

Article

# Metabolite Profiling of *Aquilaria malaccensis* Leaf Extract Using Liquid Chromatography-Q-TOF-Mass Spectrometry and Investigation of Its Potential Antilipoxygenase Activity In-Vitro

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**Abstract:** The *Aquilaria malaccensis* species of the genus *Aquilaria* is an abundant source of agarwood resin and many bioactive phytochemicals. Recent data regarding the chemical constituents and biological activities of *Aquilaria* leaves led us to attempt to qualitatively profile the metabolites of *Aquilaria malaccensis* leaves from a healthy, noninoculated tree through phytochemical screening, GC-MS, and LC/Q-TOF-MS. The present work is also the first to report the antilipoxygenase activity of *A. malaccensis* leaves from healthy noninoculated tree and investigate its toxicity on oral mucosal cells. A total of 53 compounds were tentatively identified in the extract, some of which have been described in literature as exhibiting anti-inflammatory activity. A number of compounds were identified for the first time in the extract of *A. malaccensis* leaf, including quercetin, quercetin-*O*-hexoside, kaempferol-*O*-dirhamnoside, isorhamnetin-*O*-hexoside, syringetin-*O*-hexoside, myricetin, tetrahydroxyflavanone, hesperetin, sissotrin, and lupeol. The antilipoxygenase assay was used to determine the lipoxygenase (LOX) inhibitory potential of the extract, while a WST-1 assay was conducted to investigate the effect of the extract on oral epithelial cells (OEC). The extract implied moderate anti-LOX activity with IC<sub>50</sub> value of 71.6 µg/mL. Meanwhile, the cell viability of OEC ranged between 92.55% (10 µg/mL)–76.06% ± (100 µg/mL) upon treatment, indicating some potential toxicity risks. The results attained encourage future studies of the isolation of bioactive compounds from *Aquilaria malaccensis* leaves, as well as further investigation on the anti-inflammatory mechanisms and toxicity associated with their use.

**Keywords:** *Aquilaria malaccensis*; agarwood; extract; LCMS; GCMS; anti-inflammatory; toxicity; OEC; WST-1

## 1. Introduction

*Aquilaria* (Thymelaeaceae) is an agarwood-producing genus that includes 21 species [1]. The *Aquilaria* species predominating in the literature are *Aquilaria malaccensis* (synonym to *Aquilaria agallocha*),

*Aquilaria sinensis*, and *Aquilaria crassna* [2]. The *A. malaccensis* tree is geographically distributed mainly in Indomalaysian rainforests, and is known for its production of agarwood resin. Its economically valuable fragrant resin led to illegitimate and irrational harvesting of the *Aquilaria* tree, causing a severe decline in their natural population. The tree is labeled as endangered, and is currently protected under CITES regulations and the IUCN Red List [1].

In addition to its commercial value, different parts of the tree have been investigated for their therapeutic effects [2]. *Aquilaria* leaves are consumed as a herbal tea drink in Indonesia [3], and were traditionally used in the treatment of trauma, constipation, headache, elevated blood pressure, and diabetes [3,4]. Scientific information on the bioactivities of *A. malaccensis* leaves in particular reported their ability to exhibit antimicrobial [3,5–7], antioxidant [3,6,8], antidiabetic [6], antiarthritic [9], hepatoprotective [10], antitrypanosomal [11], and potential anti-inflammatory [12] effects. Moreover, the leaf extract demonstrated the potential to improve fertility and induce embryogenesis [13], as well as increasing healthy sperm production [14]. Furthermore, a recent study [15] indicated the role of *A. malaccensis* leaves in the treatment of immune suppression-related disorders. The chemical constituents of the leaves of *A. malaccensis* are reportedly compounds like phenolic acids, benzophenones, xanthonoids, flavonoids, terpenoids, steroids, quinones, lignans, and fatty acids [2], in addition to hydroxyacetanilide [16]. The presence of multiple bioactive constituents in the leaves of *A. malaccensis* earned it the potential to become a source for the development of natural therapy.

Inflammation is an adaptive response to noxious stimuli and conditions in order to protect the body tissues through a cascade of complex events that involves cellular and vascular reactions [17]. Yet, chronic unresolved inflammation leads to various chronic diseases such as Alzheimer's, cardiovascular disorders, and cancer [18]. Many types of herbs have been used traditionally for the treatment and management of inflammatory disorders [19]. Recently *A. malaccensis* leaves have been explored for their anti-inflammatory activity, and were reported to inhibit albumin denaturation in vitro [9,12]. However, other anti-inflammatory pathways ought to be investigated.

*A. sinensis* and *A. crassna* leaves have demonstrated that they can be consumed orally as herbal drinks without major toxicity concerns [4]. An *A. malaccensis* leaf extract was tested for its oral acute toxicity on Wistar albino rats and was found to be safe [15]. Nevertheless, additional studies are needed to gain a better understanding of the extract's toxicity.

The diverse pharmacological activities of *A. malaccensis* leaves have drawn increasing attention from researchers worldwide, and prompted the authors of this paper to characterize the ethanolic extract of *A. malaccensis* leaves from a healthy noninoculated tree using gas chromatography-mass spectroscopy (GC-MS) and high resolution liquid chromatography-quadrupole time-of-flight-mass spectrometry (LC/Q-TOF-MS), in order to evaluate its toxicity and potential antilipoxygenase activity.

## 2. Materials and Methods

### 2.1. Sample, Chemicals, and Reagents

Fresh leaves of a healthy noninoculated *Aquilaria malaccensis* tree were collected from a local plantation in Semenyih, Selangor. The leaves were identified according to their morphology and voucher specimen deposited at the KAED Herbarium at the International Islamic University Malaysia. Analytical grade (95%) ethanol was purchased from Fisher Chemical. Methanol, Acetonitrile, and formic acid were provided by Sigma-Aldrich. Deionized water was used for LC/Q-TOF-MS analysis. Sodium phosphate buffer, soyabean lipoxygenase enzyme, sodium linoleic acid, Nordihydroguaiaretic acid (NDGA), and dimethyl sulfoxide (DMSO) were provided by FRIM (Forest Research Institute Malaysia). The OEC (abm<sup>®</sup> Cat. Number: T4074), WST-1 reagent and DMEM complete media were obtained from Nawah Scientific, Egypt.

## 2.2. Sample Preparation

Leaves of *A. malaccensis* were washed and dried in a drying chamber at a temperature of 40 °C overnight. The sample was ground into a powder, 50 g of which was extracted with 95% ethanol. Excess ethanol was evaporated under reduced pressure at 40 °C using a rotary evaporator (Heidolph-instruments, Rotavapor, Germany). Dried ethanolic extract (3.25 g) was stored in a dry container until further analysis.

## 2.3. Phytochemical Screening

Qualitative chemical screening was conducted to test the presence of organic compounds such as alkaloids, phenols, tannins, steroids, triterpenoids, flavonoids, and saponins. Phytochemical screening tests were done using the procedures provided by [20]. The results of the phytochemical screening can provide information about the potential therapeutic uses of the sample.

## 2.4. GC-MS

The volatile constituents of the sample were analyzed using a gas chromatography system, the Agilent 7890A (Agilent Technologies Inc., Santa Clara, CA, USA) coupled with an Agilent 5975C quadrupole mass spectrometer and an auto sampler Hewlett Packard HP-5MS ultra inert silica capillary column (30 m × 0.25 mm; 0.25 μm) [21]. The detected peaks from the total ion chromatography (TIC) and mass chromatograms were identified based on National Institute of Standards and Technology (NIST) 2008 mass spectral library.

## 2.5. LC/Q-TOF-MS

Samples were submitted to the Proteomics and Metabolomics lab, Cancer Children Hospital Egypt (CCHE) 57357, Cairo, Egypt. The sample was prepared by dissolving 50 mg of the extract in 1 mL mobile phase working solution (MP-WS), i.e., Deionized Water, Methanol, and Acetonitrile in a ratio of 50:25:25 respectively. The mixture was vortexed for 2 min followed by ultra-sonication for 10 min, and centrifuged for 5 min at 10,000 rpm. Then, 20 μL stock (50/1000 μL) was diluted with 1000 μL reconstitution solvent. Finally, the injected concentration was 1 μg/μL; 25 μL of the sample was injected on the positive and negative mode along with injection of 25 μL MP-WS as blank sample. The constituents of the extract were analyzed by using LC/Q-TOF-MS in both positive and negative modes. The analysis was carried out using a system equipped with a column (Waters, Xbridge C-18, 50 mm × 2.1 mm, 3 μm particle size) operated at 40 °C, and precolumn (In-Line filter disks, Phenomenex, 0.5 μm × 3.0 mm). The mobile phases were solvent A (Deionized water containing 0.1% Formic acid) for positive mode, solvent B (5 mM ammonium formate buffer pH8 containing 1% methanol) for negative mode, and solvent C (100% acetonitrile) for both positive and negative modes. The following gradient elution at a rate of 0.3 mL/min was used: isocratic 90% A or B and 10% C (0–1 min), linear from 90% to 10% A or B to 10% to 90% C (1.1–20.9 min), isocratic 10% A or B and 90% C (21–25 min), and finally, isocratic 90% A or B and 10% C (25.1–28 min). The injection volume was 10 μL. Only compounds having PCDL Score > 80 and a mass error within 5 ppm of the known mass of the detected molecules were tentatively identified based on their masses and by comparing their fragmentation patterns obtained by Peakview software (SCIEX, Framingham, MA, USA) with fragmentation patterns of similar compounds in reference literature and in the MassBank i.e., MoNA (Massbank of North America), HMDB (Human Metabolome Database) and NIST (National Institute of Standards and Technology) libraries.

