

Article

Combining Multiconfiguration and Perturbation Methods: Perturbative Estimates of Core–Core Electron Correlation Contributions to Excitation Energies in Mg-Like Iron

Stefan Gustafsson ¹, Per Jönsson ^{1,*}, Charlotte Froese Fischer ² and Ian Grant ^{3,4}

¹ Materials Science and Applied Mathematics, Malmö University, SE-205 06 Malmö, Sweden; stefan.gustafsson@mah.se

² Department of Computer Science, University of British Columbia, Vancouver, BC V6T 1Z4, Canada; charlotte.f.fischer@comcast.net

³ Mathematical Institute, University of Oxford, Woodstock Road, Oxford OX2 6GG, UK; iangrant15@btinternet.com

⁴ Department of Applied Mathematics and Theoretical Physics, Centre for Mathematical Sciences, Wilberforce Road, Cambridge CB3 0WA, UK

* Correspondence: per.jonsson@mah.se; Tel.: +46-40-66-57251

Academic Editor: Joseph Reader

Received: 25 November 2016; Accepted: 6 January 2017; Published: 12 January 2017

Abstract: Large configuration interaction (CI) calculations can be performed if part of the interaction is treated perturbatively. To evaluate the combined CI and perturbative method, we compute excitation energies for the $3l3l'$, $3l4l'$ and $3s5l$ states in Mg-like iron. Starting from a CI calculation including valence and core–valence correlation effects, it is found that the perturbative inclusion of core–core electron correlation halves the mean relative differences between calculated and observed excitation energies. The effect of the core–core electron correlation is largest for the more excited states. The final relative differences between calculated and observed excitation energies is 0.023%, which is small enough for the calculated energies to be of direct use in line identifications in astrophysical and laboratory spectra.

Keywords: excitation energies; multiconfiguration Dirac–Hartree–Fock; configuration interaction

1. Introduction

Transitions from highly charged ions are observed in the spectra of astrophysical sources as well as in Tokamak and laser-produced plasmas, and they are routinely used for diagnostic purposes [1]. Often, transitions between configurations in the same complex are used, but transitions from higher lying configurations are also important (see, e.g., [2] for a discussion of the higher lying states in the case of Mg-like iron). Transition energies are available from experiments for many ions and collected in various data bases [3], but large amounts of data are still lacking. Although experimental work is aided by a new generation of light sources such as EBITs [4], spectral identifications are still a difficult and time-consuming task. A way forward is provided by theoretical transition energies that support line identification and render consistency checks for experimental level designations.

Much work has been done to improve both multiconfiguration methods and perturbative methods, each with their strengths and weaknesses, in order to provide theoretical transition energies of spectroscopic accuracy, i.e., transition energies with uncertainties of the same order as the ones obtained from experiments and observations using Chandra, Hinode or other space based missions in the X-ray and EUV spectral ranges [5–8]. Further advancements for complex systems with several

electrons outside a closed atomic core calls for a combination of multiconfiguration and perturbative methods [9] and also for methods based on new principles [10,11].

In this paper, we describe how the multiconfiguration Dirac–Hartree–Fock (MCDHF) and relativistic configuration interaction (CI) methods can be modified to include perturbative corrections that account for core–core electron correlation. Taking Mg-like iron as an example, we show how the corrections improve excitation energies for the more highly excited states.

2. Relativistic Multiconfiguration Methods

2.1. Multiconfiguration Dirac–Hartree–Fock and Configuration Interaction

In the MCDHF method [12,13], as implemented in the GRASP2K program package [14], the wave function $\Psi(\gamma P J M_J)$ for a state labeled $\gamma P J M_J$, where J and M_J are the angular quantum numbers and P is the parity, is expanded in antisymmetrized and coupled configuration state functions (CSFs)

$$\Psi(\gamma P J M_J) = \sum_{j=1}^M c_j \Phi(\gamma_j P J M_J). \quad (1)$$

The labels $\{\gamma_j\}$ denote other appropriate information of the configuration state functions, such as orbital occupancy and coupling scheme. The CSFs are built from products of one-electron orbitals, having the general form

$$\psi_{n\kappa,m}(\mathbf{r}) = \frac{1}{r} \begin{pmatrix} P_{n\kappa}(r) \chi_{\kappa,m}(\theta, \varphi) \\ i Q_{n\kappa}(r) \chi_{-\kappa,m}(\theta, \varphi) \end{pmatrix}, \quad (2)$$

where $\chi_{\pm\kappa,m}(\theta, \varphi)$ are 2-component spin-orbit functions. The radial functions $\{P_{n\kappa}(r), Q_{n\kappa}(r)\}$ are numerically represented on a grid.

Wave functions for a number of targeted states are determined simultaneously in the extended optimal level (EOL) scheme. Given initial estimates of the radial functions, the energies E and expansion coefficients $\mathbf{c} = (c_1, \dots, c_M)^t$ for the targeted states are obtained as solutions to the configuration interaction (CI) problem

$$\mathbf{H}\mathbf{c} = E\mathbf{c}, \quad (3)$$

where \mathbf{H} is the CI matrix of dimension $M \times M$ with elements

$$H_{ij} = \langle \Phi(\gamma_i P J M_J) | H | \Phi(\gamma_j P J M_J) \rangle. \quad (4)$$

In relativistic calculations, the Hamiltonian H is often taken as the Dirac–Coulomb Hamiltonian. Once the expansion coefficients have been determined, the radial functions are improved by solving a set of differential equations results from applying the variational principle on a weighted energy functional of the targeted states together with additional terms needed to preserve orthonormality of the orbitals. The CI problem and the solution of the differential equations are iterated until the radial orbitals and the energy are converged to a specified tolerance.

The MCDHF calculations are often followed by CI calculations where terms representing the transverse photon interaction are added to the Dirac–Coulomb Hamiltonian and the vacuum polarization effects are taken into account by including the Uehling potential. Electron self-energies are calculated with the screened hydrogenic formula [12,15]. Due to the relative simplicity of the CI method, often much larger expansions are included in the final CI calculations compared to the MCDHF calculations.

