

## Supplementary Materials

# Crystal structure and coordination of B-cations in the Ruddlesden-Popper phases $\text{Sr}_{3-x}\text{Pr}_x(\text{Fe}_{1.25}\text{Ni}_{0.75})\text{O}_{7-\delta}$ ( $0 \leq x \leq 0.4$ )

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### 1. XRD studies

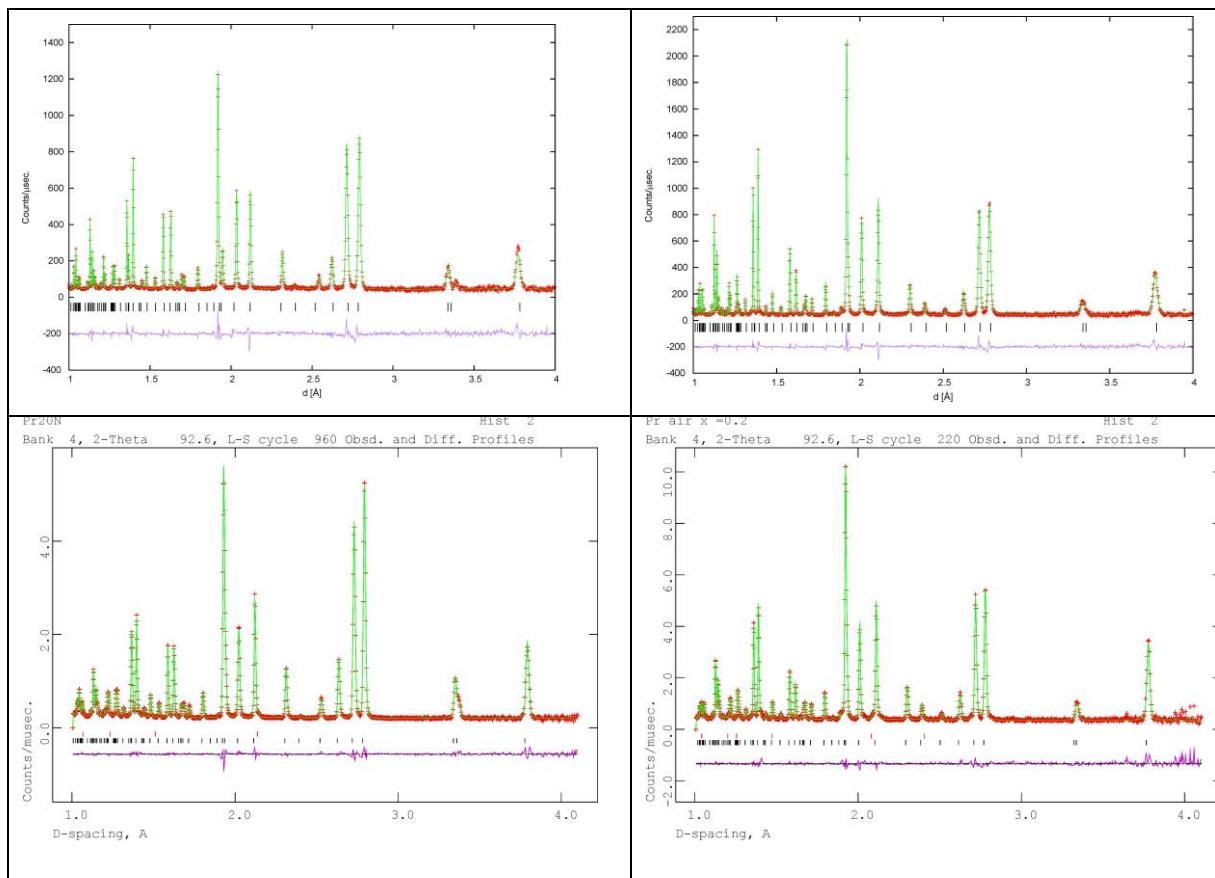
Room-temperature XRD patterns were recorded using a PIXCEL detector and variable slits with a  $2 \times 2 \text{ cm}^2$  illuminated area. The powder samples were evenly spread unto zero-background Si plates. In the structure refinements, 30–35 parameters were refined, and the  $2\theta$  range used covered ~120 theoretical reflections. The background was modeled by linear interpolation between ~20 specified background points. The peak widths were observed to show an anisotropic (i.e.  $hkl$  dependent) strain broadening. The extent and  $hkl$  dependence of this anisotropic strain broadening were furthermore found to vary with composition and between as-prepared and air-annealed samples. In general, the largest extent in variation was observed in the low  $x$ -values for as-prepared samples. This was taken into account in the refinements when refining  $S_{HKL}$  micro-strain parameters using the model by Stephens.<sup>1</sup> Corrections for surface roughness micro-absorption were made by refining the  $P_0$  and  $C_p$  parameters for the model by Pitschke.<sup>2</sup> For the N<sub>2</sub>-annealed samples, anisotropic atomic displacement parameters (ADPs) were refined for all metal atoms and a collective isotropic ADP refined for O1, O2, and O3. For the air-annealed samples, ADPs were refined for O3 and a collective isotropic ADP refined for O1 and O2.