## 2.6. Antilipoxygenase Assay

Lipoxygenase inhibiting activity was measured by slightly modifying the spectrometric method developed by [22]. In the assay protocol, 160 μL of 100mM sodium phosphate buffer (pH 8.0), 10 μL of test-compound solution, and 20 μL of soybean lipoxygenase enzyme solution were mixed and

incubated for 10 min at 25 °C. The enzyme reaction was initiated by the addition of 10 µL sodium linoleic acid (substrate) solution, and was incubated once more in spectrophotometer at 25 °C for 10 min. NDGA was used as reference standard. DMSO was used as negative control. The extract and the reference standard were dissolved in DMSO. All the reactions were performed in triplicate in a 96-well micro-plate, and the absorbance was read at 234 nm in a microplate reader. The percentage inhibition was calculated from the following equation:

$$\% \text{ inhibition} = (\text{O.D of control} - \text{O.D of sample}) / \text{O.D of control} \times 100$$

### 2.7. Toxicity Screening Assay

The cell viability of oral epithelial cells (OEC) was tested by WST-1 assay using Abcam® Kit (ab155902 WST-1 Cell Proliferation Reagent). The assay is a colorimetric assay that measures the cell viability based on the cleavage of tetrazolium salts by mitochondrial dehydrogenase in viable cells. Cells were maintained in DMEM media supplemented with 100 mg/mL of streptomycin, 100 units/mL of penicillin, and 10% (v/v) of heat-inactivated fetal bovine serum. The cells were incubated in 5% CO<sub>2</sub> at 37 °C for 24 h. A stock solution of the extract (1000 mg/mL) was prepared in serum-free media containing 0.1% DMSO (v/v), and serial dilutions of 10 µg/mL and 100 µg/mL were obtained by dilution with serum-free media. The toxicity screening assay was performed by first seeding  $3 \times 10^3$  cells (aliquots of 50 µL cell suspension) in 96-well plate prior to incubation for 24 h. The old media was then discarded and the cells were treated with an aliquot of 50 µL media containing extract in a concentration of 10 µg/mL and 100 µg/mL, respectively. The control group comprised media and 0.1% DMSO. After 48 h, 10 µL WST-1 reagent was added. After 1 h, the absorbance was measured at  $\lambda = 450 \text{ nm}$  using BMG LABTECH®-FLUOstar Omega microplate reader (Allmendgrün, Ortenberg, Germany). The method was adopted from [23,24] with slight modifications. Toxicity results were obtained from three experiments with three well plates for each concentration in a separate 96-well plate. The cell viability was calculated using the equation:

$$\text{Cell viability \%} = [(\text{Abs. of Treatment} - \text{Abs. of Blank}) / (\text{Abs. of Control} - \text{Abs. of Blank})] * 100$$

### 2.8. Statistical Methods

Data are presented as mean  $\pm$  standard deviation of three independent experiments.

## 3. Results and Discussion

### 3.1. Phytochemical Screening

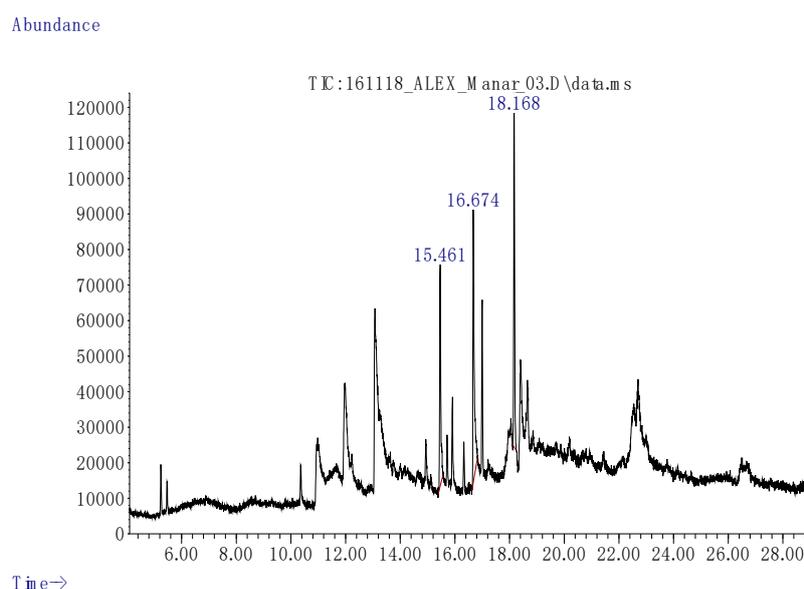
The results presented in Table 1 show that the ethanolic extract of *A. malaccensis* leaves contains flavonoids, saponins, steroids, terpenoids, tannins, and phenols, but is free from alkaloids. The results are consistent with a previous study [25]. However, other research has reported the presence of alkaloids in *A. malaccensis* leaves [7,26,27]. This can be attributed to the different geographical origin of the sample, since environmental factors such as land and weather play a role in phytochemical profile variations [8], or due to the utilization of different extracting solvents or extracting parameters. It has been pointed out that the type and age of organ are among the factors that affect the occurrence of phytochemicals [3]. Phenolic compounds, flavonoids, sterols, and terpenoids have been closely related to anti-inflammatory activity.

**Table 1.** Summary of the phytochemical screening of the ethanolic extract of *A. malaccensis* leaves.

Phytochemical	Test	Result	Observation
Flavonoids	NaOH test	+	Yellow color
	Shinoda test	+	Orange red color
	Pb acetate test	+	Formation of yellow ppt
Saponins	Froth Test	+	Formation of froth
Steroids	Liebermann-Burchard test	+	Change in color
Terpenoids	Salkowski test	+	Reddish brown color
Tannins	Ferric chloride test	+	Dark green color
Alkaloids	Dragendroff's test	–	No reddish brown ppt
	Mayer's test	–	No creamy ppt
Phenols	Ferric chloride test	+	Dark blue color

### 3.2. GC-MS

Table 2 shows the results of the identified compounds in the underivatized crude leaf ethanolic extract. A total of 20 compounds were identified, with the major constituents being the monoterpene (Bicyclo [3.1.1] heptane, 2,6,6-trimethyl-, [1R-(1.alpha.,2.beta.,5.alpha.)]), the acyclic diterpene (Phytol), and the fatty acid (Hexadecanoic acid), as shown in Figure 1. Both phytol and hexadecanoic acid were detected in the GC-MS analysis performed by [26]. In addition, fatty acids, esters, and aliphatic compounds were also observed. Of the identified compounds, hexadecanoic acid and phytol are known for their anti-inflammatory activity, according to [28,29]. Resemblances and diversities occur in comparative GC-MS results in the literature, depending on the species, the extraction method, the extracted part of the tree, and environmental factors. For instance, sterols were not observed in this study, while they were present in the GC-MS analysis of agarwood oil obtained from *A. malaccensis* wood [30]. According to [31], sterols occur in GC-MS analysis after sample derivatization. Squalene has also been identified previously in *Aquilaria* leaves extracts obtained by hydrodistillation, and by methanolic and *n*-hexane extraction [32]. In the GC-MS analysis of a Borneo *A. malaccensis* leaf ethanolic extract performed by [26], the presence of alkaloids predominated, in contrast to the present study, where only traces of imidazole were noted.

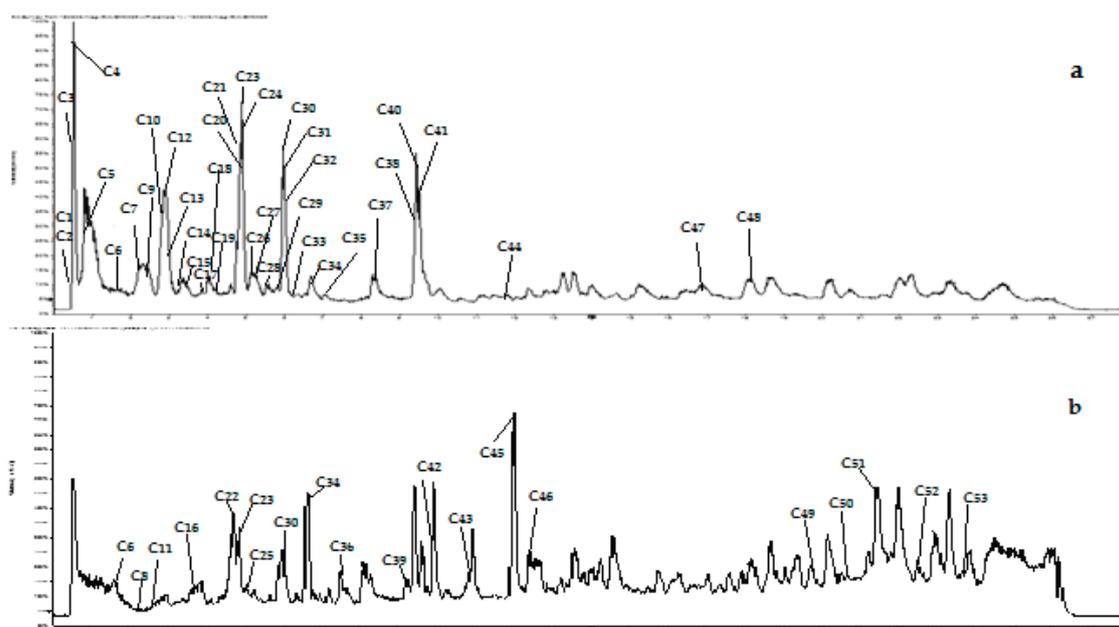
**Figure 1.** GC-MS chromatogram of *A. malaccensis* ethanolic leaf extract.

