

Ce-MOF with intrinsic haloperoxidase-like activity for ratiometric colorimetric detection of hydrogen peroxide

Supplementary Material

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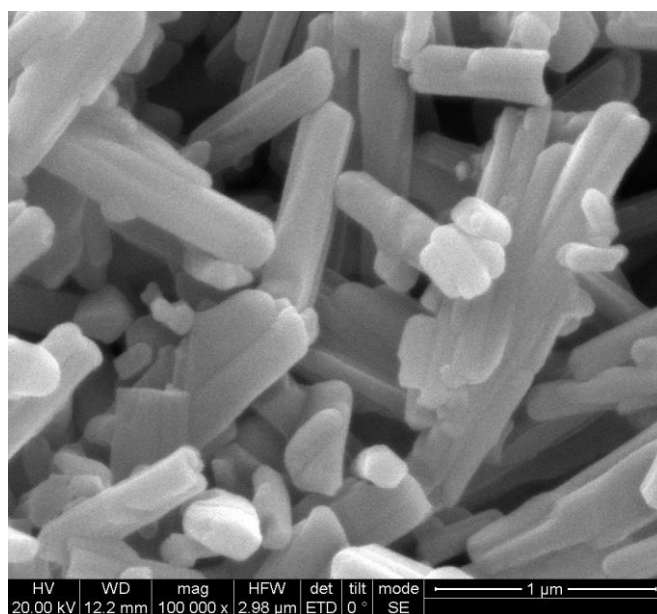


Figure S1 SEM image of the original Ce-MOF.

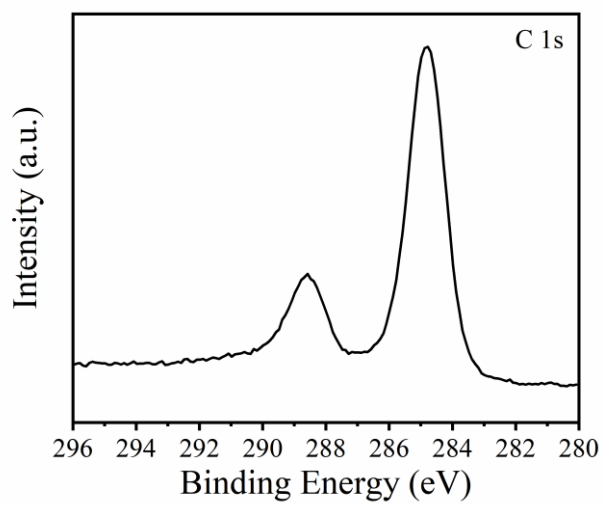


Figure S2 C1s XPS high-resolution spectra of MVCM.

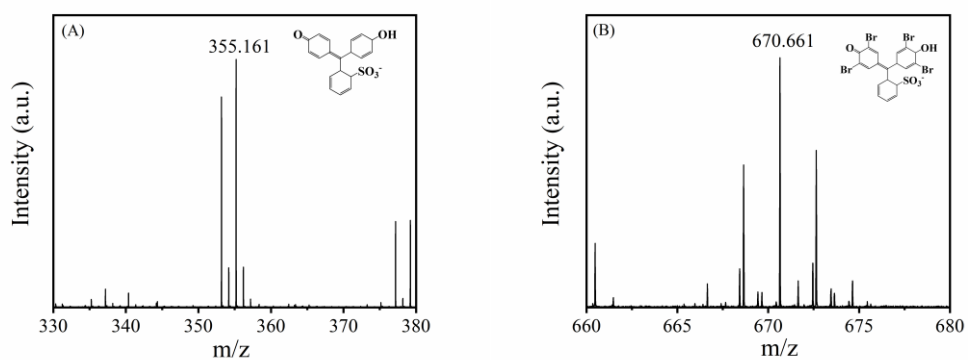


Figure S3 MALDI-TOF MS spectra (positive ion modes) of (A) phenol red and (B) bromophenol blue.