The XRD refinement revealed the presence of small amounts of NiO in the air-annealed samples. This impurity was most probably present as Ni(s) in the corresponding N<sub>2</sub>-annealed sample, but in too small or badly crystallized amounts, seen in the XRD pattern.

The problems associated with refining positions/occupancies of light elements together with heavier ones like Sr/Pr/Fe/Ni are well-known.

### 2. NPD Studies

The NPD patterns of the investigated  $x = 0$  and 0.2, N<sub>2</sub>(g)- and air-annealed samples are shown in Figure S1. Since the data are from two state-of-the-art instruments, the number of refined structural parameters is limited, low  $R$ - and  $\chi^2$  values, we have high trust in our results. The Rietveld refinements yield residual  $R_F$  factors of 4.2 or lower for all refinements. The refinements revealed the presence of minor impurity phases such as 2 wt% Pr<sub>6</sub>O<sub>11</sub>, and 1 wt% NiO for air-annealed Pr  $x = 0.2$ . The refined unit-cell and structural parameters are given in Table 1, and selected bond distances and angles in Table 2, in the main text.



**Figure S1.** Measured, calculated, and differences in NPD patterns ( $147^\circ$  bank): (a)  $N_2$ -annealed  $Sr_3Fe_{1.25}Ni_{0.75}O_{5.587(7)}$ , (b) air-annealed  $Sr_3Fe_{1.25}Ni_{0.75}O_{6.552(5)}$ , (c)  $N_2$ -annealed (tics; upper-magnetic structure, mid-NiO,  $Sr_{2.80}Pr_{0.20}Fe_{1.25}Ni_{0.75}O_{5.682(8)}$ ) and (d) air-annealed (tics; upper-magnetic structure, mid-NiO,  $Sr_{2.80}Pr_{0.20}Fe_{1.25}Ni_{0.75}O_{6.640(6)}$ ).

**Table S1.** Anisotropic thermal parameters ( $U_{ij}^* \times 100$ ) (NPD-data) for  $N_2$ - and air-annealed  $Sr_3-xPr_xFe_{1.25}Ni_{0.75}O_{7-\delta}$  samples.

$x = 0.0$						
	N <sub>2</sub> -annealed			air-annealed		
Site	U <sub>11</sub>	U <sub>22</sub>	U <sub>33</sub>	U <sub>11</sub>	U <sub>22</sub>	U <sub>33</sub>
A1	2.39(5)	= U <sub>11</sub>	0.78(5)	1.17(3)	= U <sub>11</sub>	0.351(4)
A2	1.57(3)	= U <sub>11</sub>	6.92(3)	0.60(2)	= U <sub>11</sub>	0.49(3)
B	1.22(2)	= U <sub>11</sub>	1.33(3)	0.23(1)	= U <sub>11</sub>	0.85(3)
O1	3.5(2)	= U <sub>11</sub>	0.74(2)	0.87(6)	= U <sub>11</sub>	0.13(8)
O2	1.96(3)	= U <sub>11</sub>	0.81(5)	1.01(2)	= U <sub>11</sub>	0.55(4)
O3	3.42(6)	1.82(5)	2.80(6)	0.70(2)	0.66(2)	1.84(4)
$x = 0.2$						
	N <sub>2</sub> -annealed			air-annealed		
Site	U <sub>11</sub>	U <sub>22</sub>	U <sub>33</sub>	U <sub>11</sub>	U <sub>22</sub>	U <sub>33</sub>
A1	2.01(5)	= U <sub>11</sub>	0.48(6)	1.21(5)	= U <sub>11</sub>	0.32(6)
A2	1.37(3)	= U <sub>11</sub>	0.93(4)	0.54(3)	= U <sub>11</sub>	0.65(4)
B	1.09(2)	= U <sub>11</sub>	0.78(4)	0.16(1)	= U <sub>11</sub>	0.83(3)

