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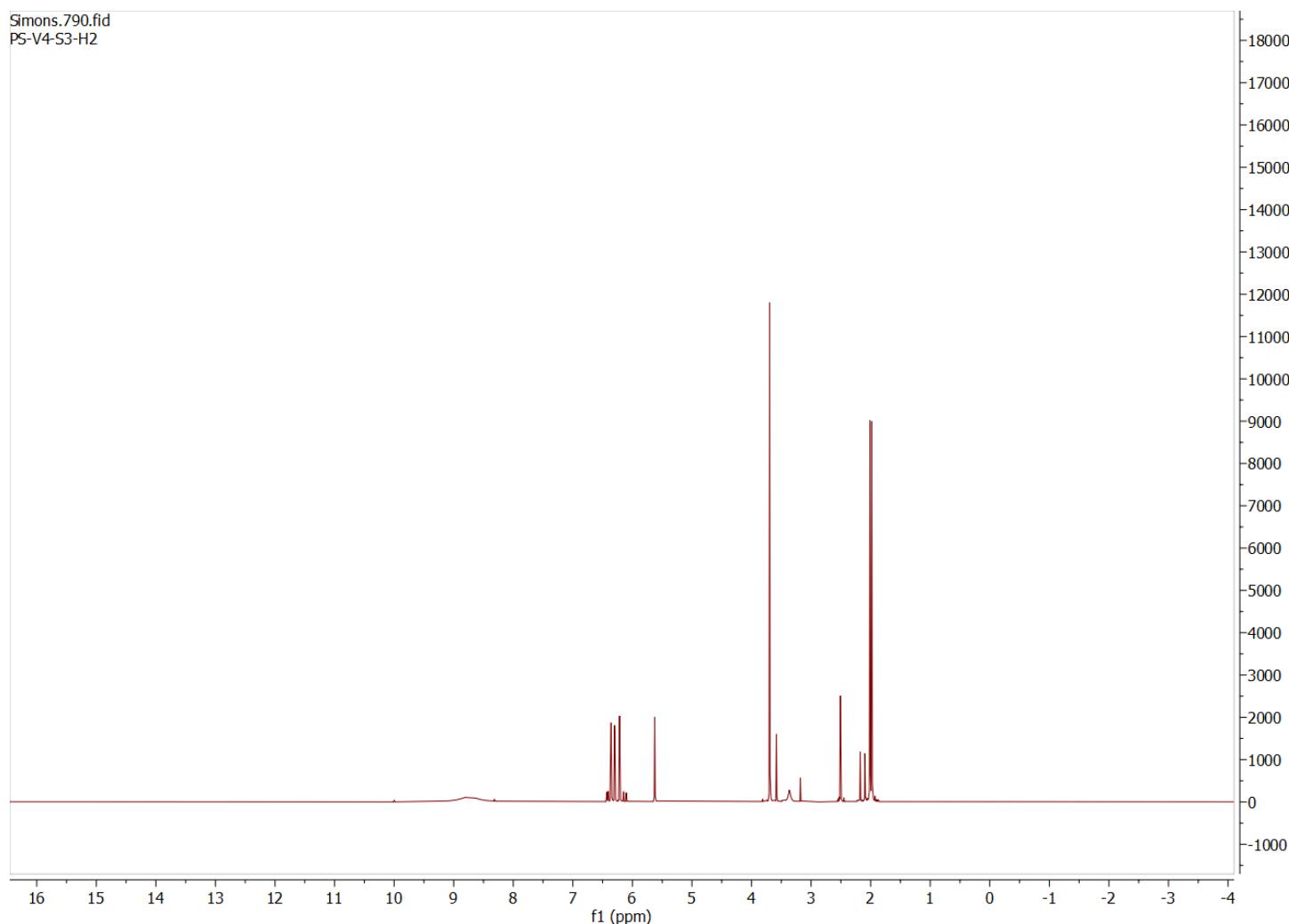