**Table 2.** Chemical constituents identified in ethanolic extract of *A. malaccensis* by GCMS.

PK	RT	Area Pct	Library/ID	Ref	Qual
1	5.2411	1.7647	3,4-dihydroxy-3,4-dimethylhexane-2,5-dione	41,472	40
2	5.4741	1.1581	N-Hydroxycarbamic acid,2-(isopropoxycarbonylamino)ethyl ester	66,362	25
3	6.8827	0.3698	Trimethylurea	4204	27
4	8.699	0.1852	2R,3S-9-[1,3,4-Trihydroxy-2-butoxymethyl]guanine	131,542	17
5	9.991	0.1905	1H-Imidazole-4-acrylamide	16,605	38
6	10.3564	1.7308	1,1,3,3-Tetramethyl-1,3-disiletane	20,576	38
7	10.6	0.2157	3,4,5,6,7-pentahydroxy-1-phenoxyheptan-2-one	132,320	50
8	10.7853	0.2232	methyl N-amino-N-methylcarbamidithioate	15,931	35
9	10.9601	3.438	N-Butylacetamide	7803	37
10	11.9556	3.5325	4-Cyclopropylmethylbenzotrile	29,225	43
11	13.0729	17.2609	4-Methyldecane	28,433	22
12	14.9368	1.9729	5-Methyl-2-(1-methylethylidene)cyclohexan-1-one oxime	35,916	27
13	15.4611	10.8317	Pinane	17,016	50
14	15.7152	2.0621	1,2-Dipentylcyclopropene	45,727	50
15	15.9112	3.0504	1-Ethynyl-1-cyclohexanol	10,508	38
16	16.3242	2.0009	Methyl hexadecanoate	119,407	93
17	16.6684	17.9539	n-Hexadecanoic acid	107,547	90
18	16.9967	6.1465	Ethyl hexadecanoate	131,288	98
19	18.167	15.9946	Phytol	141,393	91
20	18.4	9.9175	9-octadecenoic acid	129,341	94

### 3.3. LC/Q-TOF-MS

LC/Q-TOF-MS represents an important approach to identifying unknown constituents in plant extracts. The chemical constituents of *A. malaccensis* leaves extracted with 95% ethanol were analyzed via LC/Q-TOF-MS in both positive and negative ionization modes, since flavonoid fragmentations can be recorded on positive and negative ion modes [33]. However, the negative ion mode in the current study provided a wider range of metabolites than the positive ion mode. A total of 53 peaks were tentatively identified as belonging to various classes, including phenolic acids (5), benzophenones (4), flavonoids (27), xanthenes (2), sterols (2), terpenoids (5), coumarins (1), lignans (1), fatty acids (2), and tocopherols (1), in addition to three unknown compounds, as shown in Table 3. The LC/Q-TOF-MS peaks chromatograms of both negative and positive modes are shown in Figure 2.



**Figure 2.** LC/Q-TOF-MS Total Ion Chromatogram (TIC) of the ethanolic extract of *A. malaccensis* leaf on the negative ion mode (a) and positive ion mode (b). Chromatographic conditions are presented in the materials and methods section. The tentatively identified compounds are listed in Table 3.

#### 3.3.1. Phenolic Acids

Prior studies have revealed the existence of multiple phenolic acids in *Aquilaria* leaves [2]. In the present study, four new phenolic acids were introduced regarding the chemical constituents of *Aquilaria* species. The MS/MS spectra of peaks C1 ( $m/z$  133.0152,  $C_4H_5O_5^-$ ), C2 ( $m/z$  117.0182,  $C_4H_5O_4^-$ ), C3 ( $m/z$  153.0199,  $C_7H_5O_4^-$ ), and C5 ( $m/z$  163.0390,  $C_9H_7O_3^-$ ) all exhibited a fragmentation pattern of  $[M-H-CO_2]^-$ , followed by  $[M-H-H_2O]^-$  on negative ionization mode due to the elimination of  $CO_2$  group ( $-44$  Da) and  $H_2O$  molecule ( $-18$  Da) from the deprotonated precursor ion (Supplementary Figures S1–S4). The compounds were identified as Malic acid (C1), Succinic acid (C2), Protocatechuic acid (C3), and Coumaric acid (C5), and their MS/MS spectra were reported by [34]. The glycosylated phenolic acid Glucopyranosyl sinapate (C7) [ $m/z$  385.1896,  $C_{17}H_{21}O_{10}^-$ ] was assigned to peak (C7); its fragmentation was described previously by [35] (Supplementary Figure S5). Protocatechuic acid (C3) was previously identified in *A. sinensis* leaves and demonstrates antidiabetic activity [36].

#### 3.3.2. Benzophenones

A total of four benzophenones were identified in the sample. A benzophenone aglycone occurred at peak C13, while three other benzophenone glycosides occurred at peaks C12, C23, and C30. The precursor ion  $[M-H]^-$  of peak C13 ( $m/z$  245.0472,  $C_{13}H_9O_5^-$ ) produced fragments at  $m/z$

151  $[M-H-C_6H_6O]^-$  and  $m/z$  107  $[M-H-C_6H_6O-CO_2]^-$ , which is a common fragmentation pattern of benzophenones, as mentioned in [37]. The mass, together with its produced fragment ions, suggest that C13 is Iriflophenone [38] (Supplementary Figure S6), which was isolated earlier from *A. sinensis* leaves and has demonstrated significant anti-inflammatory activity [39].

Peaks C12, C23, and C30 presented a significant fragment ion at  $m/z$  245, which is the same as compound C13, leading us to speculate that the three compounds are Iriflophenone derivatives. The product ion of peak C12 ( $m/z$  391.1029,  $C_{19}H_{19}O_9^-$ ) occurred at  $m/z$  245 due to cleavage of deoxyhexose moiety  $[M-H-146]^-$ . This suggests that C12 is possibly Iriflophenone-2-*O*- $\alpha$ -*L*-rhamnopyranoside (Supplementary Figure S7). Similarly, peak C23 ( $m/z$  433.1136,  $C_{21}H_{21}O_{10}^-$ ) exhibited a mass is higher than Iriflophenone by 188 Da, supporting the identification of the compound as Iriflophenone, [2-(2-*O*-actyl-*L*-Rhamnopyranosyl) oxy] (Supplementary Figure S8a). The compound also occurred on the positive ion mode at  $m/z$  435.1296 (Supplementary Figure S8b). Aquilarisinin is another Iriflophenone derivative that was assigned to peak C30 ( $m/z$  553.1350,  $C_{25}H_{29}O_{14}^-$ ). Product ions occurred at  $m/z$  391 and  $m/z$  245 due to consecutive loss of hexose (−162 Da) and deoxyhexose (−146 Da) moiety (Supplementary Figure S9a). The presence of the compound was affirmed by its occurrence at  $m/z$  555.1508 ( $C_{25}H_{31}O_{14}^+$ ) on the positive ion mode (Supplementary Figure S2b). The fragmentation pattern of Aquilarisinin is proposed in Figure 3. It is worth noting that the fragmentation patterns of C12 and C23 were identified earlier in a LC-MS analysis of *A. sinensis* leaves extract performed by [38]. Aquilarisinin was also reported in *A. sinensis* leaves as an  $\alpha$ -glucosidase inhibitor [40].

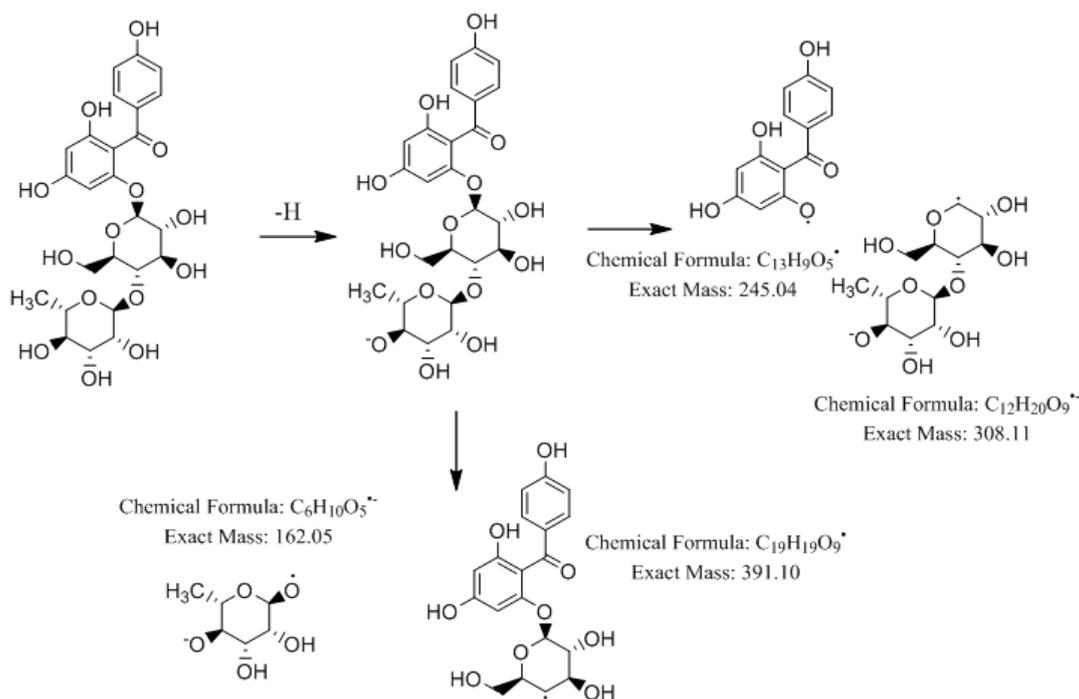
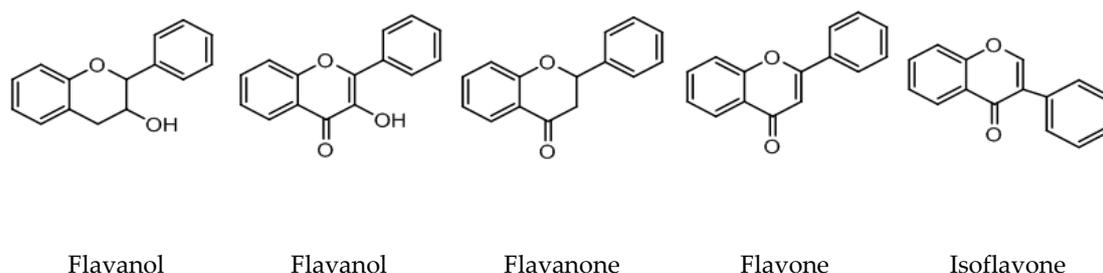


Figure 3. Fragmentation pattern of Aquilarisinin on negative ion mode.