2.2. Large Expansions and Perturbative Corrections

The number of CSFs in the wave function expansions depend on the shell structure of the ionic system as well as the model for electron correlation (to be discussed in Section 3). For accurate calculations, a large number of CSFs are required, leading to very large matrices. To handle these large matrices, the CSFs can a priori be divided into two groups. The first group, P , with m elements ($m \ll M$) contains CSFs that account for the major parts of the wave functions. The second group, Q , with $M - m$ elements contains CSFs that represent minor corrections. Allowing interaction between CSFs in group P , interaction between CSFs in group P and Q and diagonal interactions between CSFs in Q gives a matrix

$$\begin{pmatrix} H^{(PP)} & H^{(PQ)} \\ H^{(QP)} & H^{(QQ)} \end{pmatrix}, \quad (5)$$

where $H_{ij}^{(QQ)} = \delta_{ij}E_i^Q$. The restriction of $H^{(QQ)}$ to diagonal elements results in a huge reduction in the total number of matrix elements and corresponding computational time. The assumptions of the approximation and the connections to the method of deflation in numerical analysis are discussed in [13]. This form of the CI matrix, which has been available in the non-relativistic and relativistic multiconfiguration codes for a long time [16,17], yields energies that are similar to the ones obtained by applying second-order perturbation theory (PT) corrections to the energies of the smaller $m \times m$ matrix. The method is therefore referred to here as CI combined with second-order Brillouin–Wigner perturbation theory [18]. Note, however, that the CI method with restrictions on the interactions gives, in contrast to ordinary perturbative methods, wave functions that can be directly used to evaluate expectation values such as transition rates.

3. Calculations

Calculations were performed for states belonging to the $3s^2$, $3p^2$, $3s3d$, $3d^2$, $3s4s$, $3s4d$, $3p4p$, $3p4f$, $3d4s$, $3d4d$, $3s5s$, $3s5d$, $3s5g$ even configurations and the $3s3p$, $3p3d$, $3s4p$, $3p4s$, $3s4f$, $3p4d$, $3d4p$, $3d4f$, $3s5p$, $3s5f$ odd configurations of Mg-like iron. For $3d4f$, only states below the $3p5s$ configuration were included. The above configurations define the multireference (MR) for the even and odd parities, respectively. Following the procedure in [19], an initial MCDHF calculation for all even and odd reference states was done in the EOL scheme. The initial calculation was followed by separate calculations in the EOL scheme for the even and odd parity states. The MCDHF calculations for the even states were based on CSF expansions obtained by allowing single (S) and double (D) substitutions of orbitals in the even MR configurations to an increasing active set of orbitals. In a similar way, the calculations for the odd states were based on CSF expansions obtained by allowing single (S) and double (D) substitutions of orbitals in the odd MR configurations to an increasing active set of orbitals. To prevent the CSF expansions from growing unmanageably large and in order to obtain orbitals that are spatially localized in the valence and core–valence region, at most, single substitutions were allowed from the $2s^22p^6$ core. The $1s^2$ shell was always closed. The active sets of orbitals for the even and odd parity states were extended by layers to include orbitals with quantum numbers up to $n = 8$ and $l = 6$, at which point the excitation energies are well converged.

To investigate the effects of electron correlation, three sets of CI calculations were done. In the first set of CI calculations, one calculation was done for the even states and one calculation for the odd states, the SD substitutions were only allowed from the valence shells of the MR, and the CSFs account for valence–valence correlation. In the second set of calculations, SD substitutions were such that there was at most one substitution from the $2s^22p^6$ core, and the CSFs account for valence–valence and core–valence correlation. In the final set of calculations, all SD substitutions were allowed, and the CSFs account for valence–valence, core–valence and core–core correlation. When all substitutions are allowed, the number of CSFs grows very large. For this reason, we apply CI with second-order perturbation corrections. The CSFs describing valence–valence and core–valence effects (SD substitutions with at most one substitution from the $2s^22p^6$ core) were included in group P ,

whereas the CSFs accounting for core–core correlation (D substitutions from $2s^2 2p^6$) were included in group Q and treated in second-order perturbation theory. The number of CSFs for the different CI calculations are given in Table 1.

Table 1. Number of CSFs for the even and odd parity expansions for the different sets of CI calculations. VV are the expansions accounting for valence–valence correlation, VV+CV are the expansions accounting for valence–valence and core–valence correlation and VV+CV+CC are the expansions accounting for valence–valence, core–valence and core–core correlation.

	VV	VV+CV	VV+CV+CC
even	2738	644,342	5,624,158
odd	2728	630,502	6,214,393

4. Results

The excitation energies from the different CI calculations, along with observed energies from the NIST database [3], are displayed in Table 2. From the table, we see that states belonging to $3l3l'$, with the exception of $3s3p\ ^3P_{0,1,2}$, are too high for the valence–valence correlation calculation. The states belonging to $3l4l'$ and $3s5l$, on the other hand, are too low. When including also the core–valence correlation, the states belonging to $3l3l'$ go down in energy and approach the observed excitation energies. The states belonging to $3l4l'$ and $3s5l$ go up and are now too high. Including also the core–core correlation results in a rather small energy change for the states belonging to $3l3l'$. The main effect of the core–core correlation is to lower the energies of the states belonging to $3l4l'$ and $3s5l$, bringing them in very good agreement with observations. The labeling of levels is normally done by looking at the quantum designation of the leading component in the CSF expansion [20]. There are two levels (67 and 69) with $3p4d\ ^3D_3$ as the leading component in the corresponding CSF expansion. To distinguish these levels, we added subscripts A and B to the labels of the dominant component. In a similar way, subscripts A and B were added to distinguish levels 78 and 80, both with $3p4f\ ^3F_3$ as the leading component.

Table 2 indicates that there are a few states that are either misidentified or assigned with a label that is inconsistent with the labels of the current calculation. The observed energy for $3p4f\ ^3D_2$ (level 84) is 2417 cm^{-1} too low compared to the calculated value and the observed energy for $3s5s\ ^3S_1$ (level 92) is $33,948\text{ cm}^{-1}$ too high. There seem to be no other computed energy levels that match the observed energies. The observed energy for $3s5p\ ^1P_1^o$ (level 100) is 3733 cm^{-1} too low. The observed energy matches the computed energy of $3s5p\ ^3P_1^o$ (level 97), and, thus, it seems like an inconsistency in the labeling. Finally, $3s5f\ ^1F_3^o$ (level 117) is $101,545\text{ cm}^{-1}$ too high and there is no other computed energy level that matches. Removing the energy outliers above, the mean relative energy differences are, respectively, 0.217%, 0.051%, 0.023% for the valence, the valence and core–valence and the valence, core–valence and core–core calculations. The energy differences are mainly due to higher-order electron correlation effects that have not been accounted for in the calculations. At the same time, one should bear in mind that the observed excitation energies are also associated with uncertainties as reflected in the limited number of valid digits displayed in the NIST tables.