Figure S1. ¹H-NMR spectrum (DMSO-d₆ – 600 MHz) of Compound A

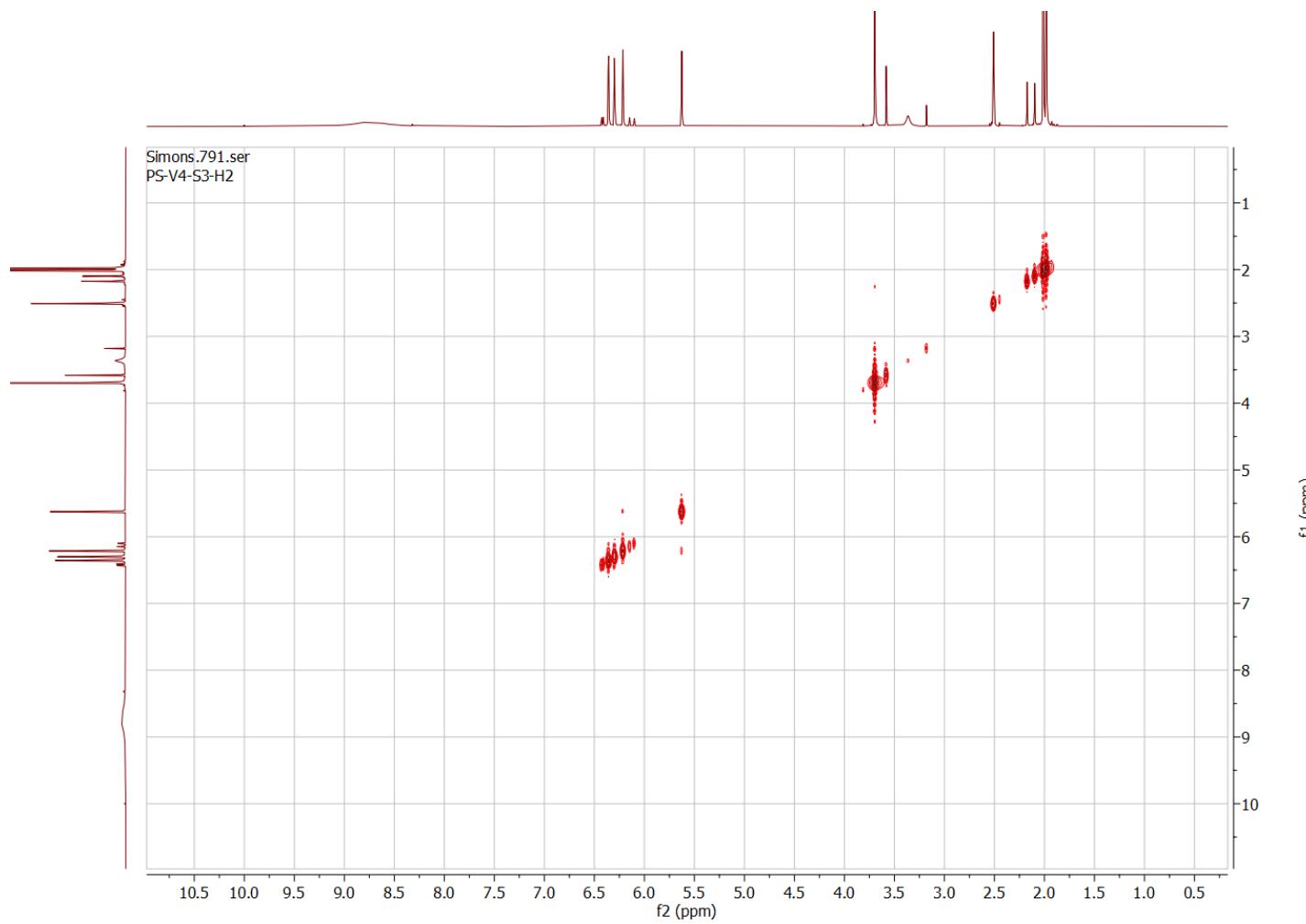


Figure S2. COSY NMR spectrum (DMSO-d₆ – 600 MHz) of compound A

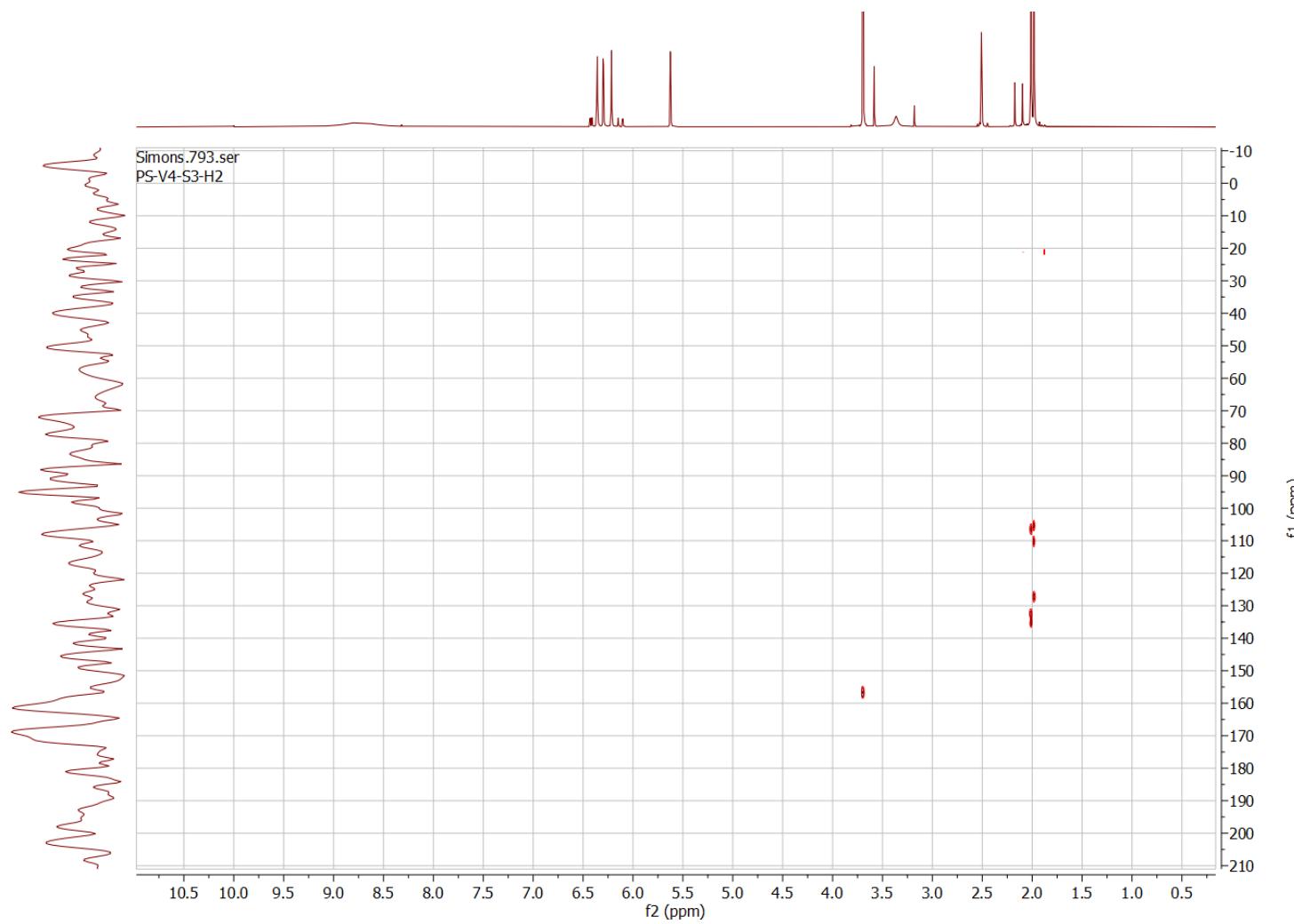


Figure S3. HMBC NMR spectrum (DMSO-d₆ – 600 MHz) of compound A

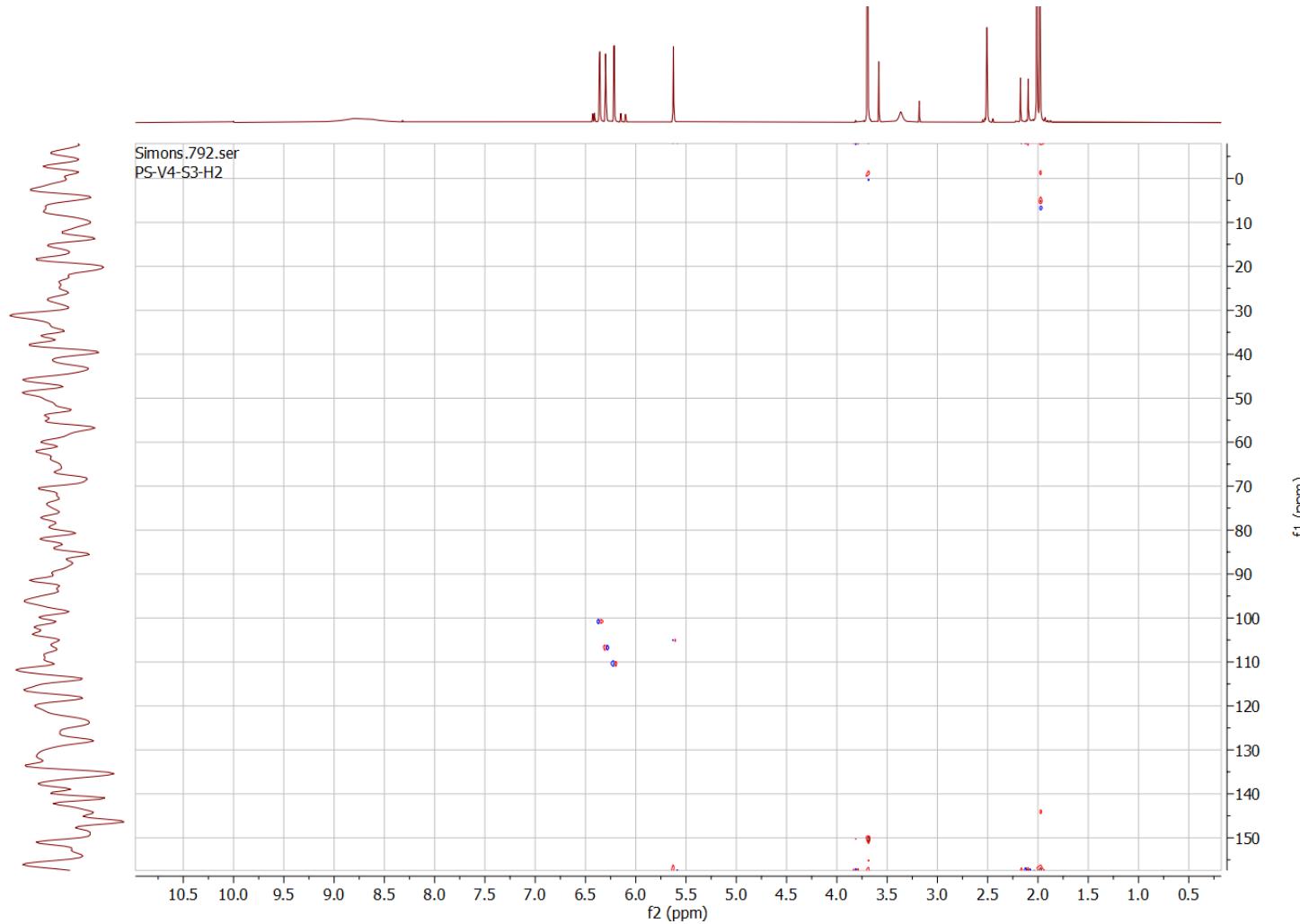


Figure S4. HSQC NMR spectrum (DMSO-d₆ – 600 MHz) of compound A

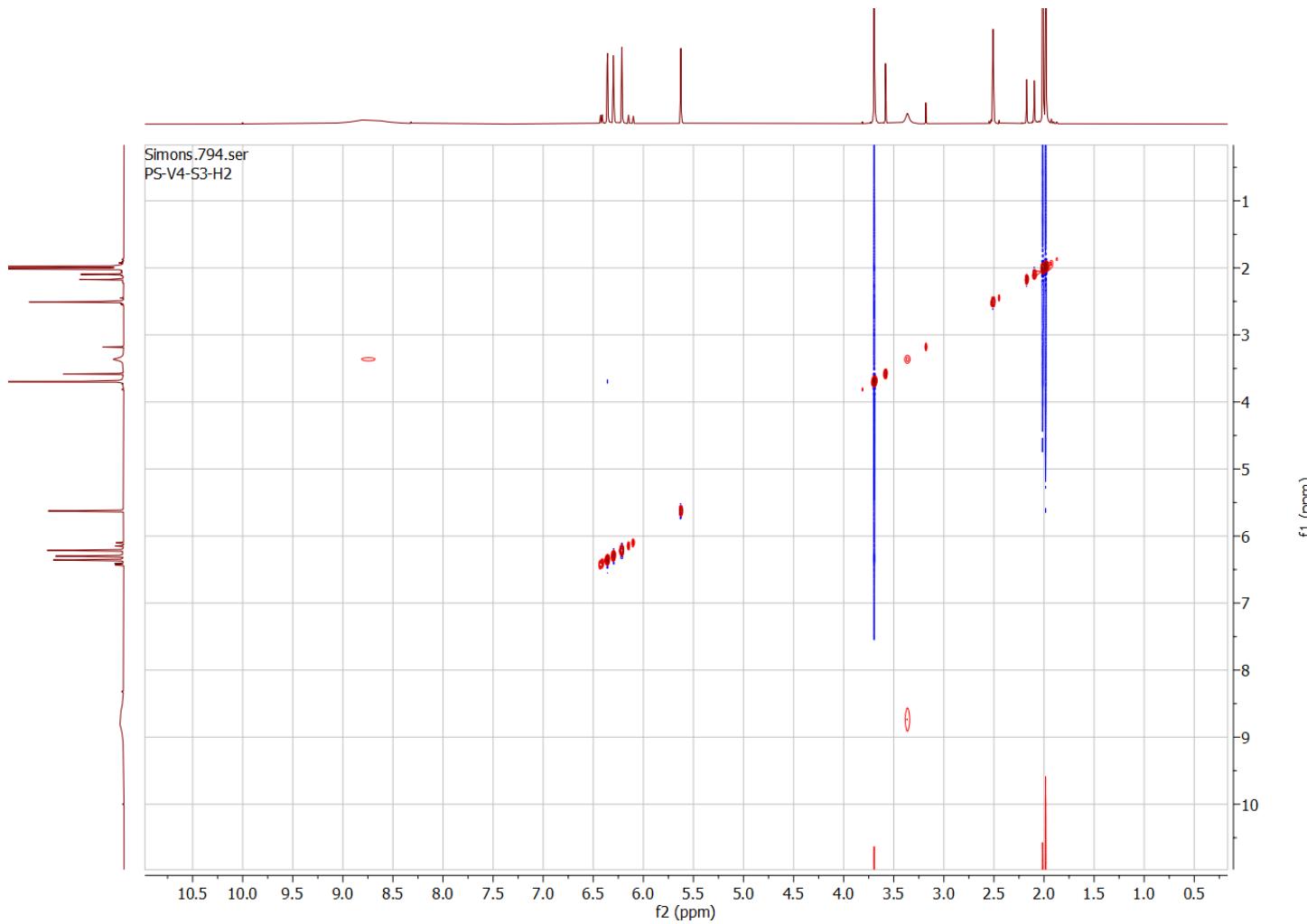


Figure S5. NOESY NMR spectrum (DMSO-d₆ – 600 MHz) of compound A

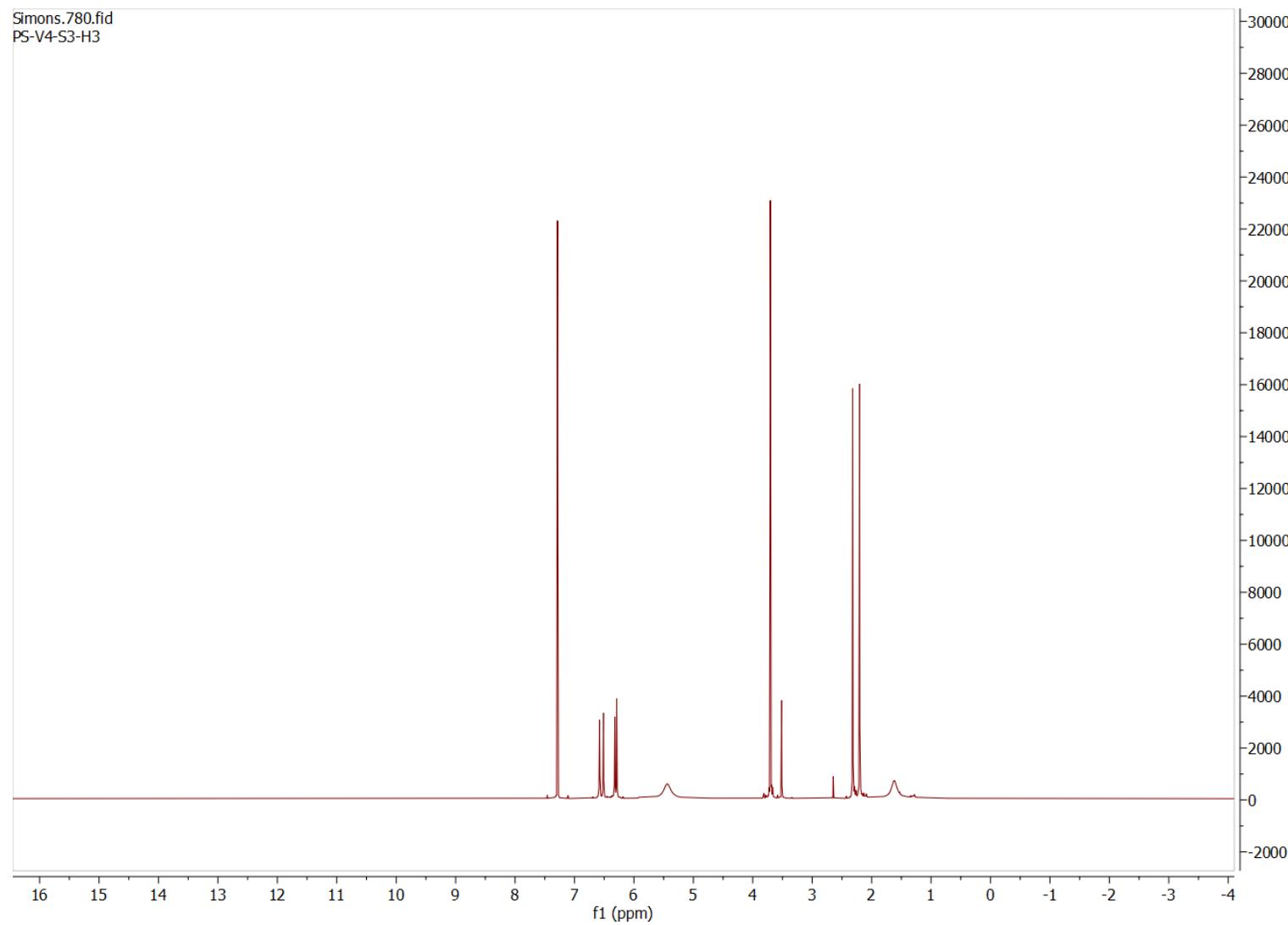


Figure S6. ¹H NMR spectrum (CDCl_3 , 600 MHz) of compound B

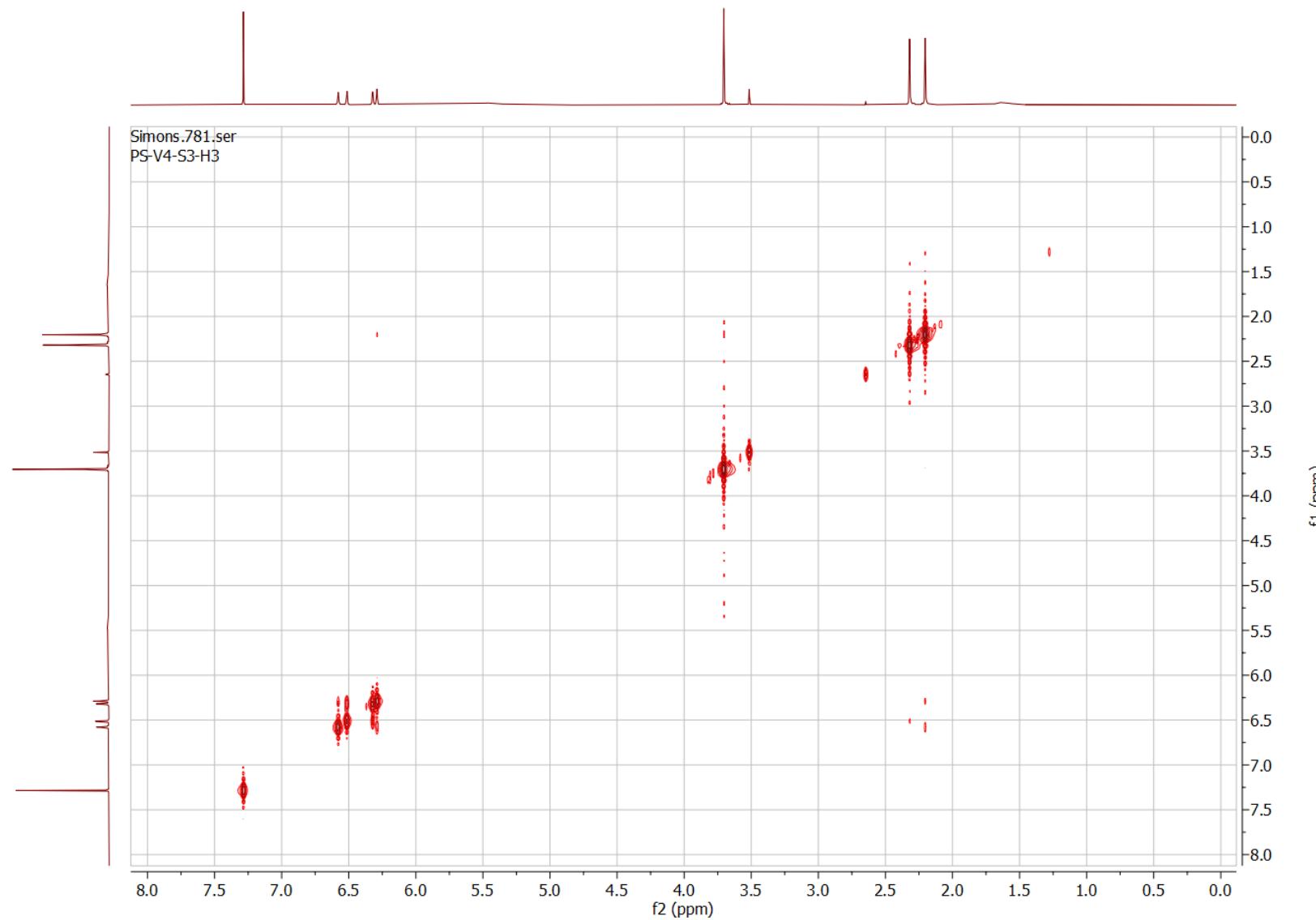


Figure S7. COSY NMR spectrum (CDCl_3 , 600 MHz) of compound B