### 3.3.3. Flavonoids

In the ethanolic extract under study, flavonoids occurred predominately over other bioactive compounds. Flavonoids are among the most common phenolic compounds existing in nature. Flavonoid aglycones consist of aromatic rings A and B joined by ring C. The substitution on ring C determines the type of the aglycone, i.e., flavanol, flavonol, flavone, isoflavone, and flavanone, as shown in Figure 4. The common features of flavonoid aglycones fragmentation is the production of fragment ions as a result of the cleavage of  $CH_3$  (−15 Da),  $H_2O$  (−18 Da),  $CO$  (−28 Da),  $CH_2O$  (−30 Da), and  $CO_2$  (−44 Da), and a series of Retro-Diels–Alder (RDA) fragmentation reactions [41].

Fragmentation pathways have been observed in this study of flavonoid derivatives that occur in plants, specifically, flavonoid-*O*-glycosides and flavonoid-*C*-glycosides. In case of flavonoids-*O*-glycosides, the major fragment is produced due to the loss of hexose  $[M\pm H-162]^\pm$ , deoxyhexose  $[M\pm H-146]^\pm$ , dideoxyhexoside  $[M\pm H-130]^\pm$ , or pentose  $[M\pm H-132]^\pm$  sugar moieties, and an ion that represents the aglycone [33]. On the other hand, a flavonoids-*C*-glycosides fragmentation pattern results from the loss of the water molecule  $[M\pm H-18]^\pm$ , cross-ring cleavages of sugar units including  $[M\pm H-120/90]^\pm$  for *C*-hexoside,  $[M\pm H-104/74]^\pm$  for *C*-deoxyhexoside or  $[M\pm H-90/60]^\pm$  for *C*-pentoside, and ions corresponding to the aglycones, plus the remaining sugars [33].



**Figure 4.** Structures of major flavonoid backbones.

- Flavanols

Peak (C6) was identified on both negative and positive ionization modes as Catechin ( $m/z$  289.0739/291.0896  $[M\pm H]^\pm$ ,  $C_{15}H_{13}O_6^-/C_{15}H_{15}O_6^+$ ). In the negative mode, C6 yielded fragment ions at  $m/z$  271  $[M-H-H_2O]^-$  after the loss of 18 Da, and at  $m/z$  245  $[M-H-CH_2CHOH]^-$  after the loss of 44 Da. The produced ion at  $m/z$  179  $[M-H-dihydroxybenzene\ moiety]^-$  was confirmed by the presence of  $m/z$  109  $[M-H-179]^-$ . Also,  $m/z$  137 occurred after the cleavage of ring C by retro-Diels-Alder (RDA) reaction, which was also confirmed by the presence of the ion at  $m/z$  151 [42] (Supplementary Figure S10a). Likewise, the protonated precursor ion was subjected to heterocyclic ring fission (HRF) that resulted in production of fragment ion  $m/z$  147. In addition, the protonated precursor ion was also subjected to heterocyclic ring fission (HRF), benzofuran ring fission (BFF), and Retro-Diels Alder cleavage (RDA), that resulted in the production of ions at  $m/z$  165,  $m/z$  123, and  $m/z$  139 [43] (Supplementary Figure S10b). Catechins are expected to be present in *A. malaccensis* leaves as a major constituent of tea popularly consumed in Asian countries [3]. For instance, catechin derivatives such as epigallocatechin gallate and epicatechin gallate were previously isolated from *A. crassna* leaves [44]. Recent studies have revealed that catechins exhibit anti-inflammatory activity via wide range of mechanisms [45,46].

- Flavonols

Peak C29 ( $m/z$  301.0360,  $C_{15}H_9O_7^-$ ) demonstrated further neutral losses of 28 Da (-CO), 122 Da (-ring B) and 150 Da (-ring B and -CO) to produce the fragment ions of the  $[M-H-CO]^-$  ion at  $m/z$  273,  $[M-H-ring\ B]^-$  ion at  $m/z$  179 and  $[M-H-ring\ B-CO]^-$  ion at  $m/z$  151. Peak C15 ( $m/z$  463.0889,  $C_{21}H_{19}O_{12}^-$ ) demonstrated similar fragmentation pattern as observed in peak C29, but with an additional ion resulting from the loss of the hexoside group (-162 Da). Based on a comparison between this fragmentation pattern and the MS/MS spectra in the previous studies [47], peak C29 was identified as quercetin (Supplementary Figure S11), while peak C15 was identified as quercetin-*O*-hexoside (Supplementary Figure S12). Peak C10 ( $m/z$  461.0711,  $C_{21}H_{17}O_{12}^-$ ) and C33 ( $m/z$  593.1518,  $C_{27}H_{29}O_{15}^-$ ) both produced deprotonated aglycone fragment at  $m/z$  285, which corresponds to kaempferol aglycone. The precursor ion of C10 produced the fragment ions at  $m/z$  417  $[M-H-44]^-$  and  $m/z$  285  $[M-H-176]^-$ ; this was attributed to the loss of  $CO_2$  and glucuronic acid, respectively. C10 is most likely kaempferol-*O*-glucuronide (Supplementary Figure S13), as proposed by [48]. The product ion of C33 occurred at  $m/z$  285  $[M-H-308]^-$ , and the compound was tentatively identified

as kaempferol-*O*-neohesperidoside (Supplementary Figure S14). The fragmentation pattern of the compound was provided by [49]. Peak C22 ( $m/z$  579.1703,  $C_{27}H_{31}O_{14}^+$ ) was identified on the positive ion mode as kaempferol-*O*-dirhamnoside based on the occurrence of ion at  $m/z$  285, which suggests the loss of two deoxyhexose moieties (Supplementary Figure S15). The fragmentation pattern was also reported by [50]. Peaks C18 ( $m/z$  477.1034,  $C_{22}H_{21}O_{12}^-$ ) and C32 ( $m/z$  623.1359,  $C_{28}H_{31}O_{16}^-$ ) exhibited a product ion at  $m/z$  315, which resembles isorhamnetin aglycone. The occurrence of the abundant ion  $[M-H-162]^-$  in the fragmentation pattern of C18 suggests the loss of hexoside moiety. Nevertheless, the fragmentation pattern of C32 showed the major fragment ion at  $[M-H-308]^-$ . Accordingly, these two peaks were identified as isorhamnetin-*O*-hexoside (Supplementary Figure S16) and isorhamnetin-*O*-rutinoside (Supplementary Figure S17), respectively. This identification is supported by the previous suggestion of their fragmentation patterns by [51]. Peak C19 is another flavonol-*O*-hexoside ( $m/z$  507.1489,  $C_{23}H_{23}O_{13}^-$ ) that showed a daughter fragment ion at  $m/z$  345 caused by the cleavage of hexose unit and is corresponding to syringetin aglycone, which suggests that C19 is syringetin-*O*-hexoside [52] (Supplementary Figure S18). The MS/MS spectra of peak C20 ( $m/z$  317.0315,  $C_{15}H_9O_8^-$ ) presented two major fragment ions at  $m/z$  179 [M-ring A] $^-$  and  $m/z$  151 [M-H-ring B] $^-$  which result from typical RDA reaction in flavonol aglycones [41]. A similar fragmentation pattern was observed in [53]; the compound was tentatively characterized as myricetin (Supplementary Figure S19). It is worth noting that the mentioned flavonols were identified in *Aquilaria* for the first time, to the best of our knowledge.

