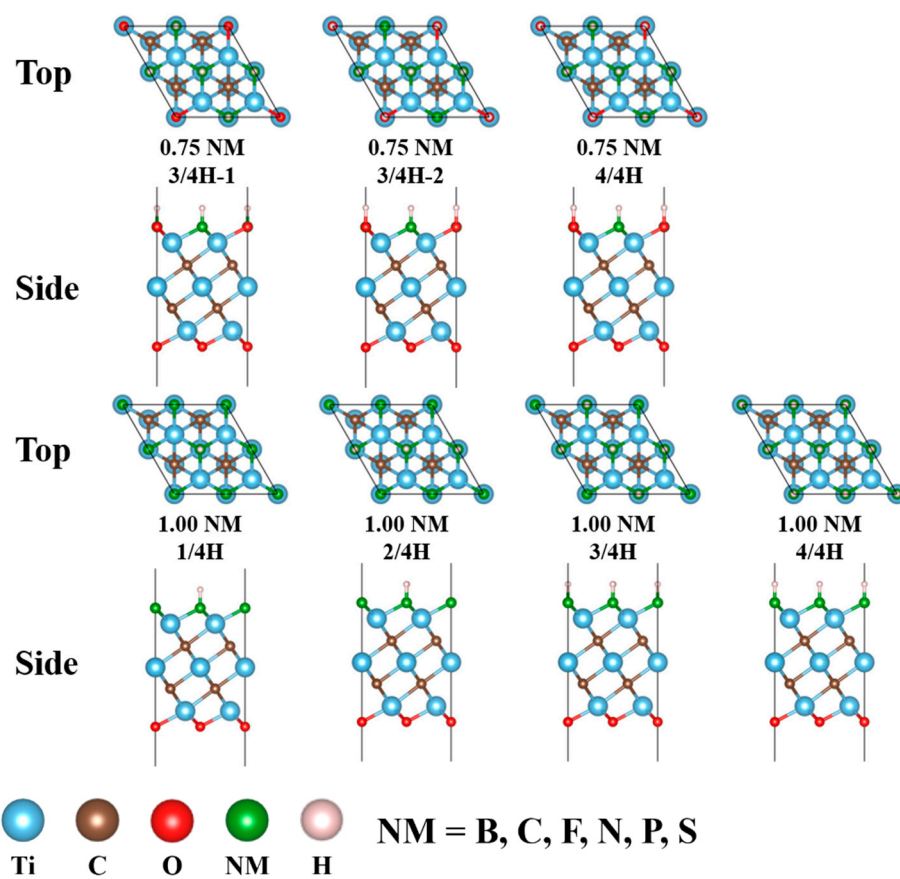
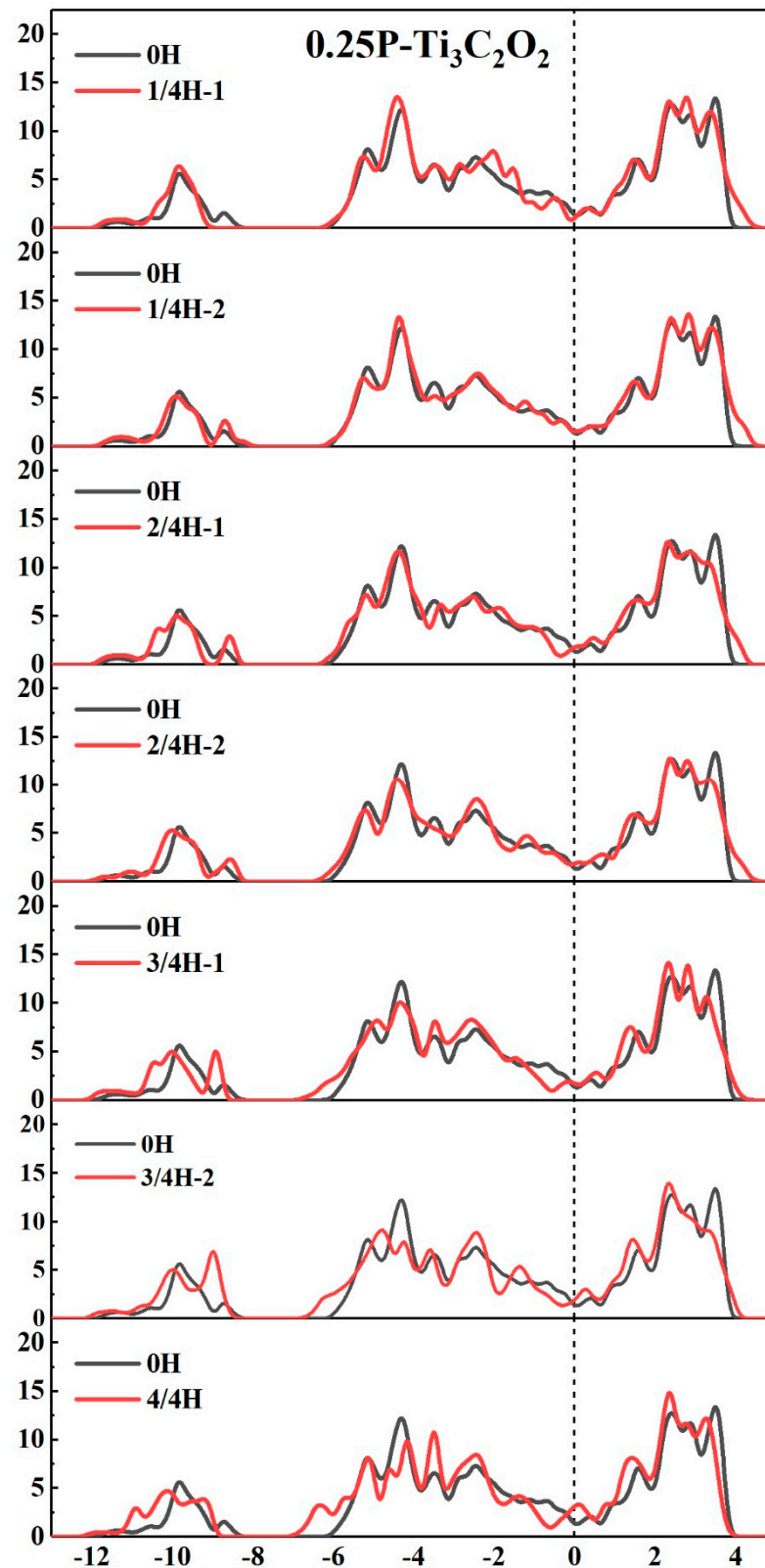
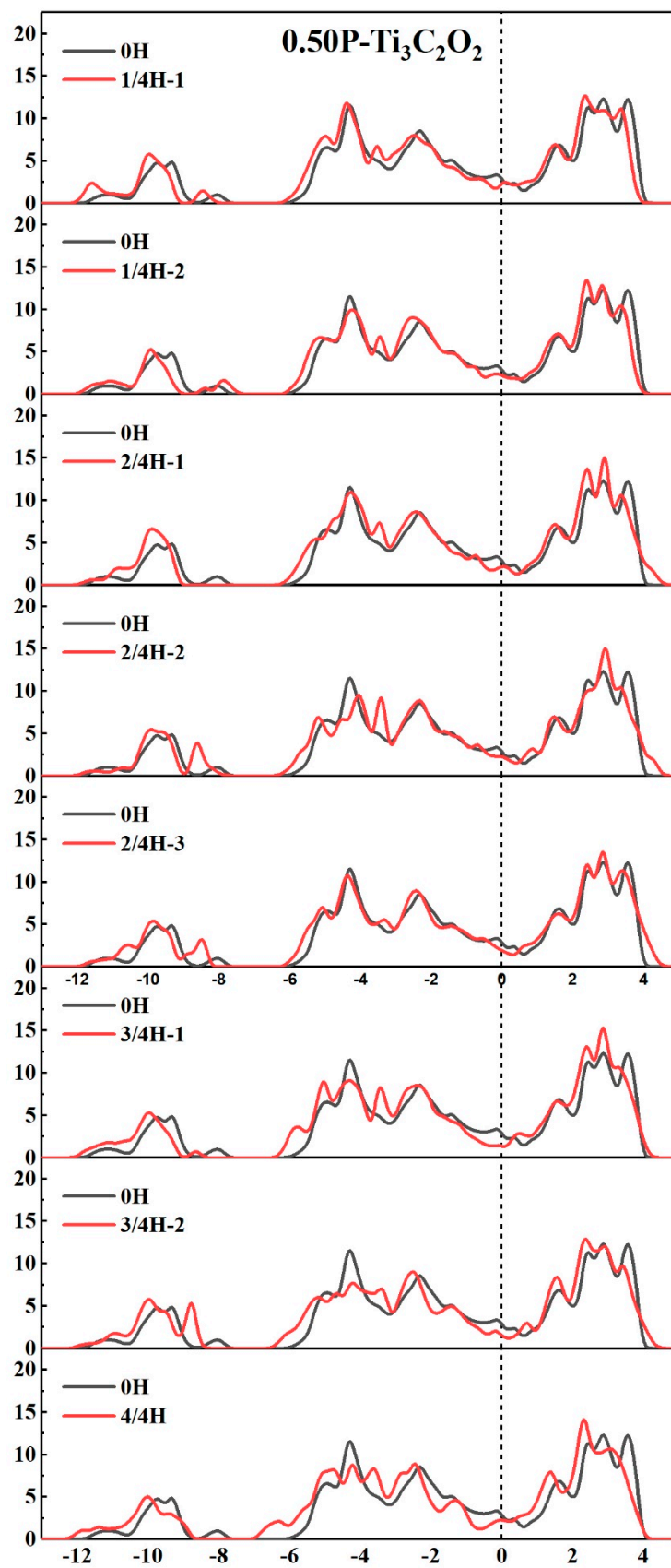