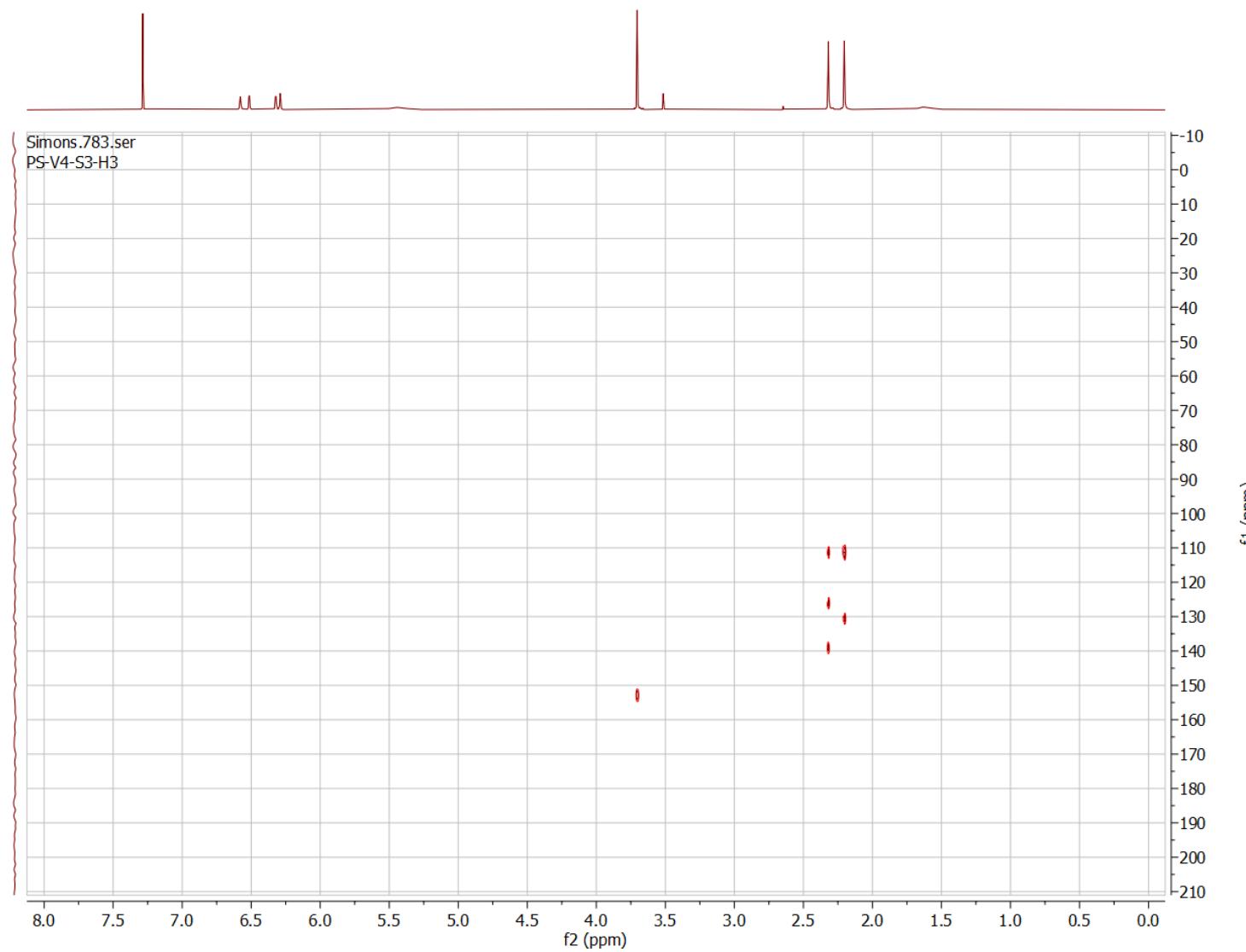


Figure S8. HMBC NMR spectrum (CDCl_3 , 600 MHz) of compound B

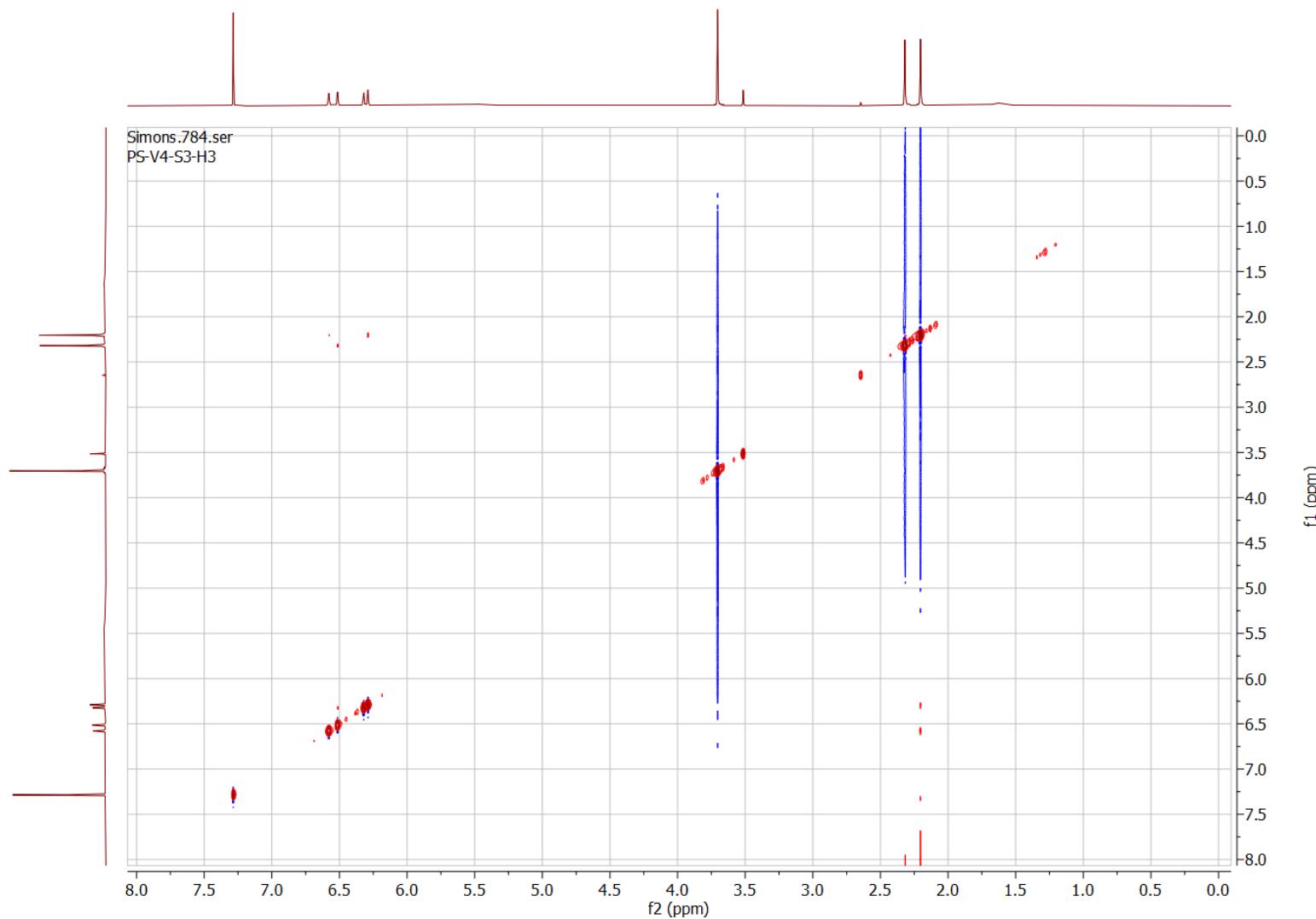


Figure S9. NOESY NMR spectrum (CDCl_3 , 600 MHz) of compound B

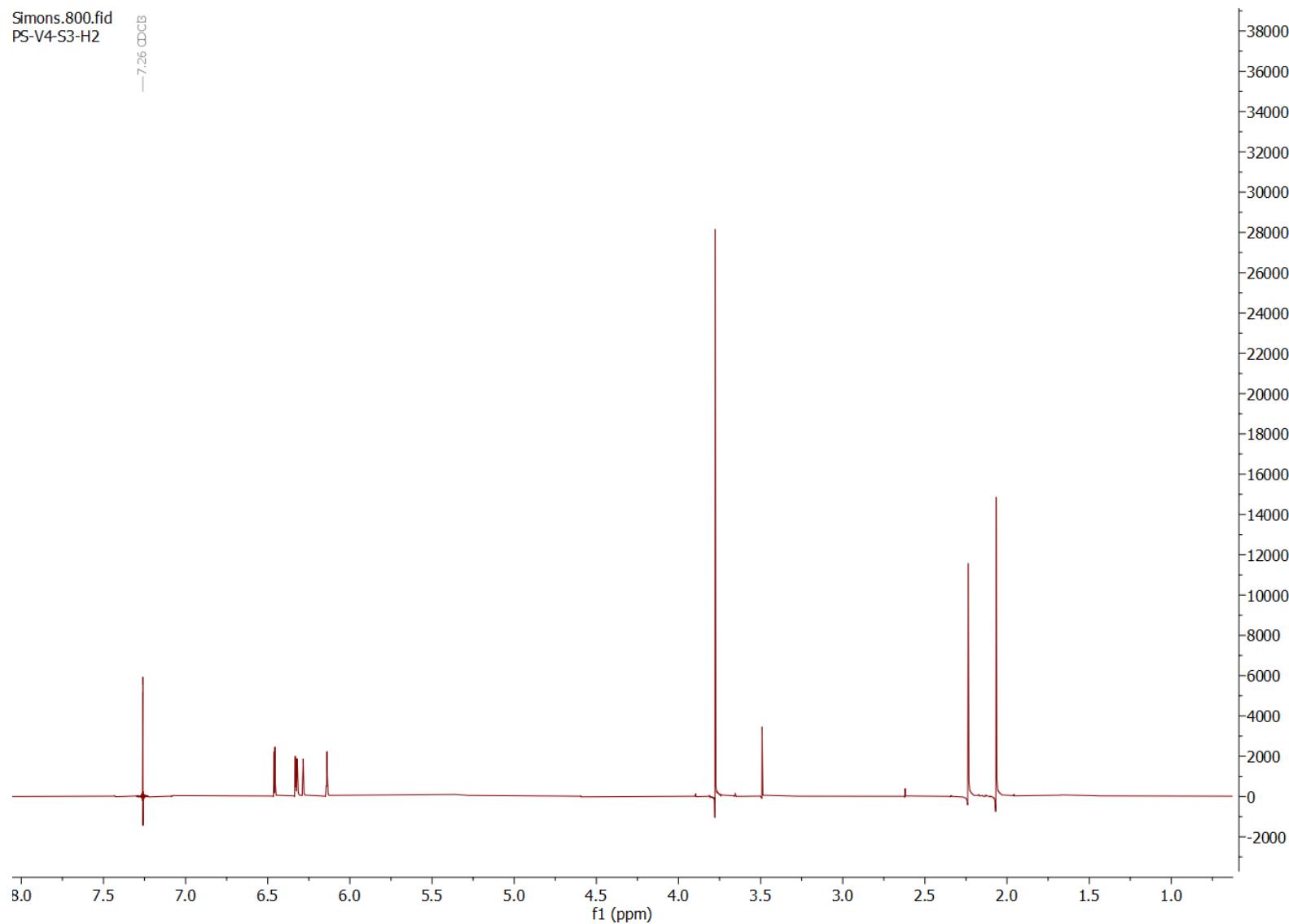


Figure S10. ¹H NMR spectrum (CDCl₃ · 600 MHz) of compound C

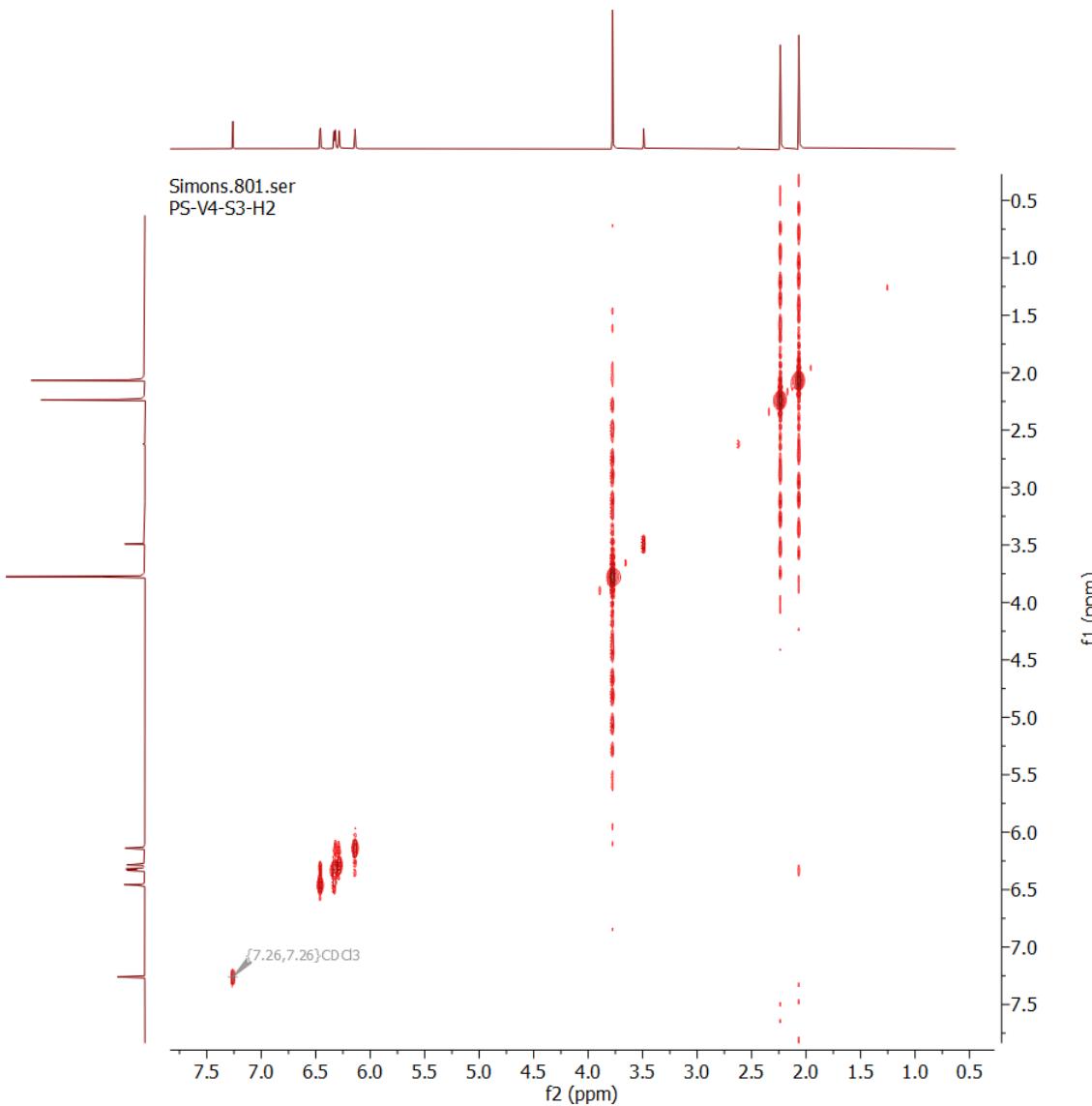


Figure S11. COSY NMR spectrum (CDCl_3 , 600 MHz) of compound C

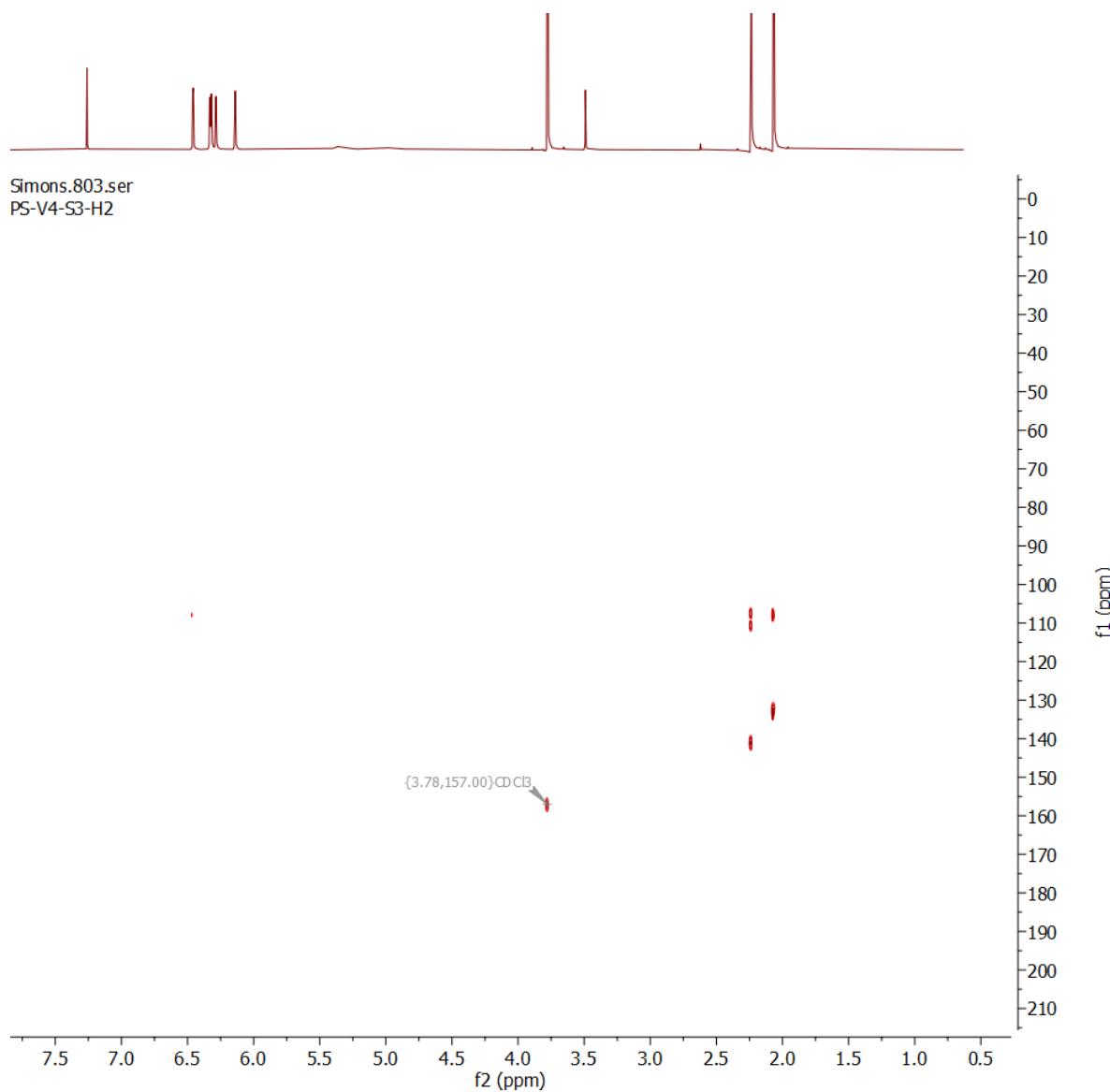


Figure S12. HMBC NMR spectrum (CDCl_3 , 600 MHz) of compound C

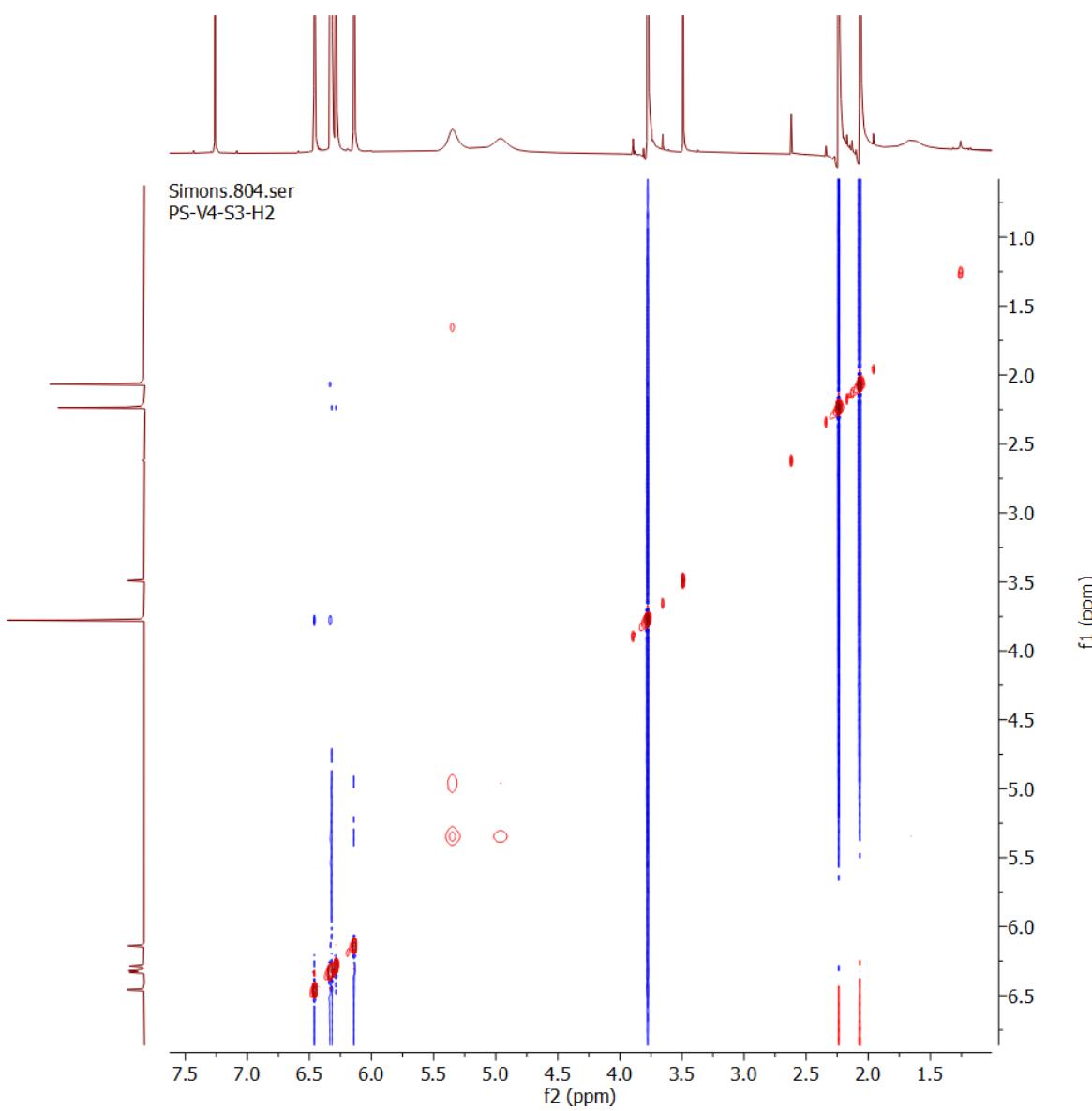


Figure S13. NOESY NMR spectrum (CDCl_3 , 600 MHz) of compound C

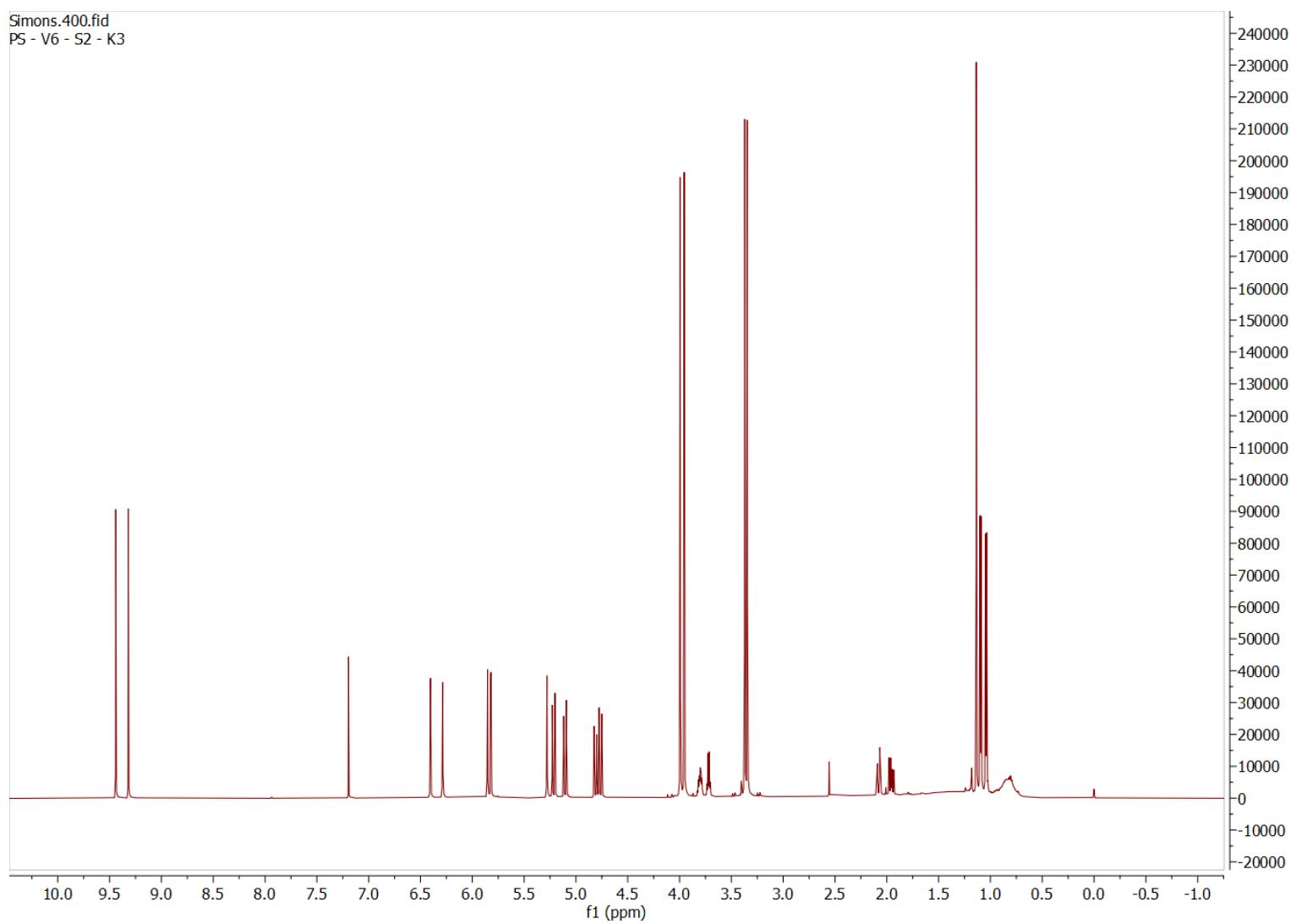