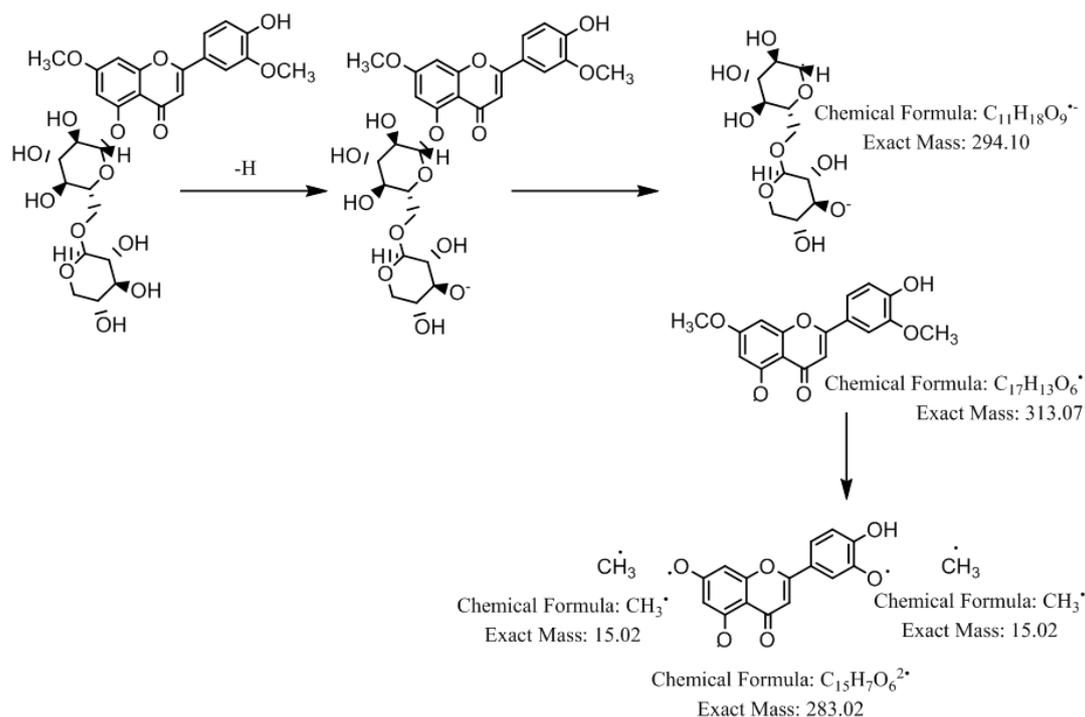
- Flavanones

Three flavanones were detected in the sample, namely, peak C27 ( $m/z$  287.0579,  $C_{15}H_{11}O_6^-$ ), peak C35 ( $m/z$  301.0588,  $C_{16}H_{13}O_6^-$ ), and peak C38 ( $m/z$  285.0774,  $C_{16}H_{13}O_5^-$ ). The tentative identification of C27 as tetrahydroxyflavanone is based on the fragmentation pathway, as shown by [54], that include the cleavage of ring A at  $m/z$  151 and ring B at  $m/z$  135 (Supplementary Figure S20). Hesperetin flavanone was assigned to peak C35, and the precursor ion demonstrated a fragment ion at  $m/z$  286 caused by the loss of one methyl group ( $-15$  Da) (Supplementary Figure S21). The fragmentation pattern of hesperetin was reported by [55,56]. C38 exhibited fragmentation ion at  $m/z$  165 caused by breakdown of ring B in addition to  $m/z$  119 due to the cleavage of ring C followed by another ion at  $m/z$  93 due to subsequent loss of ethynyl group, which suggests that C38 may be sakuranetin (Supplementary Figure S22), which is supported by similar fragmentation pattern displayed by [57]. Sakuranetin was previously reported in *A. sinensis* stem barks to express anti-inflammatory activity through noticeable inhibition of superoxide anion generation and elastase release [58]. The other detected flavanones (i.e., tetrahydroxyflavanone and hesperetin) were detected in *Aquilaria* for the first time.

- Flavones

The flavone aglycone luteolin was assigned to peak C28 ( $m/z$  285.0409,  $C_{15}H_9O_6^-$ ), and was identified based on its fragmentation pattern that showed an ion at  $m/z$  133 caused by the cleavage of ring B (Supplementary Figure S23). The fragmentation pattern of Luteolin was proposed earlier by [57]. Luteolin was formerly isolated from *A. sinensis* flowers [59] and *A. sinensis* leaves, where it revealed potent anti-inflammatory activity [39]. Nevertheless, its glycoside derivative occurred at peak C17 ( $m/z$  447.0946,  $C_{21}H_{19}O_{11}^-$ ) and produced an abundant fragment ion at  $m/z$  285 [M-H-Hexose] $^-$ , which suggests that C17 is luteolin-*O*-hexoside (Supplementary Figure S24), as formerly reported by [51]. C34 was annotated on the negative mode ( $m/z$  269.0473,  $C_{15}H_9O_5^-$ ), and is characterized by the presence of unique fragment ion at  $m/z$  149 characteristic for apigenin aglycone according to [51] (Supplementary Figure S25a). The presence of apigenin is also confirmed on the positive ion mode ( $m/z$  271.0601,  $C_{15}H_{11}O_5^+$ ), whereas its fragmentation pattern matches with the MS/MS spectra of apigenin reported by [60] (Supplementary Figure S25b). Apigenin was isolated earlier from *A. sinensis* flowers [59]. However, the identification of peak C21 ( $m/z$  431.0995,  $C_{21}H_{19}O_{10}^-$ ) relied on the

formation of ion fragments at  $m/z$  311 caused by the loss of 90 Da, which corresponds to the cross-ring cleavage of the sugar unit [33] and suggesting an aglycone-C-hexoside. The occurrence of another fragment ion at  $m/z$  269 (apigenin aglycone) is caused by the loss of hexose moiety (-162 Da). Based on the above information, C21 was tentatively identified as apigenin-C-hexoside, known as Vitexin (Supplementary Figure S26). Vitexin was identified earlier in *A. sinensis* leaves [4]; its fragmentation pattern was reported in literature by [57]. Peak C39 was identified on positive ionization mode ( $m/z$  285.0748,  $C_{16}H_{13}O_5^+$ ), and exhibits similar fragmentation pattern of peak C34 with extra  $CH_3$  group (15 Da), which suggests that the compound is methyl apigenin, known as acacetin (Supplementary Figure S27). The fragmentation pattern of acacetin was previously identified by [56]. Furthermore, peak C11 ( $m/z$  593.1263,  $C_{28}H_{33}O_{14}^+$ ) on the positive mode showed a neutral loss of 146 Da which is caused by the cleavage of deoxyhexose unit to give the fragment ion with  $m/z$  477, followed by a loss of hexose unit (-162 Da) to produce the fragment ion with  $m/z$  285, which suggest a disaccharide structure. The molecular weight of the compound after the elimination of two sugar units matches with that of acacetin, leading us to conclude that C11 is diglycosylated acacetin (acacetin-*O*-rutinoside) (Supplementary Figure S28) [56]. Likewise, peak C41 ( $m/z$  313.0726,  $C_{17}H_{13}O_6^-$ ) upon fragmentation gave an ion at  $m/z$  283; after the loss of two methyl groups, it was identified as velutin (Supplementary Figure S29). The fragmentation pattern of velutin was previously reported by [61], and the compound was recognized in *A. sinensis* stem barks and demonstrated potent anti-inflammatory activity [58]. The MS spectrum of peak C24 ( $m/z$  607.1660,  $C_{28}H_{31}O_{15}^-$ ) exhibited highly significant peak at  $m/z$  313, which indicates the loss of 294 Da, which can be attributed to the cleavage of two pentose moieties. In contrast, the fragmentation ion at  $m/z$  313 generated the product ions  $m/z$  298 (- $CH_3$ ) and  $m/z$  283 (- $2CH_3$ ), which is in agreement with the fragmentation of velutin. Such fragmentation suggests that C24 is a flavonoid glycoconjugate, whereas the aglycone is velutin. The compound is annotated as aquisiflavoside, which was isolated previously from *A. sinensis* leaves and was reported as potent nitric oxide inhibitor [62]. The fragmentation pattern of tentatively identified aquisiflavoside (Supplementary Figure S30) is introduced for the first time, as shown in Figure 5. Peak C40 ( $m/z$  283.0617,  $C_{16}H_{11}O_5^-$ ), showing a fragment ion at  $m/z$  268 [M-H- $CH_3$ ] $^-$ , was assigned to genkwanin (Supplementary Figure S31), as previously reported in different *Aquilaria* species [63,64], while peak C37 ( $m/z$  299.0561,  $C_{16}H_{11}O_6^-$ ) expressed the presence of extra hydroxyl group to compound 40, which suggests that C37 is Hydroxygenkwanin (Supplementary Figure S32). The fragmentation patterns of both peak C40 and C37 were previously reported in the ms/ms analysis of *A. sinensis* leaves [38]. Peak C26 ( $m/z$  445.1136,  $C_{21}H_{17}O_{11}^-$ ) produced a fragment ion at  $m/z$  268 [M-2H-176] caused by the loss of the glucuronide group, as reported by [65], and was nominated as baicalein-*O*-glucuronide (Supplementary Figure S33). Violanthin is a C-glycoside flavone that has been lately isolated from *A. malaccensis* leaves, and which has revealed antioxidant activity [66]. Violanthin was assigned to peak C44 ( $m/z$  577.2710,  $C_{27}H_{29}O_{14}^-$ ) with the molecular weight of violanthin (578.2 g/mol) as reported in the literature, and a fragment ion produced at  $m/z$  268 (Supplementary Figure S34).



**Figure 5.** Fragmentation pattern of Aquisiflavoside on negative ion mode.

- Isoflavones

Two isoflavones were detected on the positive mode ionization. Peak C25 ( $m/z$  447.1280,  $C_{22}H_{23}O_{10}^+$ ) was identified as sissotrin, which produced fragment ions at  $m/z$  285 and  $m/z$  270, caused by loss of the hexose moiety (162 Da), followed by the loss of the methyl group (15 Da) (Supplementary Figure S35), while peak C45 ( $m/z$  269.0810,  $C_{16}H_{13}O_4^+$ ) was assigned to formononetin, and exhibited a fragment ion at  $m/z$  254 after methyl group dissociation (Supplementary Figure S36). Sissotrin is identified in *Aquilaria* species for the first time. Meanwhile, formononetin was isolated earlier from *A. sinensis* stems [67]. The fragmentation patterns of sissotrin and formononetin were reported by [68,69], respectively.