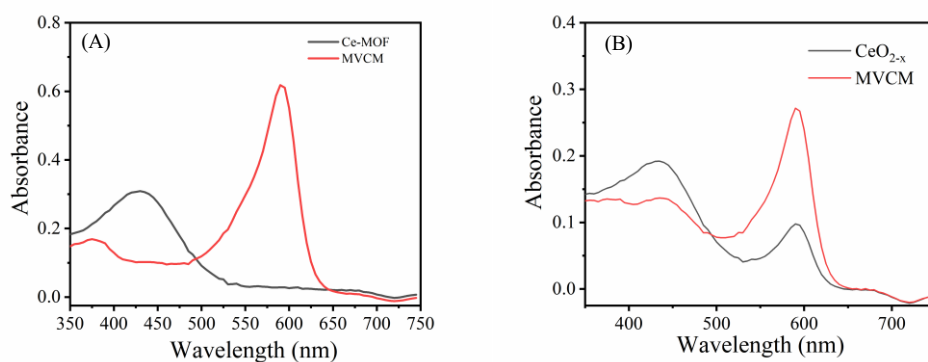


Figure S4 The absorption spectrum of MVCM and original Ce-MOF (A), CeO_{2-x} and MVCM (B) under the same reaction condition.

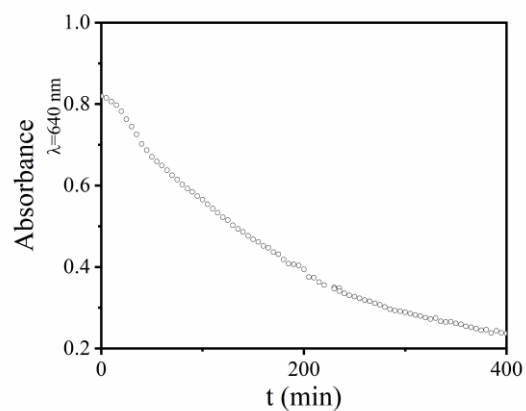


Figure S5 The bleaching of celestine blue (CB) at 640 nm indicates the formation of oxidized bromine species (e.g., OBr^-).

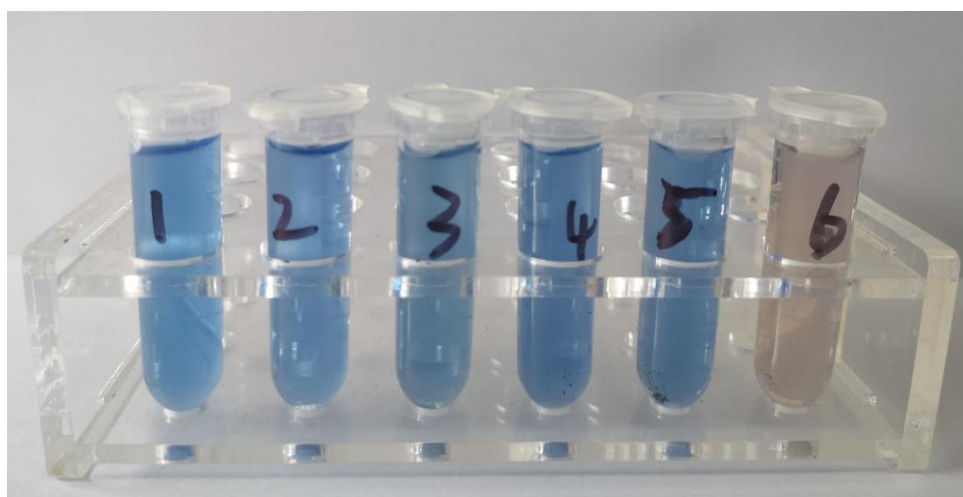


Figure S6 Celestine blue (CB) reacts in different systems. (1. $\text{MVCM} + \text{NH}_4\text{Br} + \text{CB}$; 2. $\text{H}_2\text{O}_2 + \text{NH}_4\text{Br} + \text{CB}$; 3. $\text{MVCM} + \text{H}_2\text{O}_2 + \text{CB}$; 4. $\text{MVCM} + \text{CB}$; 5. $\text{H}_2\text{O}_2 + \text{CB}$; 6. $\text{MVCM} + \text{NH}_4\text{Br} + \text{H}_2\text{O}_2 + \text{CB}$).