Figure S14. ^1H NMR spectrum (CDCl_3 , 600 MHz) of compound D

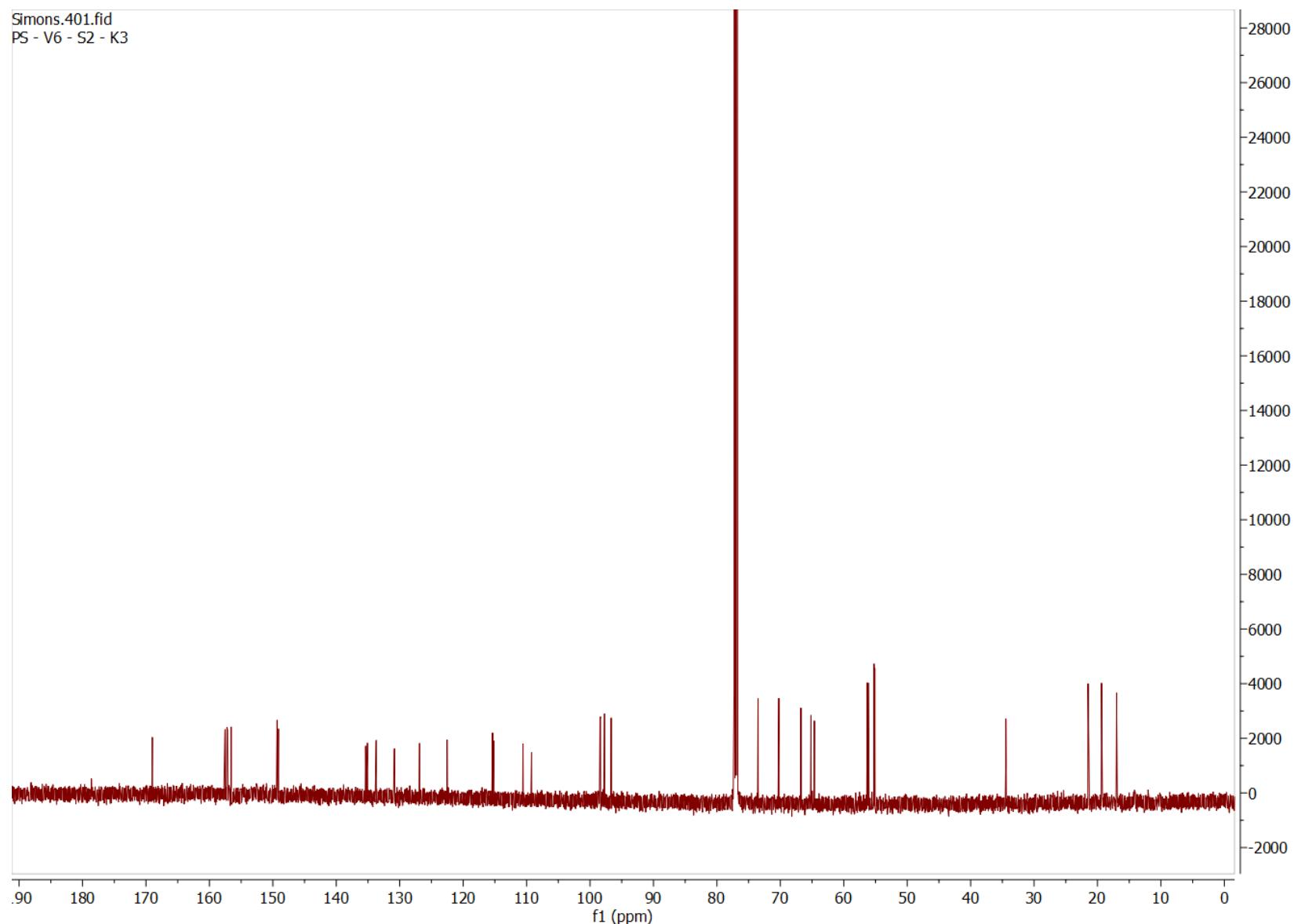


Figure S15. ¹³C NMR spectrum (CDCl_3 , 150 MHz) of compound D



Figure S16. COSY NMR spectrum (CDCl_3 - 600 MHz) of compound D

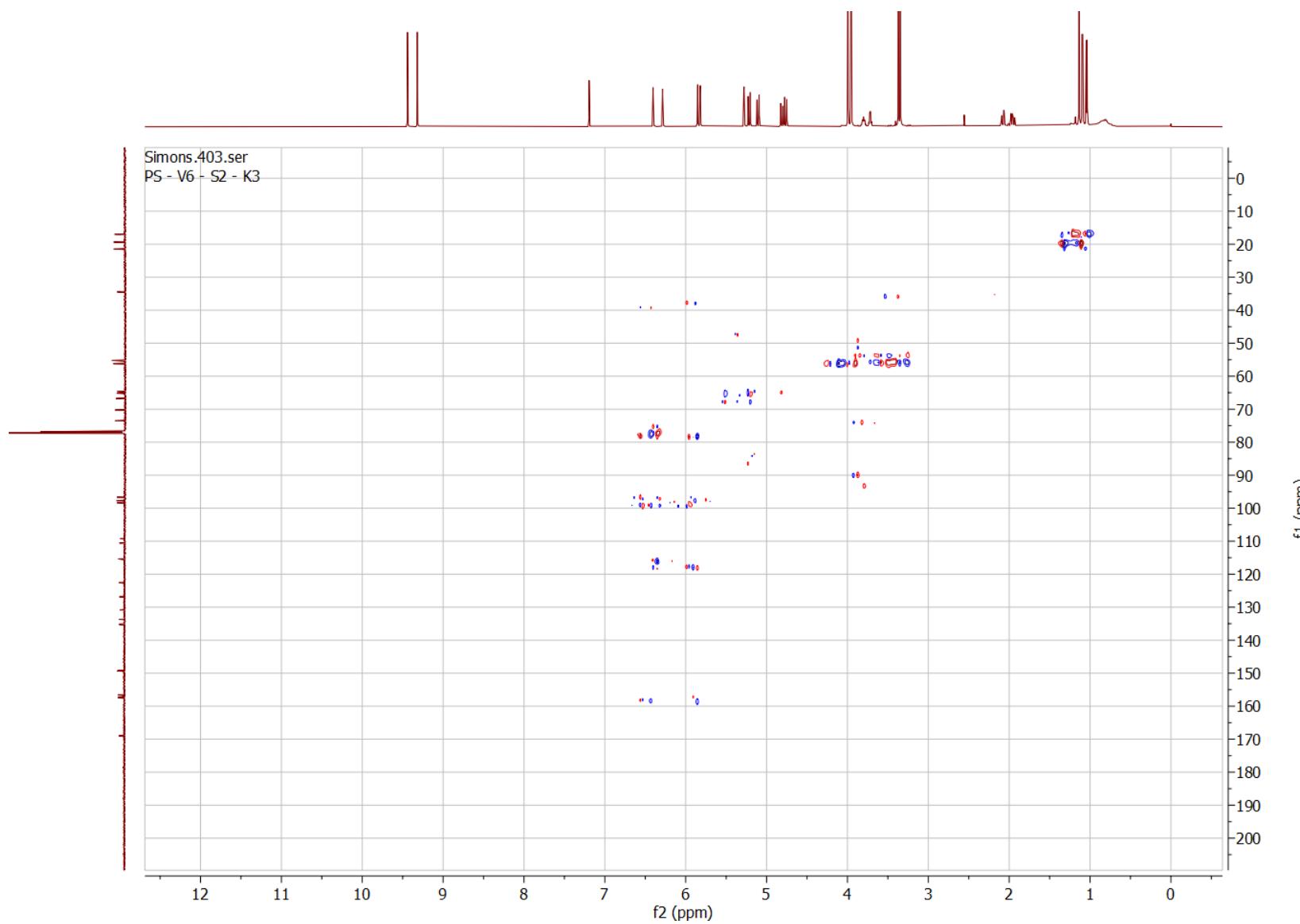


Figure S17. HSQC NMR spectrum (CDCl_3 - 600 MHz) of compound D

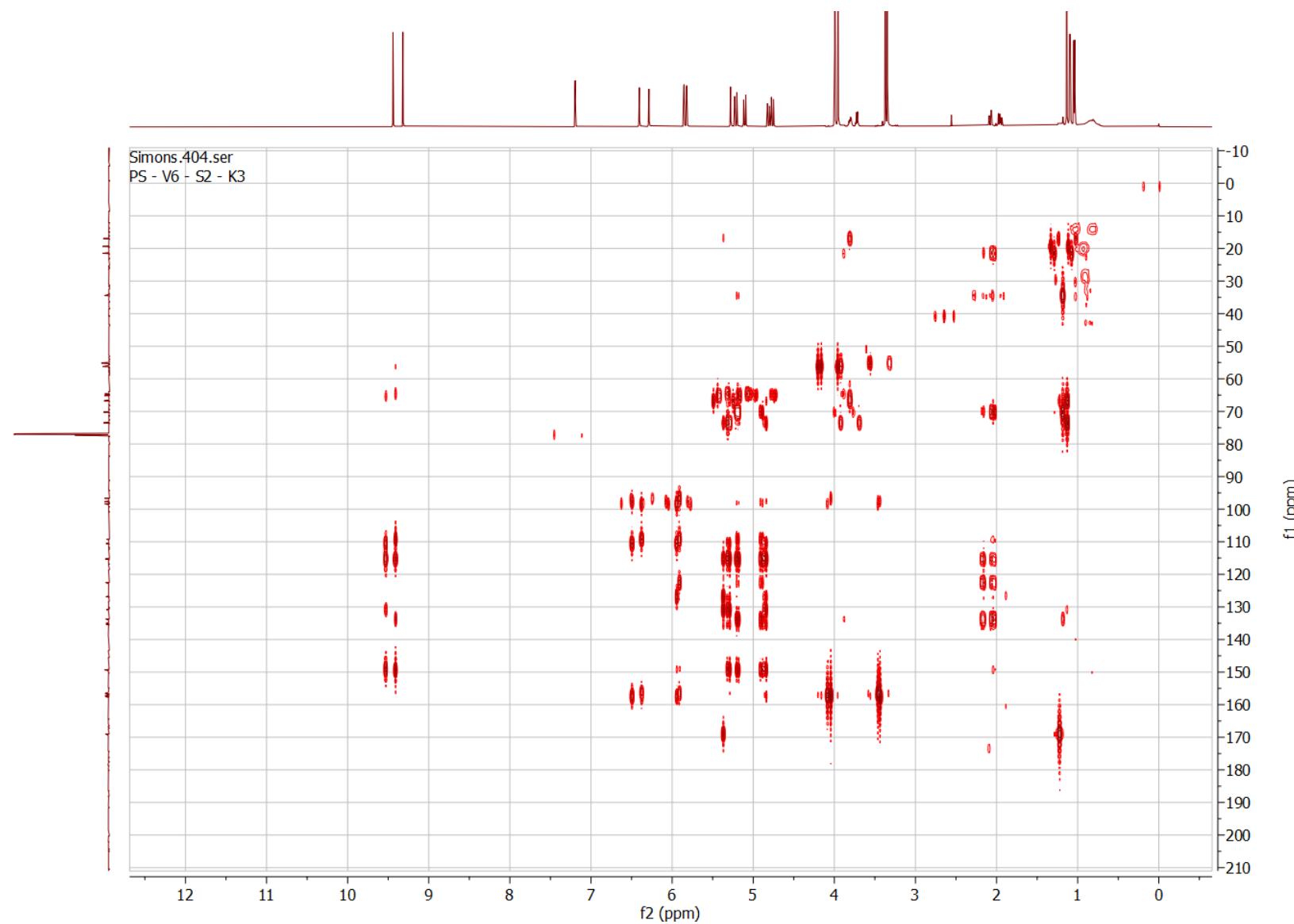


Figure S18. HMBC NMR spectrum (CDCl_3 - 600 MHz) of compound D

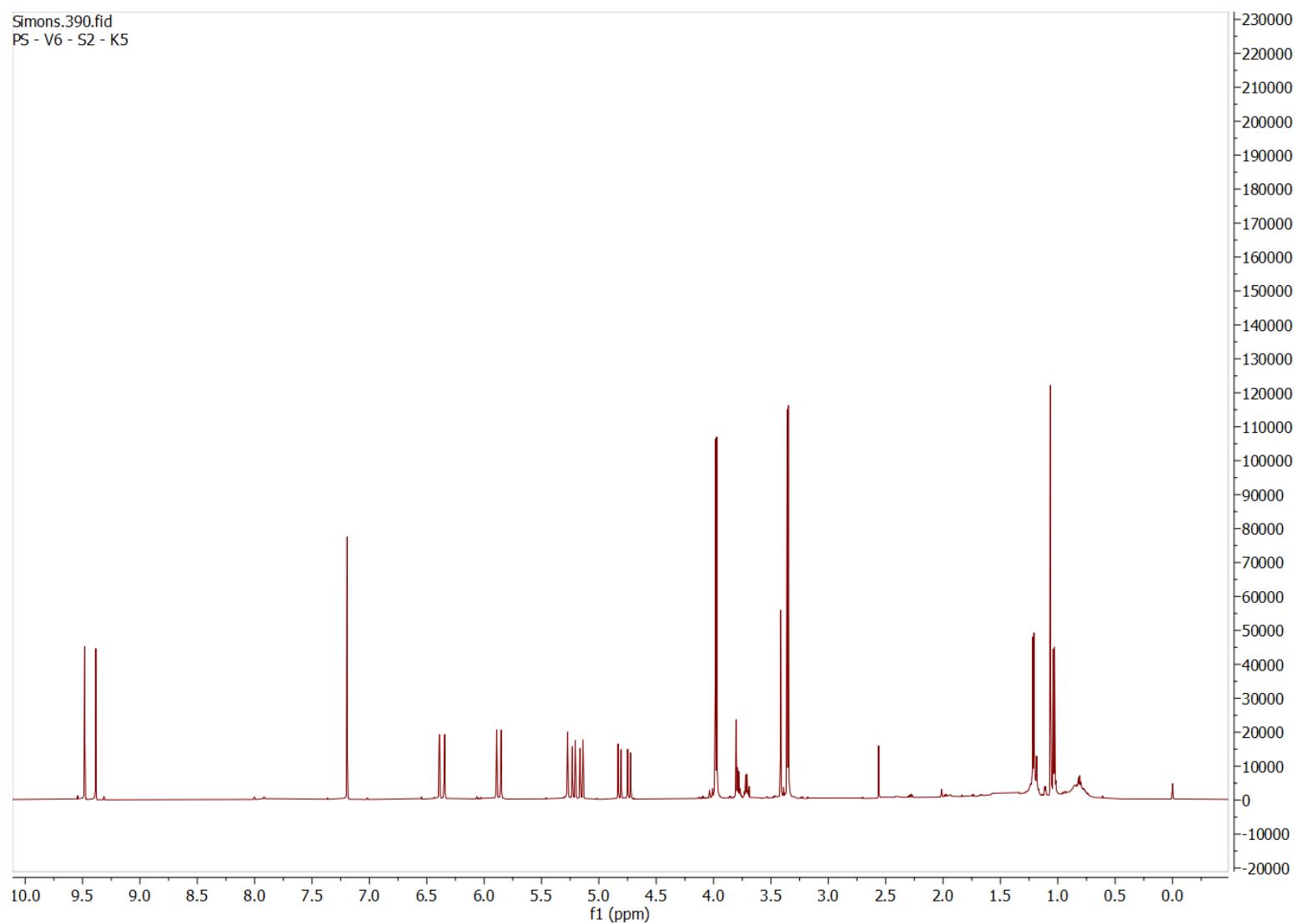


Figure S19. ^1H NMR spectrum (CDCl_3 - 600 MHz) of compound E

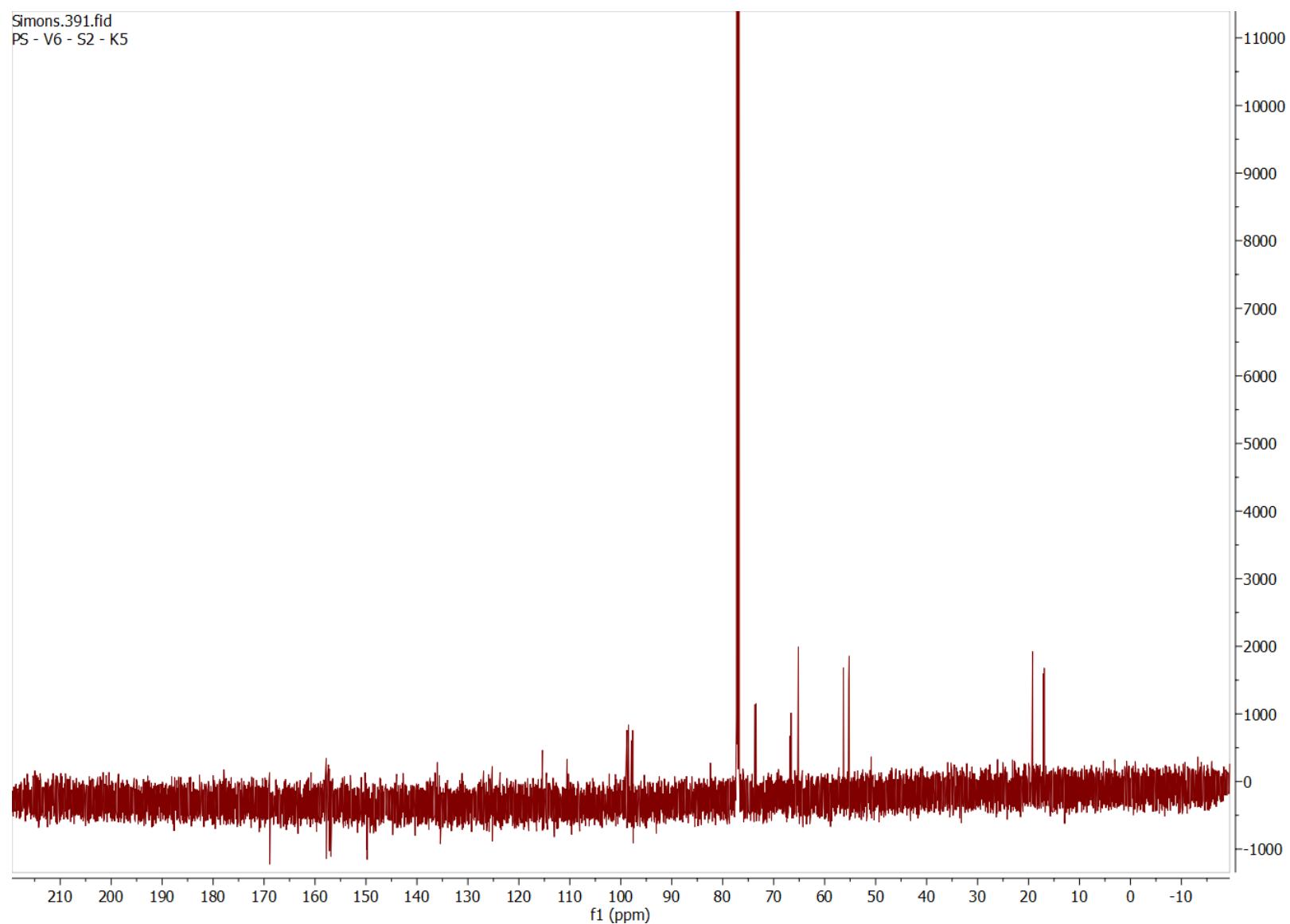


Figure S20. ¹³C NMR spectrum (CDCl_3 , 150 MHz) of compound E

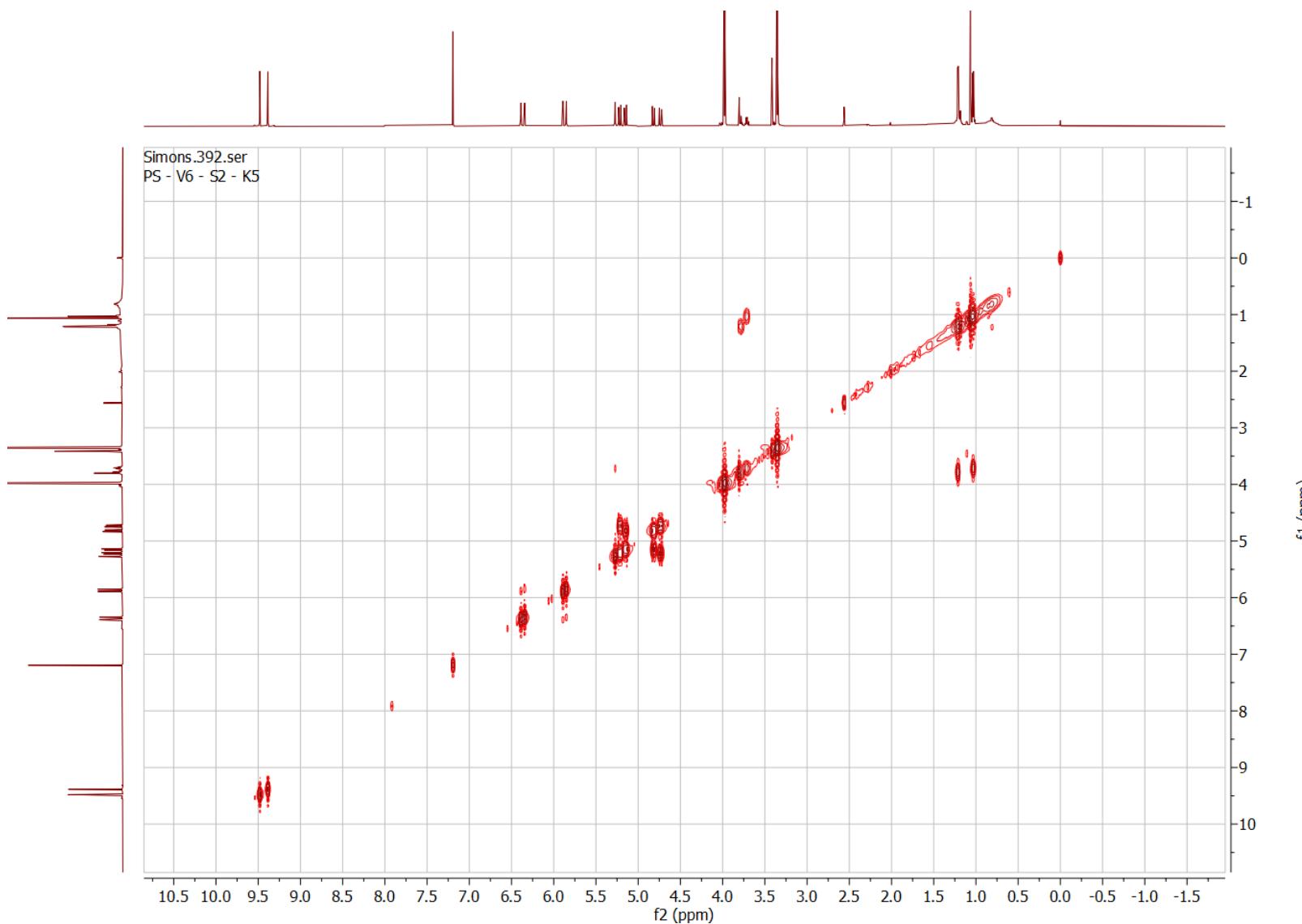