### 3.3.4. Xanthones

Xanthones-C-glycosides were observed in the sample. The MS/MS spectra of peak C4 ( $m/z$  421.0770,  $C_{19}H_{17}O_{11}^-$ ) showed a fragment ion at  $m/z$  403 which may be attributed to the loss of  $H_2O$  molecule (18 Da). The fragment ions  $m/z$  331 [M-H-90] $^-$  and  $m/z$  301 [M-H-120] $^-$  suggest xanthone-C-glycosides compounds. C4 was tentatively identified as mangiferin or isomangiferin (Supplementary Figure S37). Mangiferin was previously identified in *Aquilaria* leaves [36,63,64]. Peak C9 ( $m/z$  435.0979,  $C_{20}H_{19}O_{11}^-$ ) produced a fragment ion at  $m/z$  417 due to the loss of the  $H_2O$  molecule, and was tentatively identified as homomangiferin (Supplementary Figure S38). The MS/MS spectra of C4 and C9 were previously reported in the LC-MS analysis of *A. sinensis* leaves [38].

### 3.3.5. Sterols

Two plant-derived sterols were detected in the sample on the positive ionization mode. The characteristic MS/MS spectra that have been obtained from free sterols occur as [M+H-H<sub>2</sub>O] $^+$  [70].

Peaks C50 ( $m/z$  415.3518,  $C_{29}H_{51}O^+$ ) and C51 ( $m/z$  413.3785,  $C_{29}H_{49}O^+$ ) yielded significant fragmentation ions at  $m/z$  397 and  $m/z$  395 respectively, which was attributed to the loss of the  $H_2O$  molecule (18 Da). The detected phytosterols were identified as  $\beta$ -sitosterol (Supplementary Figure S39) and stigmasterol (Supplementary Figure S40). The fragmentation patterns of C50 and C51

are reported in [71].  $\beta$ -sitosterol was formerly isolated from *A. sinensis* stem barks and evaluated for its anti-inflammatory activity [58], and was also present in the GC-MS profile of the methanolic extract of *A. sinensis* leaves [32]. However, it was absent in the GC-MS analysis of the methanolic extract of *A. malaccensis* leaves [32]. Stigmasterol was detected in *A. malaccensis* wood using GC-MS, and is also known for its potent anti-inflammatory activity [30].

### 3.3.6. Terpenoids

The complex organic compounds known as terpenoids consist of five carbon isoprene units, and are classified according to the number of isoprene units into monoterpenes, diterpenes, triterpenes, and sesquiterpenes [72]. Five terpenoids were identified in the extract on the positive ion mode.

Peak C8 ( $m/z$  137.0596,  $C_{10}H_{17}^+$ ) produced a major fragment ion at  $m/z$  122 due to the loss of  $CH_3$  group, and was tentatively annotated as the monoterpene sabinene (Supplementary Figure S41). The fragmentation pattern of sabinene is presented in Figure 6. Phytol diterpene was assigned to peak C36 ( $m/z$  297.1127,  $C_{20}H_{41}O^+$ ), and yielded a fragment ion at  $m/z$  281 (Supplementary Figure S42). Phytol has been identified in the GC-MS run of *A. malaccensis* ethanolic leaf extract, as in Table 2. Peaks C46 ( $m/z$  223.1681,  $C_{15}H_{27}O^+$ ) and 49 ( $m/z$  427.3582,  $C_{30}H_{51}O^+$ ) were tentatively recognized as the sesquiterpenes nerolidol (Supplementary Figure S43) and  $\alpha$ -myrin (Supplementary Figure S44) according to the fragmentation patterns suggested by [71,73] respectively. The triterpenoid Lupeol was assigned to peak C52 ( $m/z$  427.3911,  $C_{30}H_{51}O^+$ ), which yielded a major fragment ion at  $m/z$  409 due to the cleavage of  $H_2O$  molecule ( $-18$  Da) (Supplementary Figure S45) [71]. Lupeol was reported in the literature to exhibit anti-inflammatory and antiarthritic activities [74]. Although the aforementioned terpenoids have not been isolated from *Aquilaria* species to date, sabinene, phytol, and nerolidol were detected previously in the GC-MS profile of an *A. malaccensis* leaf extract [32].

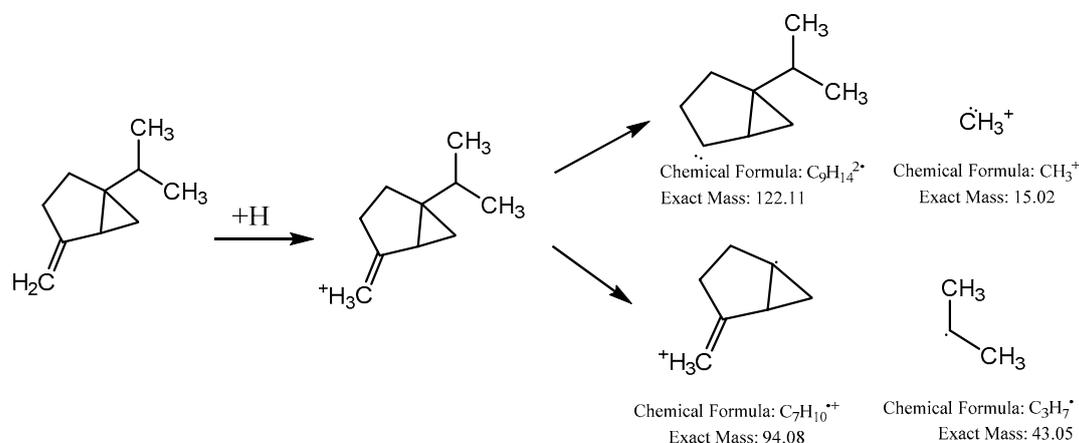


Figure 6. Fragmentation pattern of Sabinene on positive ion mode.

**Table 3.** Metabolites identified in *A. malaccensis* leaf ethanolic extract via LC/Q-TOF-MS in negative (N) and positive (P) ionization modes.

Peak	Mode	RT (min)	M <sup>+/-</sup>	Formula	Error	Fragmentation Pattern	Identification	Class	Literature Data
C1	N	0.452	133.0152	C <sub>4</sub> H <sub>5</sub> O <sub>5</sub> <sup>-</sup>	-2.2	115, 89, 71	Malic acid	Phenolic acid	[34], MoNA
C2	N	0.452	117.0182	C <sub>4</sub> H <sub>5</sub> O <sub>4</sub> <sup>-</sup>	2.9	99, 73	Succinic acid	Phenolic acid	[34], HMDB
C3	N	0.499	153.0199	C <sub>7</sub> H <sub>5</sub> O <sub>4</sub> <sup>-</sup>	-0.7	109, 91, 81	Protocatechuic acid	Phenolic acid	[34], MoNA
C4	N	0.515	421.0770	C <sub>19</sub> H <sub>17</sub> O <sub>11</sub> <sup>-</sup>	1.2	403, 331, 301, 259	Mangiferin/Isomangiferin	Xanthone	[38], MoNA
C5	N	0.659	163.0390	C <sub>9</sub> H <sub>7</sub> O <sub>3</sub> <sup>-</sup>	1.7	147, 119	Coumaric acid	Phenolic acid	[34], HMDB
C6	P	1.673	291.0896	C <sub>15</sub> H <sub>15</sub> O <sub>6</sub> <sup>+</sup>	7.8	165, 147, 139, 123	Catechin	Flavanol	[42], MoNA
	N	1.709	289.0739	C <sub>15</sub> H <sub>13</sub> O <sub>6</sub> <sup>-</sup>	1.1	271, 245, 179, 151, 137, 109			[43], MoNA
C7	N	2.202	385.1896	C <sub>17</sub> H <sub>21</sub> O <sub>10</sub> <sup>-</sup>	-0.6	223, 205, 179	Glucopyranosyl sinapate	Phenolic acid	[35]
C8	P	2.408	137.0596	C <sub>10</sub> H <sub>17</sub> <sup>+</sup>	-0.1	122, 94, 81, 77	Sabinene	Monoterpene	—
C9	N	2.484	435.0979	C <sub>20</sub> H <sub>19</sub> O <sub>11</sub> <sup>-</sup>	0.7	417, 345, 315, 272	Homomangiferin	Xanthone	[38]
C10	N	2.627	461.0711	C <sub>21</sub> H <sub>17</sub> O <sub>12</sub> <sup>-</sup>	0.8	285, 175	Kaempferol-O-Glucuronide	Flavonol	[48], MoNA
C11	P	2.728	593.1263	C <sub>28</sub> H <sub>33</sub> O <sub>14</sub> <sup>+</sup>	0.8	477, 285	Acacetin-O-rutinoside	Flavone	[56], NIST
C12	N	2.864	391.1029	C <sub>19</sub> H <sub>19</sub> O <sub>9</sub> <sup>-</sup>	0.4	245, 151	Iriflophenone-2-O- $\alpha$ -L-rhamnopyranoside	Benzophenone	[38]
C13	N	3.067	245.0472	C <sub>13</sub> H <sub>9</sub> O <sub>5</sub> <sup>-</sup>	3.9	151, 107	Iriflophenone	Benzophenone	[38]
C14	N	3.362	549.1127	—	1.1	531, 115	Unknown	—	—
C15	N	3.424	463.0889	C <sub>21</sub> H <sub>18</sub> O <sub>12</sub> <sup>-</sup>	3.6	301, 271	Quercetin-O-hexoside	Flavonol	[47], HMDB
C16	P	3.681	179.1070	C <sub>9</sub> H <sub>7</sub> O <sub>4</sub> <sup>+</sup>	0.5	161, 146, 133	Dihydroxycoumarin (Esculetin)	Coumarin	[75], HMDB
C17	N	3.807	447.0946	C <sub>21</sub> H <sub>19</sub> O <sub>11</sub> <sup>-</sup>	1.6	285	Luteolin-O-hexoside	Flavone	[56], HMDB
C18	N	4.242	477.1034	C <sub>22</sub> H <sub>21</sub> O <sub>12</sub> <sup>-</sup>	-0.9	315, 271	Isorhamnetin-O-hexoside	Flavonol	[51], MoNA
C19	N	4.282	507.1551	C <sub>23</sub> H <sub>23</sub> O <sub>13</sub> <sup>-</sup>	-0.5	344	Syringetin-O-hexoside	Flavonol	[52], MoNA
C20	N	4.512	317.0315	C <sub>15</sub> H <sub>9</sub> O <sub>8</sub> <sup>-</sup>	0.7	179, 151	Myricetin	Flavonol	[53], HMDB
C21	N	4.550	431.0995	C <sub>21</sub> H <sub>19</sub> O <sub>10</sub> <sup>-</sup>	-0.7	311, 269	Apigenin-C-hexoside (Vitexin)	Flavone	[57], MoNA
C22	P	4.664	579.1703	C <sub>27</sub> H <sub>31</sub> O <sub>14</sub> <sup>+</sup>	-1.7	285	Kaempferol-O-dirhamnoside	Flavonol	[50]
C23	P	4.820	435.1296	C <sub>21</sub> H <sub>23</sub> O <sub>10</sub> <sup>+</sup>	-1.7	247	Iriflophenone, [2-(2-O-actyA-L-Rhamnopyranosyl) oxy]	Benzophenone	—
	N	4.833	433.1136	C <sub>21</sub> H <sub>21</sub> O <sub>10</sub> <sup>-</sup>	0.3	245, 151			[38]
C24	N	4.954	607.1660	C <sub>28</sub> H <sub>31</sub> O <sub>15</sub> <sup>-</sup>	3.4	313, 298	Aquisiflavoside	Flavone	—
C25	P	5.022	447.1280	C <sub>22</sub> H <sub>23</sub> O <sub>10</sub> <sup>+</sup>	0.6	285, 270	Sissotrin	Isoflavone	[68], NIST
C26	N	5.097	445.1136	C <sub>21</sub> H <sub>17</sub> O <sub>11</sub> <sup>-</sup>	0.2	268	Baicalein-O-glucuronide	Flavone	[65], MoNA