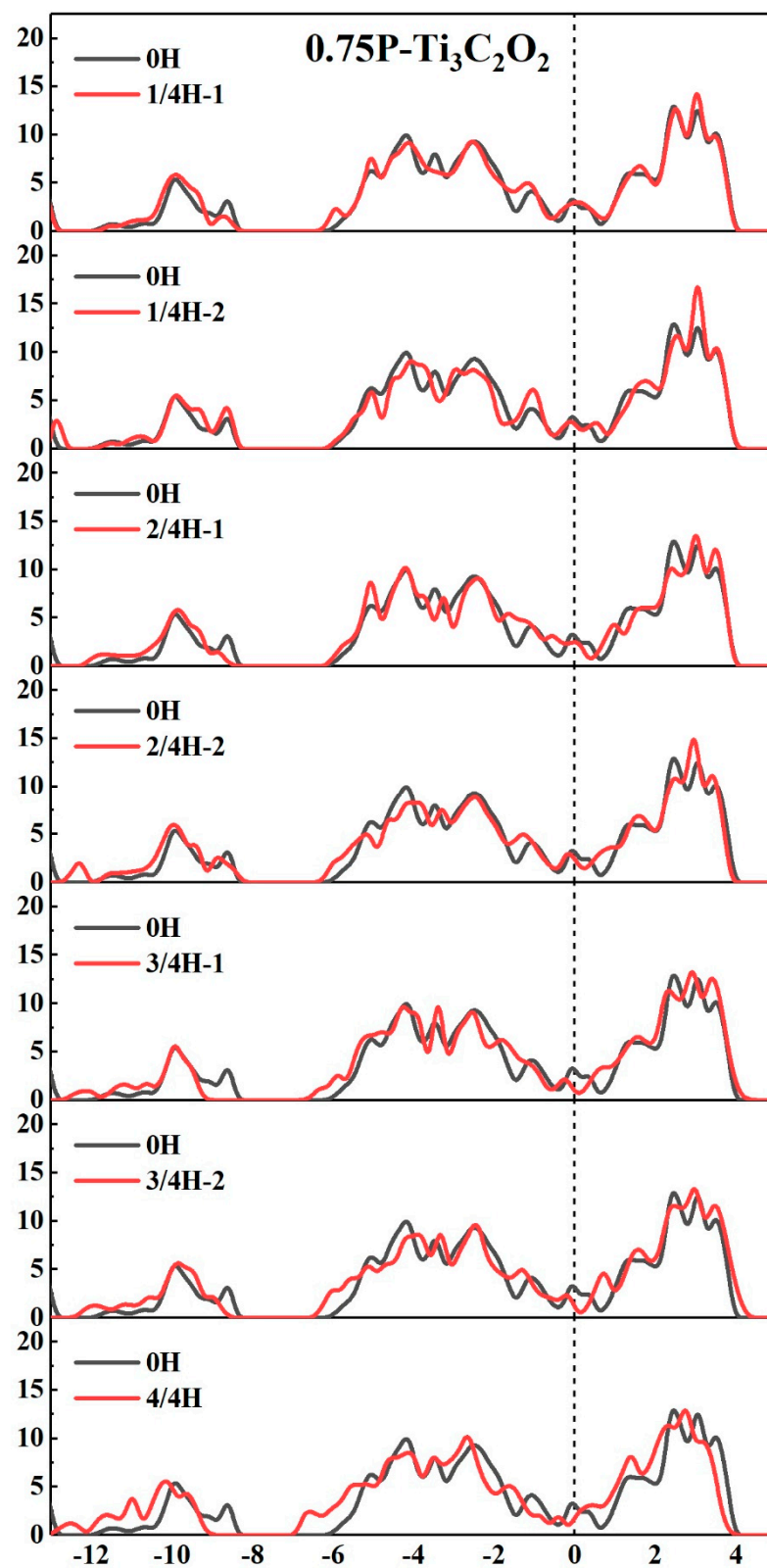


Figure S2. Different concentrations nonmetallic element doping at different H absorption concentrations configurations.