O1	4.4(3)	= U <sub>11</sub>	3.1(4)	0.97(7)	= U <sub>11</sub>	0.97(6)
O2	1.87(3)	= U <sub>11</sub>	0.75(5)	1.12(3)	= U <sub>11</sub>	0.55(5)
O3	2.71(5)	1.51(5)	2.3(6)	0.77(3)	0.53(3)	1.82(5)

**Table S2.** Unit cell parameters and results from the refinement using XRD data for N<sub>2</sub>-annealed Sr<sub>3-x</sub>Pr<sub>x</sub>Fe<sub>1.25</sub>Ni<sub>0.75</sub>O<sub>7-δ</sub> samples.

<i>x</i>	$\chi^2$	R <sub>F</sub>	<i>a</i> /Å	<i>c</i> /Å	<i>V</i> /Å <sup>3</sup>
0.00	3.50	7.0	3.8394	20.3468	299.93
0.10	2.32	7.0	3.8474	20.2687	300.03
0.20	1.99	8.8	3.8584	20.1818	300.45
0.30	1.55	5.0	3.8675	20.1276	301.06
0.40	1.35	5.1	3.8797	20.0175	301.30

**Table S3.** Unit cell parameters, NiO content (mass %) and results from the refinement using XRD data for air-annealed Sr<sub>3-x</sub>Pr<sub>x</sub>Fe<sub>1.25</sub>Ni<sub>0.75</sub>O<sub>7-δ</sub> samples.

<i>x</i>	$\chi^2$	R <sub>F</sub>	NiO/mass %	<i>a</i> /Å	<i>c</i> /Å	<i>V</i> /Å <sup>3</sup>
0.00	3.86	7.5	-	3.8433	20.1059	296.98
0.10	2.64	6.4	0.22	3.8437	20.0809	296.68
0.20	2.58	7.2	0.10	3.8438	20.0621	296.41
0.30	1.93	7.0	0.19	3.8441	20.0635	296.48
0.40	1.62	6.5	-	3.8435	20.0538	296.24

**Table S4** Atomic coordinates (z-value) from the refinement using XRD data for N<sub>2</sub>-annealed Sr<sub>3-x</sub>Pr<sub>x</sub>Fe<sub>1.25</sub>Ni<sub>0.75</sub>O<sub>7-δ</sub> samples.

<i>x</i>	A2	B	O2	O3
0.00	0.6832(1)	0.1009(2)	0.1925(7)	0.0885(7)
0.10	0.6828(1)	0.1015(2)	0.1930(8)	0.0874(8)
0.20	0.6823(1)	0.1012(2)	0.1954(7)	0.0821(8)
0.30	0.6819(1)	0.1009(2)	0.1940(6)	0.0806(4)
0.40	0.6812(1)	0.1005(1)	0.1958(4)	0.0818(3)

**Table S5** Atomic coordinates (z-value) from the refinement using XRD data for air-annealed Sr<sub>3-x</sub>Pr<sub>x</sub>Fe<sub>1.25</sub>Ni<sub>0.75</sub>O<sub>7-δ</sub> samples.

<i>x</i>	A2	B	O2	O3
0.00	0.6815(1)	0.1000(2)	0.1935(6)	0.0903(4)
0.10	0.6816(1)	0.0992(1)	0.1925(5)	0.0909(5)
0.20	0.6818(1)	0.0993(2)	0.1928(6)	0.0913(6)
0.30	0.6818(1)	0.0989(1)	0.1941(5)	0.0905(5)
0.40	0.6820(1)	0.0985(1)	0.1939(5)	0.0914(4)

**Table S6** A-O bond lengths (Å) from the refinement using XRD data for N<sub>2</sub>-annealed Sr<sub>3-x</sub>Pr<sub>x</sub>Fe<sub>1.25</sub>Ni<sub>0.75</sub>O<sub>7-δ</sub> samples.

<i>x</i>	A1-O3, 8x	A1–O1, 4x	A2-O3, 4x	A2-O2, 4x	A2-O2, 1x
0.00	2.632(9)	2.7150	2.720(9)	2.722(1)	2.53(1)

0.10	2.617(10)	2.7209	2.729(10)	2.729(1)	2.52(1)
0.20	2.544(10)	2.7272	2.796(10)	2.740(1)	2.47(1)
0.30	2.523(5)	2.7345	2.810(7)	2.745(1)	2.50(1)
0.40	2.538(4)	2.7430	2.779(5)	2.758(1)	2.46(1)

**Table S7** A-O bond lengths ( $\text{\AA}$ ) from the refinement using XRD data for air-annealed  $\text{Sr}_{3-x}\text{Pr}_x\text{Fe}_{1.25}\text{Ni}_{0.75}\text{O}_{7-\delta}$  samples.