Figure S21. COSY NMR spectrum (CDCl_3 , 600 MHz) of compound E

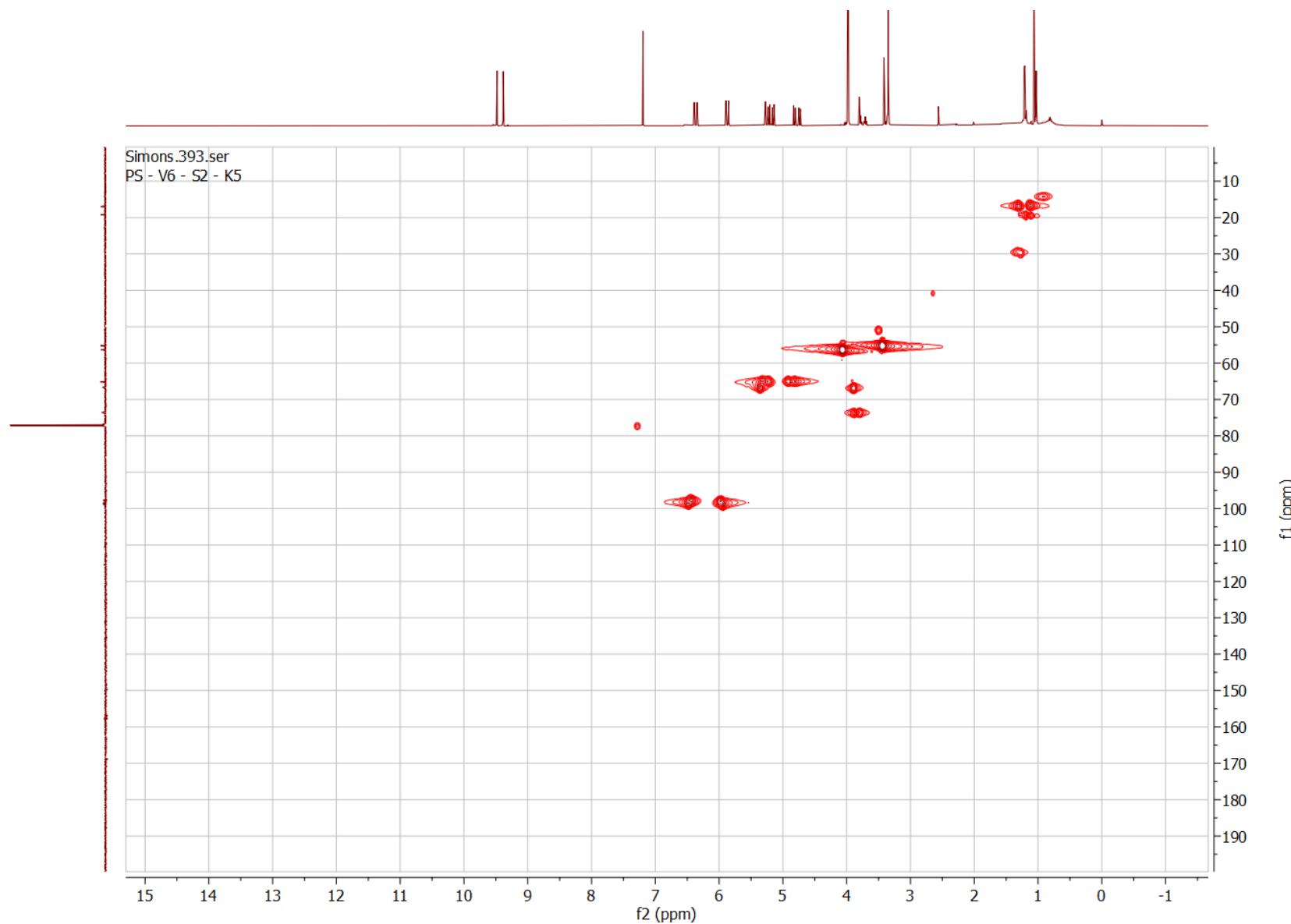


Figure S22. HSQC NMR spectrum (CDCl_3 - 600 MHz) of compound E

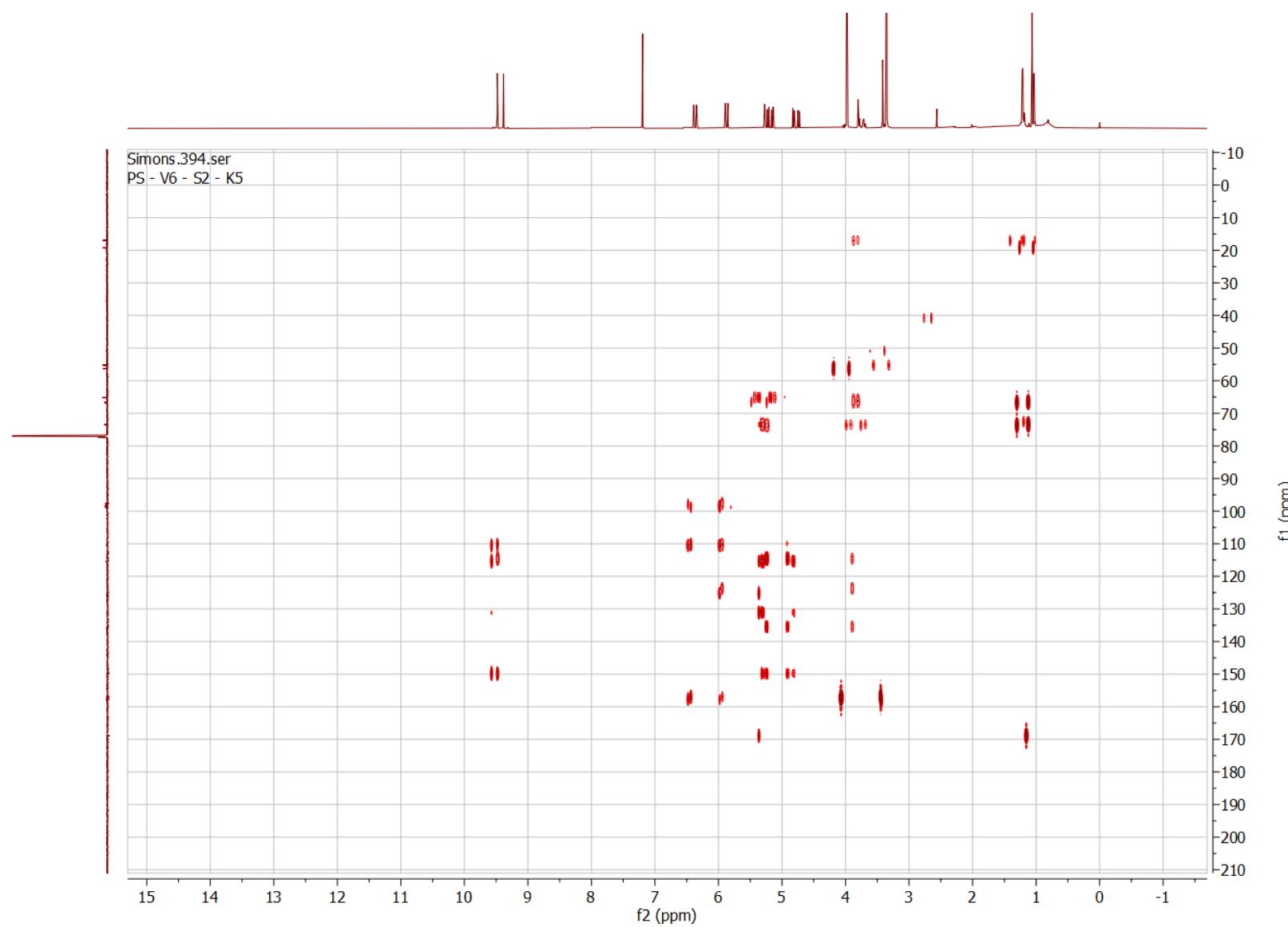


Figure S23. HMBC NMR spectrum (CDCl_3 - 600 MHz) of compound E

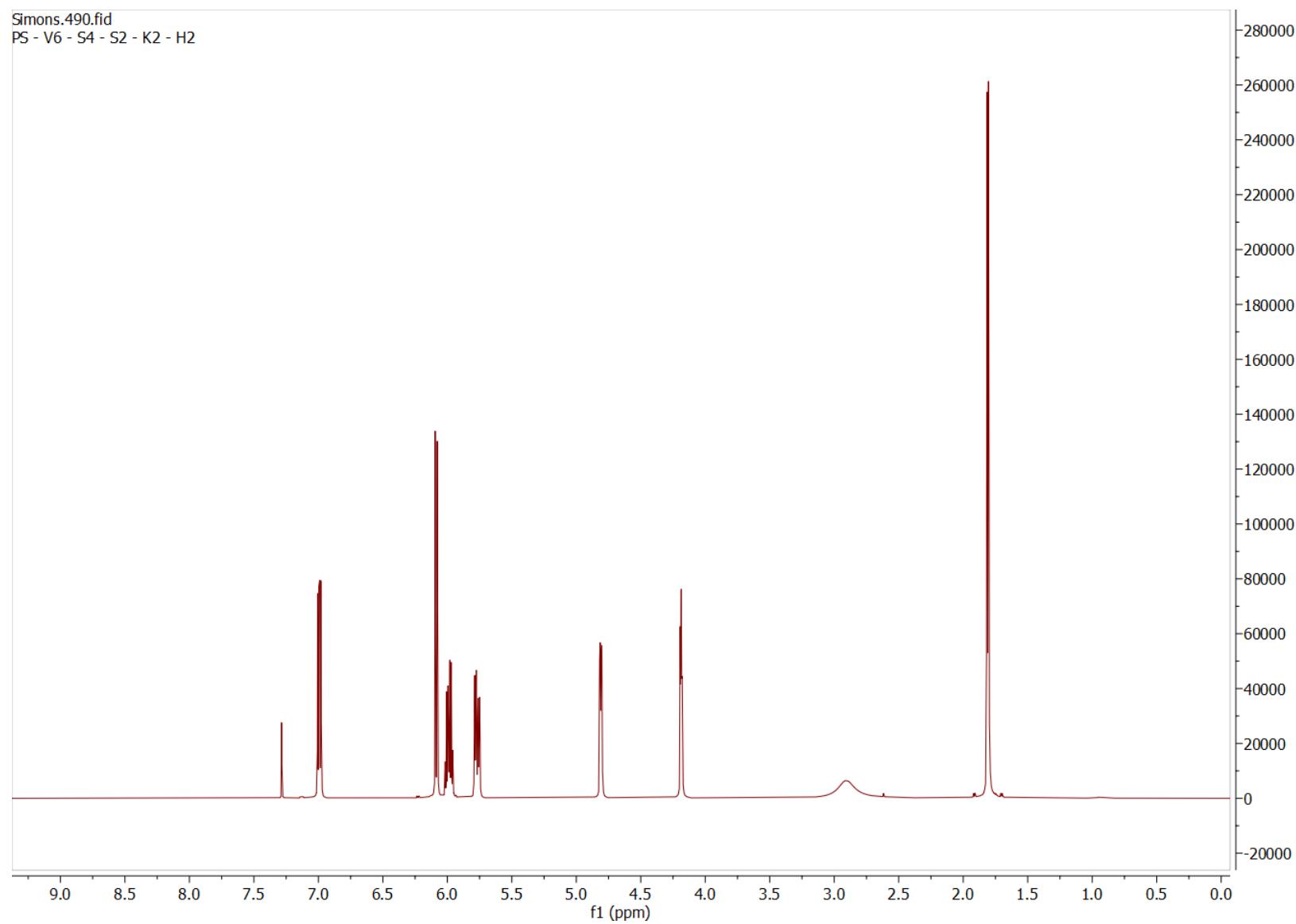


Figure S24. ¹H NMR spectrum (CDCl_3 - 600 MHz) of compound F

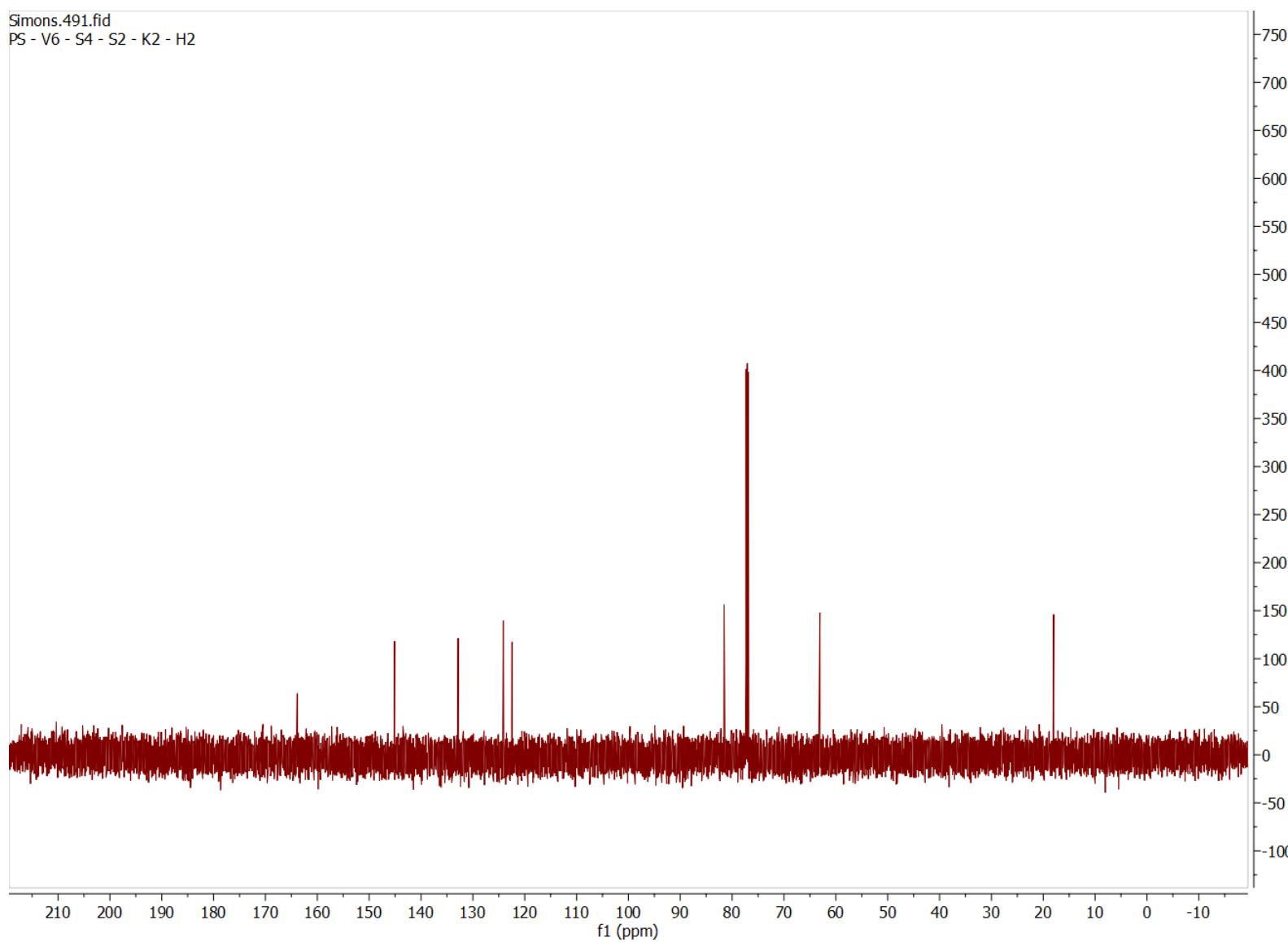


Figure S25. ¹³C NMR spectrum (CDCl_3 , 150 MHz) of compound F

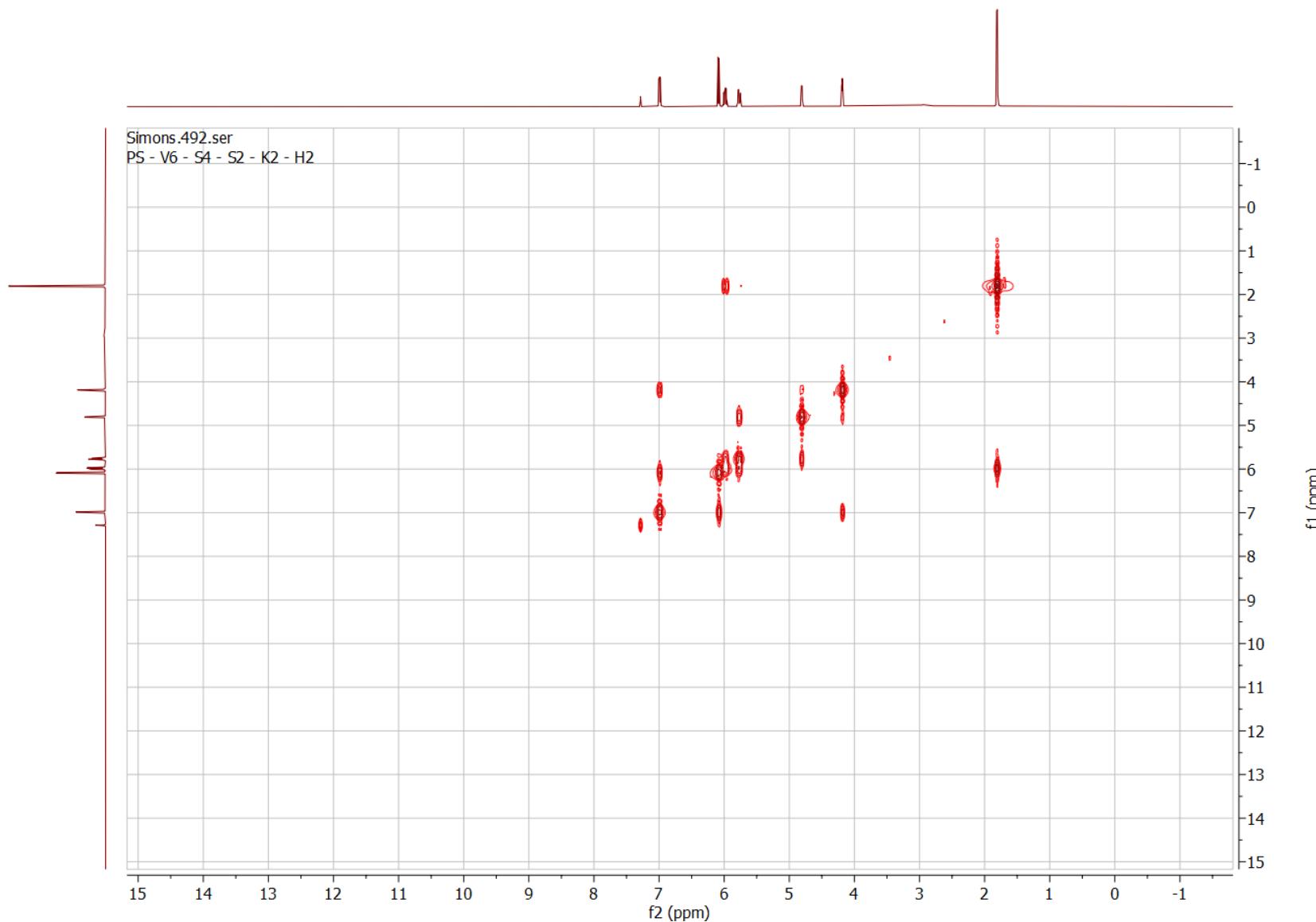


Figure S26. COSY NMR spectrum (CDCl_3 - 600 MHz) of compound F

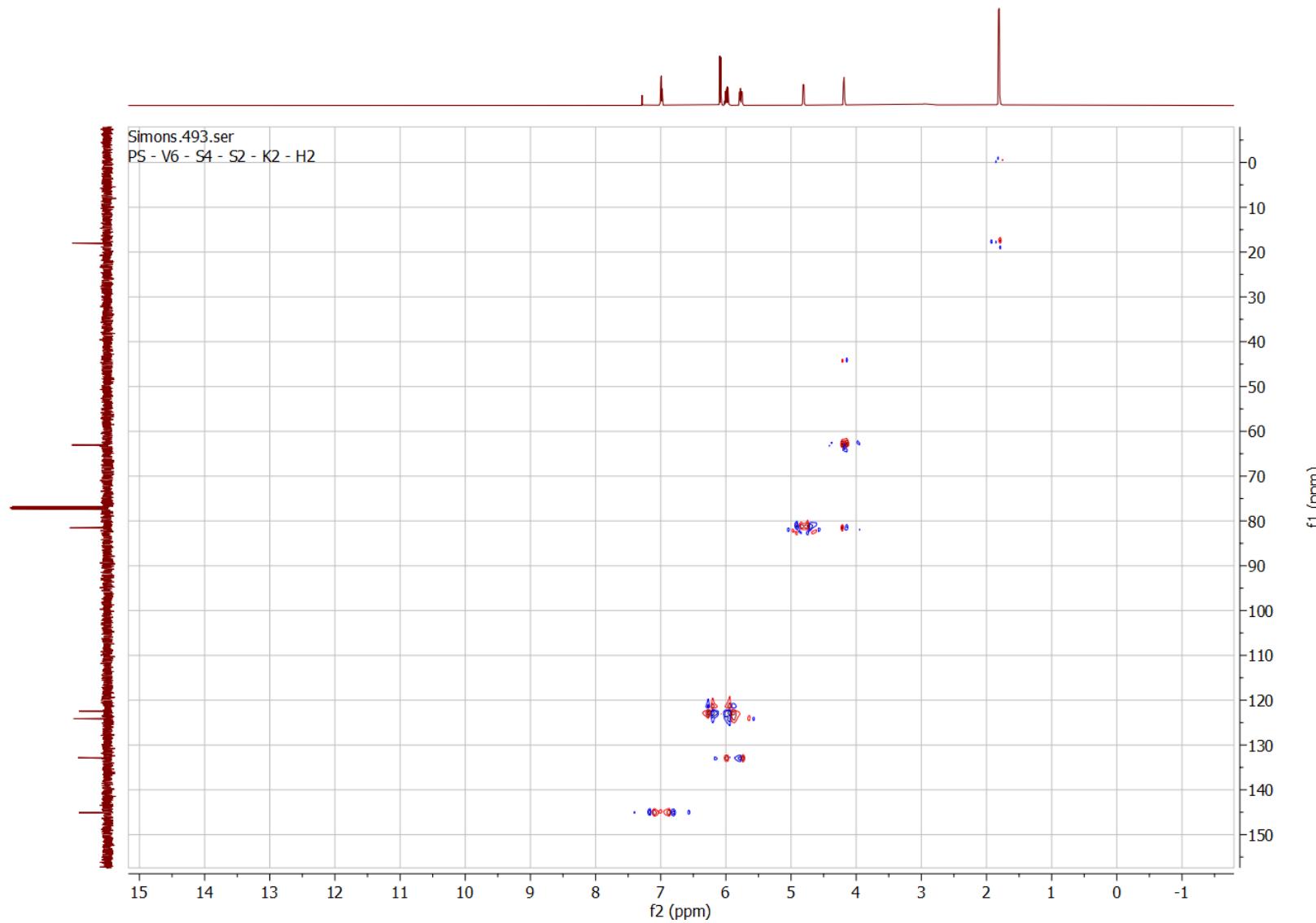
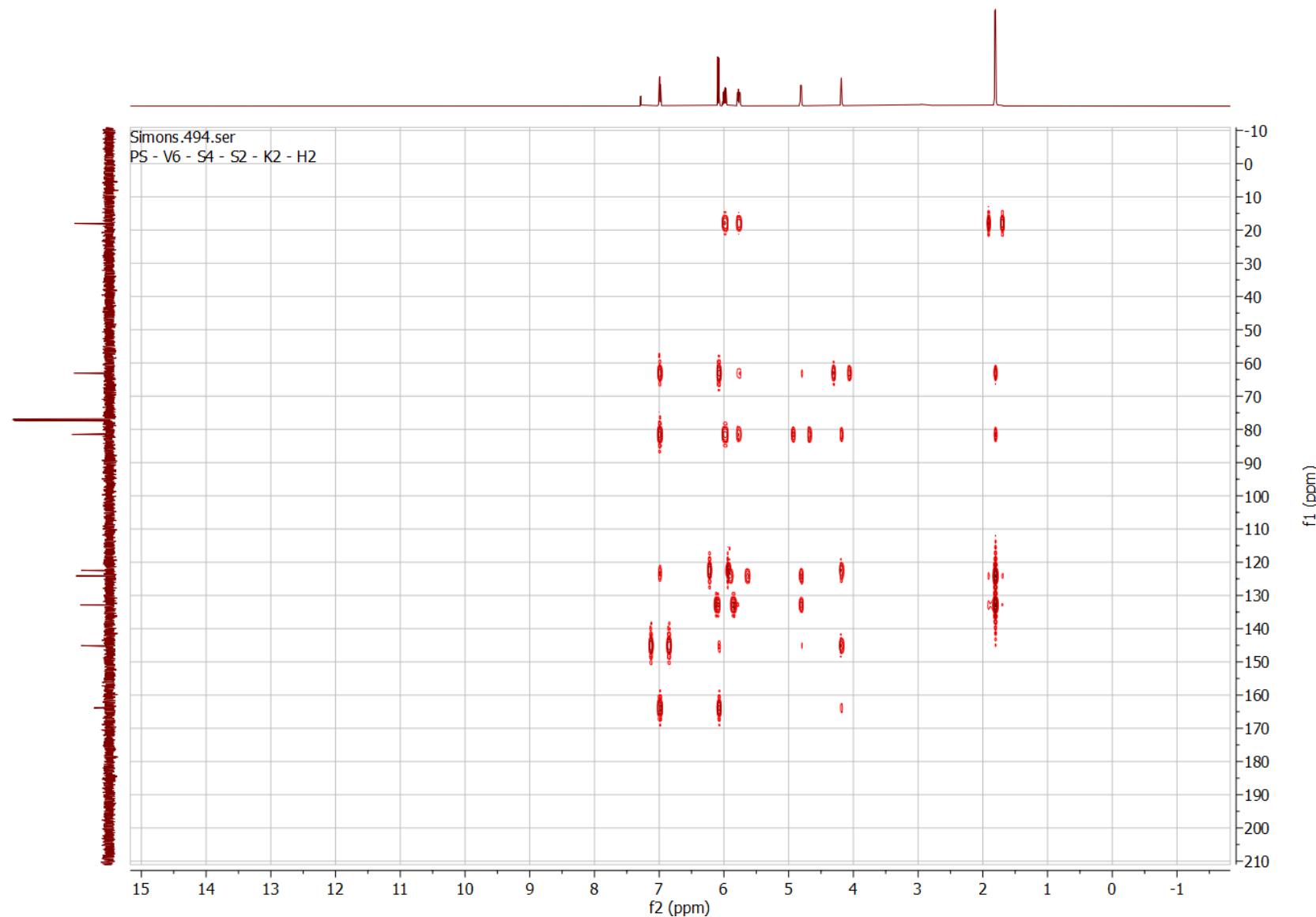


