

Optical and Thermal Investigations of Eutectic Metallomesogen Mixtures Based on Salicylaldiaminates Metal Complexes with a Large Nematic Stability Range

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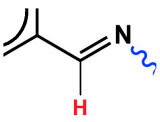
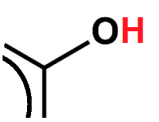
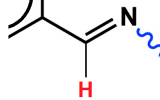
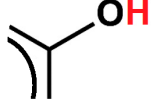
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Supporting Information

Table S1. More relevant ¹H NMR and IR data for ligand and related metal complexes.

Compound				
	δ /ppm ^a	δ /ppm ^b	wavenumber / cm ⁻¹ ^c	wavenumber / cm ⁻¹ ^d
12-8N	3.68	10.54	1718	3422
12-8NCu			1726	-
A11O-6ON	3.68	10.54	1708	3431
A11O-6ONNi	4.11	-	1728	-
A11O-6ONPd	4.13	-	1735	-

^a TMS chemical shift in ppm of proton of imine; ^b TMS chemical shift in ppm of phenolic proton; ^c IR wavenumber in cm⁻¹ of proton of imine; ^d TMS chemical shift in ppm of phenolic proton

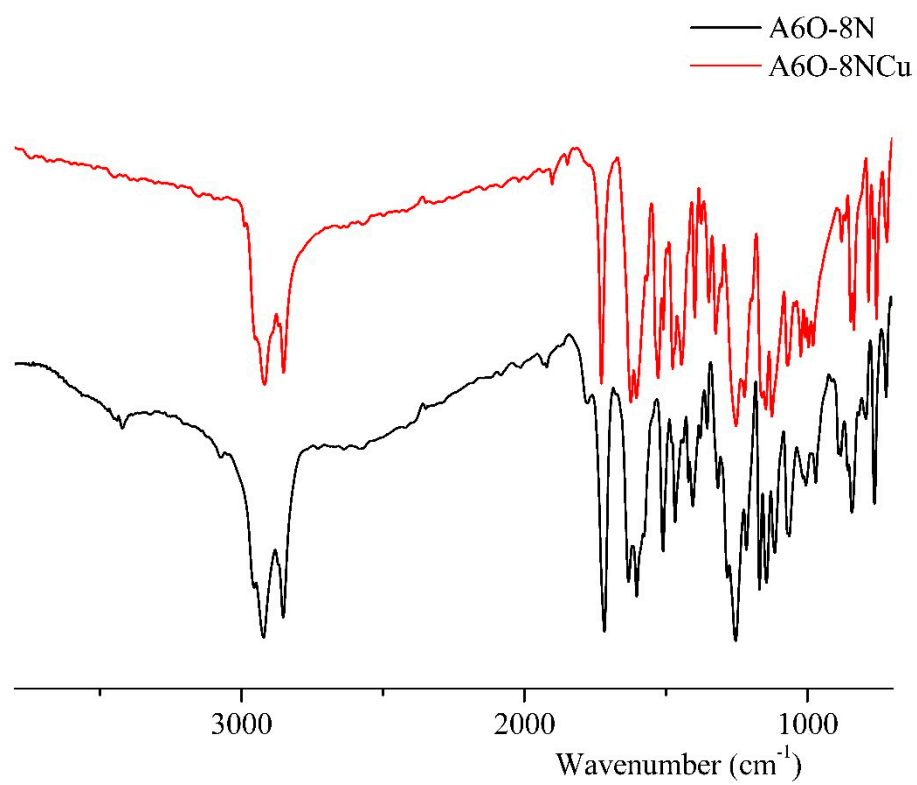


Figure S1. FTIR curves for of A6O-8N (black) and A6O-8NCu (red)

