

Electronic structure calculation of Cr^{3+} and Fe^{3+} in phosphor host materials based on relaxed structures by molecular dynamics simulation

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Table S1. Comparison of the lattice parameters simulated by MD using the potential parameters (A_{ij} and ρ_{ij}) in **Table 1** with those experimentally determined (Exp.). MSD represents the “mean square displacement” in MD.

Figure S1. Crystal structures of (a) LiAl_5O_8 , (b) $\alpha\text{-Al}_2\text{O}_3$, and (c) $\gamma\text{-LiAlO}_2$.

Figure S2. Emission and excitation spectra of (a) $\text{LiAl}_5\text{O}_8\text{:Cr}^{3+}$, (b) $\alpha\text{-Al}_2\text{O}_3\text{:Cr}^{3+}$, and (c) $\gamma\text{-LiAlO}_2\text{:Fe}^{3+}$.

Figure S3. Calculation clusters for isolated Cr^{3+} and Fe^{3+} in (a) $\text{LiAl}_5\text{O}_8\text{:Cr}^{3+}$ [$(\text{Li}_2\text{Al}_{10}\text{CrO}_{38})^{41-}$], (b) $\alpha\text{-Al}_2\text{O}_3\text{:Cr}^{3+}$ [$(\text{Al}_{13}\text{CrO}_{45})^{48-}$], and (c) $\gamma\text{-LiAlO}_2\text{:Fe}^{3+}$ [$(\text{Li}_9\text{Al}_6\text{FeO}_{32})^{34-}$].

Figure S4. The clusters for the calculation of the electronic interaction in the Fe–Fe pairs on (a) first-, (b) second-, and (c) third-nearest cation sites.

*The crystal structures in **Figure S1** and the calculation clusters in **Figure S3** and **S4** are illustrated by VESTA [i].

i Momma, K.; Izumi, F. VESTA 3 for three-dimensional visualization of crystal, volumetric and morphology data. *J. Appl. Crystallogr.* **2011**, *44*, 1272-1276.

Table S1 Comparison of the lattice parameters simulated by MD using the potential parameters (A_{ij} and ρ_{ij}) in **Table 1** with those experimentally determined (Exp.). MSD represents the “mean square displacement” in MD.

Material	Method	Lattice parameter (Å)			MSD (Å ²)	Ref.
		*The values in parentheses in the MD rows are difference (%) between Exp. and MD.				
		<i>a</i>	<i>b</i>	<i>c</i>		
α -Al ₂ O ₃	Exp.	4.7540	4.7540	12.990	–	21
	MD	4.76 (+0.13%)	4.75 (−0.08%)	12.87 (−0.92%)	Al: 0.012 O: 0.006	–
Li ₂ O	Exp.	4.610	4.610	4.610	–	24
	MD	4.67 (+1.30%)	4.67 (+1.30%)	4.67 (+1.30%)	Li: 0.025 O: 0.013	–
α -LiAlO ₂	Exp.	2.8003	2.8003	14.216	–	25
	MD	2.85 (+1.77%)	2.85 (+1.77%)	14.48 (+1.86%)	Al: 0.008 Li: 0.029 O: 0.015	–
γ -LiAlO ₂	Exp.	5.1687	5.1687	6.2679	–	22
	MD	5.23 (+1.19%)	5.23 (+1.19%)	6.22 (−0.76%)	Al: 0.007 Li: 0.032 O: 0.037	–
LiAl ₅ O ₈	Exp.	7.908	7.908	7.908	–	20
	MD	7.87 (−0.48%)	7.87 (−0.48%)	7.87 (−0.48%)	Al: 0.006 Li: 0.011 O: 0.010	–
Cr ₂ O ₃	Exp.	4.9507	4.9507	13.5656	–	26
	MD	4.94 (−0.22%)	4.94 (−0.22%)	13.37 (−1.44%)	Cr: 0.034 O: 0.008	–
LiCrO ₂	Exp.	2.9010	2.9010	14.4311	–	27
	MD	2.95 (+1.69%)	2.95 (+1.69%)	14.62 (+1.31%)	Cr: 0.006 Li: 0.029 O: 0.031	–
Fe ₂ O ₃	Exp.	5.0355	5.0355	13.7471	–	28
	MD	5.01 (−0.51%)	5.02 (−0.31%)	13.49 (−1.87%)	Fe: 0.013 O: 0.009	–
LiFe ₅ O ₈	Exp.	8.319	8.319	8.319	–	29
	MD	8.25 (−0.83%)	8.25 (−0.83%)	8.25 (−0.83%)	Fe: 0.018 Li: 0.023 O: 0.036	–