In Table 3, the excitation energies obtained by including core–core correlation in the CI calculations are compared with energies from calculations by Landi [2] using the FAC code and with energies by Aggarwal et al. [21] using CIV3 in the Breit–Pauli approximation. The uncertainties of the excitation energies for the latter calculations are substantially larger. The calculations by Landi support the conclusion that some of the levels in the NIST database are misidentified. One may note that Landi gives levels 78 and 80 the labels $3p4f\ ^3F_3$ and $3p4f\ ^1F_3$, respectively, whereas Aggarwal et al. reverse the labels. This illustrates that labeling is dependent on the calculation and that the labeling process is far from straightforward [20].

Table 2. Comparison of calculated and observed excitation energies in Mg-like iron (Fe XV). E_{VV} are energies from CI calculations that account for valence–valence correlation. E_{VV+CV} are energies from CI calculations that account for valence–valence and core–valence electron correlation. $E_{VV+CV+CC}$ are energies that account for valence–valence and core–valence electron correlation and where core–core electron correlation effects have been included perturbatively. E_{NIST} are observed energies from the NIST database ([3]). ΔE are energy differences with respect to E_{NIST} . All energies are in cm^{-1} .

No.	Level	E_{VV}	ΔE	E_{VV+CV}	ΔE	$E_{VV+CV+CC}$	ΔE	E_{NIST}
1	$3s^2\ ^1S_0$	0	0	0	0	0	0	0
2	$3s3p\ ^3P_0^o$	233,087	−755	233,828	−14	233,928	86	233,842
3	$3s3p\ ^3P_1^o$	238,936	−724	239,668	8	239,741	81	239,660
4	$3s3p\ ^3P_2^o$	253,017	−803	253,829	9	253,773	−47	253,820
5	$3s3p\ ^1P_1^o$	354,941	3030	352,169	258	352,091	180	351,911
6	$3p^2\ ^3P_0$	556,594	2070	554,643	119	554,895	371	554,524
7	$3p^2\ ^1D_2$	559,900	300	559,834	234	559,661	61	559,600
8	$3p^2\ ^3P_1$	566,524	1922	564,663	61	564,674	72	56,4602
9	$3p^2\ ^3P_2$	583,327	1524	581,933	130	581,870	67	581,803
10	$3p^2\ ^1S_0$	662,999	3372	660,269	642	660,229	602	659,627
11	$3s3d\ ^3D_1$	680,522	1750	678,954	182	678,329	−443	678,772
12	$3s3d\ ^3D_2$	681,520	1735	679,986	201	679,381	−404	679,785
13	$3s3d\ ^3D_3$	683,080	1664	681,603	187	680,952	−464	681,416
14	$3s3d\ ^1D_2$	766,690	4597	762,729	636	762,176	83	762,093
15	$3p3d\ ^3F_2^o$	929,158	917	928,565	324	928,086	−155	928,241
16	$3p3d\ ^3F_3^o$	938,885	759	938,469	343	938,068	−58	938,126
17	$3p3d\ ^1D_2^o$	950,226	1713	948,768	255	948,383	−130	948,513
18	$3p3d\ ^3F_4^o$	950,300	642	949,990	332	949,451	−207	949,658
19	$3p3d\ ^3D_1^o$	986,221	3353	983,077	209	982,740	−128	982,868
20	$3p3d\ ^3P_2^o$	986,499	2985	983,765	251	983,350	−164	983,514
21	$3p3d\ ^3D_3^o$	998,324	3472	995,088	236	994,712	−140	994,852
22	$3p3d\ ^3P_0^o$	998,597	2708	996,218	329	995,835	−54	995,889
23	$3p3d\ ^3P_1^o$	999,166	2923	996,547	304	996,127	−116	996,243
24	$3p3d\ ^3D_2^o$	999,755	3132	996,892	269	996,449	−174	996,623
25	$3p3d\ ^1F_3^o$	1,066,906	4391	1,063,163	648	1,062,704	189	1,062,515
26	$3p3d\ ^1P_1^o$	1,078,913	4026	1,075,795	908	1,075,306	419	1,074,887
27	$3d^2\ ^3F_2$	1,373,374	3043	1,370,858	527	1,369,758	−573	1,370,331
28	$3d^2\ ^3F_3$	1,374,983	2948	1,372,527	492	1,371,407	−628	1,372,035
29	$3d^2\ ^3F_4$	1,376,965	2909	1,374,580	524	1,373,475	−581	1,374,056
30	$3d^2\ ^1D_2$	1,405,702	3110	1,403,474	882	1,402,237	−355	1,402,592
31	$3d^2\ ^3P_0$	1,409,066		1,406,328		1,405,381		
32	$3d^2\ ^3P_1$	1,409,639		1,406,926		1,405,672		
33	$3d^2\ ^1G_4$	1,409,702	2644	1,407,974	916	1,406,831	−227	1,407,058
34	$3d^2\ ^3P_2$	1,411,053	3280	1,408,467	694	1,407,210	−563	1,407,773
35	$3d^2\ ^1S_0$	1,489,913	2859	1,488,993	1939	1,487,460	406	1,487,054
36	$3s4s\ ^3S_1$	1,761,471	−2229	1,764,876	1176	17,63,699	−1	1,763,700
37	$3s4s\ ^1S_0$	1,785,265	−1735	1,788,455	1455	1,787,322	322	1,787,000
38	$3s4p\ ^3P_0^o$	1,880,014		1,883,187		1,882,236		
39	$3s4p\ ^3P_1^o$	1,880,440		1,883,595		1,882,588		
40	$3s4p\ ^3P_2^o$	1,887,508		1,890,703		1,889,632		
41	$3s4p\ ^1P_1^o$	1,887,872	−2098	1,891,051	1081	1,890,042	72	1,889,970
42	$3s4d\ ^3D_1$	2,029,659	−1651	2,032,907	1597	2,031,683	373	2,031,310
43	$3s4d\ ^3D_2$	2,030,413	−1607	2,033,653	1633	2,032,413	393	2,032,020

Table 2. Cont.