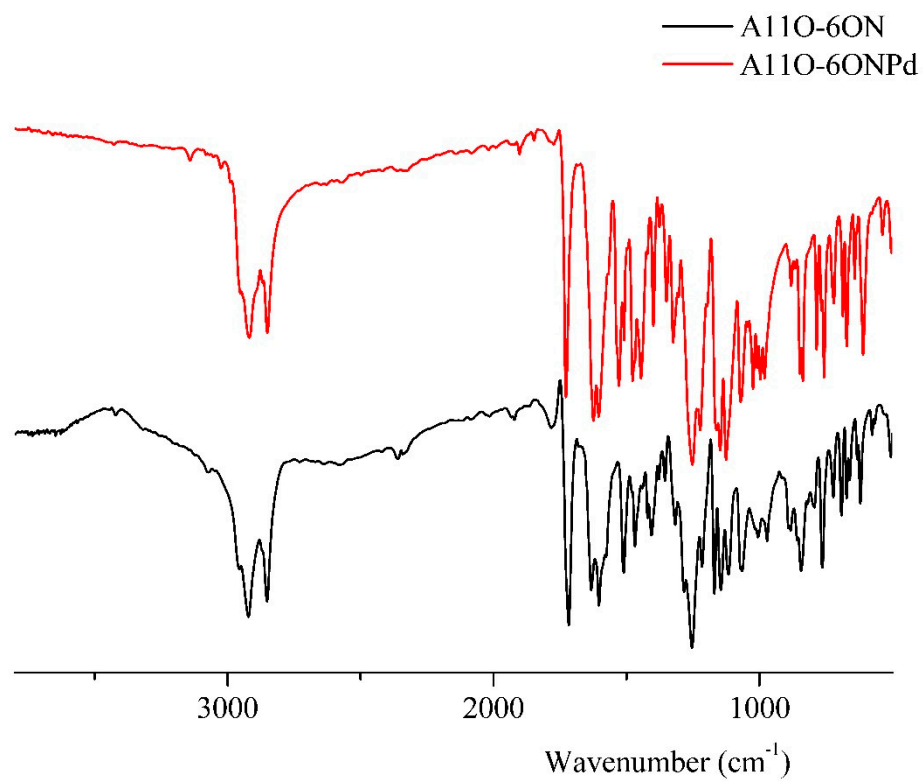


Figure S2. FTIR curves for of A11O-6ON (black) and A11O-6ONCu (red)