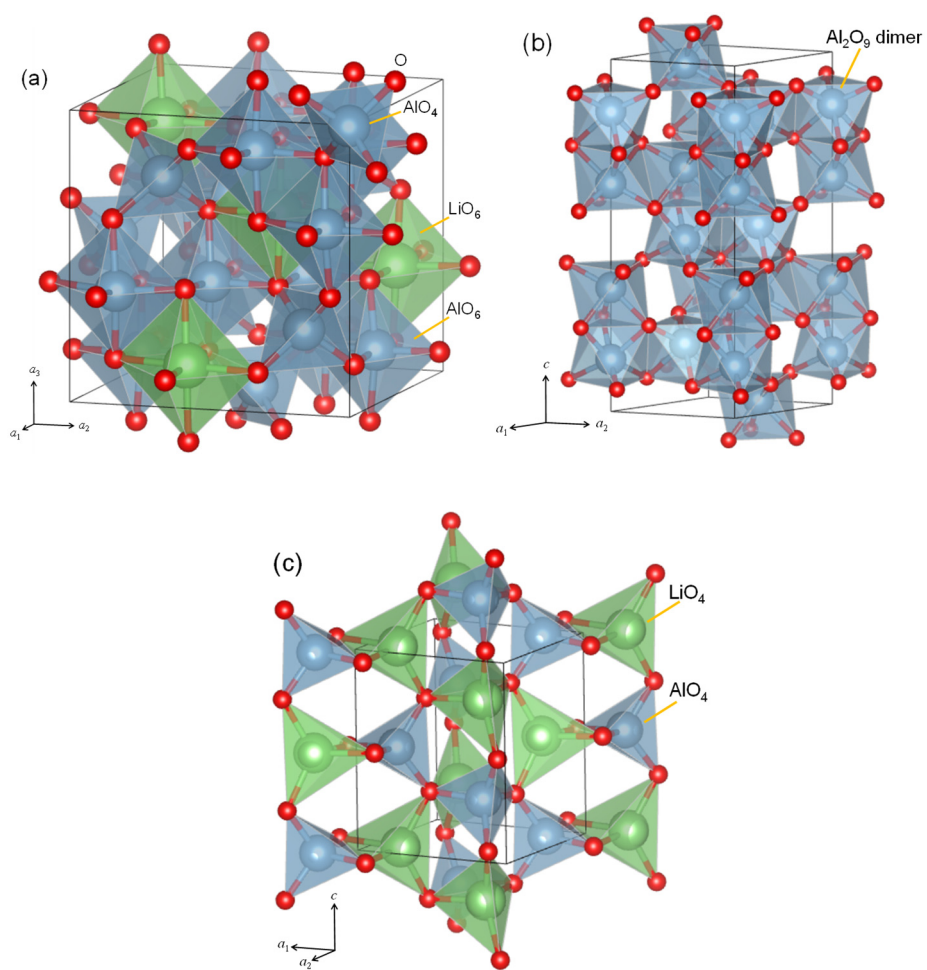


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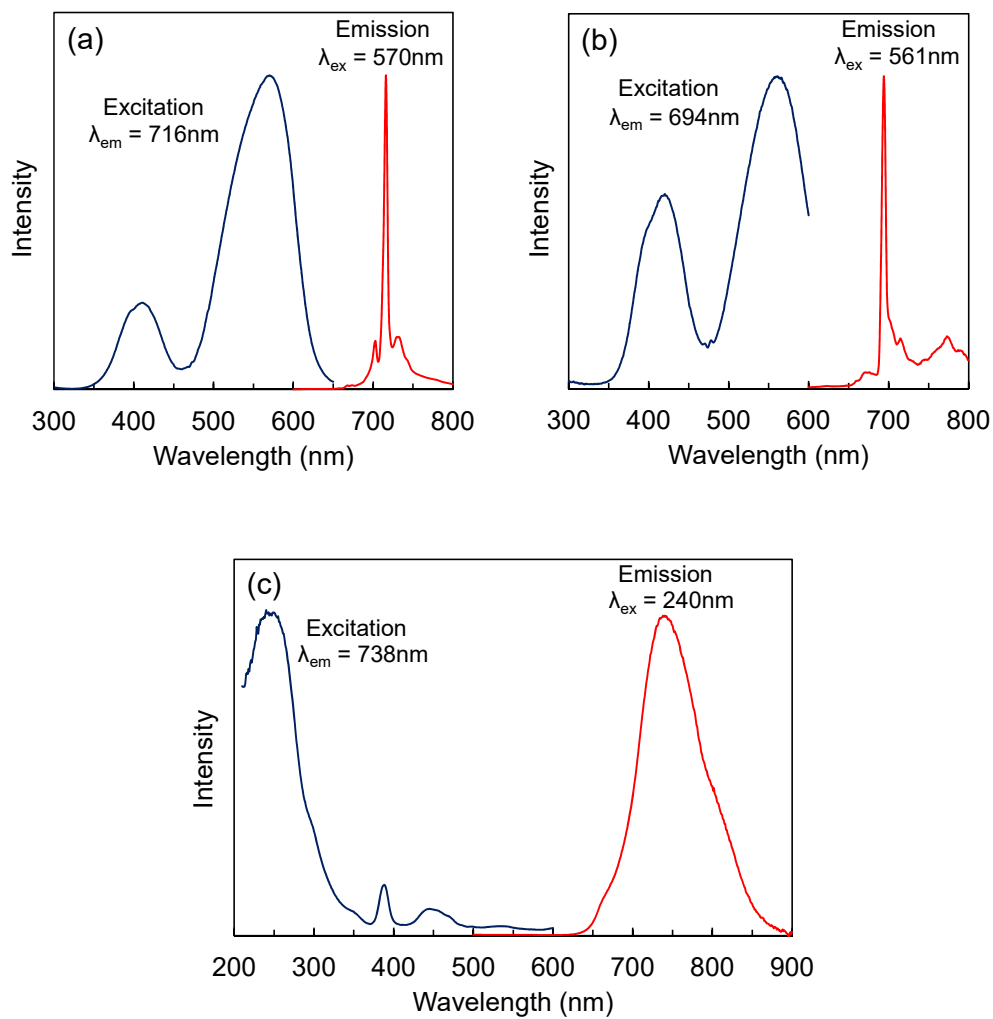


Figure S2. Emission and excitation spectra of (a) $\text{LiAl}_5\text{O}_8:\text{Cr}^{3+}$, (b) $\alpha\text{-Al}_2\text{O}_3:\text{Cr}^{3+}$, and (c) $\gamma\text{-LiAlO}_2:\text{Fe}^{3+}$.