No.	Level	E_{VV}	ΔE	E_{VV+CV}	ΔE	$E_{VV+CV+CC}$	ΔE	E_{NIST}
44	3s4d 3D_3	2,031,636	−1544	2,034,880	1700	2,033,623	443	2,033,180
45	3s4d 1D_2	2,032,991	−2289	2,036,318	1038	2,035,053	−227	2,035,280
46	3p4s $^3P_0^o$	2,051,314		2,053,909		2,053,031		
47	3p4s $^3P_1^o$	2,054,922		2,057,446		2,056,493		
48	3p4s $^3P_2^o$	2,071,700		2,074,376		2,073,372		
49	3p4s $^1P_1^o$	2,085,097		2,087,237		2,086,235		
50	3s4f $^3F_2^o$	2,105,597	−2923	2,109,821	1301	2,108,281	−239	2,108,520
51	3s4f $^3F_3^o$	2,105,804	−2816	2,110,029	1409	2,108,503	−117	2,108,620
52	3s4f $^3F_4^o$	2,106,098	−2782	2,110,327	1447	2,108,798	−82	2,108,880
53	3s4f $^1F_3^o$	2,120,519	−2631	2,124,654	1504	2,123,180	30	2,123,150
54	3p4p 1P_1	2,152,851		2,155,266		2,154,244		
55	3p4p 3D_1	2,167,018		2,169,386		2,168,341		
56	3p4p 3D_2	2,168,756		2,171,070		2,170,006		
57	3p4p 3P_0	2,173,624		2,175,566		2,174,583		
58	3p4p 3P_1	2,181,779		2,183,914		2,182,831		
59	3p4p 3D_3	2,184,022		2,186,457		2,185,350		
60	3p4p 3P_2	2,189,341		2,191,385		2,190,270		
61	3p4p 3S_1	2,192,119		2,194,460		2,193,367		
62	3p4p 1D_2	2,206,894		2,208,893		2,207,746		
63	3p4p 1S_0	2,235,724		2,237,406		2,236,314		
64	3p4d $^3D_1^o$	2,311,660		2,314,071		2,313,090		
65	3p4d $^1D_2^o$	2,311,989		2,314,331		2,313,312		
66	3p4d $^3D_2^o$	2,312,449		2,314,882		2,313,865		
67	3p4d $^3D_{3A}^o$	2,313,908		2,316,401		2,315,387		
68	3p4d $^3F_2^o$	2,329,261		2,331,722		2,330,678		
69	3p4d $^3D_{3B}^o$	2,330,539		2,333,084		2,332,039		
70	3p4d $^3F_4^o$	2,337,384		2,339,922		2,338,857		
71	3p4d $^1F_3^o$	2,337,651		2,340,302		2,339,278		
72	3p4d $^3P_2^o$	2,341,803		2,344,120		2,343,033		
73	3p4d $^3P_1^o$	2,342,778		2,345,091		2,344,049		
74	3p4d $^3P_0^o$	2,346,915		2,349,198		2,348,199		
75	3p4d $^1P_1^o$	2,350,169		2,352,543		2,351,513		
76	3p4f 3G_3	2,377,507	−2653	2,381,283	1123	2,379,714	−446	2,380,160
77	3p4f 3G_4	2,384,217	−2483	2,387,976	1276	2,386,434	−266	2,386,700
78	3p4f $^3F_{3A}$	2,384,435		2,388,118		2,386,537		
79	3p4f 3F_2	2,388,049	−2051	2,391,670	1570	2,390,091	−9	2,390,100
80	3p4f $^3F_{3B}$	2,397,860		2,401,630		2,400,029		
81	3p4f 3G_5	2,399,542	−2558	2,403,453	1353	2,401,876	−224	2,402,100
82	3p4f 3F_4	2,400,524	−1576	2,404,286	2186	2,402,697	597	2,402,100
83	3p4f 3D_3	2,411,680	−1320	2,415,368	2368	2,413,758	758	2,413,000
84	3p4f 3D_2	2,414,633	333	2,418,319	4019	2,416,717	2417	2,414,300
85	3p4f 3D_1	2,417,852	−2248	2,421,557	1457	2,419,975	−125	2,420,100
86	3p4f 1G_4	2,426,828	−1872	2,430,497	1797	2,429,063	363	2,428,700
87	3p4f 1D_2	2,433,430	−2570	2,437,039	1039	2,435,534	−466	2,436,000
88	3d4s 3D_1	2,458,614		2,460,640		2,458,997		
89	3d4s 3D_2	2,459,450		2,461,503		2,459,846		
90	3d4s 3D_3	2,461,283		2,463,415		2,461,742		
91	3d4s 1D_2	2,468,780		2,470,737		2,469,163		
92	3s5s 3S_1	2,507,700	−37,100	2,512,036	−32,764	2,510,852	−33,948	2,544,800

Table 2. Cont.