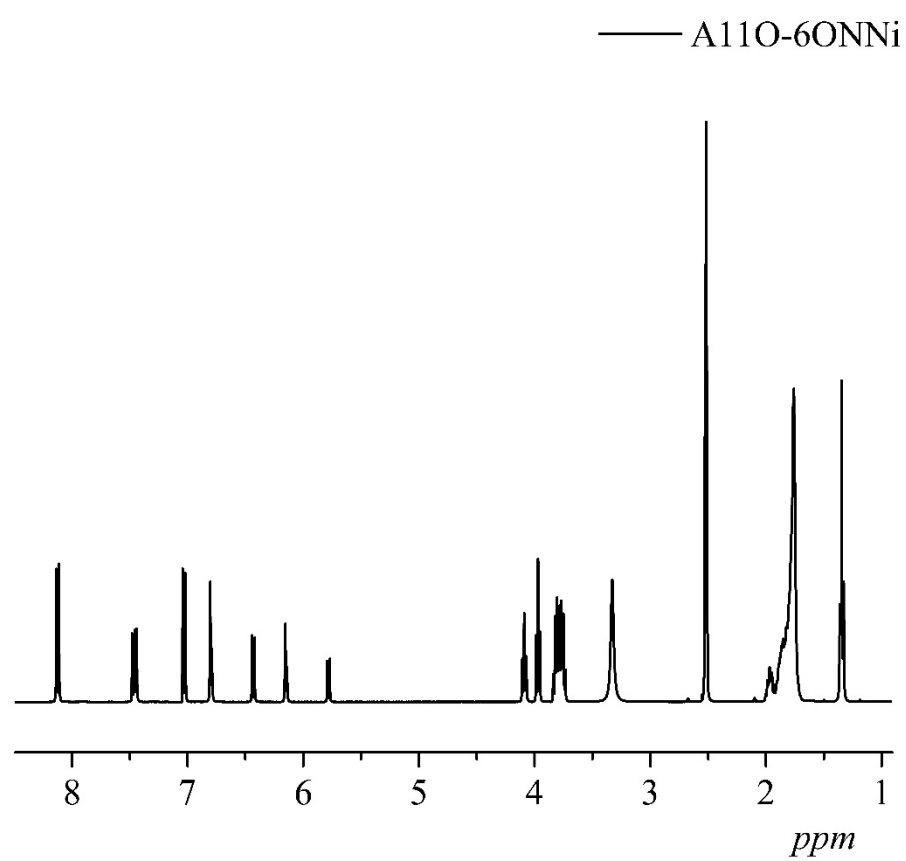


Figure S3. ^1H NMR curves of A11O-6ONNi

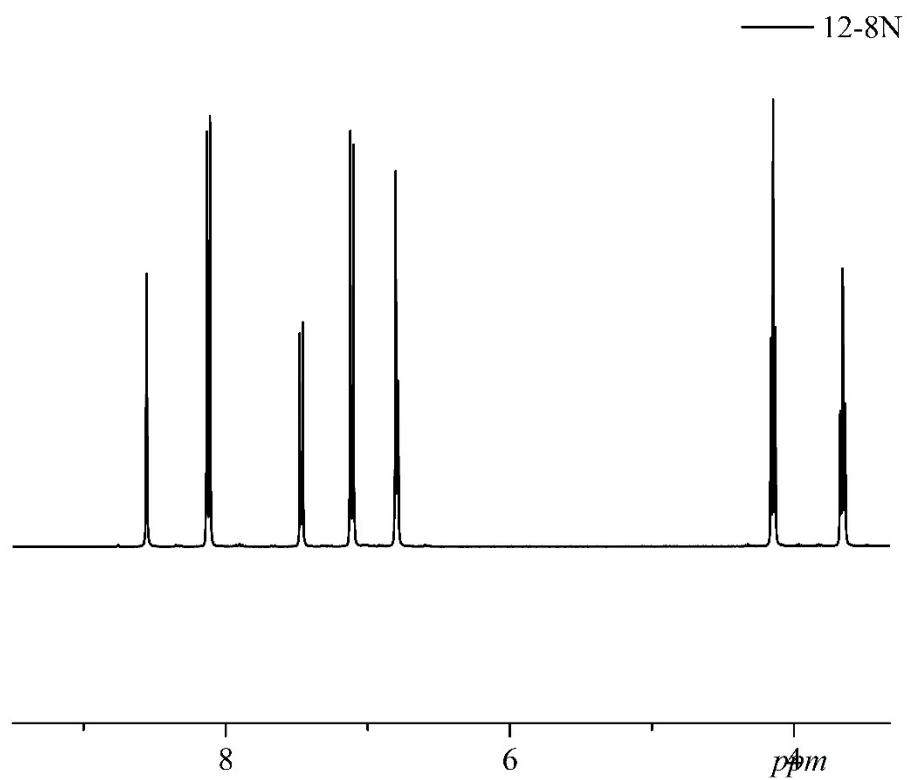


Figure S4. ^1H NMR curves for of 12-8N

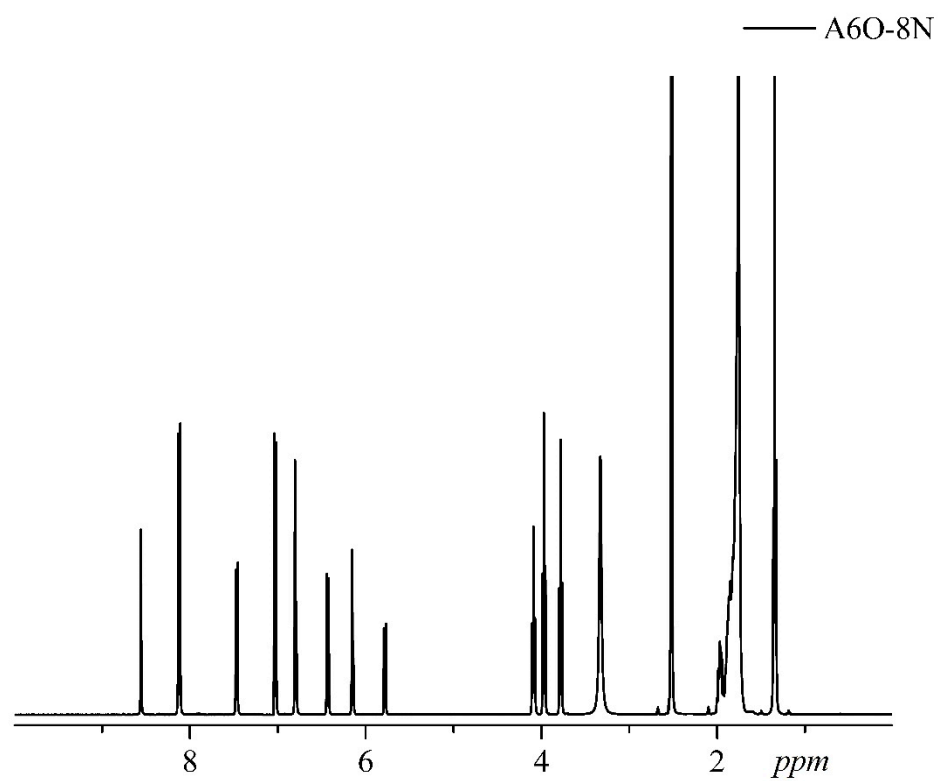


Figure S5. ^1H NMR curves for of A6O-8N