$x$	A1-O3, 8x	A1-O1, 4x	A2-O3, 4x	A2-O2, 4x	A2-O2, 1x
0.00	2.644(8)	2.7176	2.655(8)	2.728(1)	2.51(1)
0.10	2.651(5)	2.7178	2.648(8)	2.727(1)	2.53(1)
0.20	2.654(8)	2.7180	2.646(8)	2.727(1)	2.52(1)
0.30	2.644(8)	2.7179	2.654(8)	2.729(1)	2.49(1)
0.40	2.655(5)	2.7175	2.645(5)	2.728(1)	2.49(1)

**Table S8** B-O bond lengths ( $\text{\AA}$ ) and angles (in degrees) from the refinement using XRD data for  $\text{N}_2$ -annealed  $\text{Sr}_{3-x}\text{Pr}_x\text{Fe}_{1.25}\text{Ni}_{0.75}\text{O}_{7-\delta}$  samples.

$x$	B-O3, 4x	B-O2, 1x	B-O1, 1x	O3-B-O3
0.00	1.936(2)	1.87(1)	2.052(4)	165.10
0.10	1.945(2)	1.86(1)	2.058(5)	163.16
0.20	1.967(3)	1.90(1)	2.046(4)	157.30
0.30	1.976(2)	1.87(1)	2.030(3)	156.13
0.40	1.976(1)	1.91(1)	2.012(3)	158.09

**Table S9** B-O bond lengths ( $\text{\AA}$ ) and angles (in degrees) from the refinement using XRD data for air-annealed  $\text{Sr}_{3-x}\text{Pr}_x\text{Fe}_{1.25}\text{Ni}_{0.75}\text{O}_{7-\delta}$  samples.

$x$	B-O3, 4x	B-O2, 1x	B-O1, 1x	O3-B-O3
0.00	1.931(11)	1.88(1)	2.010(3)	168.48
0.10	1.929(8)	1.87(1)	1.994(3)	170.04
0.20	1.929(11)	1.88(1)	1.993(1)	170.45
0.30	1.929(8)	1.91(1)	1.985(3)	169.95
0.40	1.927(8)	1.91(1)	1.974(3)	171.56

**Table S10** Bond valence sums based on the crystal structures refined from XRD data for  $\text{N}_2$ -annealed  $\text{Sr}_{3-x}\text{Pr}_x\text{Fe}_{1.25}\text{Ni}_{0.75}\text{O}_{7-\delta}$  samples.

$x$	A1	A2	B
0.00	1.52	1.90	2.62
0.10	1.79	1.87	2.84
0.20	1.81	1.77	2.33
0.30	2.43	1.71	2.76
0.40	2.51	1.77	2.84

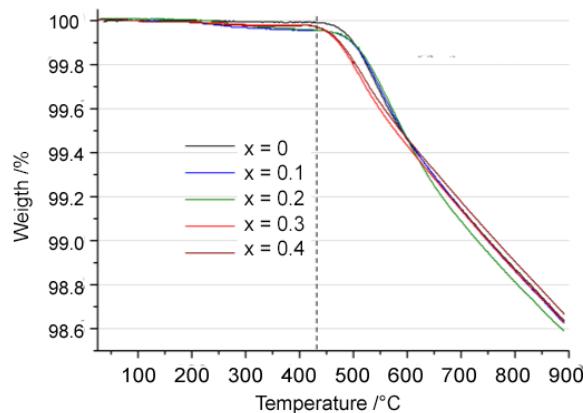
**Table S11** Bond valence sums based on the crystal structures refined from XRD data for air-annealed  $\text{Sr}_{3-x}\text{Pr}_x\text{Fe}_{1.25}\text{Ni}_{0.75}\text{O}_{7-\delta}$  samples.