No.	Level	E_{VV}	ΔE	E_{VV+CV}	ΔE	$E_{VV+CV+CC}$	ΔE	E_{NIST}
93	3s5s 1S_0	2,516,613		2,520,681		2,519,752		
94	3d4p $^1D_2^o$	2,561,358		2,563,408		2,561,899		
95	3d4p $^3D_1^o$	2,564,069		2,567,301		2,565,949		
96	3s5p $^3P_0^o$	2,564,472		2,568,582		2,567,624		
97	3s5p $^3P_1^o$	2,565,848		2,568,791		2,567,639		
98	3d4p $^3D_2^o$	2,567,134		2,569,092		2,567,703		
99	3d4p $^3D_3^o$	2,568,154		2,571,175		2,569,693		
100	3s5p $^1P_1^o$	2,568,200	1200	2,571,834	4834	2,570,733	3733	2,567,000
101	3s5p $^3P_2^o$	2,569,213		2,572,157		2,570,743		
102	3d4p $^3F_2^o$	2,570,296		2,572,316		2,571,126		
103	3d4p $^3F_3^o$	2,573,116		2,575,101		2,573,592		
104	3d4p $^3F_4^o$	2,576,139		2,578,374		2,576,829		
105	3d4p $^3P_1^o$	2,583,286		2,585,242		2,583,862		
106	3d4p $^3P_2^o$	2,583,400		2,585,407		2,583,960		
107	3d4p $^3P_0^o$	2,583,734		2,585,658		2,584,322		
108	3d4p $^1F_3^o$	2,592,868		2,594,519		2,593,236		
109	3d4p $^1P_1^o$	2,603,279		2,605,145		2,604,533		
110	3s5d 3D_1	2,637,190	−2910	2,641,400	1300	2,640,247	147	2,640,100
111	3s5d 3D_2	2,637,419	−2481	2,641,630	1730	2,640,442	542	2,639,900
112	3s5d 3D_3	2,637,852	−2448	2,642,072	1772	2,640,870	570	2,640,300
113	3s5d 1D_2	2,639,773		2,643,981		2,642,888		
114	3s5f $^3F_2^o$	2,672,676	−3724	2,677,360	960	2,675,889	−511	2,676,400
115	3s5f $^3F_3^o$	2,672,770	−3630	2,677,455	1055	2,675,988	−412	2,676,400
116	3s5f $^3F_4^o$	2,672,907	−3693	2,677,594	994	2,676,123	−477	2,676,600
117	3s5f $^1F_3^o$	2,678,041	−104,659	2,682,597	−100,103	2,681,155	−101,545	2,782,700
118	3s5g 3G_3	2,682,487		2,687,368		2,685,680		
119	3s5g 3G_4	2,682,654		2,687,556		2,685,877		
120	3s5g 3G_5	2,682,855		2,687,777		2,686,099		
121	3s5g 1G_4	2,685,580		2,690,506		2,688,841		
122	3d4d 1F_3	2,699,116		2,701,602		2,699,874		
123	3d4d 3D_1	2,703,542		2,705,972		2,704,354		
124	3d4d 3D_2	2,704,742		2,707,218		2,705,580		
125	3d4d 3D_3	2,706,116		2,708,636		2,706,964		
126	3d4d 3G_3	2,707,934		2,710,522		2,708,828		
127	3d4d 1P_1	2,709,315		2,711,813		2,710,163		
128	3d4d 3G_4	2,709,360		2,711,928		2,710,264		
129	3d4d 3G_5	2,711,220		2,713,878		2,712,174		
130	3d4d 3S_1	2,720,698		2,723,175		2,721,783		
131	3d4d 3F_2	2,726,309		2,728,092		2,726,350		
132	3d4d 3F_3	2,727,568		2,729,398		2,727,634		
133	3d4d 3F_4	2,729,029		2,730,908		2,729,156		
134	3d4d 1D_2	2,741,839		2,743,862		2,742,627		
135	3d4d 3P_0	2,744,213		2,746,022		2,744,706		
136	3d4d 3P_1	2,744,807		2,746,626		2,745,163		
137	3d4d 3P_2	2,745,935		2,747,809		2,746,300		
138	3d4d 1G_4	2,748,985		2,751,121		2,749,474		
139	3d4f $^3H_4^o$	2,765,833		2,770,098		2,768,443		
140	3d4f $^1G_4^o$	2,767,533		2,771,821		2,770,030		
141	3d4f $^3H_5^o$	2,767,692		2,771,943		2,770,434		
142	3d4d 1S_0	2,775,538		2,779,275		2,777,362		

Table 2. Cont.

No.	Level	E_{VV}	ΔE	E_{VV+CV}	ΔE	$E_{VV+CV+CC}$	ΔE	E_{NIST}
143	$3d4f\ ^3F_2^o$	2,776,151		2,779,298		2,778,011		
144	$3d4f\ ^3F_3^o$	2,776,264		2,779,933		2,778,867		
145	$3d4f\ ^3F_4^o$	2,776,981		2,780,796		2,780,729		
146	$3d4f\ ^1D_2^o$	2,786,768		2,790,305		2,788,248		

Table 3. Comparison of calculated and observed excitation energies in Mg-like iron (Fe XV). $E_{VV+CV+CC}$ are energies that account for valence–valence and core–valence electron correlation and where core–core electron correlation effects have been included perturbatively. E_{FAC} are energies by Landi [2] using the FAC code. E_{CIV3} are energies by Aggarwal et al. [21] using the CIV3 code. E_{NIST} are observed energies from the NIST database ([3]). ΔE are energy differences with respect to E_{NIST} . All energies are in cm^{-1} .

No.	Level	$E_{VV+CV+VV}$	ΔE	E_{FAC}	ΔE	E_{CIV3}	ΔE	E_{NIST}
1	$3s^2\ ^1S_0$	0	0	0	0	0	0	0
2	$3s3p\ ^3P_0^o$	233,928	86	233,068	−774	235,013	1171	233,842
3	$3s3p\ ^3P_1^o$	239,741	81	238,900	−760	240,511	851	239,660
4	$3s3p\ ^3P_2^o$	253,773	−47	252,917	−903	253,548	−272	253,820
5	$3s3p\ ^1P_1^o$	352,091	180	356,126	4215	356,262	4351	351,911
6	$3p^2\ ^3P_0$	554,895	371	556,994	2470	560,275	5751	554,524
7	$3p^2\ ^1D_2$	559,661	61	560,266	666	563,216	3616	559,600
8	$3p^2\ ^3P_1$	564,674	72	566,832	2230	569,295	4693	564,602
9	$3p^2\ ^3P_2$	581,870	67	583,564	1761	584,856	3053	581,803
10	$3p^2\ ^1S_0$	660,229	602	665,768	6141	665,260	5633	659,627
11	$3s3d\ ^3D_1$	678,329	−443	680,146	1374	687,680	8908	678,772
12	$3s3d\ ^3D_2$	679,381	−404	681,129	1344	688,733	8948	679,785
13	$3s3d\ ^3D_3$	680,952	−464	682,667	1251	690,401	8985	681,416
14	$3s3d\ ^1D_2$	762,176	83	769,369	7276	774,295	12,202	762,093
15	$3p3d\ ^3F_2^o$	928,086	−155	928,786	545	938,265	10,024	928,241
16	$3p3d\ ^3F_3^o$	938,068	−58	938,555	429	947,307	9181	938,126
17	$3p3d\ ^1D_2^o$	948,383	−130	949,447	934	958,402	9889	948,513
18	$3p3d\ ^3F_4^o$	949,451	−207	949,927	269	957,820	8162	949,658
19	$3p3d\ ^3D_1^o$	982,740	−128	986,082	3214	995,526	12,658	982,868
20	$3p3d\ ^3P_2^o$	983,350	−164	986,407	2893	995,767	12,253	983,514
21	$3p3d\ ^3D_3^o$	994,712	−140	997,944	3092	1,007,026	12,174	994,852
22	$3p3d\ ^3P_0^o$	995,835	−54	998,762	2873	1,006,708	10,819	995,889
23	$3p3d\ ^3P_1^o$	996,127	−116	999,173	2930	1,007,366	11,123	996,243
24	$3p3d\ ^3D_2^o$	996,449	−174	999,578	2955	1,008,124	11,501	996,623
25	$3p3d\ ^1F_3^o$	1,062,704	189	1,070,794	8279	1,077,456	14,941	1,062,515
26	$3p3d\ ^1P_1^o$	1,075,306	419	1,083,826	8939	1,089,691	14,804	1,074,887
27	$3d^2\ ^3F_2$	1,369,758	−573	1,372,400	2069	1,388,111	17,780	1,370,331
28	$3d^2\ ^3F_3$	1,371,407	−628	1,373,988	1953	1,389,834	17,799	1,372,035
29	$3d^2\ ^3F_4$	1,373,475	−581	1,375,938	1882	1,391,941	17,885	1,374,056
30	$3d^2\ ^1D_2$	1,402,237	−355	1,407,428	4836	1,421,702	19,110	1,402,592
31	$3d^2\ ^3P_0$	1,405,381		1,409,507		1,424,577		
32	$3d^2\ ^3P_1$	1,405,672		1,410,109		1,425,246		
33	$3d^2\ ^1G_4$	1,406,831	−227	1,412,127	5069	1,425,872	18,814	1,407,058
34	$3d^2\ ^3P_2$	1,407,210	−563	1,411,643	3870	1,426,815	19,042	1,407,773
35	$3d^2\ ^1S_0$	1,487,460	406	1,498,668	11,614	1,508,954	21,900	1,487,054
36	$3s4s\ ^3S_1$	1,763,699	−1	1,760,910	−2790	1,764,005	305	1,763,700