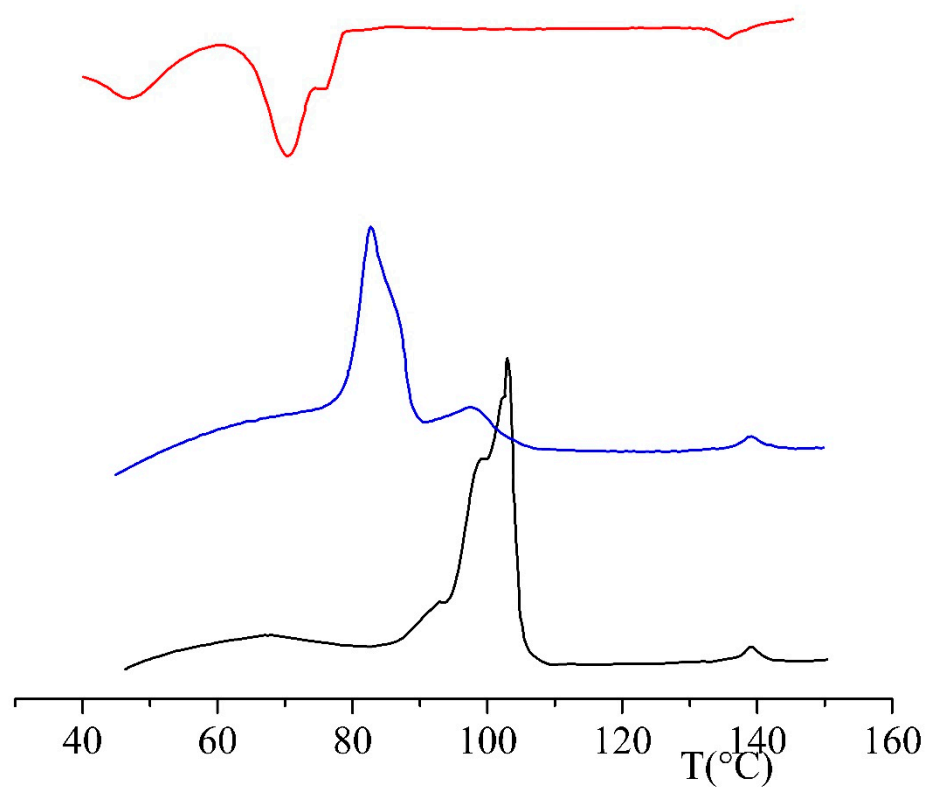


Figure S6. DSC thermograms of for A11O-6ONNi/A11O-6ONPd blends with composition of 0.207 wt./wt.: (black) first heating; (red) first cooling run and (blue) second heating run at 10 K/min scanning rate.