Figure S27. HSQC NMR spectrum (CDCl_3 - 600 MHz) of compound F

Figure S28. HMBC NMR spectrum (CDCl_3 - 600 MHz) of compound F

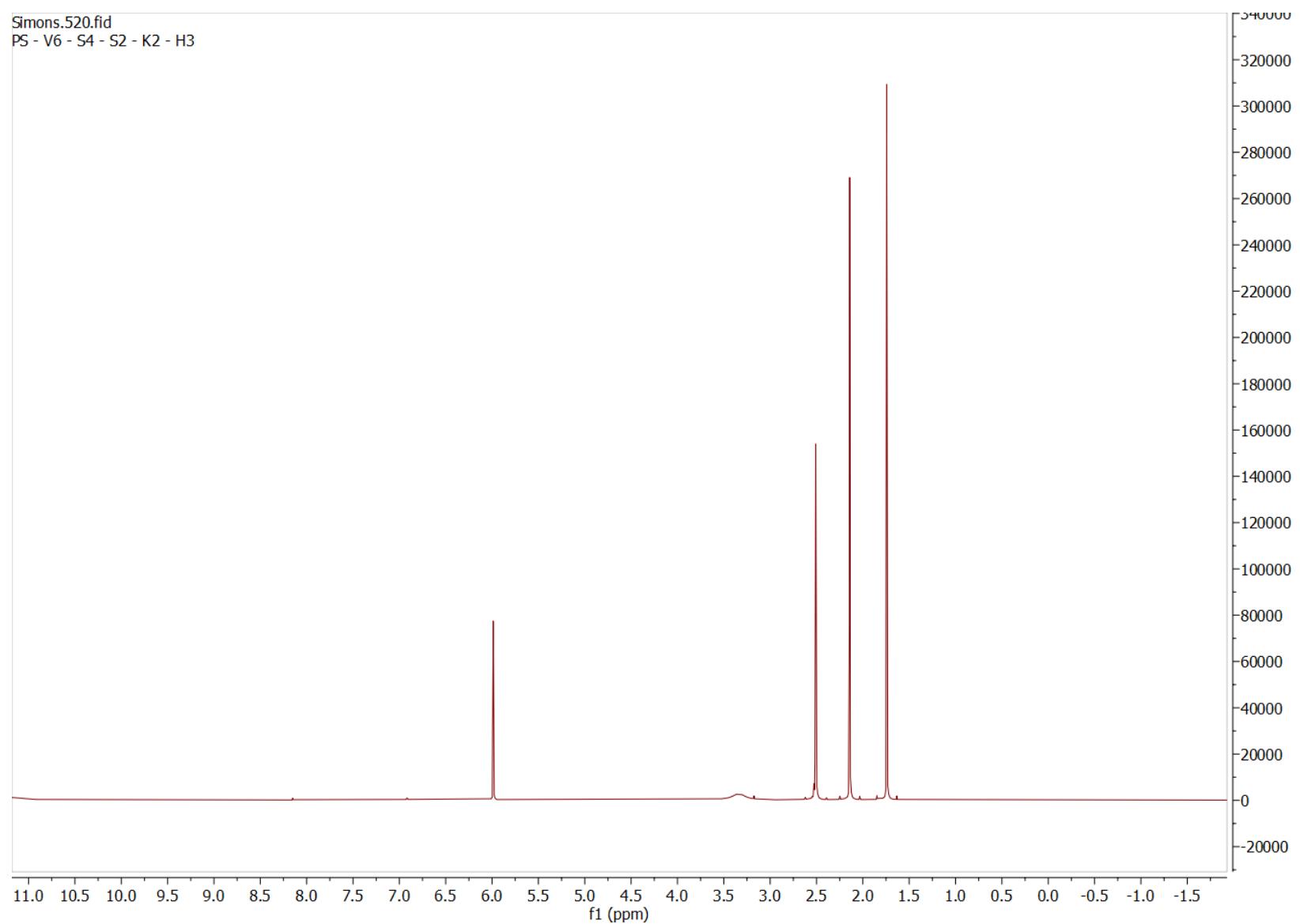


Figure S29. ¹H NMR spectrum (DMSO-d₆ – 600 MHz) of compound G

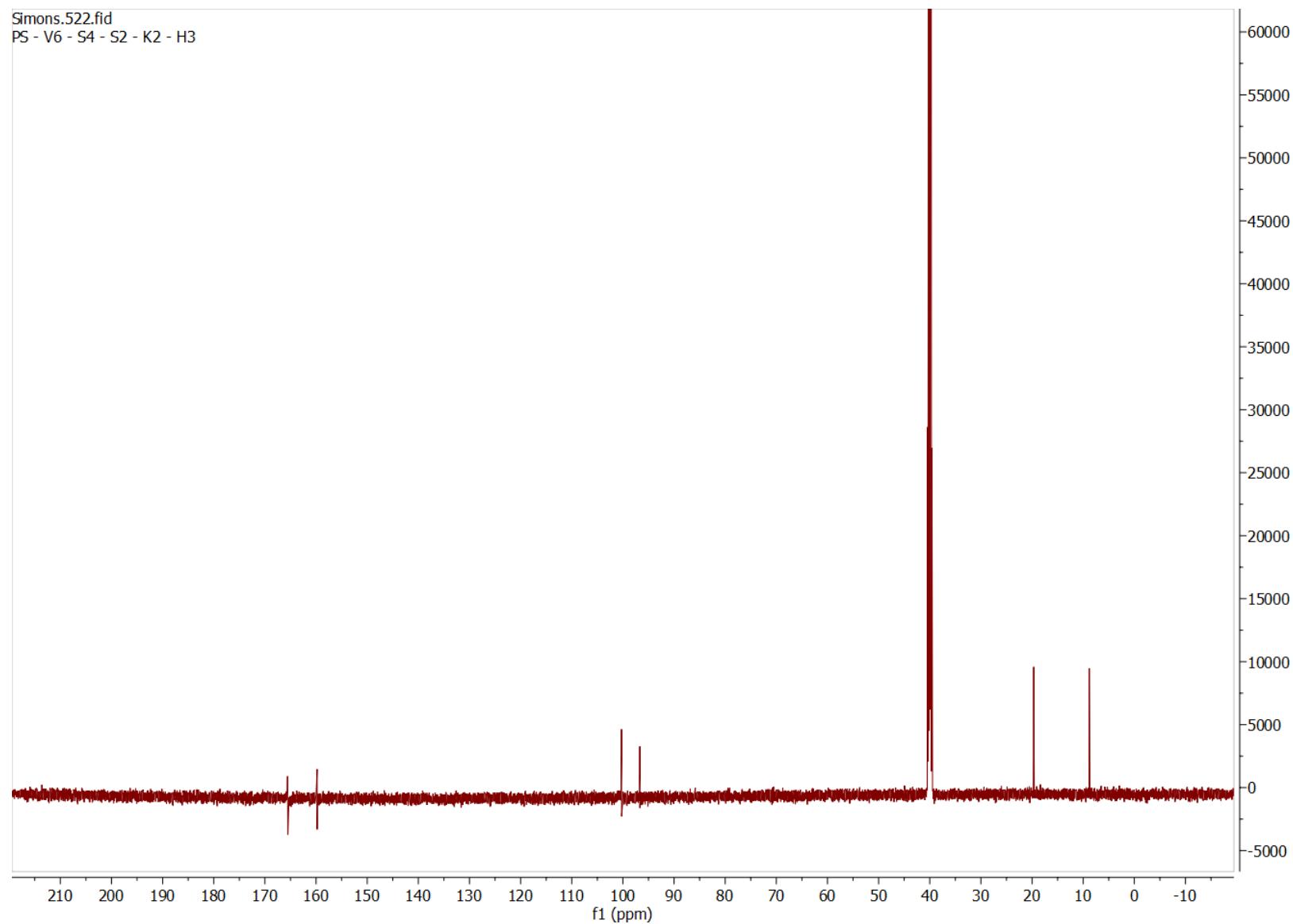


Figure S30. ¹³C NMR spectrum (DMSO-d₆ – 150 MHz) of compound G

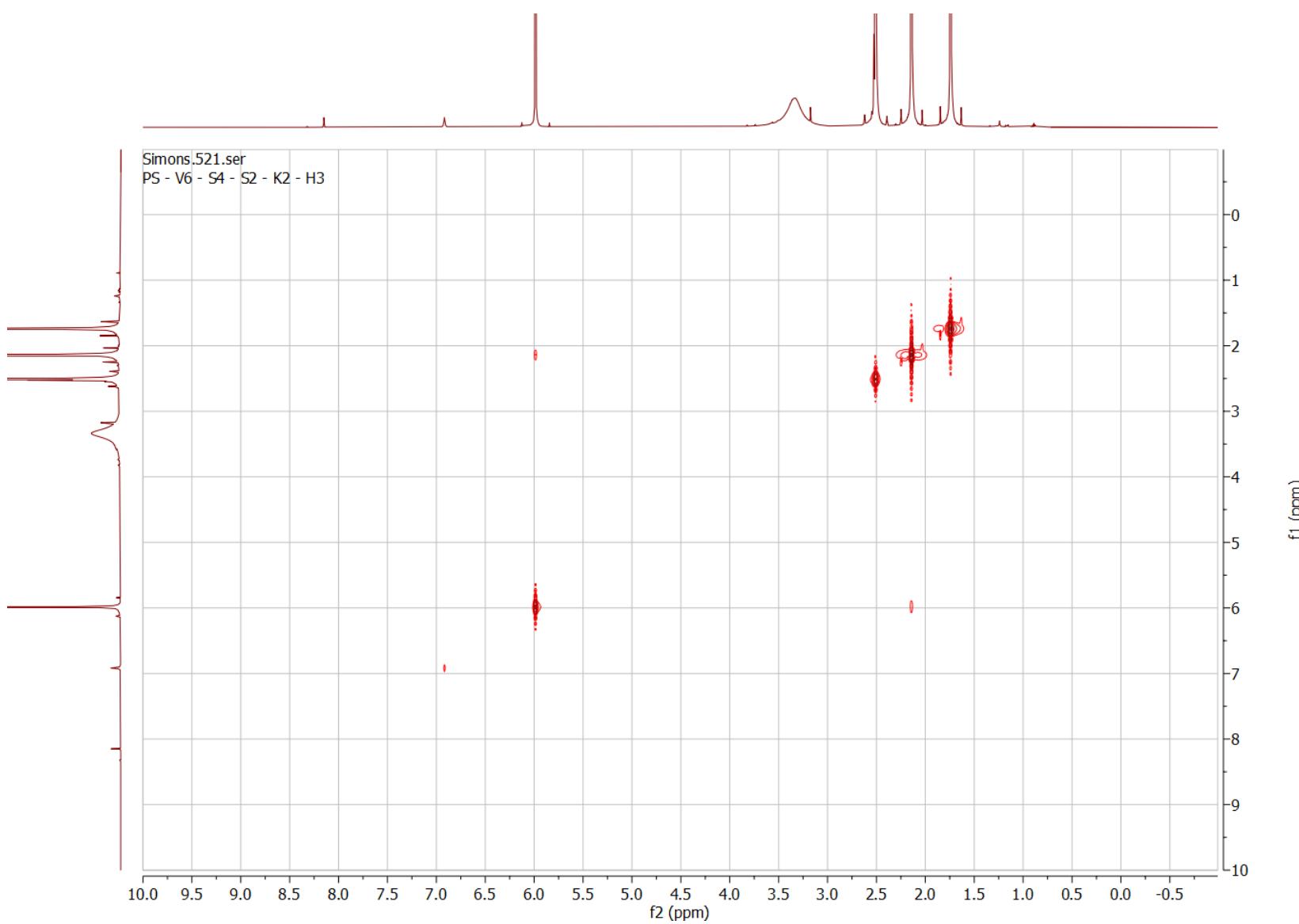


Figure S31. COSY NMR spectrum (DMSO-d₆ – 600 MHz) of compound G

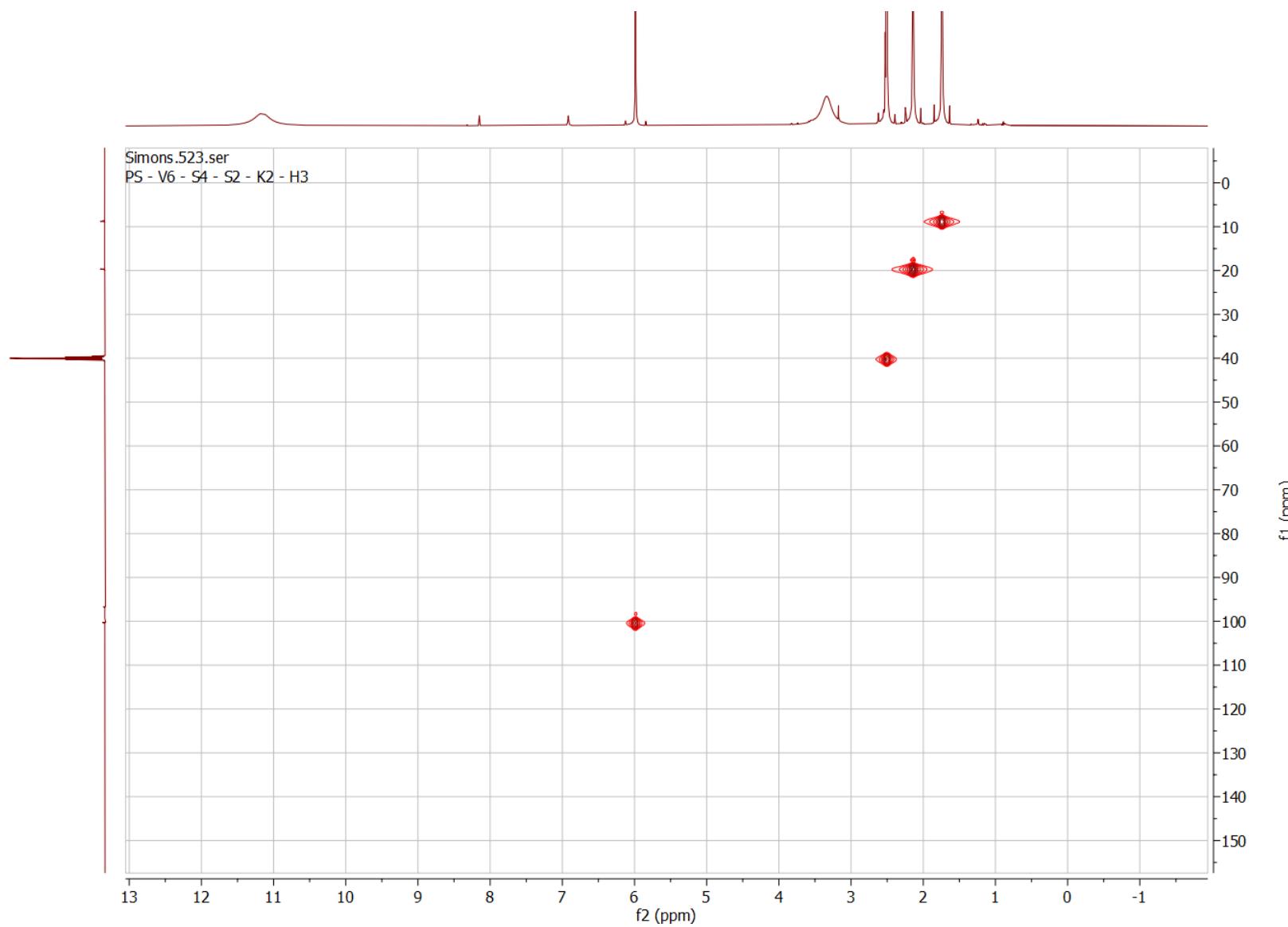


Figure S32. HSQC NMR spectrum (DMSO-d₆ – 600 MHz) of compound G

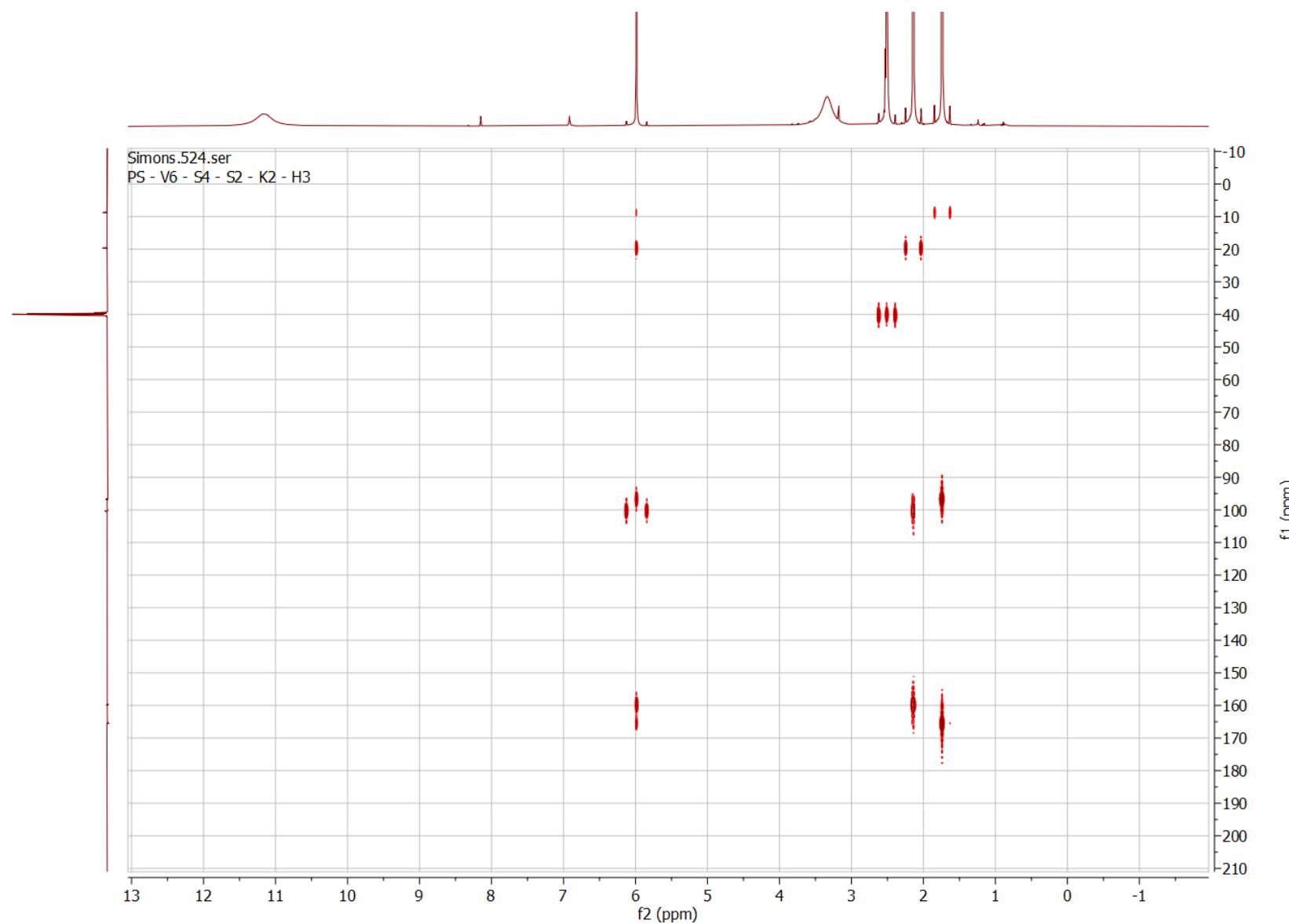


Figure S33. HMBC NMR spectrum (DMSO-d₆ – 600 MHz) of compound G

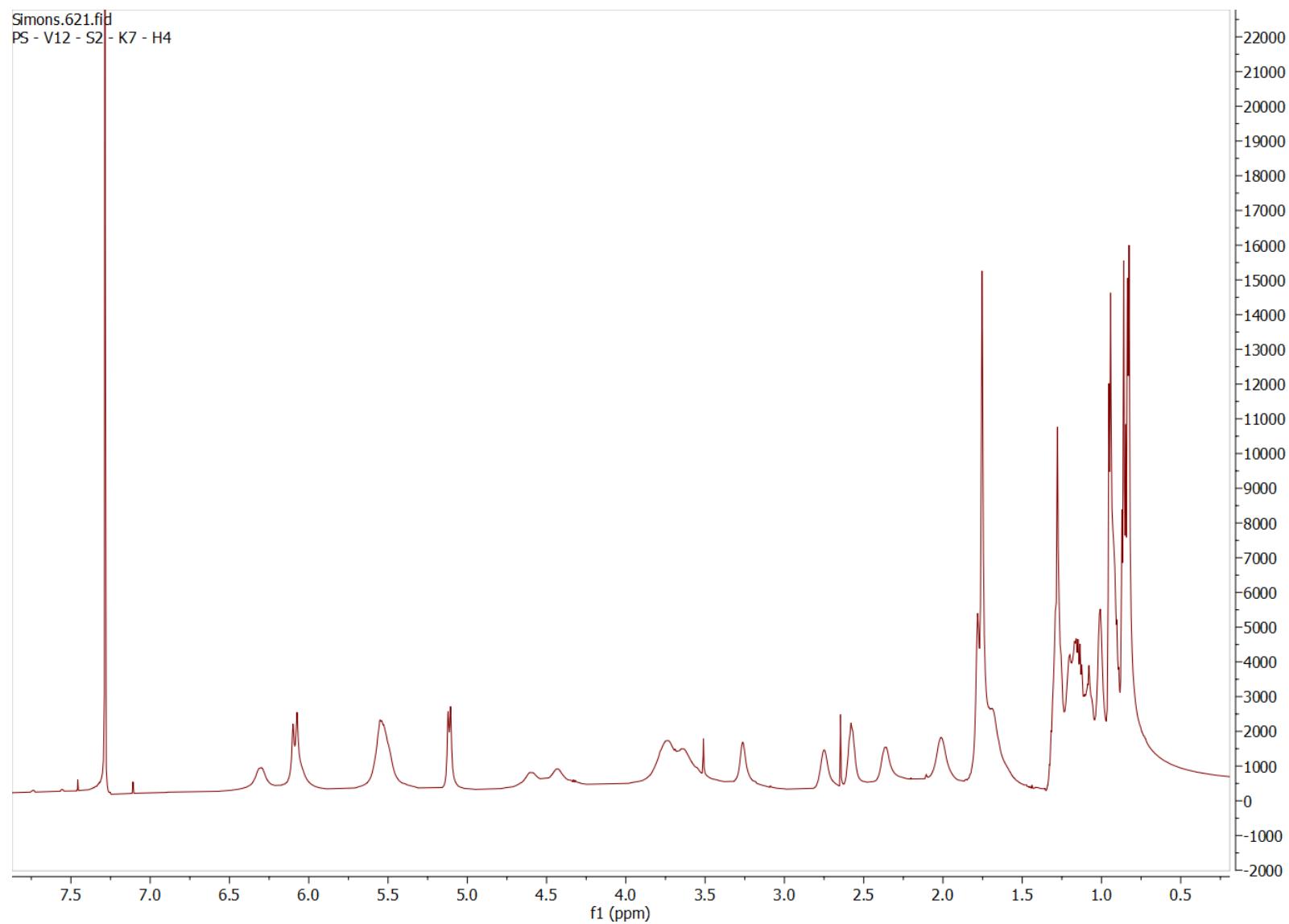
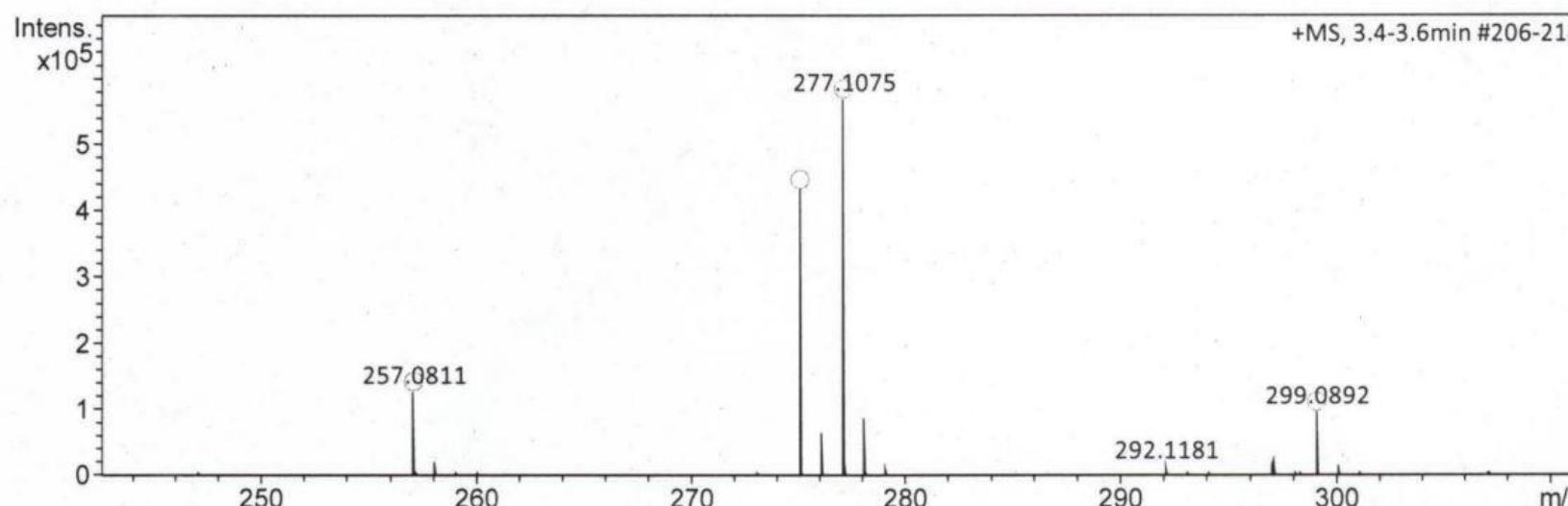


Figure S34. ^1H NMR spectrum (CDCl_3 - 600 MHz) of compound H

Acquisition Parameter

Source Type	ESI	Ion Polarity	Positive	Set Nebulizer	0.3 Bar
Focus	Not active	Set Capillary	4000 V	Set Dry Heater	180 °C
Scan Begin	50 m/z	Set End Plate Offset	-500 V	Set Dry Gas	4.0 l/min
Scan End	1500 m/z	Set Collision Cell RF	600.0 Vpp	Set Divert Valve	Source

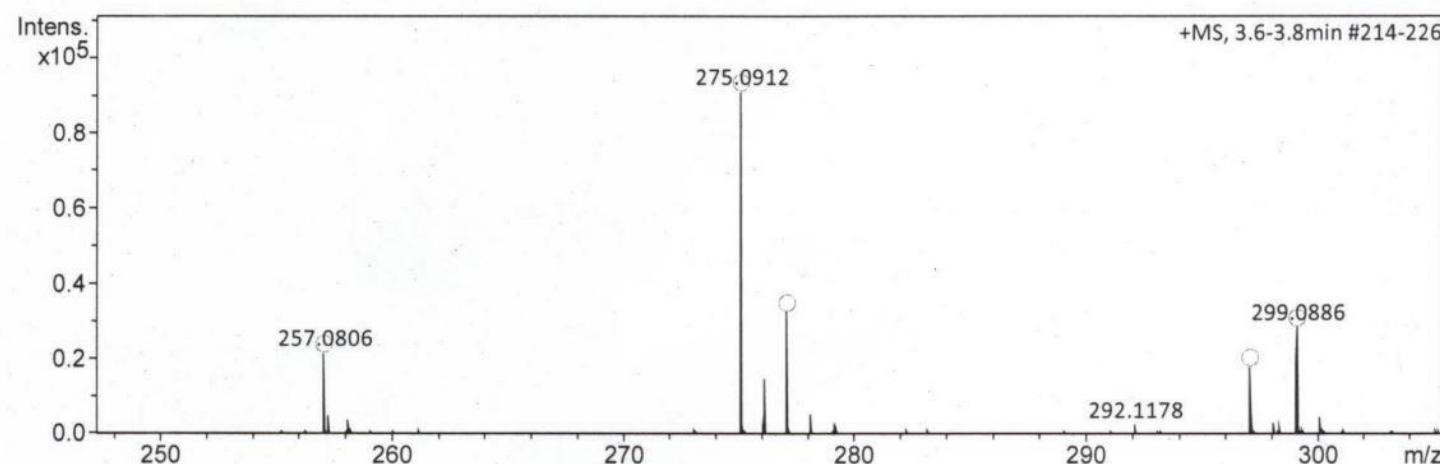


Meas. m/z	#	Ion Formula	m/z	err [ppm]	mSigma	# mSigma	Score	rdb	e⁻ Conf	N-Rule
257.0811	1	C15H13O4	257.0808	-1.1	7.2	1	100.00	9.5	even	ok
275.0919	1	C15H15O5	275.0914	-1.8	581.8	1	100.00	8.5	even	ok
277.1075	1	C15H17O5	277.1071	-1.6	8.4	1	100.00	7.5	even	ok
299.0892	1	C15H16NaO5	299.0890	-0.8	0.6	1	100.00	7.5	even	ok

Figure S35. HRESIMS (MeOH) of compound A

Acquisition Parameter

Source Type	ESI	Ion Polarity	Positive	Set Nebulizer	0.3 Bar
Focus	Not active	Set Capillary	4000 V	Set Dry Heater	180 °C
Scan Begin	50 m/z	Set End Plate Offset	-500 V	Set Dry Gas	4.0 l/min
Scan End	1500 m/z	Set Collision Cell RF	600.0 Vpp	Set Divert Valve	Source



Meas. m/z	#	Ion Formula	m/z	err [ppm]	mSigma	# mSigma	Score	rdb	e⁻ Conf	N-Rule
257.0806	1	C15H13O4	257.0808	1.1	5.7	1	100.00	9.5	even	ok
275.0912	1	C15H15O5	275.0914	0.7	192.5	1	100.00	8.5	even	ok
277.1065	1	C15H17O5	277.1071	2.1	4.8	1	98.95	7.5	even	ok
	2	C14H14N4NaO	277.1060	-1.8	6.5	2	100.00	9.5	even	ok
297.0730	1	C13H9N6O3	297.0731	0.3	11.1	1	100.00	12.5	even	ok
	2	C15H14NaO5	297.0733	1.2	602.8	2	0.00	8.5	even	ok
299.0886	1	C13H11N6O3	299.0887	0.4	3.4	1	100.00	11.5	even	ok
	2	C15H16NaO5	299.0890	1.3	4.2	2	87.30	7.5	even	ok

Figure S36. HRESIMS (MeOH) of compound B

Acquisition Parameter

Source Type	ESI	Ion Polarity	Positive	Set Nebulizer	0.3 Bar
Focus	Not active	Set Capillary	4000 V	Set Dry Heater	180 °C
Scan Begin	50 m/z	Set End Plate Offset	-500 V	Set Dry Gas	4.0 l/min
Scan End	1500 m/z	Set Collision Cell RF	600.0 Vpp	Set Divert Valve	Source

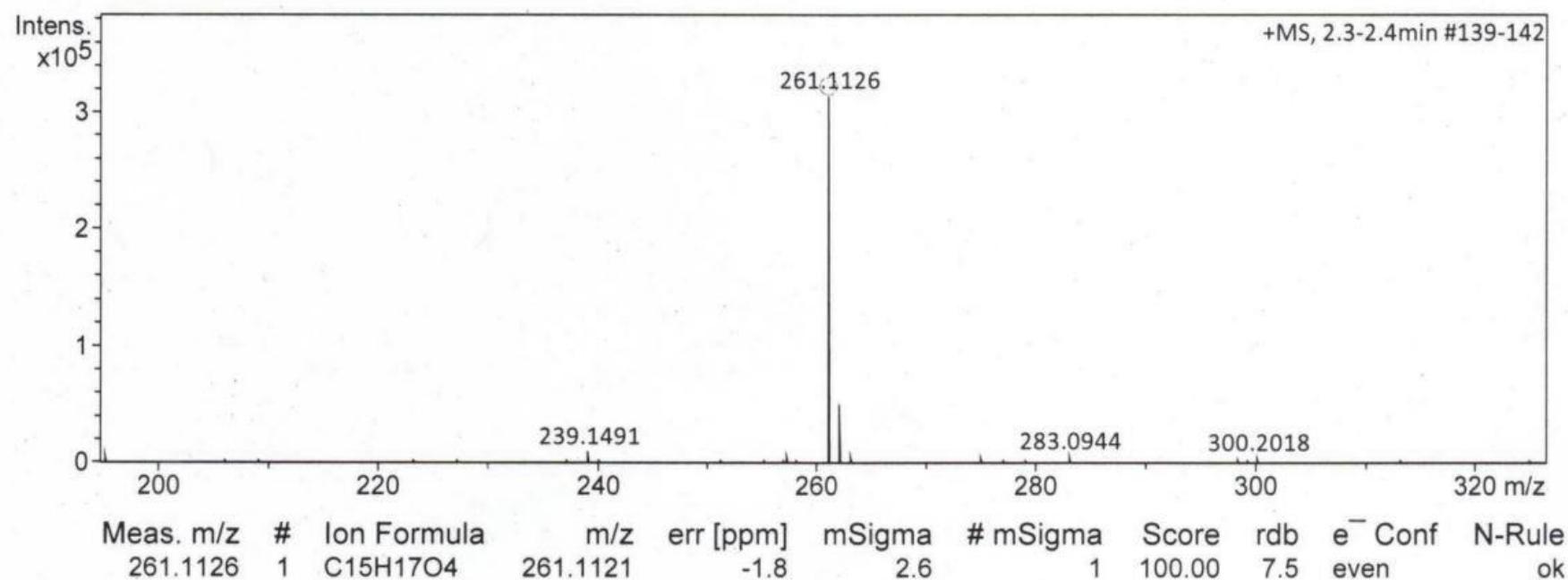
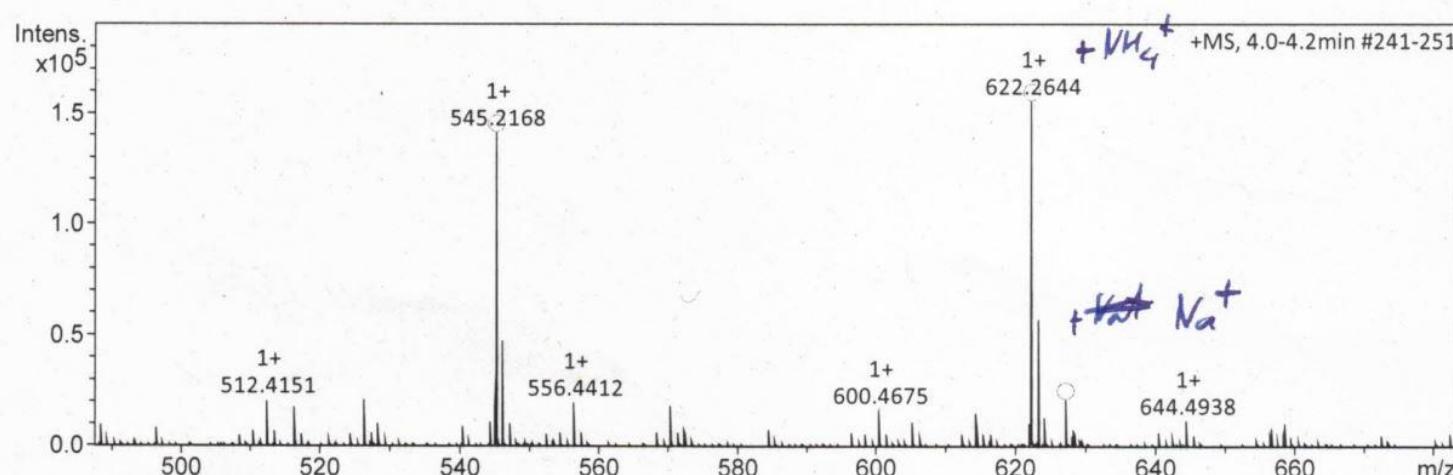


Figure S37. HRESIMS (MeOH) of compound C

Acquisition Parameter

Source Type	ESI	Ion Polarity	Positive	Set Nebulizer	0.3 Bar
Focus	Not active	Set Capillary	4000 V	Set Dry Heater	180 °C
Scan Begin	50 m/z	Set End Plate Offset	-500 V	Set Dry Gas	4.0 l/min
Scan End	1500 m/z	Set Collision Cell RF	600.0 Vpp	Set Divert Valve	Source



Meas. m/z	#	Ion Formula	m/z	err [ppm]	mSigma	# mSigma	Score	rdb	e ⁻ Conf	N-Rule
545.2168	1	C32H33O8	545.2170	0.4	8.6	1	100.00	16.5	even	ok
	2	C29H25N10O2	545.2156	-2.1	10.0	2	60.03	22.5	even	ok
	3	C33H29N4O4	545.2183	2.8	20.2	3	38.17	21.5	even	ok
622.2644	1	C34H40NO10	622.2647	0.4	6.3	1	100.00	15.5	even	ok
	2	C31H32N11O4	622.2633	-1.8	7.7	2	63.41	21.5	even	ok
	3	C35H36N5O6	622.2660	2.6	17.9	3	38.03	20.5	even	ok
	4	C32H28N15	622.2647	0.4	18.6	4	79.33	26.5	even	ok
627.2193	1	C34H36NaO10	627.2201	1.2	11.0	1	100.00	16.5	even	ok
	2	C30H32N6NaO8	627.2174	-3.1	11.3	2	45.90	17.5	even	ok
	3	C31H28N10NaO4	627.2187	-1.0	14.8	3	98.78	22.5	even	ok
	4	C32H24N14Na	627.2201	1.2	24.2	4	76.55	27.5	even	ok

Figure S38. HRESIMS (MeOH) of compound D

Acquisition Parameter

Source Type	ESI	Ion Polarity	Positive	Set Nebulizer	0.3 Bar
Focus	Not active	Set Capillary	4000 V	Set Dry Heater	180 °C
Scan Begin	50 m/z	Set End Plate Offset	-500 V	Set Dry Gas	4.0 l/min
Scan End	1500 m/z	Set Collision Cell RF	600.0 Vpp	Set Divert Valve	Source

