

Supplementary material

Catalytic Ability of K- and Co-Promoted Oxo-Re and Oxo-ReMo Nanosized Compositions for Water–Gas Shift Reaction

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Table S1. Catalyst composition determined by XRF.

Sample/spot	Element concentration (wt %)							
	Re	Co	Mo	K	Re ₂ O ₇	CoO	MoO ₃	K ₂ O
CoRe								
spot 1	86.11	13.89	-	-			-	-
spot 2	85.68	14.32	-	-			-	-
Average	85.90	14.11	-	-	15.76	2.53	-	-
CoReMo								
spot 1	47.27	15.82	36.81	-				
spot 2	50.75	19.60	29.65	-				
Average	49.01	17.71	33.23	-	12.24	4.32	9.57	-
K-CoRe								
spot 1	54.20	17.22	-	28.58				
spot 2	46.04	20.09	-	33.87				
Average	50.12	18.66	-	31.23	11.90	4.34	-	6.86
K-CoReMo								
spot 1	22.41	14.20	32.34	31.05				
spot 2	29.45	13.62	27.02	29.91				
Average	25.93	13.91	29.68	30.48	7.87	4.13	10.40	8.58

After deposition of potassium, deviation from nominal values is related to the effect of X-ray depth penetration.

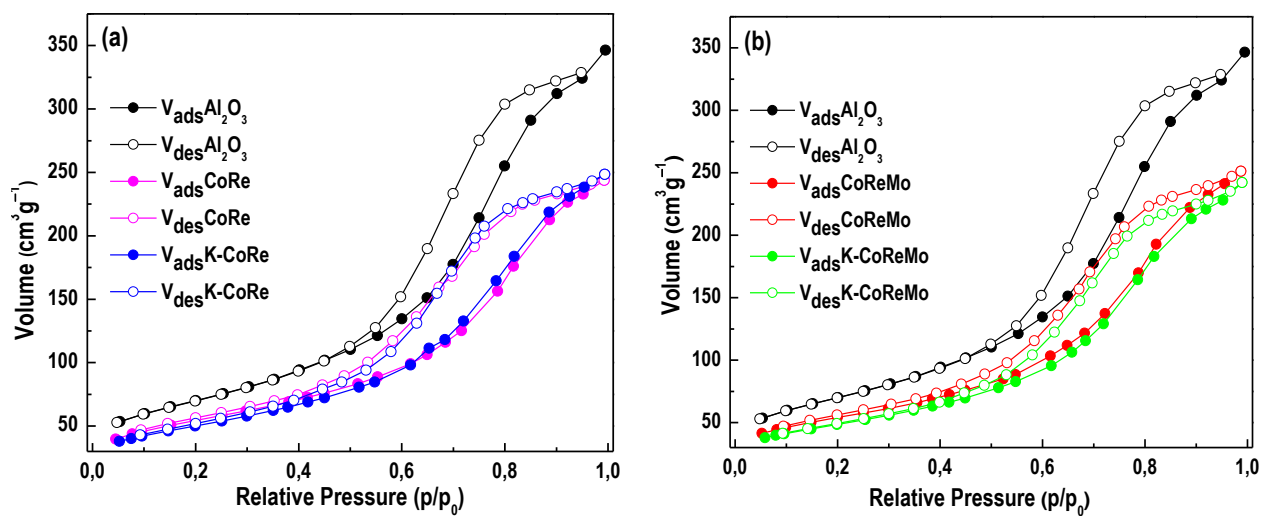


Figure S1. N₂ adsorption-desorption isotherms: (a) Al₂O₃ support, CoRe and K-CoRe catalysts; (b) Al₂O₃ support, CoReMo and K-CoReMo catalysts.