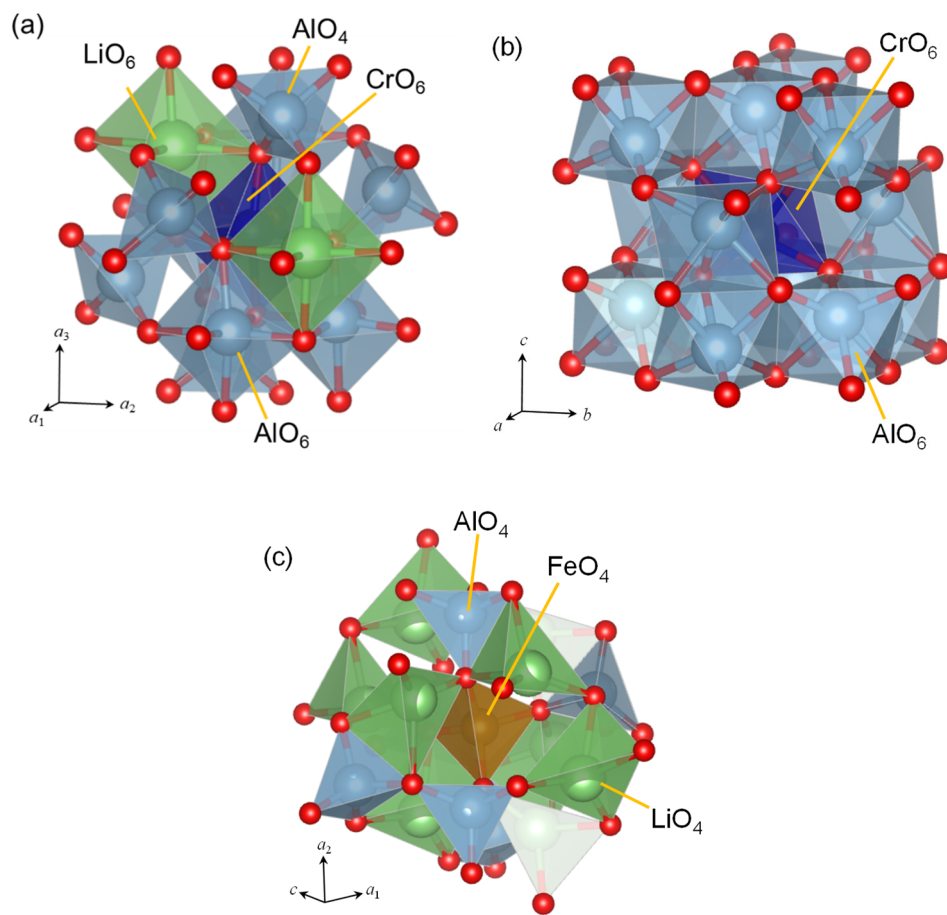


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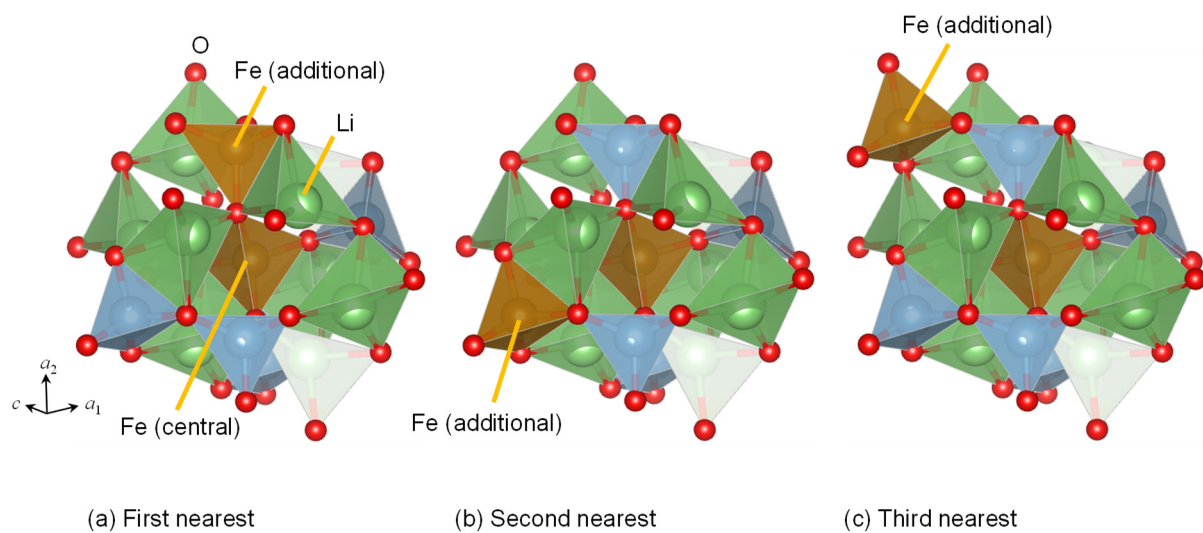


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