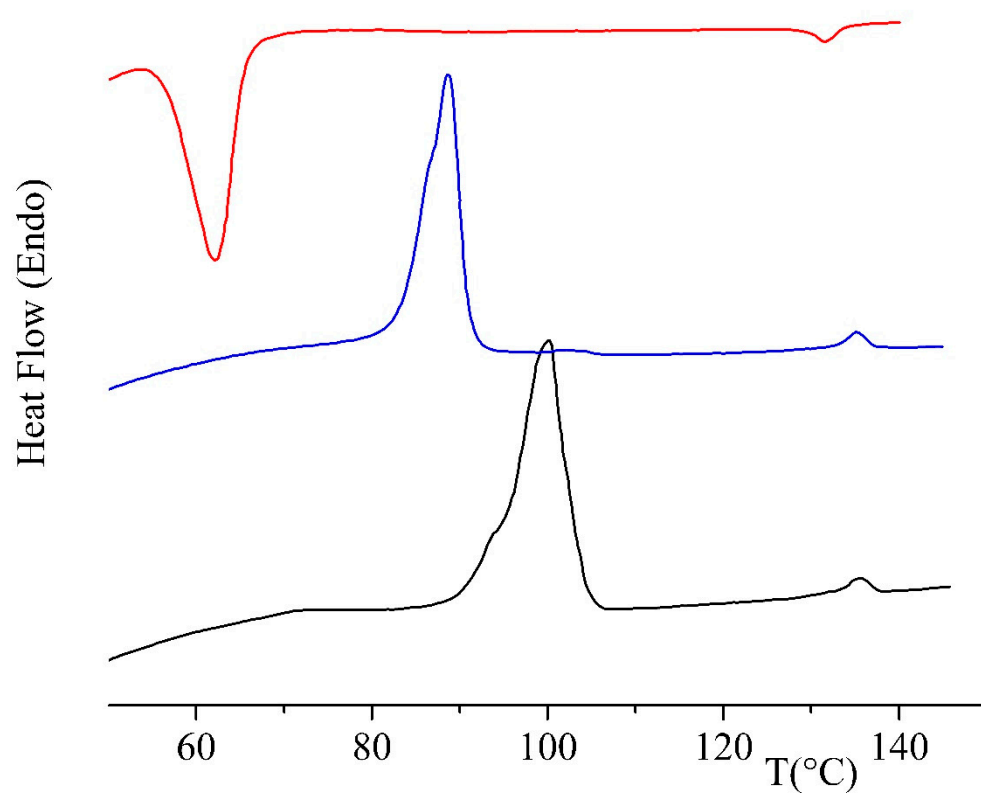


Figure S7. DSC thermograms of for A11O-6ONNi/A11O-6ONPd blends with composition of 0.401 wt./wt.: (black) first heating; (red) first cooling run and (blue) second heating run at 10 K/min scanning rate.