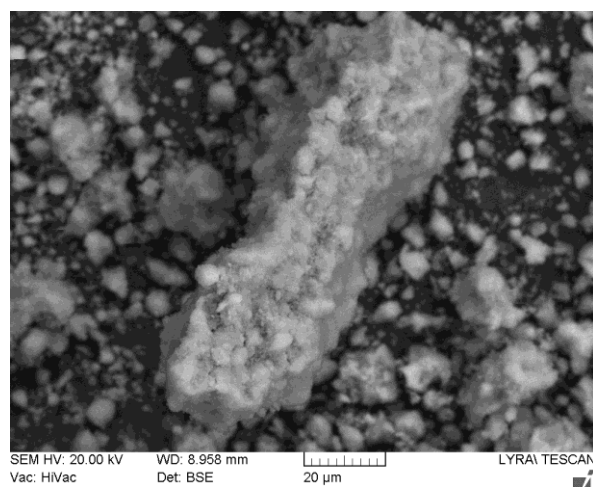


Figure S2. SEM image of γ-Al₂O₃ support

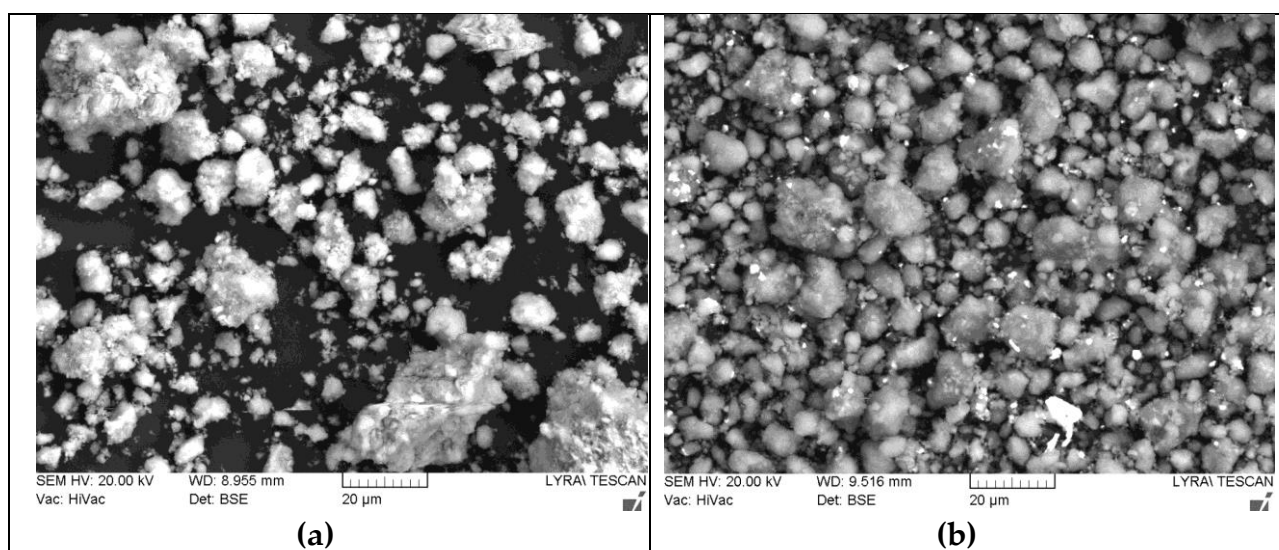


Figure S3. SEM image of: (a) CoRe; (b) K-CoRe.

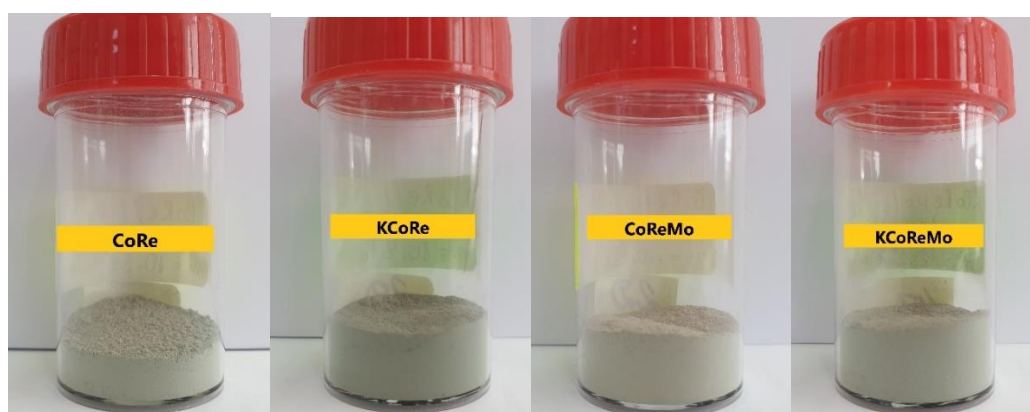


Figure S4. Photos displaying colors of oxide CoRe, K-CoRe, CoReMo, and K-CoReMo catalyst systems.

Table S2. Data on the formed oxide structures

Sample	Formed metal-oxide structures
CoRe	ReO_4^- , non-stoichiometric spinel-like mixed $\text{Co}^{2+}(\text{Co}^{3+}, \text{Al}^{3+})_2\text{O}_4$ oxide, bed formed Co_3O_4 ($\text{CoO} \cdot \text{Co}_2\text{O}_3$) and CoAl_2O_4 spinel structures; $\text{ReO}_3\text{--O--ReO}_3$ and Co--Re--O--(Al) surface species
K-CoRe	better formed Co_3O_4 ($\text{CoO} \cdot \text{Co}_2\text{O}_3$) and CoAl_2O_4 spinel structures; $\text{K}(\text{ReO}_4)$; $\text{K--Co--Re--O--(Al)}$ surface species
CoReMo	mixture of isolated molybdate MoO_4^- and polymolybdate --MoO_6^- structures; Co_3O_4 , CoAl_2O_4 , $\text{Co--Re--Mo--O--(Al)}$ and CoMoO_4 structures
K-CoReMo	mixture of isolated molybdate MoO_4^- and polymolybdate --MoO_6^- structures; K_2MoO_4 , $\text{K--Co--Re--Mo--O--(Al)}$ and CoMoO_4 structures