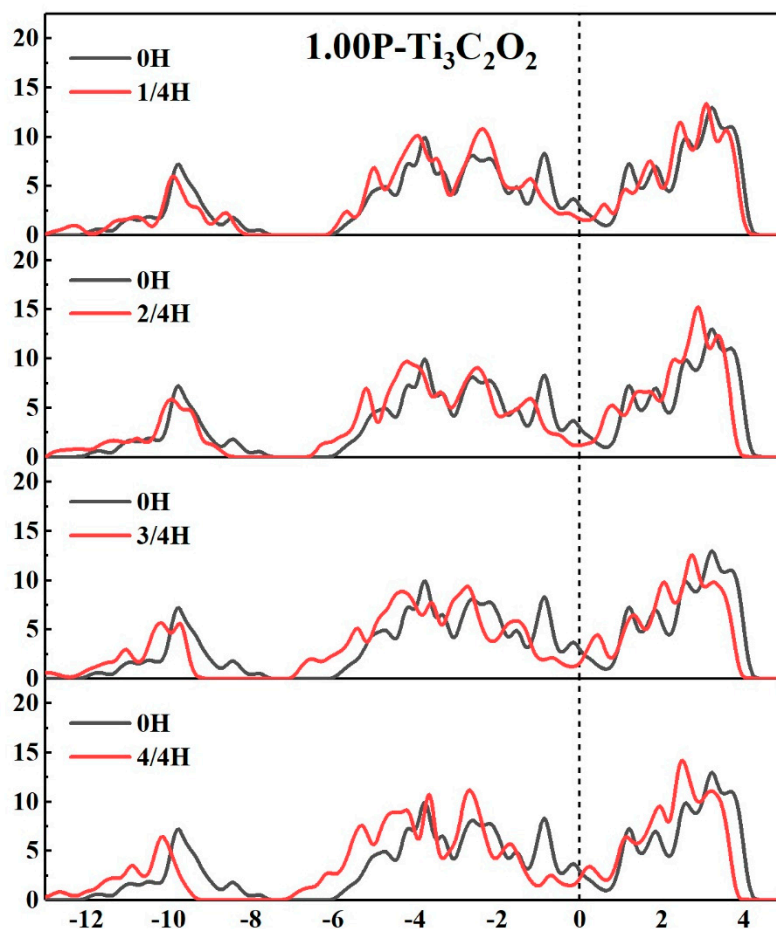


Figure S3. TDOS of different concentrations P element at different H element absorption concentrations. The black lines represent H atoms does not absorbed on the surface, the red lines represent H atoms absorbed on the surface.

