Supplementary Materials: Effect of [Zr(α-PW₁₁O₃₉)₂]¹⁰⁻ Polyoxometalate on the Self-Assembly of Surfactant Molecules in Water Studied by Fluorescence and DOSY NMR Spectroscopy

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Figure S1. The emission spectra of: (**a**) saturated aqueous pyrene solution (2–3 μ M) buffered at pH 7.4 by a 10 mM sodium phosphate solution. The vibronic peaks which are situated at 372, 378, 283, 389 and 392 nm are labeled I, II, III, IV and IV, respectively; (**b**) The normalized emission of a saturated pyrene solution (2–3 μ M) buffered at pH 7.4 by a 10 mM sodium pohosphate solutions in the presence of increasing concentrations of Zw3-12. Notice the coalescence of peak II and III and peak IV and V.



Figure S2. ¹H DOSY spectra of: (a) CHAPS/1 mixture; (b) CHAPS/Et₂NH₂Cl mixture; (c) CHAPS.



Figure S3. CHAPS/1—³¹P DOSY.



Figure S4. ¹H DOSY spectra of: (a) Zw3-12/1 mixture; (b) Zw3-12/Et₂NH₂Cl mixture; (c) Zw3-12.



Figure S5. Zw3-12/1 mixture—³¹P DOSY.



Figure S6. ¹H DOSY spectra of: (a) SDS/1 mixture; (b) SDS/Et2NH2Cl mixture; (c) SDS.



Figure S7. SDS/1 mixture $-^{31}$ P DOSY.



Figure S8. ¹H DOSY spectra of: (a) TX100/1 mixture; (b) TX100/Et₂NH₂Cl mixture; (c) TX100.



Figure S9. TX100/1 mixture $-^{31}$ P DOSY.



Figure S10. ¹H DOSY Et₂NH₂Cl.



Figure S11. ¹H DOSY of 1.



Figure S12. ³¹P DOSY of 1.