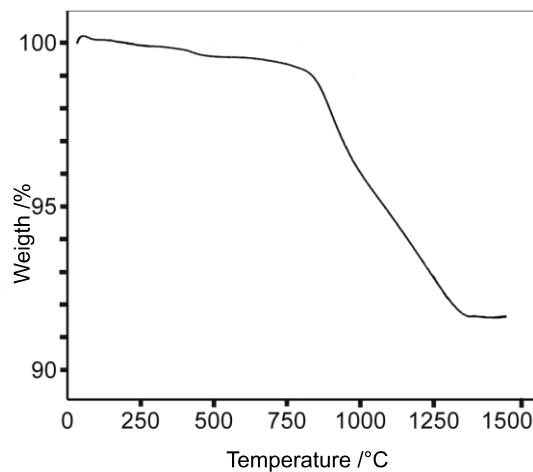
$x$	A1	A2	B
0.00	1.99	2.05	3.30
0.10	2.07	2.05	3.50
0.20	2.03	2.07	3.45
0.30	2.19	2.07	3.55
0.40	2.14	2.10	3.58

**Table S12** Oxygen content of  $\text{Sr}_{3-x}\text{Pr}_x\text{Fe}_{1.25}\text{Ni}_{0.75}\text{O}_{7-\delta}$  calculated from the occupancies of oxygen atomic positions using XRD, NPD (*in italics*), and TG data for  $\text{N}_2$ - and air-annealed samples.  $\Delta\delta$  is the difference in oxygen content per formula unit between  $\text{N}_2$ - and air-annealed samples.

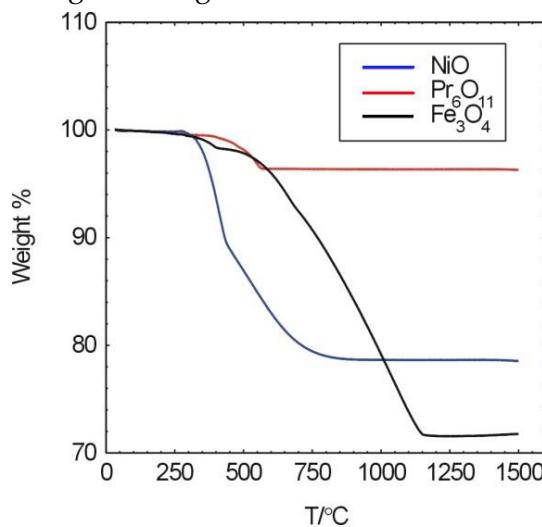
$x$	N <sub>2</sub> - annealed			Air - annealed		Change in oxygen content upon annealing in air	
	O3	O1	O	O1	O	$\Delta\delta$	$\Delta\text{TG}$
0.00	0.73(3), 0.813(3)	0.17(5), 0.335(5)	5.1(1), 5.587(7)	0.14(5), 0.552(6)	6.13(5), 6.552(5)	1.0(1), 1.03(1)	0.80
0.10*	0.85(3)	0.07(6)	5.5(1)	0.44(4)	6.44(4)	1.0(1)	0.75
0.20	0.74(3), 0.852(2)	0.20(5), 0.274(4)	5.2(1), 5.6(1)	0.38(5), 0.66(4)	6.38(5), 6.64(4)	1.2(1), 1.0(1)	0.70
0.30	0.90(2)	0.03(2)	5.6(1)	0.66(4)	6.66(4)	1.1(1)	0.82
0.40	0.98(2)	0.02(3)	5.9(1)	0.68(4)	6.68(4)	0.8(1)	-

\*5.46 according to TG

**Figure S2.** TG curves for air-annealed  $\text{Sr}_{3-x}\text{Pr}_x\text{Fe}_{1.25}\text{Ni}_{0.75}\text{O}_{7-\delta}$ .



**Figure S3.** TG curves for  $\text{Sr}_{3-x}\text{Pr}_x\text{Fe}_{1.25}\text{Ni}_{0.75}\text{O}_{7-\delta}$   $x = 0.1$  recorded in 4%  $\text{H}_2/\text{Ar}$  atmosphere and using a heating rate  $5 \text{ }^{\circ}\text{C}\cdot\text{min}^{-1}$ .



**Figure S4.** TG curves for pure oxides recorded in 4%  $\text{H}_2/\text{Ar}$  atmosphere and using a heating rate of  $10 \text{ }^{\circ}\text{C}\cdot\text{min}^{-1}$ , based on supplementary information in Reference [14].

## References

14. Samain, L., Amshoff, P., Biendicho, J.J., Tietz, F., Mahmoud, A., Hermann, R.P., Istomin, S.Y. Grins, J., Svensson, G., *J. Solid State Chem.* **2015**, *227*, 45–55, DOI: 10.1016/j.jssc.2015.03.018