Table S1. The hydrogen adsorption free energy(ΔG_H) of nonmetal doping $Ti_3C_2O_2$. 1-NM, 2-NM, 3-NM and 4-NM represent 1, 2, 3 and 4 H atoms absorbed on the doping nonmetallic atoms; 1-O, 2-O, 3-O represent 1, 2 and 3 H atoms absorbed on the surface O atoms. Bold values selected in the paper.

Doping Concentration	H coverage	configuration	$\Delta G(H)/eV$				
			C	F	Doping Element		
					N	P	S
25%	1/4	1-NM	-1.387	-1.532	-1.035	-0.377	0.552
		1-O	-4.230	-3.705	-0.269	-0.308	-0.237
	1/2	1-NM,1-O	-0.792	-3.324	-0.530	-0.256	0.173
		2-O	-0.271	-3.324	-0.089	-0.205	-0.112
	3/4	1-NM,2-O	-0.512	-1.233	-0.270	-0.142	0.298
		3-O	-0.081	-3.054	0.077	-0.084	0.060
	1	1-NM,3-O	-0.284	-2.509	-0.028	0.034	0.384
50%	1/4	1-NM	-1.635	-3.448	-1.346	-0.908	0.520
		1-O	-0.571	-3.105	-0.526	-0.959	-0.304
	1/2	2-NM	-1.468	0.467	-1.170	-0.524	0.567
		2-O	-0.441	0.466	-0.314	-0.349	-0.635
	3/4	1-NM,1-O	-1.018	1.547	-1.327	-0.406	0.284
		2-NM,1-O	-1.065	-	-0.768	-0.337	0.530
	1	1-NM,2-O	-0.724	1.151	-0.458	-0.303	0.314
75%	1/4	2-NM,2-O	-0.821	1.539	-0.495	-0.190	0.586
		1-NM	-5.521	0.488	-1.854	0.149	0.556
	1/2	1-O	-4.652	0.488	-0.941	-0.086	-0.284
	1/2	2-NM	-5.193	2.878	-1.540	0.092	0.719

100%		1-NM,1-O	-4.719	1.507	-1.124	0.076	0.278
	3/4	3-NM	-5.934	1.000	-	0.012	0.904
		2-NM,1-O	-5.777	2.233	-1.072	0.034	0.612
	1	3-NM,1-O	-1.233	2.011	-	0.117	-
	1/4	1-NM	-6.704	-0.879	-2.132	-0.303	3.640
	1/2	2-NM	-4.637	-3.270	-1.924	-0.132	2.247
	3/4	3-NM	-3.363	-2.432	-3.222	0.008	1.927
	1	4-NM	-2.665	-0.894	-0.329	0.119	1.833

Table S2. Doped structure lattice constant, doping atomic radius and O-H bond length of 2D Ti₃C₂O₂ and nonmetal doping Ti₃C₂O₂.

NM	concentration	a	r	b(NM-Ti)
O	-	6.028	0.740	1.968
	0.25	6.055	0.810	2.204
	0.5	6.060	0.810	2.209
B	0.75	6.025	0.810	2.129
	1	6.000	0.810	2.285
	0.25	6.063	0.770	1.982
C	0.5	6.080	0.770	2.016
	0.75	6.072	0.770	2.013
	1	6.028	0.770	2.092
F	0.25	6.032	0.720	2.140
	0.5	6.036	0.720	2.125
	0.75	6.038	0.720	2.117
N	1	6.040	0.720	2.149
	0.25	0.740	6.048	1.921
	0.5	0.740	6.068	1.901
P	0.75	0.740	6.076	1.923
	1	0.740	6.093	1.937
	0.25	1.100	6.071	2.393
S	0.5	1.100	6.083	2.482
	0.75	1.100	6.038	2.581
	1	1.100	6.005	2.608
	0.25	1.040	6.058	2.391
	0.5	1.040	6.079	2.393
	0.75	1.040	6.093	2.399
	1	1.040	6.114	2.392