Table 3. Cont.

Peak	Mode	RT (min)	M <sup>+/-</sup>	Formula	Error	Fragmentation Pattern	Identification	Class	Literature Data
C27	N	5.214	287.0579	C <sub>15</sub> H <sub>11</sub> O <sub>6</sub> <sup>-</sup>	-0.1	151, 135, 107	tetrahydroxyflavanone	Flavanone	[54]
C28	N	5.565	285.0409	C <sub>15</sub> H <sub>9</sub> O <sub>6</sub> <sup>-</sup>	1.1	133	Luteolin	Flavone	[57], HMDB
C29	N	5.728	301.0360	C <sub>15</sub> H <sub>9</sub> O <sub>7</sub> <sup>-</sup>	-0.9	273, 179, 151	Quercetin	Flavonol	[47], HMDB
C30	N	5.895	553.1350	C <sub>25</sub> H <sub>29</sub> O <sub>14</sub> <sup>-</sup>	-0.9	391, 245, 151	Aquilarisinin	Benzophenone	—
	P	5.939	555.1508	C <sub>25</sub> H <sub>31</sub> O <sub>14</sub> <sup>+</sup>	0.7	291, 247, 121, 83			
C31	N	6.094	417.1582	C <sub>22</sub> H <sub>25</sub> O <sub>8</sub> <sup>-</sup>	-2.6	402, 181, 166, 151	Syringaresinol	Lignan	[76], MoNA
C32	N	6.167	623.1359	C <sub>28</sub> H <sub>31</sub> O <sub>16</sub> <sup>-</sup>	3	315	Isorhamnetin-O-rutinoside (narcissoside)	Flavonol	[51], MoNA
C33	N	6.254	593.1518	C <sub>27</sub> H <sub>29</sub> O <sub>15</sub> <sup>-</sup>	0.8	299, 285	Kaempferol-O-neohesperidoside	Flavonol	[49], MoNA
C34	N	6.499	269.0473	C <sub>15</sub> H <sub>9</sub> O <sub>5</sub> <sup>-</sup>	0.1	151, 149, 117	Apigenin	Flavone	[51], MoNA
	P	6.638	271.0601	C <sub>15</sub> H <sub>11</sub> O <sub>5</sub> <sup>+</sup>	3.8	229, 153			
C35	N	6.931	301.0588	C <sub>16</sub> H <sub>13</sub> O <sub>6</sub> <sup>-</sup>	0.2	286, 255	Hesperetin	Flavanone	[55,56], MoNA
C36	P	8.019	297.1127	C <sub>20</sub> H <sub>41</sub> O <sup>+</sup>	1.9	281, 265, 248	Phytol	Diterpene	MoNA
C37	N	8.352	299.0561	C <sub>16</sub> H <sub>11</sub> O <sub>6</sub> <sup>-</sup>	-4.3	284, 256	Hydroxygenkwanin	Flavone	[38], HMDB
C38	N	9.302	285.0774	C <sub>16</sub> H <sub>13</sub> O <sub>5</sub> <sup>-</sup>	4.6	165, 119, 93	Sakuranetin	Flavanone	[57], HMDB
C39	P	9.310	285.0748	C <sub>16</sub> H <sub>13</sub> O <sub>5</sub> <sup>+</sup>	1.1	270, 242, 176, 124	Acacetin	Flavone	[56] MoNA
C40	N	9.445	283.0617	C <sub>16</sub> H <sub>11</sub> O <sub>5</sub> <sup>-</sup>	2.5	268	Genkwanin	Flavone	[38], MoNA
C41	N	9.683	313.0726	C <sub>17</sub> H <sub>13</sub> O <sub>6</sub> <sup>-</sup>	1.8	298, 283, 270, 255	Velutin	Flavone	[61], MoNA
C42	P	9.891	307.2488	—	2.3	289, 271, 220	Unknown	—	—
C43	P	10.856	329.1022	—	-0.4	313, 285	Unknown	—	—
C44	N	11.757	577.2710	C <sub>27</sub> H <sub>29</sub> O <sub>14</sub> <sup>-</sup>	-0.3	268	Violanthin	Flavone	MoNA
C45	P	11.878	269.0810	C <sub>16</sub> H <sub>13</sub> O <sub>4</sub> <sup>+</sup>	1.3	254, 197	Formononetin	Isoflavone	[69]
C46	P	12.302	223.1681	C <sub>15</sub> H <sub>27</sub> O <sup>+</sup>	-0.6	207, 191, 177, 149	Nerolidol	Sesquiterpenes	[73], MoNA
C47	N	16.856	279.2330	C <sub>18</sub> H <sub>31</sub> O <sub>2</sub> <sup>-</sup>	4	261, 234	Linoleic acid	Fatty acid	MoNA
C48	N	17.976	255.2334	C <sub>16</sub> H <sub>31</sub> O <sub>2</sub> <sup>-</sup>	-0.9	237	Hexadecanoic acid	Fatty acid	MoNA
C49	P	19.775	427.3582	C <sub>30</sub> H <sub>51</sub> O <sup>+</sup>	-3	409, 367, 217, 191	$\alpha$ -amyrin	Sesquiterpenes	[71]
C50	P	20.870	415.3518	C <sub>29</sub> H <sub>51</sub> O <sup>+</sup>	1	397, 189, 161, 119	$\beta$ -sitosterol	Sterols	[71], HMDB
C51	P	21.572	413.3785	C <sub>29</sub> H <sub>49</sub> O <sup>+</sup>	-1.7	395, 255, 159	Stigmasterol	Sterols	[71], HMDB
C52	P	22.407	427.3911	C <sub>30</sub> H <sub>51</sub> O <sup>+</sup>	-1.6	409, 185	Lupeol	Triterpenoid	[71], NIST
C53	P	23.782	431.3790	C <sub>29</sub> H <sub>51</sub> O <sub>2</sub> <sup>+</sup>	0.7	165	$\alpha$ -Tocopherol	Tocopherols	[77], NIST

MoNA: MassBank of North America, NIST: National Institute of Standards and Technology, HMDB: Human Metabolome Database.

### 3.3.7. Coumarins

Peak C16 ( $m/z$  179.1070,  $C_9H_7O_4^+$ ) initially produced the ion at  $m/z$  161 due loss of  $H_2O$  molecule ( $-18$  Da) followed by a fragment ion at  $m/z$  133 caused by the dissociation of another  $H_2O$  and CO molecules. The presence of the fragment ion at  $m/z$  146, which is the similar molecular weight of coumarin (146 g/mol) in the literature, suggests that the compound is a coumarin derivative. Accordingly, it can be anticipated that the compound is dihydroxycoumarin, known as esculetin [75] (Supplementary Figure S46).