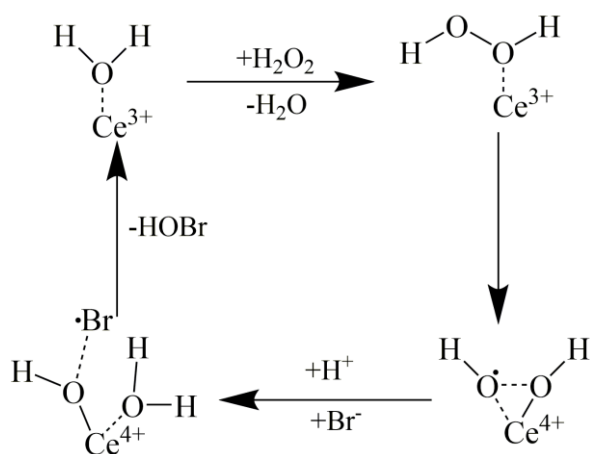


Figure S7 Probable catalytic mechanism of the MVCM as haloperoxidase mimic.

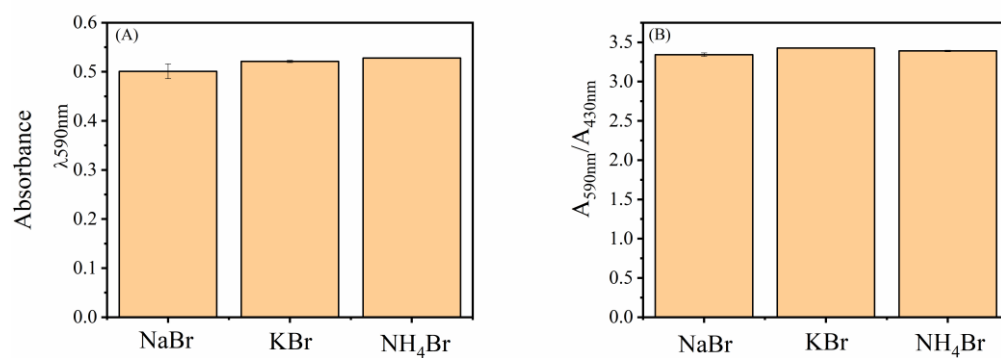


Figure S8 Dependence on the bromine source.

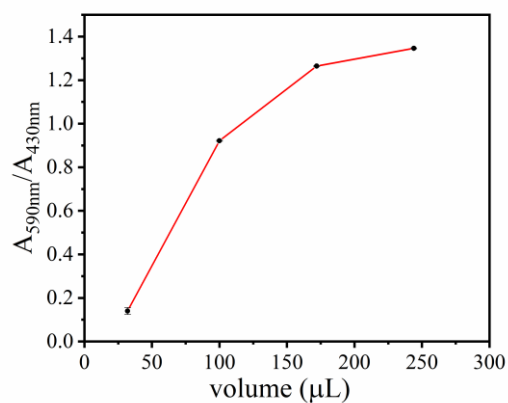


Figure S9 Ce-MOF was treated with different NaOH/H₂O₂ volume.