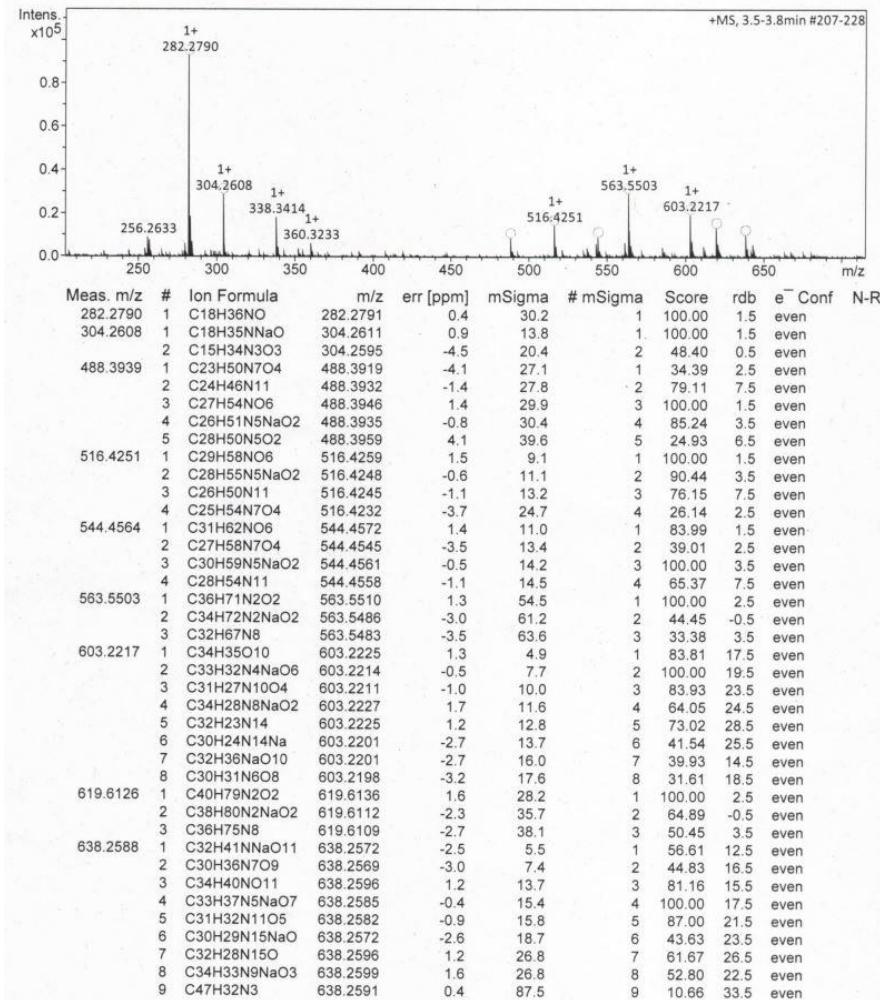


Figure S39. HRESIMS (MeOH) of compound E

Acquisition Parameter

Source Type	ESI	Ion Polarity	Positive	Set Nebulizer	0.3 Bar
Focus	Not active	Set Capillary	4000 V	Set Dry Heater	180 °C
Scan Begin	50 m/z	Set End Plate Offset	-500 V	Set Dry Gas	4.0 l/min
Scan End	1500 m/z	Set Collision Cell RF	600.0 Vpp	Set Divert Valve	Source

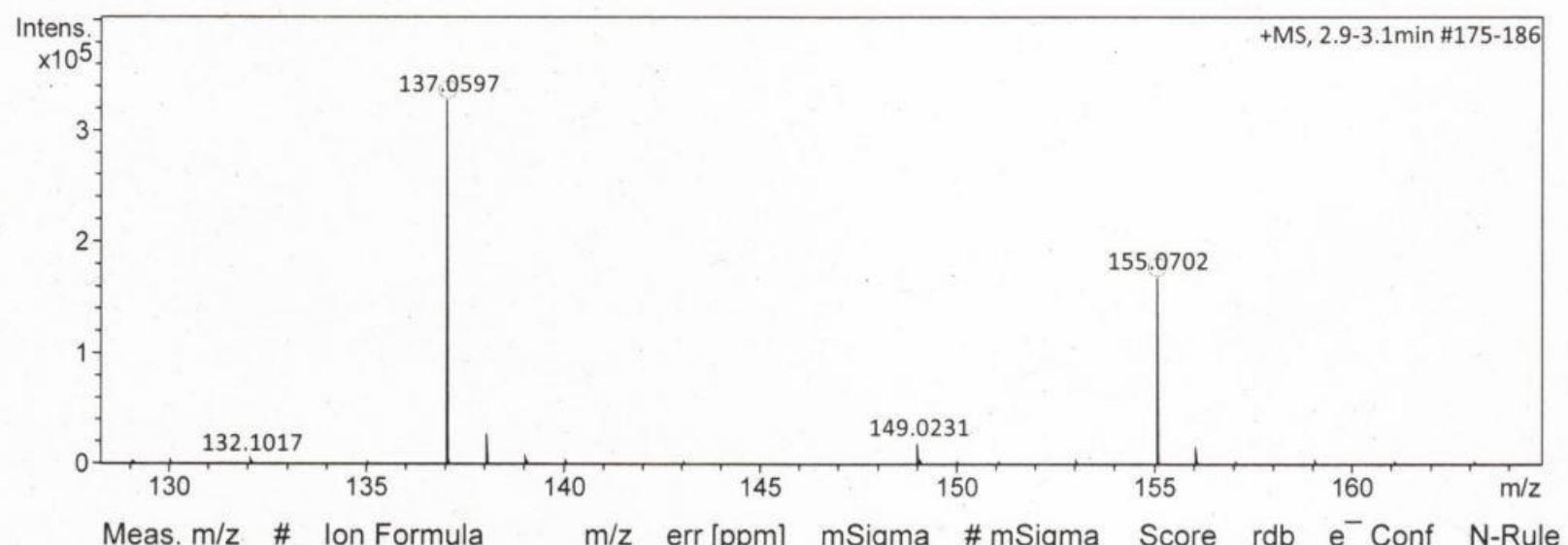


Figure S40. HRESIMS (MeOH) of compound F

Acquisition Parameter

Source Type	ESI	Ion Polarity	Positive	Set Nebulizer	0.3 Bar
Focus	Not active	Set Capillary	4000 V	Set Dry Heater	180 °C
Scan Begin	50 m/z	Set End Plate Offset	-500 V	Set Dry Gas	4.0 l/min
Scan End	1500 m/z	Set Collision Cell RF	600.0 Vpp	Set Divert Valve	Source

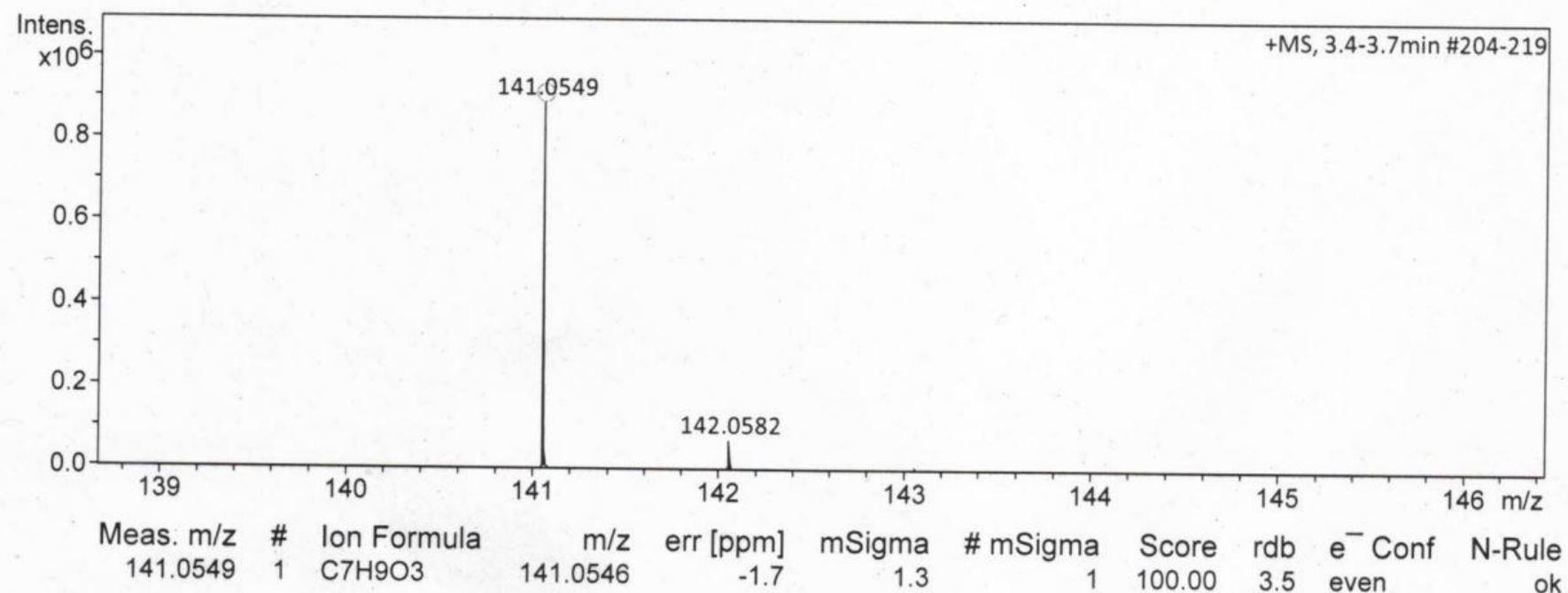


Figure S41. HRESIMS (MeOH) of compound G

Acquisition Parameter

Source Type	ESI	Ion Polarity	Positive	Set Nebulizer	0.3 Bar
Focus	Not active	Set Capillary	4000 V	Set Dry Heater	180 °C
Scan Begin	50 m/z	Set End Plate Offset	-500 V	Set Dry Gas	4.0 l/min
Scan End	1500 m/z	Set Collision Cell RF	600.0 Vpp	Set Divert Valve	Source