### 3.3.8. Lignans

The MS/MS spectra of peak C31 ( $m/z$  417.1753,  $C_{22}H_{25}O_8^-$ ) showed fragmentation ions at  $m/z$  402 and  $m/z$  387 due to the successive loss of two methyl groups. A major fragment ion also occurred at  $m/z$  181, representing the syringyl moiety, which subsequently lost two  $CH_3$  molecules, thereby forming a guaiacyl moiety at  $m/z$  151. The demonstrated fragmentation pattern matches with the fragmentation pattern of syringaresinol [76]. Hence, C31 is possibly syringaresinol (Supplementary Figure S47). The lignan syringaresinol has been determined in *A. sinensis* stem barks, and has been shown to demonstrate moderate anti-inflammatory activity [58].

### 3.3.9. Fatty Acids

Two fatty acids, namely, linoleic and hexadecanoic fatty acids, were detected in the present sample. Linoleic acid was assigned to peak C47 ( $m/z$  279.2330,  $C_{18}H_{31}O_2^-$ ) (Supplementary Figure S48), while Hexadecanoic acid was allocated to peak C48 ( $m/z$  255.2334,  $C_{16}H_{31}O_2^-$ ) (Supplementary Figure S49). Both compounds yielded major fragment ions due to the breakdown of  $H_2O$  molecules at  $m/z$  261 and  $m/z$  237 respectively. Both linoleic acid and hexadecanoic acid appeared in the GC-MS analysis of *A. malaccensis* leaves [32]. Hexadecanoic acid was also recognized in the GC-MS analysis of *A. malaccensis* leaves in the present study, as shown in Table 2.

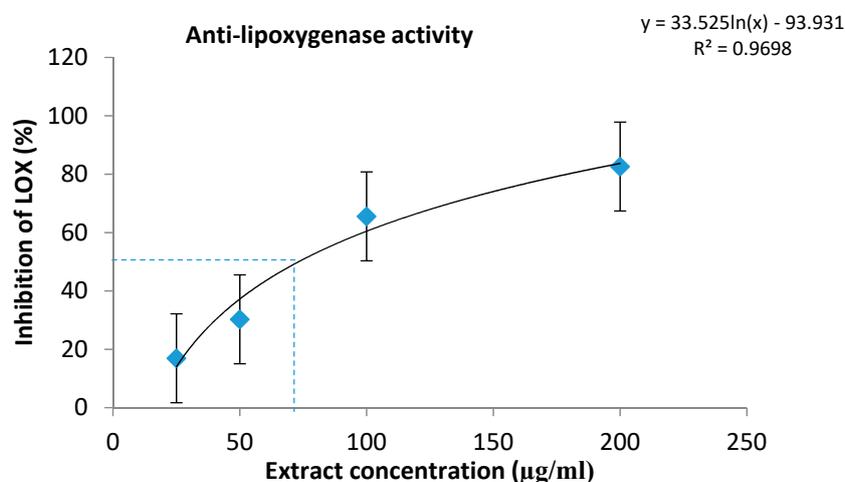
### 3.3.10. Tocopherols

Peak C53 ( $m/z$  431.3790,  $C_{29}H_{51}O_2^+$ ) was designated to  $\alpha$ -Tocopherol with a major fragment ion at  $m/z$  165, caused by the cleavage of the tocopherol phytyl tail (Supplementary Figure S50), which is consistent with its fragmentation pattern presented by [77].  $\alpha$ -Tocopherol, which has been previously identified in *A. sinensis* stem bark, has potential as an anti-inflammatory agent [58].

## 3.4. The Antilipoxygenase Assay

The anti-inflammatory potential of the *A. malaccensis* ethanolic extract was evaluated using the inhibition percentage of lipoxygenase enzyme in correlation to its  $IC_{50}$  value. The inhibition of lipoxygenase activity was demonstrated in a dose-dependent manner, whereas the maximum lipoxygenase inhibition occurred at a dose of 200  $\mu\text{g/mL}$  ( $82.61\% \pm 1.44\%$ ). Data were exported to MS Excel 2010 and the  $IC_{50}$  was determined. The extract was able to inhibit the lipoxygenase enzyme (LOX) with an  $IC_{50}$  of  $71.6 \pm 5.44$   $\mu\text{g/mL}$  (Figure 7). The lipoxygenase enzyme converts the Arachidonic acid in the body cells to hydro peroxy eicosa tetraenoic acids (HPETEs), which are then reduced to leukotrienes that are considered to be among the most powerful inflammatory mediators [78,79]. Lipoxygenase enzyme is associated with several inflammatory disorders such as asthma, cancer, atherosclerosis, obesity, diabetes, and neurodegenerative diseases [78,79]. Thus, the antilipoxygenase activity of the extract can be exploited to combat a range of inflammation-related diseases. In a previous study, 160 Malaysian plant extracts were screened for lipoxygenase inhibiting effects at 100  $\mu\text{g/mL}$ ; those with more than 70% inhibition were classified as highly active against lipoxygenase, those showing 41–70% inhibition as having moderate activity, and those with 0–40% as having low activity [80]. Accordingly, *A. malaccensis* leaf extract activity, with  $65.56\% \pm 2.75\%$  inhibition of lipoxygenase at a dose of 100  $\mu\text{g/mL}$ , suggests that the extract exhibits potential moderate antilipoxygenase activity in a

concentration-dependent manner. To best of our knowledge, no previous studies have been performed on the antilipoxygenase inhibitory potential of *A. malaccensis* extract.



**Figure 7.** Antilipoxygenase (anti-LOX) activity of *A. malaccensis* leaf extract ( $IC_{50}$ :  $71.6 \pm 5.44$   $\mu\text{g/mL}$ ). LOX inhibition of the positive control (NDGA) at 100  $\mu\text{g/mL}$  was  $98.21\% \pm 1.26\%$ .  $IC_{50}$  of NDGA:  $2.05 \pm 0.46$   $\mu\text{g/mL}$ .

Based on the research results of phytochemical screening, GC-MS, and LC/Q-TOF-MS, the ethanolic extract of *A. malaccensis* leaf comprises diverse phytochemicals. The anti-inflammatory activity exhibited by the extract can be attributed to several bioactive constituents detected which are expected to inhibit pro-inflammatory mediators and enzymes such as lipoxygenase enzyme. As discussed, compounds such as iriflophenone (C13), catechin (C6), sakuranetin (C38), luteolin (C28), velutin (C41), aquisiflavoside (C24),  $\beta$ -sitosterol (C50), stigmasterol (C51), lupeol (C52), phytol, hexadecanoic acid, syringarsinol (C31), and  $\alpha$ -tocopherol (C53) have been previously shown to exhibit anti-inflammatory activity through varied mechanisms. However, no scientific report exists on the identified compounds as possible specific inhibitors of lipoxygenase enzyme.

### 3.5. Toxicity Screening Assay

A WST-1 assay was used to evaluate the toxicity of *A. malaccensis* leaf ethanolic extract on oral epithelial cells (OECs). WST-1 (4-[3-(4-iodophenyl)-2-(4-nitrophenyl)-2H-5-tetrazolio]-1,3-benzene disulfonate) is a colorimetric assay that is commonly used to evaluate the toxicity of substances on mammalian cells through a reaction with the mitochondrial succinate-tetrazolium reductase forming the formazan dye in viable cells [81]. The assay is a rapid, single step assay, whereas WST-1 reagent forms soluble formazan in contrast to the MTT assay which forms insoluble formazan crystals [81]. The OECs are noncancerous cell lines derived from human oral squamous cells. The cells were selected for the present assay to represent normal cells that may be exposed to external pathogens which can benefit from anti-inflammatory therapeutic agents. Based on this preliminary finding, the cells viability % obtained from WST-1 assay falls between  $92.55\% \pm 1.26\%$  at a dose of 10  $\mu\text{g/mL}$  to  $76.06\% \pm 1.16\%$  at a dose of 100  $\mu\text{g/mL}$ . The reduction of cell viability in the preliminary screening could suggest the toxicity effects of the leaf extract. The concentration of the extract (100  $\mu\text{g/mL}$ ) that is able to exert moderate inhibition of lipoxygenase level causes a reduction in cell viability by 24%. The implication of this finding suggests that although the extract has potential anti-inflammatory activity, its use should be cautioned, as it has shown some toxicity effects. Further research is warranted to investigate the potential toxicity of the leaf extract.

#### 4. Conclusions

Based on the present research, LC/Q-TOF-MS analysis successfully identified previously-isolated and newly-identified compounds in *A. malaccensis* leaves from a healthy noninoculated tree, comprising phenolic acids, benzophenones, flavonoids, xanthenes, sterols, terpenoids, coumarins, lignans, fatty acids, and tocopherols. A number of compounds were identified for the first time in *A. malaccensis* leaves, including quercetin, quercetin-*O*-hexoside, kaempferol-*O*-dirhamnoside, isorhamnetin-*O*-hexoside, syringetin-*O*-hexoside, Myricetin, tetrahydroxyflavanone, hesperetin, sissotrin, and lupeol. The present work is the first to report on the antilipoxygenase activity of *A. malaccensis* leaves from a healthy noninoculated tree, and to investigate its toxicity on oral mucosal cells. The results from the biological assays suggest that the *A. malaccensis* leaf extract exhibits moderate antilipoxygenase activity and shows some toxicity risks regarding OEC. Future studies should be directed toward the bio-guided isolation and purification of compounds, as well as the plant mechanistic anti-inflammatory role to ensure its safety for human consumption.

**Supplementary Materials:** The following are available online at <http://www.mdpi.com/2227-9717/8/2/202/s1>, Supplementary Figures S1–S50.

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