Table 3. Cont.

No.	Level	$E_{VV+CV+VV}$	ΔE	E_{FAC}	ΔE	E_{CIV3}	ΔE	E_{NIST}
37	3s4s 1S_0	1,787,322	322	1,786,052	−948	1,787,950	950	1,787,000
38	3s4p $^3P_0^o$	1,882,236		1,880,319		1,883,685		
39	3s4p $^3P_1^o$	1,882,588		1,880,746		1,884,091		
40	3s4p $^3P_2^o$	1,889,632		1,887,756		1,890,313		
41	3s4p $^1P_1^o$	1,890,042	72	1,888,124	−1846	1,890,631	661	1,889,970
42	3s4d 3D_1	2,031,683	373	2,029,563	−1747	2,034,124	2814	2,031,310
43	3s4d 3D_2	2,032,413	393	2,030,328	−1692	2,034,848	2828	2,0320,20
44	3s4d 3D_3	2,033,623	443	2,031,544	−1636	2,036,055	2875	2,033,180
45	3s4d 1D_2	2,035,053	−227	2,033,212	−2068	2,037,569	2289	2,035,280
46	3p4s $^3P_0^o$	2,053,031		2,051,778		2,055,797		
47	3p4s $^3P_1^o$	2,056,493		2,055,514		2,059,308		
48	3p4s $^3P_2^o$	2,073,372		2,072,083		2,074,452		
49	3p4s $^1P_1^o$	2,086,235		2,086,607		2,088,795		
50	3s4f $^3F_2^o$	2,108,281	−239	2,107,228	−1292	2,110,073	1553	2,108,520
51	3s4f $^3F_3^o$	2,108,503	−117	2,107,423	−1197	2,110,281	1661	2,108,620
52	3s4f $^3F_4^o$	2,108,798	−82	2,107,701	−1179	2,110,567	1687	2,108,880
53	3s4f $^1F_3^o$	2,123,180	30	2,124,054	904	2,125,886	2736	2,123,150
54	3p4p 1P_1	2,154,244		2,167,343		2,158,599		
55	3p4p 3D_1	2,168,341		2,153,046		2,171,635		
56	3p4p 3D_2	2,170,006		2,169,173		2,173,578		
57	3p4p 3P_0	2,174,583		2,175,103		2,178,812		
58	3p4p 3P_1	2,182,831		2,182,790		2,185,901		
59	3p4p 3D_3	2,185,350		2,184,242		2,187,229		
60	3p4p 3P_2	2,190,270		2,190,674		2,193,265		
61	3p4p 3S_1	2,193,367		2,192,597		2,195,756		
62	3p4p 1D_2	2,207,746		2,209,221		2,211,163		
63	3p4p 1S_0	2,236,314		2,239,314		2,241,187		
64	3p4d $^3D_1^o$	2,313,090		2,311,999		2,318,014		
65	3p4d $^1D_2^o$	2,313,312		2,312,326		2,318,179		
66	3p4d $^3D_2^o$	2,313,865		2,312,835		2,318,826		
67	3p4d $^3D_{3A}^o$	2,315,387		23,144,663		2,320,538		
68	3p4d $^3F_2^o$	2,330,678		2,329,647		2,334,178		
69	3p4d $^3D_{3B}^o$	2,332,039		2,331,0213		2,335,726		
70	3p4d $^3F_4^o$	2,338,857		2,338,064		2,342,277		
71	3p4d $^1F_3^o$	2,339,278		2,338,703		2,343,517		
72	3p4d $^3P_2^o$	2,343,033		2,342,598		2,347,544		
73	3p4d $^3P_1^o$	2,344,049		2,343,850		2,348,795		
74	3p4d $^3P_0^o$	2,348,199		2,347,823		2,352,406		
75	3p4d $^1P_1^o$	2,351,513		2,351,661		2,356,773		
76	3p4f 3G_3	2,379,714	−446	2,379,430	−730	2,384,306	4146	2,380,160
77	3p4f 3G_4	2,386,434	−266	2,386,688	−12	2,391,198	4498	2,386,700
78	3p4f $^3F_{3A}$	2,386,537		2,386,430		2,390,473		
79	3p4f 3F_2	2,390,091	−9	2,390,112	12	2,393,842	3742	2,390,100
80	3p4f $^3F_{3B}$	2,400,029		2,399,796		2,402,786		
81	3p4f 3G_5	2,401,876	−224	2,401,746	−354	2,405,617	3517	2,402,100
82	3p4f 3F_4	2,402,697	597	2,402,507	407	2,405,496	3396	2,402,100
83	3p4f 3D_3	2,413,758	758	2,414,120	1120	2,417,151	4151	2,413,000
84	3p4f 3D_2	2,416,717	2417	2,417,276	2976	2,420,124	5824	2,414,300
85	3p4f 3D_1	2,419,975	−125	2,420,512	412	2,423,219	3119	2,420,100
86	3p4f 1G_4	2,429,063	363	2,432,908	4208	2,435,828	7128	2,428,700

Table 3. Cont.