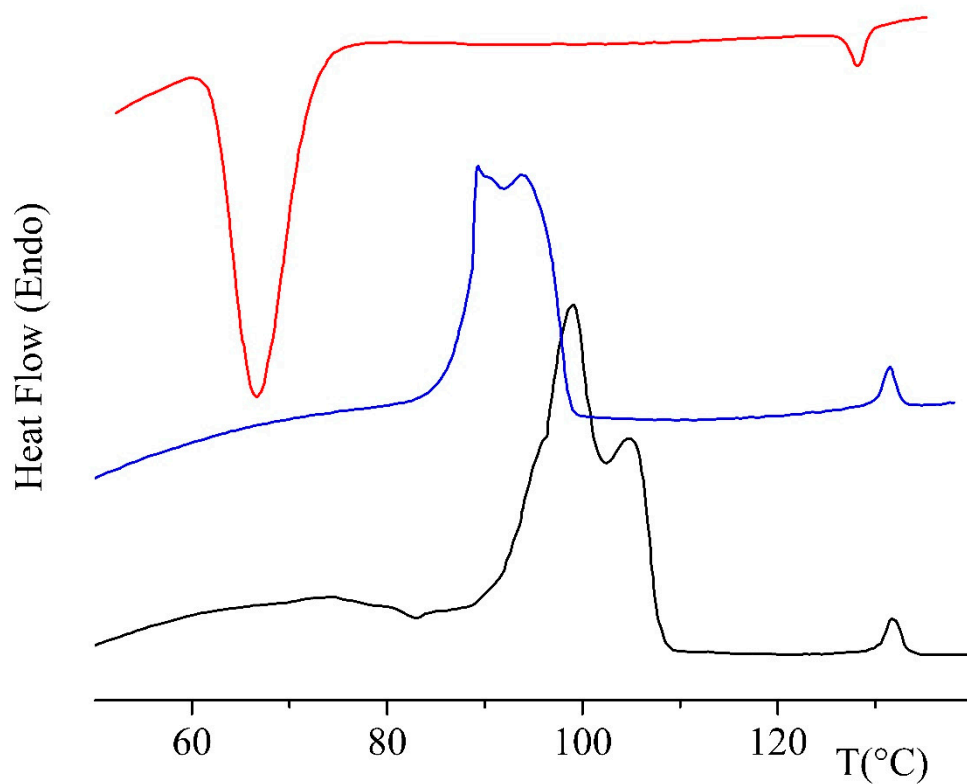


Figure S8. DSC thermograms of for A11O-6ONNi/A11O-6ONPd blends with composition of 0.606 wt./wt.: (black) first heating; (red) first cooling run and (blue) second heating run at 10 K/min scanning rate.

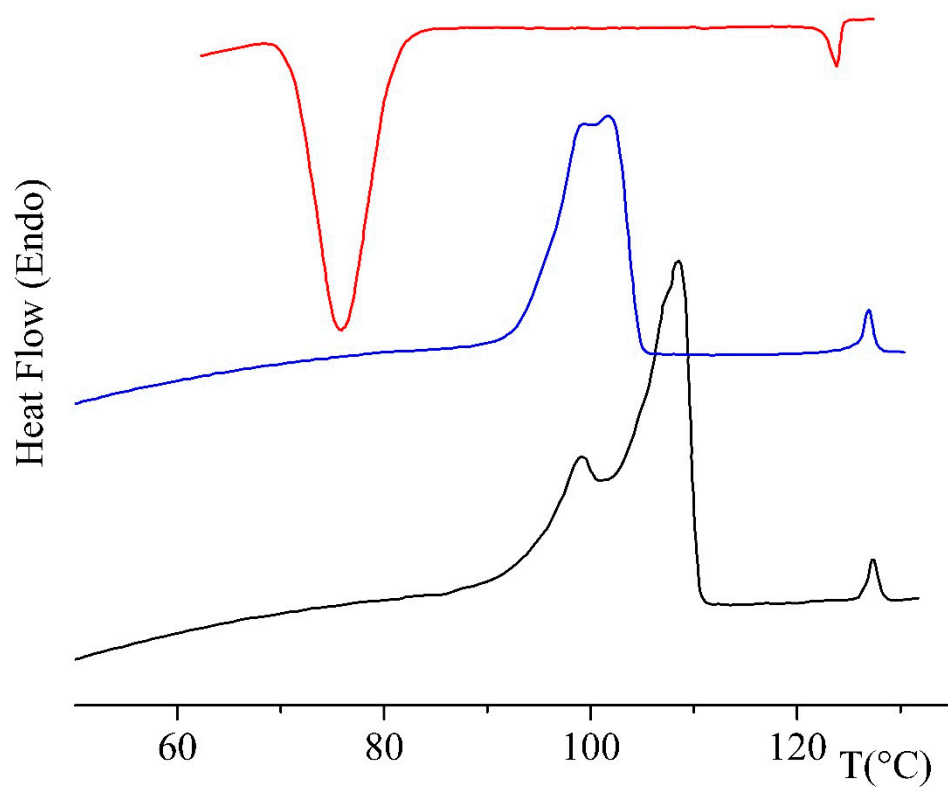


Figure S9. DSC thermograms of for A11O-6ONNi/A11O-6ONPd blends with composition of 0812 wt./wt.: (black) first heating; (red) first cooling run and (blue) second heating run at 10 K/min scanning rate.