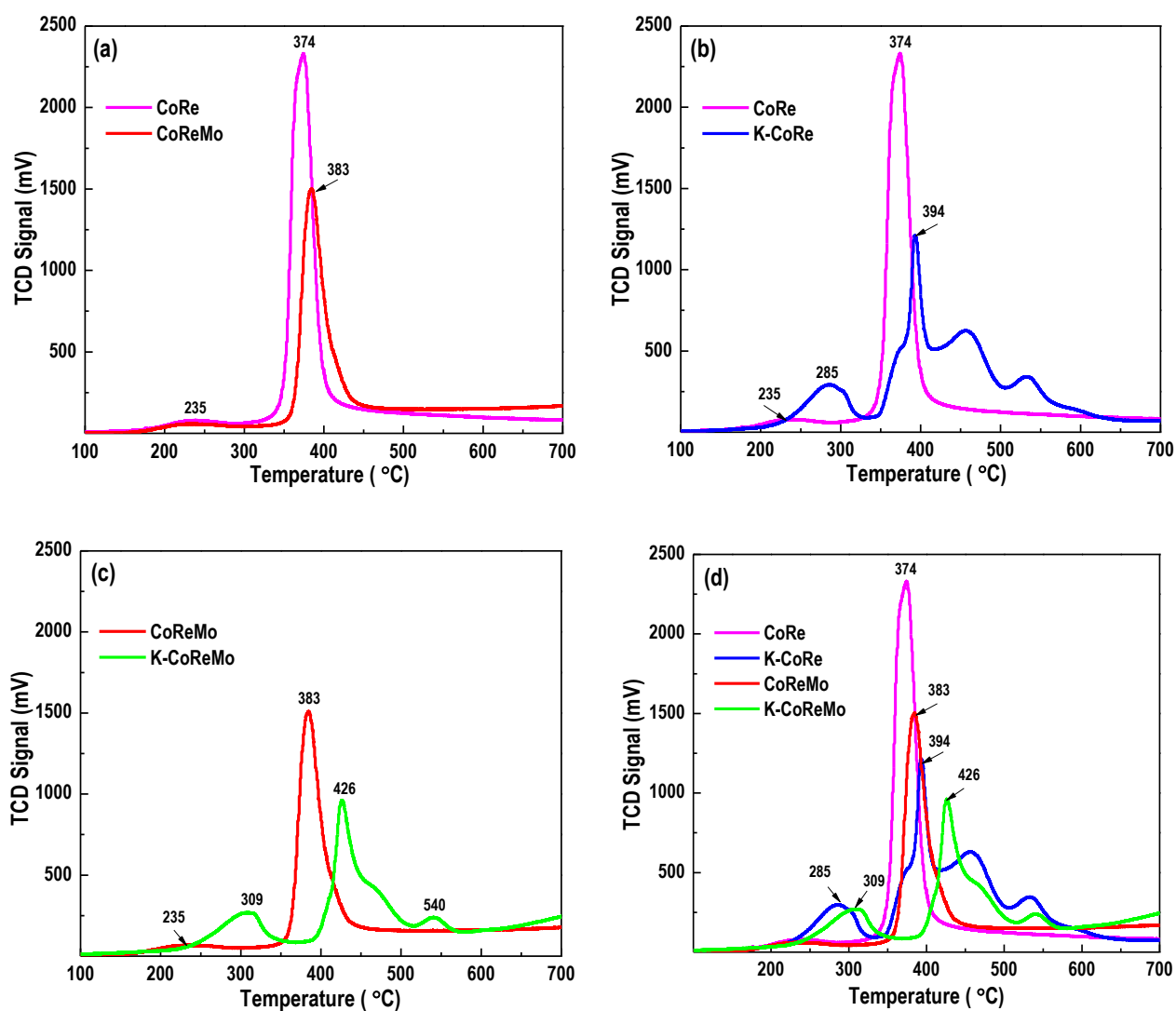


Figure S5. Comparative presentation of H₂-TPR profiles: (a) CoRe and CoReMo; (b) CoRe and K-CoRe; (c) CoReMo and K-CoReMo; (d) CoRe, K-CoRe, CoReMo, and K-CoReMo.