No.	Level	$E_{VV+CV+VV}$	ΔE	E_{FAC}	ΔE	E_{CIV3}	ΔE	E_{NIST}
87	$3p4f\ ^1D_2$	2,435,534	−466	2,438,982	2982	2,440,239	4239	2,436,000
88	$3d4s\ ^3D_1$	2,458,997		2,458,814		2,468,047		
89	$3d4s\ ^3D_2$	2,459,846		2,459,675		2,468,969		
90	$3d4s\ ^3D_3$	2,461,742		2,461,461		2,470,911		
91	$3d4s\ ^1D_2$	2,469,163		2,470,364		2,479,437		
92	$3s5s\ ^3S_1$	2,510,852	−33,948	2,507,572	−37,228			2,544,800
93	$3s5s\ ^1S_0$	2,519,752		2,517,043				
94	$3d4p\ ^1D_2^o$	2,561,899		2,561,169		2,571,814		
95	$3d4p\ ^3D_1^o$	2,565,949		2,566,041		2,576,851		
96	$3s5p\ ^3P_0^o$	2,567,624		2,564,597				
97	$3s5p\ ^3P_1^o$	2,567,639		2,564,254				
98	$3d4p\ ^3D_2^o$	2,567,703		2,567,341		2,577,905		
99	$3d4p\ ^3D_3^o$	2,569,693		2,569,518		2,583,117		
100	$3s5p\ ^1P_1^o$	2,570,733	3733	2,568,358	1358			2,567,000
101	$3s5p\ ^3P_2^o$	2,570,743		2,568,240				
102	$3d4p\ ^3F_2^o$	2,571,126		2,570,526		2,580,319		
103	$3d4p\ ^3F_3^o$	2,573,592		2,573,370		2,579,847		
104	$3d4p\ ^3F_4^o$	2,576,829		2,576,531		2,586,036		
105	$3d4p\ ^3P_1^o$	2,583,862		2,584,287		2,593,158		
106	$3d4p\ ^3P_2^o$	2,583,960		2,584,326		2,593,586		
107	$3d4p\ ^3P_0^o$	2,584,322		2,584,699		2,593,641		
108	$3d4p\ ^1F_3^o$	2,593,236		2,596,425		2,604,571		
109	$3d4p\ ^1P_1^o$	2,604,533		2,607,817		2,610,870		
110	$3s5d\ ^3D_1$	2,640,247	147	2,637,143	−2957			2,640,100
111	$3s5d\ ^3D_2$	2,640,442	542	2,637,376	−2524			2,639,900
112	$3s5d\ ^3D_3$	2,640,870	570	2,637,804	−2496			2,640,300
113	$3s5d\ ^1D_2$	2,642,888		2,640,084	0			
114	$3s5f\ ^3F_2^o$	2,675,889	−511	2,673,354	−3046			2,676,400
115	$3s5f\ ^3F_3^o$	2,675,988	−412	2,673,444	−2956			2,676,400
116	$3s5f\ ^3F_4^o$	2,676,123	−477	2,673,575	−3025			2,676,600
117	$3s5f\ ^1F_3^o$	2,681,155	−101,545	2,679,558	−103,142			2,782,700
118	$3s5g\ ^3G_3$	2,685,680		2,683,089				
119	$3s5g\ ^3G_4$	2,685,877		2,683,272				
120	$3s5g\ ^3G_5$	2,686,099		2,683,494				
121	$3s5g\ ^1G_4$	2,688,841		2,686,809				
122	$3d4d\ ^1F_3$	2,699,874		2,697,717		2,710,391		
123	$3d4d\ ^3D_1$	2,704,354		2,702,464		2,714,967		
124	$3d4d\ ^3D_2$	2,705,580		2,703,625		2,716,229		
125	$3d4d\ ^3D_3$	2,706,964		2,705,001		2,717,578		
126	$3d4d\ ^3G_3$	2,708,828		2,707,726		2,717,919		
127	$3d4d\ ^1P_1$	2,710,163		2,708,170		2,721,079		
128	$3d4d\ ^3G_4$	2,710,264		2,709,064		2,719,345		
129	$3d4d\ ^3G_5$	2,712,174		2,710,955		2,721,463		
130	$3d4d\ ^3S_1$	2,721,783		2,720,286		2,732,634		
131	$3d4d\ ^3F_2$	2,726,350		2,726,401		2,738,407		
132	$3d4d\ ^3F_3$	2,727,634		2,727,604		2,739,745		
133	$3d4d\ ^3F_4$	2,729,156		2,729,075		2,741,293		
134	$3d4d\ ^1D_2$	2,742,627		2,743,889		2,755,547		
135	$3d4d\ ^3P_0$	2,744,706		2,745,181		2,757,907		
136	$3d4d\ ^3P_1$	2,745,163		2,745,727		2,758,477		

Table 3. Cont.

No.	Level	$E_{VV+CV+VV}$	ΔE	E_{FAC}	ΔE	E_{CIV3}	ΔE	E_{NIST}
137	$3d4d\ ^3P_2$	2,746,300		2,747,024		2,759,619		
138	$3d4d\ ^1G_4$	2,749,474		2,752,675		2,761,254		
139	$3d4f\ ^3H_4^o$	2,768,443		2,766,350		2,778,483		
140	$3d4f\ ^1G_4^o$	2,770,030		2,768,154		2,780,096		
141	$3d4f\ ^3H_5^o$	2,770,434		2,768,448		2,780,831		
142	$3d4d\ ^1S_0$	2,777,362		2,781,322		2,792,233		
143	$3d4f\ ^3F_2^o$	2,778,011		2,775,995		2,787,305		
144	$3d4f\ ^3F_3^o$	2,778,867		2,776,790		2,787,964		
145	$3d4f\ ^3F_4^o$	2,780,729		2,777,446		2,788,842		
146	$3d4f\ ^1D_2^o$	2,788,248		2,787,354		2,798,312		

5. Conclusions

CI with restrictions on the interactions (CI combined with second-order Brillouin–Wigner perturbation theory) makes it possible to handle large CSF expansions. The calculations including core–core correlation take around 20 h with 10 nodes on a cluster and bring the computed and observed excitation energies into very good agreement. To improve the computed excitation energies, the orbital set would need to be further extended leading to even larger matrices. The combined CI and perturbation method can be applied to include core–valence correlation in systems with many valence electrons and calculations. Calculations including valence–valence correlation and where core–valence correlation is treated perturbatively are in progress for P-, S-, and Cl-like systems.