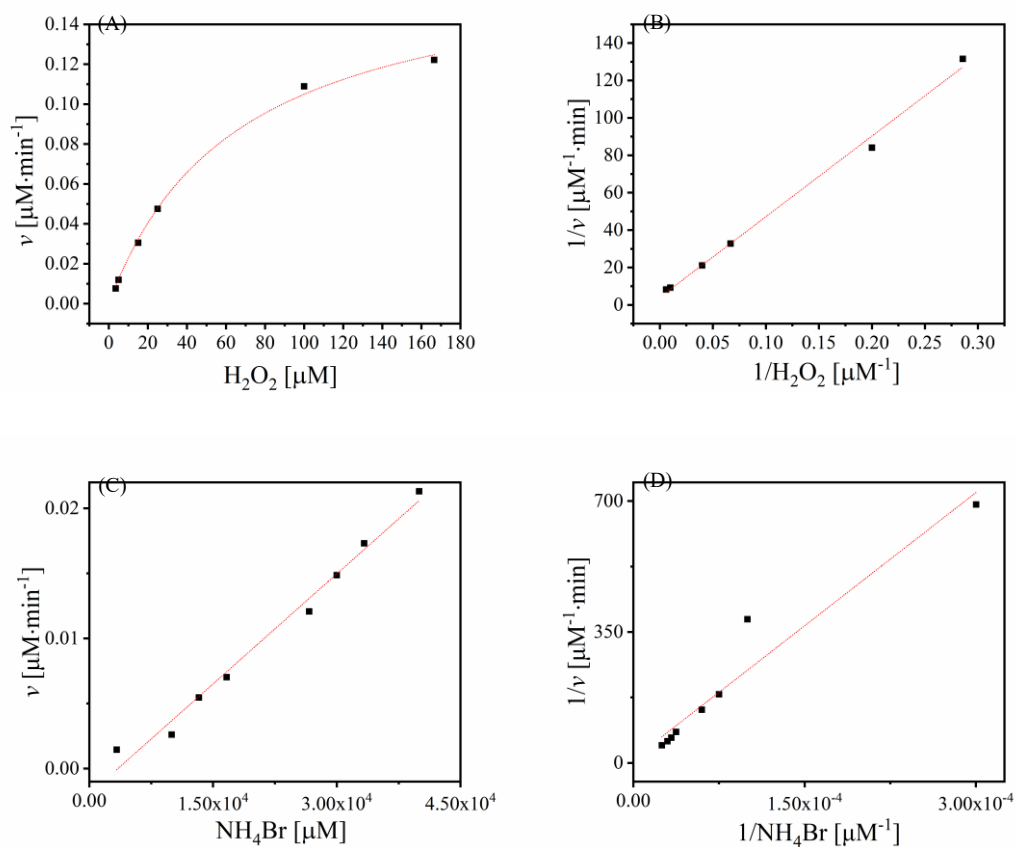


Figure S10 Steady-state kinetic assays of MVCM. The H_2O_2 concentration was varied, the concentration of NH_4Br and PR was fixed (A); the NH_4Br concentration was varied, the concentration of H_2O_2 and PR was fixed (C); and the double-reciprocal plots of haloperoxidase-like activity of MVCM with a fixed concentration of one substrate relative to varying concentration of the other substrate (B and D).

Table S1. A comparison of K_m value for materials-based haloperoxidase mimics.

Materials	Substrate	K_m (M)	Ref.
CeO ₂ @C	H ₂ O ₂	2.1×10^{-3}	[1]
	Br ⁻	0.5	
CeO _{2-x} nanorod	H ₂ O ₂	2.6×10^{-4}	[2]
	Br ⁻	0.35	
MVCM	H ₂ O ₂	1.0×10^{-4}	This work
	Br ⁻	0.22	

Table S2. Comparison detection limit in different catalyst systems by means of different methods.

Materials	Method	Detection limit (μ M)	Ref.
flake-like Cu ₂ O	Electrochemistry	90.5	[3]
ZnCrCoO ₄ /NCNTs	Electrochemistry	1	[4]
rGO/Ag NPs	Electrochemistry	31.3	[5]
Ce-MOF	Fluorescence	10	[6]
CuO nanoparticles	Fluorescence	0.34	[7][7]
NiFe LDH	Colorimetry	4.4	[8]
Ni ⁰ nanoparticle	Colorimetry	120	[9]
VS ₄	Colorimetry	5	[10]
MVCM	Colorimetry	3.25	This work

Note:

ZnCrCoO₄/NCNTs: ZnCrCoO₄ and nitrogen-doped carbon nanotubes composite

rGO/Ag NPs: reduced graphene oxide/Ag nanoparticles

NiFe LDH: NiFe layered double hydroxide nanosheets

References

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