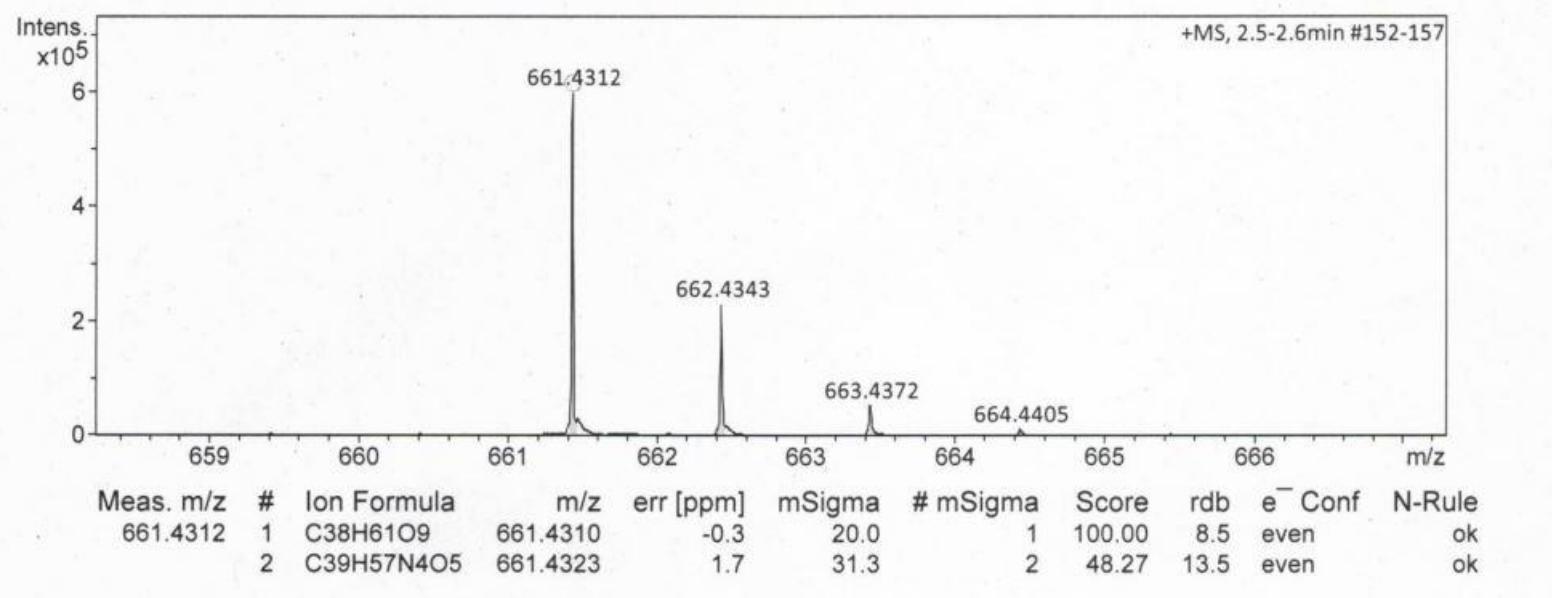


Figure S42. HRESIMS (MeOH) of compound H

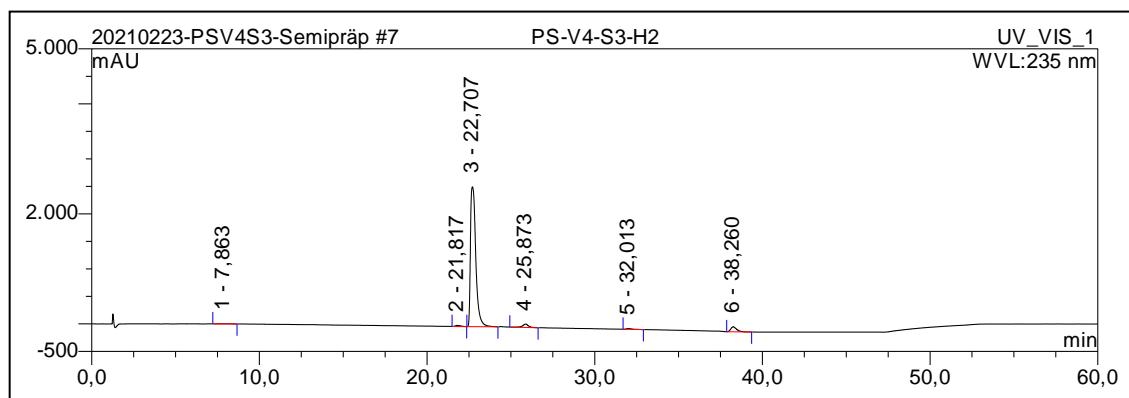


Figure S43. HPLC chromatogram (MeOH) of compound A

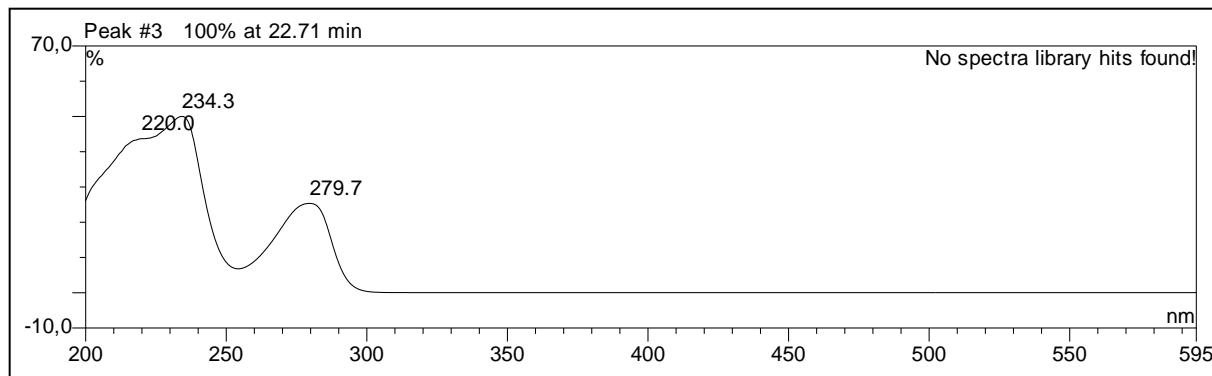


Figure S44. UV-Vis spectrum (MeOH) of compound A

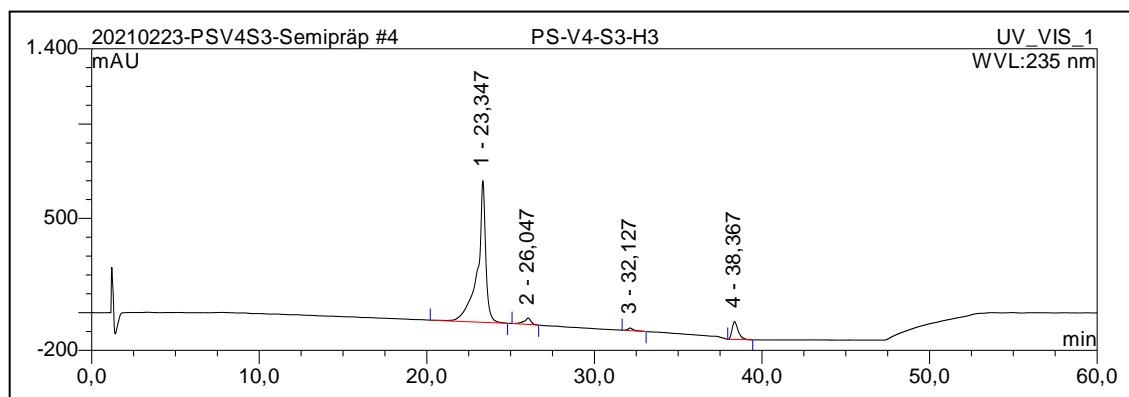


Figure S45. HPLC chromatogram (MeOH) of compound B

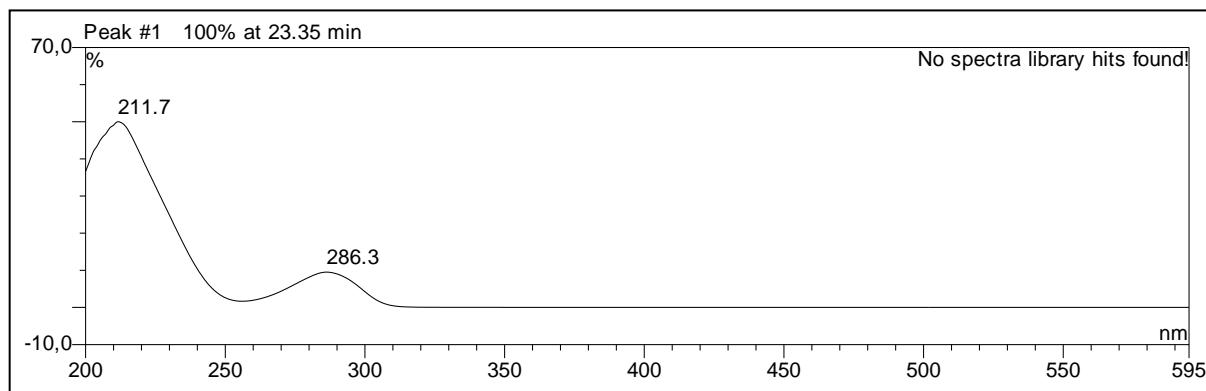


Figure S46. UV-Vis spectrum (MeOH) of compound B

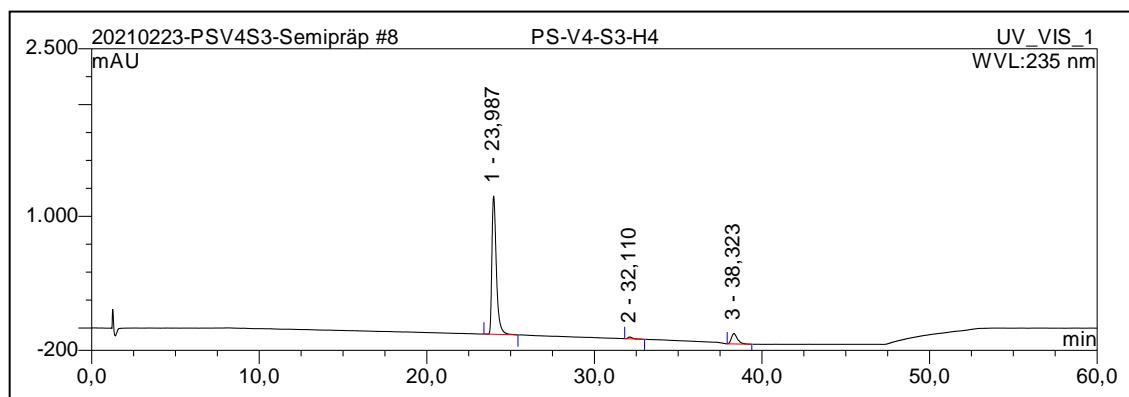


Figure S47. HPLC chromatogram (MeOH) of compound C

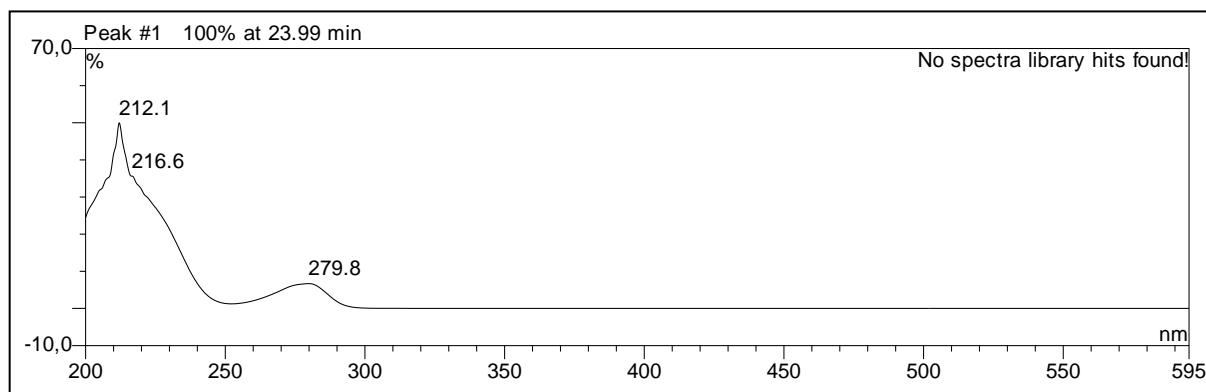


Figure S48. UV-Vis spectrum (MeOH) of compound C

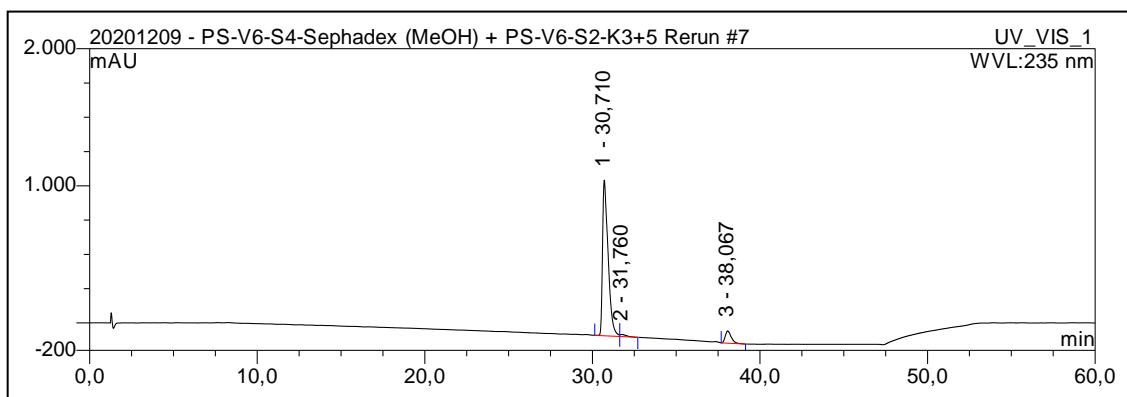


Figure S49. HPLC chromatogram (MeOH) of compound D

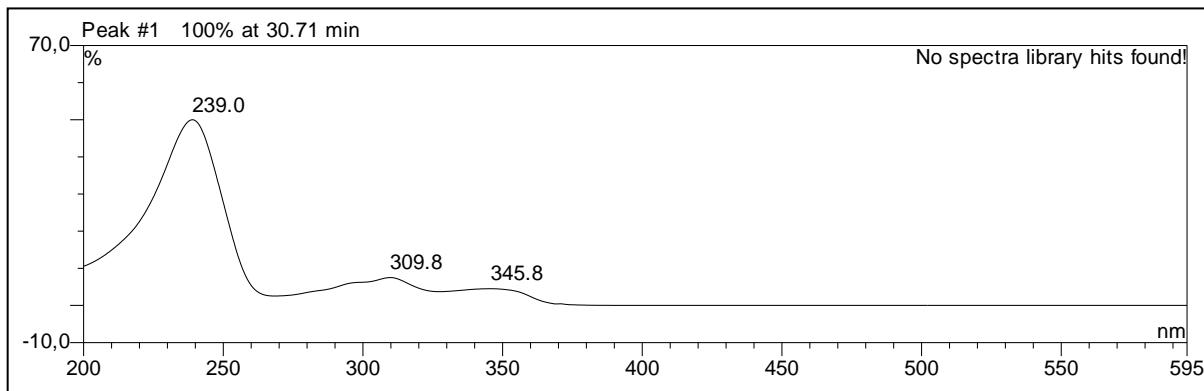


Figure S50. UV-Vis spectrum (MeOH) of compound D

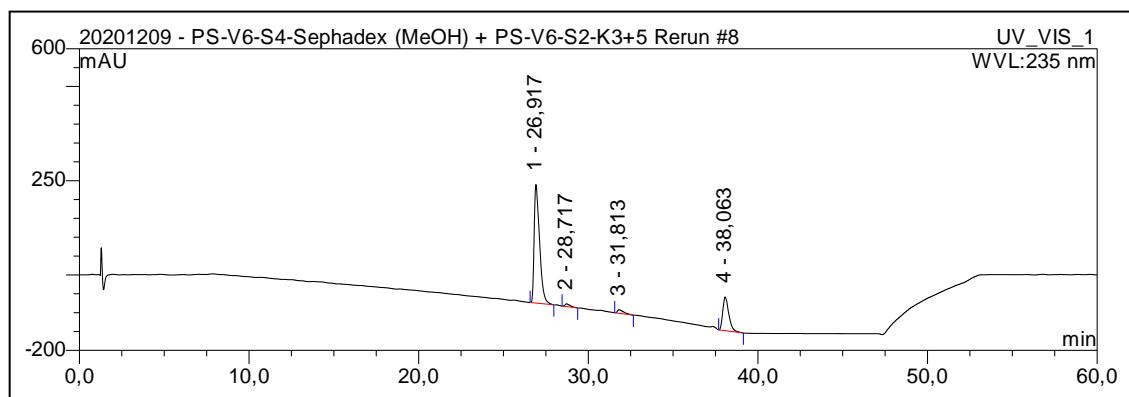


Figure S51. HPLC chromatogram (MeOH) of compound E

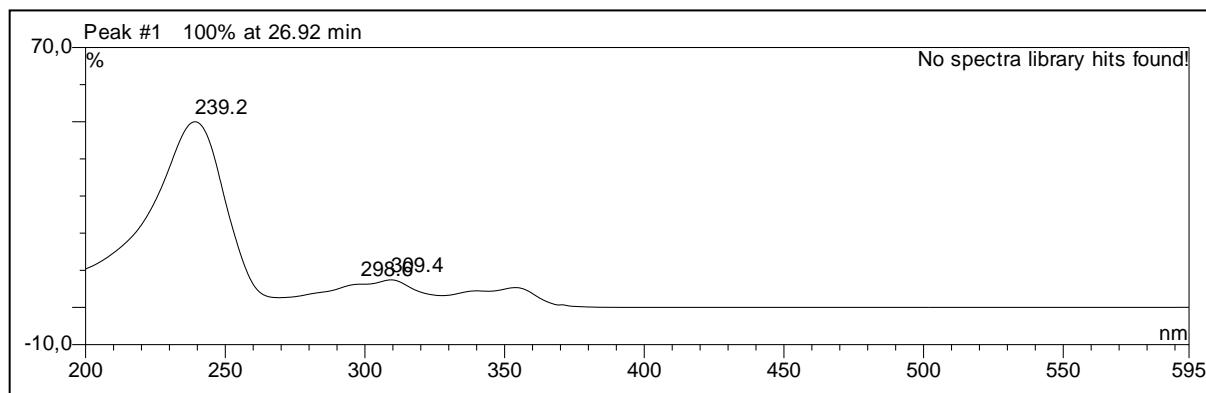


Figure S52. UV-Vis spectrum (MeOH) of compound E

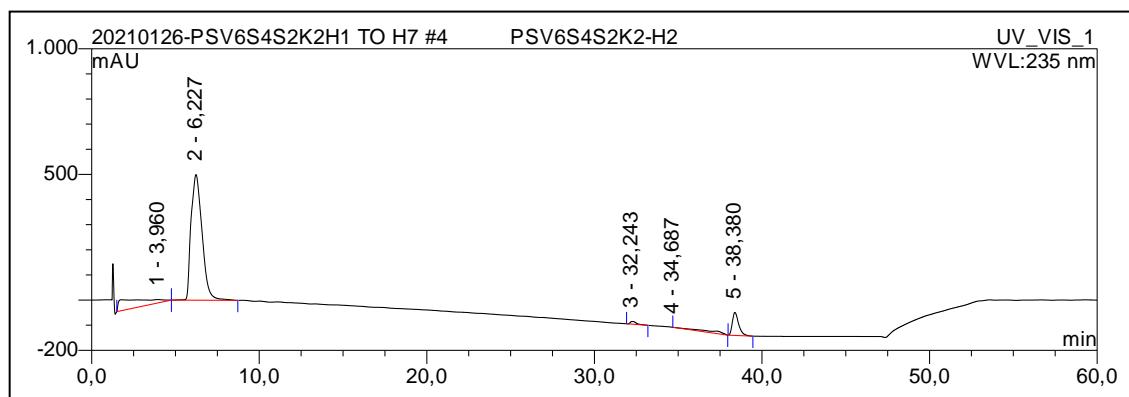


Figure S53. HPLC chromatogram (MeOH) of compound F

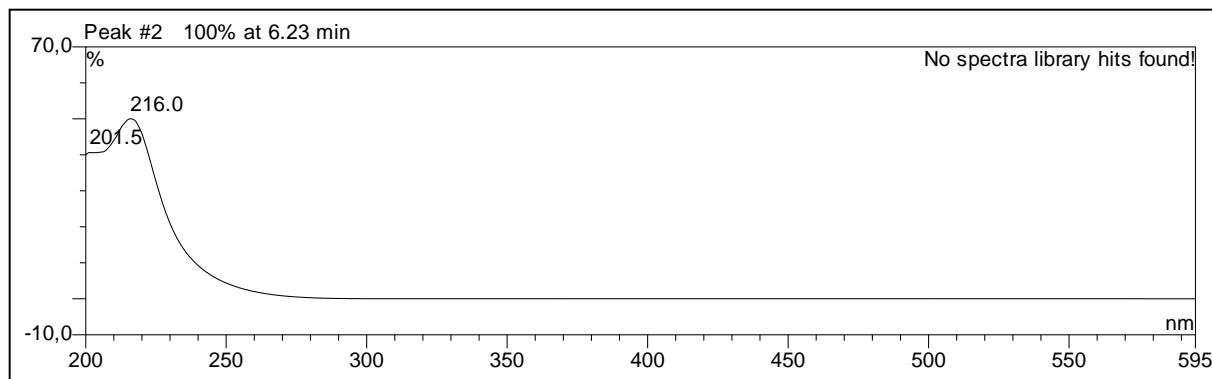


Figure S54. UV-Vis spectrum (MeOH) of compound F

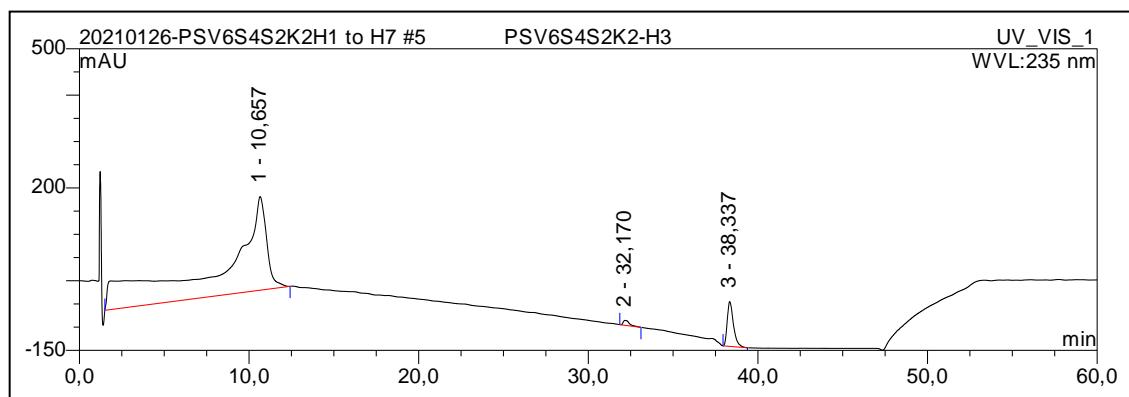


Figure S55. HPLC chromatogram (MeOH) of compound G

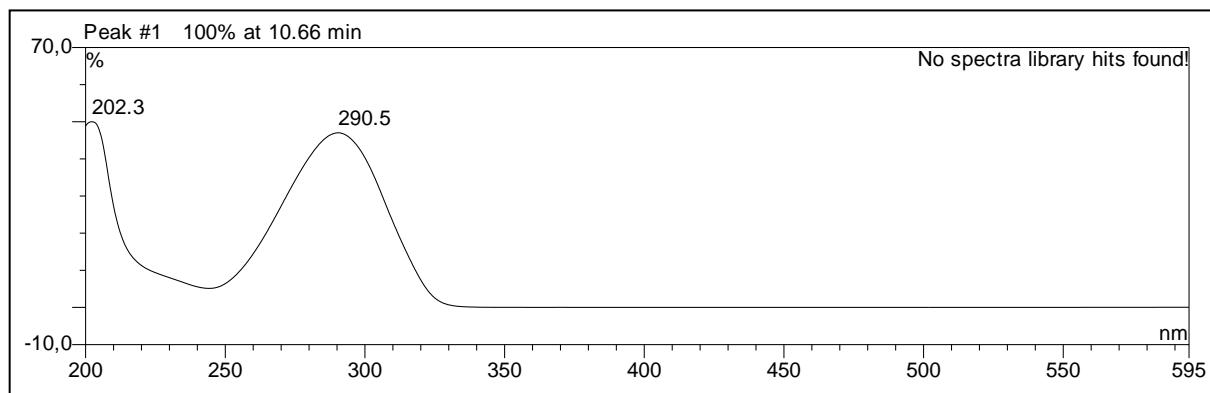


Figure S56. UV-Vis spectrum (MeOH) of compound G

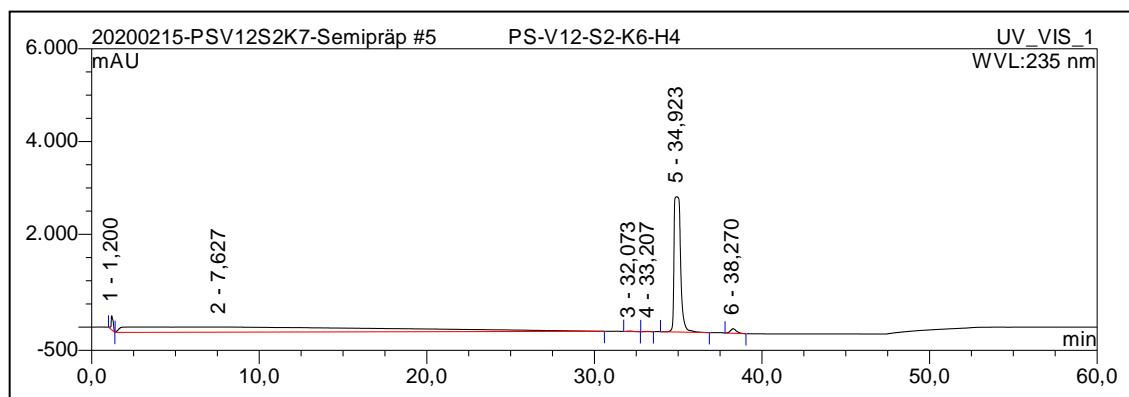


Figure S57- HPLC-chromatogram (MeOH) of compound H

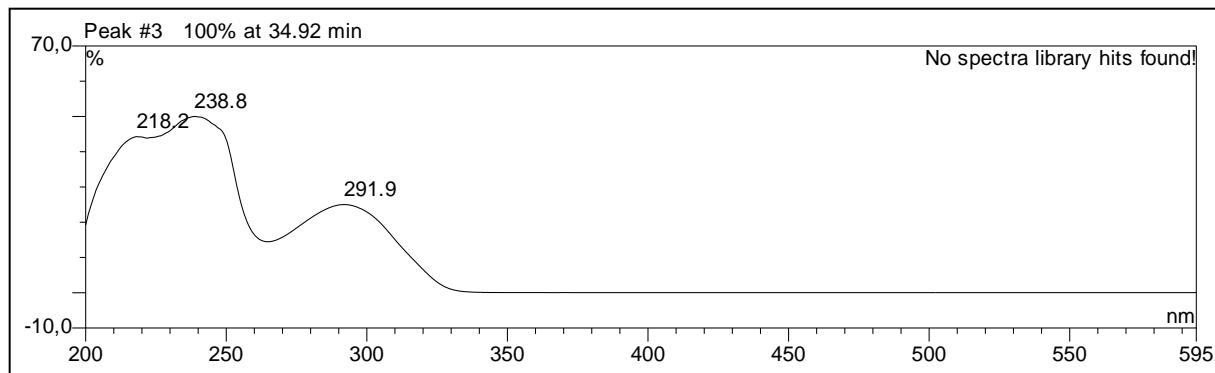


Figure S58. UV-Vis spectrum (MeOH) of compound H

Stereochemical studies of F

The ROESY spectrum of F showed correlation between the 5-H and 6-H protons of the adjacent stereogenic centers indicating their *cis* relative configuration. Therefore the conformational analysis and chiroptical calculations were performed on the arbitrarily selected *cis*-(5*S*,6*S*) stereoisomer. The Merck molecular force field (MMFF) conformational search produced 16 conformers in a 21 kJ/mol energy window which were re-optimized at the ω B97X/TZVP PCM/MeCN and the B3LYP/TZVP PCM/CHCl₃ levels, separately, yielding 10 and 8 low-energy conformers over 1% Boltzmann population, respectively (Figures S60 and S61). In the low-energy computed conformers, the 5,6-dihydro- α -pyrone moiety adopted a conformation, which oriented the C-5 hydroxyl group to axial and the C-6 prop-1-en-1-yl substituent to equatorial position. ECD spectra computed at four levels for the ω B97X conformers gave acceptable agreement with the experimental ECD spectrum (Figure S59) with CAM-B3LYP/TZVP giving the best agreement. Optical rotation calculations performed for the same conformers at four levels and PCM solvent model for EtOH reproduced the large experimental positive value $[\alpha]^{24D} +172$ in the range of +81 – +99 (Table S1). VCD spectra computed at the B3LYP/TZVP PCM/CHCl₃ level for the same level optimized conformers gave good agreement with the experimental VCD spectrum (Figure X3). That is, all three applied chiroptical methods suggested (5*S*,6*S*) absolute configuration for F allowing elucidation of the absolute configuration as (5*S*,6*S*).

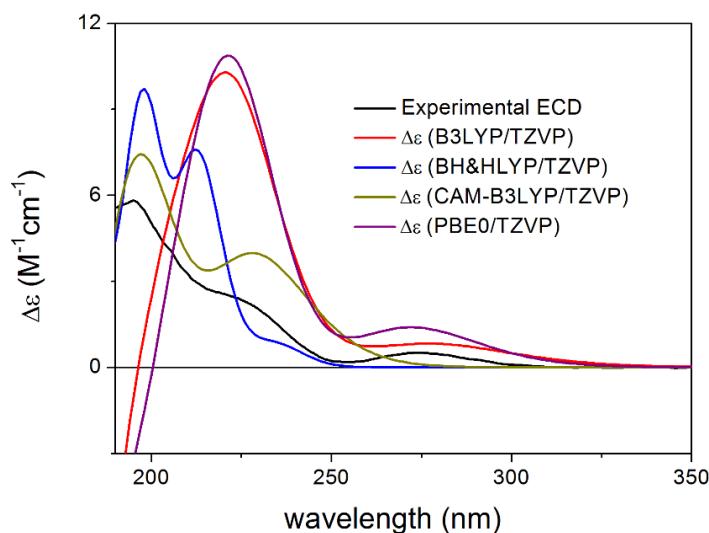


Figure S59. Comparison of the experimental ECD spectrum of F measured in MeCN and the calculated ECD spectra of (5*S*,6*S*)-F computed at various levels of theory for the 10 lowest-energy ω B97X/TZVP PCM/MeCN conformers. Black: experimental, red: B3LYP/TZVP PCM/MeCN with half-height width 4200 cm⁻¹, blue:

BH&HLYP/TZVP PCM/MeCN with half-height width 2100 cm⁻¹, olive: CAM-B3LYP/TZVP PCM/MeCN with half-height width 4200 cm⁻¹, purple: PBE0/TZVP PCM/MeCN with half-height width 4200 cm⁻¹.

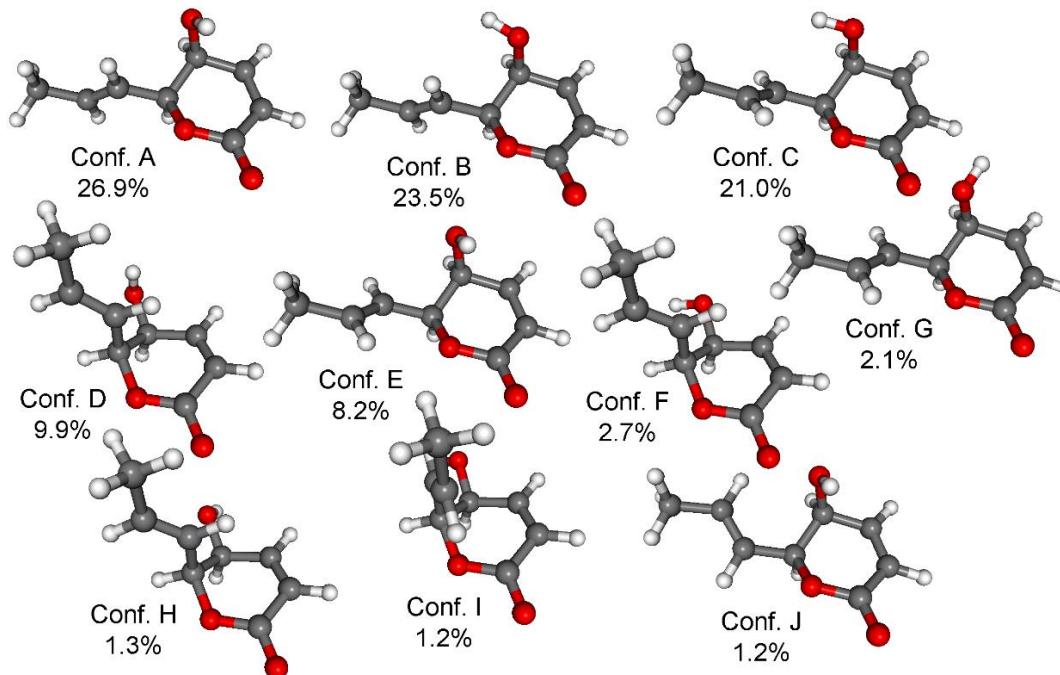


Figure S60. Geometries of the low-energy ω B97X/TZVP PCM/MeCN conformers of (5S,6S)-F.

Table S1. Boltzmann populations and specific optical rotations of the low-energy conformers of (5S,6S)-F computed at various levels for the low-energy ω B97X conformers.

Conformer	Boltzmann population	B3LYP/TZVP PCM/EtOH	BH&HLYP/TZVP PCM/EtOH	CAM-B3LYP/TZVP PCM/EtOH	PBE0/TZVP PCM/EtOH
Conf. A	26.93%	134.70	98.20	104.21	130.74
Conf. B	23.55%	29.58	5.44	9.46	29.68
Conf. C	20.95%	214.32	199.47	197.39	211.76

Conf. D	9.90%	-104.45	-80.52	-97.75	-93.21
Conf. E	8.24%	211.30	192.34	193.43	209.37
Conf. F	2.73%	-128.11	-87.40	-111.23	-111.39
Conf. G	2.09%	228.43	189.15	199.16	223.73
Conf. H	1.32%	-56.67	-26.86	-45.34	-45.81
Conf. I	1.24%	-155.48	-96.97	-122.07	-146.94
Conf. J	1.19%	263.77	211.06	223.80	260.09
Average	N/A	98.79	81.43	81.10	98.73

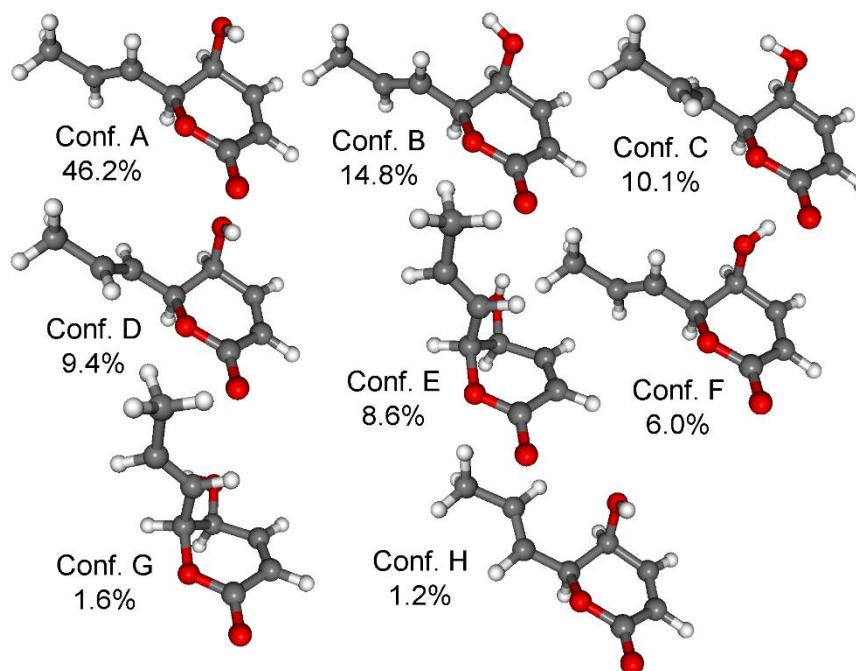


Figure S61. Geometries of the low-energy B3LYP/TZVP PCM/CHCl₃ conformers of (5*S*,6*S*)-F.

Computational details

Mixed torsional/low-frequency mode conformational searches were carried out by means of the Macromodel 10.8.011 software by using the MMFF with an implicit solvent model for CHCl₃ [1]. Geometry re-optimizations were carried out at the ωB97X/TZVP level with the PCM solvent model for MeCN and the B3LYP/TZVP level with PCM solvent model for CHCl₃. TDDFT-ECD and OR calculations were run with various functionals (B3LYP, BH&HLYP, CAMB3LYP, and PBE0) and the TZVP basis set as implemented in the Gaussian 09 package with the PCM/MeCN and the PCM/EtOH solvent models, respectively [2]. ECD spectra were generated as sums of Gaussians with 2100–4200 cm⁻¹ width at half-height, using dipole-velocity-computed rotational strength values [3]. VCD spectra were generated with 8 cm⁻¹ half-height width and scaled by a factor of 0.99. Boltzmann distributions were estimated from the ωB97X and the B3LYP energies. The Molekel software package was used for visualization of the results [4].

References

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