Acknowledgments: Per Jönsson gratefully acknowledges support from the Swedish Research Council under contract 2015-04842.

Author Contributions: All authors contributed equally to the work.

Conflicts of Interest: The authors declare no conflicts of interest.

References

- Young, P.R.; Del Zanna, G.; Mason, H.E.; Dere, K.P.; Landi, E.; Landini, M.; Doschek, G.A.; Brown, C.M.; Culhane, L.; Harra, L.K.; et al. EUV Emission Lines and Diagnostics Observed with Hinode/EIS. *Publ. Astron. Soc. Jpn.* **2007**, *59*, S857–S864.
- Landi, E. Atomic data and spectral line intensities for Fe XV. *At. Data Nucl. Data Tables* **2011**, *97*, 587–647.
- Kramida, A.; Ralchenko, Y.; Reader, J.; NIST ASD Team. *Atomic Spectra Database (ver. 5.2)*; National Institute of Standards and Technology: Gaithersburg, MD, USA, 2014. Available online: <http://physics.nist.gov/asd> (accessed on 28 december 2014).
- Brown, G.V.; Beiersdorfer, P.; Utter, S.B.; Boyce, K.R.; Gendreau, K.C.; Kelley, R.; Porter, F.S.; Gygax, J. Measurements of Atomic Parameters of Highly Charged Ions for Interpreting Astrophysical Spectra. *Phys. Scr.* **2001**, doi:10.1238/Physica.Topical.092a00130.
- Jönsson, P.; Bengtsson, P.; Ekman, J.; Gustafsson, S.; Karlsson, L.B.; Gaigalas, G.; Froese Fischer, C.; Kato, D.; Murakami, I.; Sakaue, H.A.; et al. Relativistic CI calculations of spectroscopic data for the $2p^6$ and $2p^53l$ configurations in Ne-like ions between Mg III and Kr XXVII. *At. Data Nucl. Data Tables* **2014**, *100*, 1–154.
- Wang, K.; Guo, X.L.; Li, S.; Si, R.; Dang, W.; Chen, Z.B.; Jönsson, P.; Hutton, R.; Chen, C.Y.; Yan, J. Calculations with spectroscopic accuracy: Energies and transition rates in the nitrogen isoelectronic sequence from Ar XII to Zn XXIV. *Astrophys. J. Suppl. Ser.* **2016**, *223*, 33.
- Vilkas, M.J.; Ishikawa, Y. High-accuracy calculations of term energies and lifetimes of silicon-like ions with nuclear charges $Z = 24 - 30$. *J. Phys. B At. Mol. Opt. Phys.* **2004**, *37*, 1803–1816.
- Gu, M.F. Energies of $1s^22l^q$ ($1 \leq q \leq 8$) states for $Z \leq 60$ with a combined configuration interaction and many-body perturbation theory approach. *At. Data Nucl. Data Tables* **2005**, *89*, 267–293.

9. Kozlov, M.G.; Porsev, S.G.; Safronova, M.S.; Tupitsyn, I.I. CI-MBPT: A package of programs for relativistic atomic calculations based on a method combining configuration interaction and many-body perturbation theory. *Comput. Phys. Commun.* **2015**, *195*, 199–213.
10. Verdebout, S.; Rynkun, P.; Jönsson, P.; Gaigalas, G.; Froese Fischer, C.; Godefroid, M. A partitioned correlation function interaction approach for describing electron correlation in atoms. *J. Phys. B At. Mol. Opt. Phys.* **2013**, *46*, 085003.
11. Dzuba, V.A.; Berengut, J.; Harabati, C.; Flambaum, V.V. Combining configuration interaction with perturbation theory for atoms with large number of valence electrons. **2016**, arXiv:1611.00425v1.
12. Grant, I.P. *Relativistic Quantum Theory of Atoms and Molecules*; Springer: New York, NY, USA, 2007.
13. Froese Fischer, C.; Godefroid, M.; Brage, T.; Jönsson, P.; Gaigalas, G. Advanced multiconfiguration methods for complex atoms: Part I—Energies and wave functions. *J. Phys. B At. Mol. Opt. Phys.* **2016**, *49*, 182004.
14. Jönsson, P.; Gaigalas, G.; Bieroń, J.; Froese Fischer, C.; Grant, I.P. New Version: Grasp2K relativistic atomic structure package. *Comput. Phys. Commun.* **2013**, *184*, 2197–2203.
15. McKenzie, B.J.; Grant, I.P.; Norrington, P.H. A program to calculate transverse Breit and QED corrections to energy levels in a multiconfiguration Dirac-Fock environment. *Comput. Phys. Commun.* **1980**, *21*, 233–246.
16. Froese Fischer, C. The MCHF atomic-structure package. *Comput. Phys. Commun.* **1991**, *64*, 369–398.
17. Parpia, F.A.; Froese Fischer, C.; Grant, I.P. GRASP92: A package for large-scale relativistic atomic structure calculations. *Comput. Phys. Commun.* **1996**, *94*, 249–271.
18. Kotochigova, S.; Kirby, K.P.; Tupitsyn, I. Ab initio fully relativistic calculations of x-ray spectra of highly charged ions. *Phys. Rev. A* **2007**, *76*, 052513.
19. Gustafsson, S.; Jönsson, P.; Froese Fischer, C.; Grant, I.P. MCDHF and RCI calculations of energy levels, lifetimes and transition rates for $3l3l'$, $3l4l'$ and $3s5l$ states in Ca IX—As XXII and Kr XXV. *Astron. Astrophys.* **2017**, *579*, A76.
20. Gaigalas, G.; Froese Fischer, C.; Rynkun, P.; Jönsson, P. JJ2LSJ transformation and unique labeling for energy levels. *Atoms* **2016**, submitted.
21. Aggarwal, K.M.; Tayal, V.; Gupta, G.P.; Keenan, F.P. Energy levels and radiative rates for transitions in Mg-like iron, cobalt and nickel. *At. Data Nucl. Data Tables* **2007**, *93*, 615–710.



© 2017 by the authors; licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC-BY) license (<http://creativecommons.org/licenses/by/4.0/>).