SUPPLEMENTARY INFORMATION

Solid Phase Nitrosylation of Enantiomeric Cobalt(II) Complexes

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Index **page**

Accompanying film (SI film.mp4) shows the sorption of NO into Co($L^{R,R}$) to give Co(NO)($L^{R,R}$).

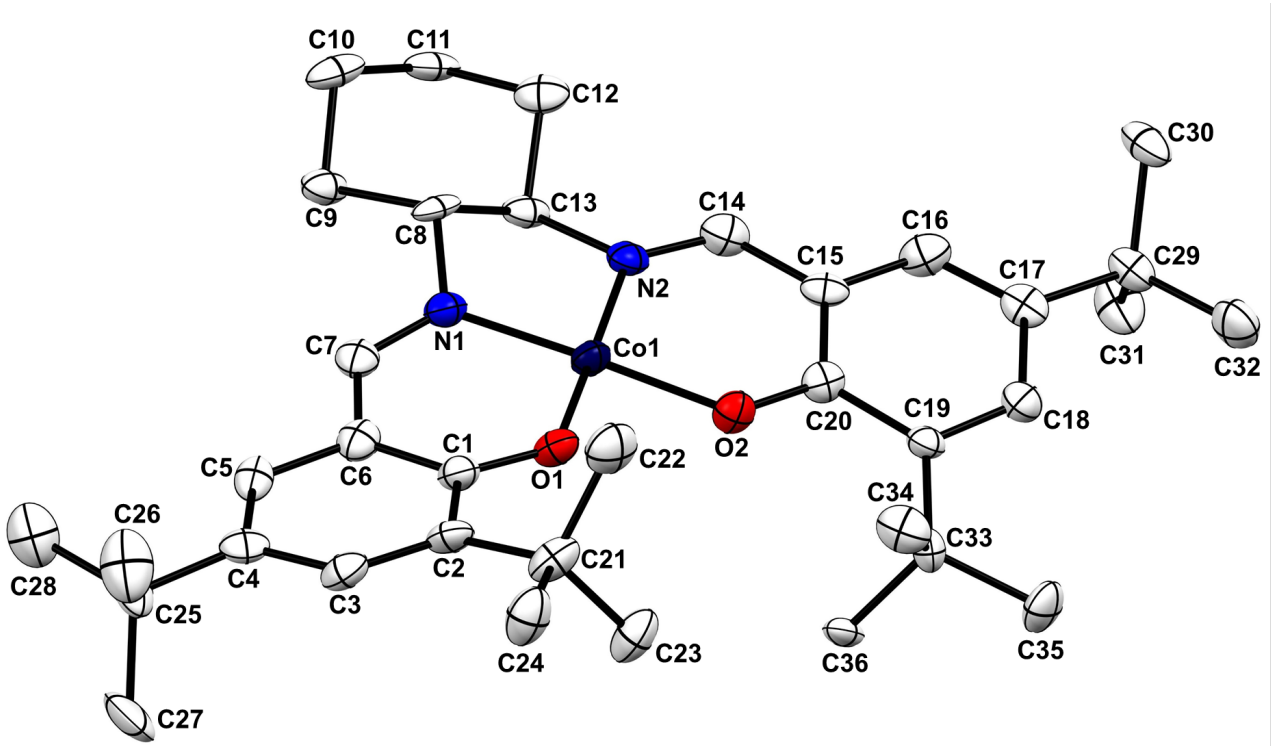

Figure S1. Single crystal X-ray structure of one of Co($L^{R,R}$). Atomic displacement ellipsoids are drawn at 50 % probability and hydrogen atoms are omitted for clarity.

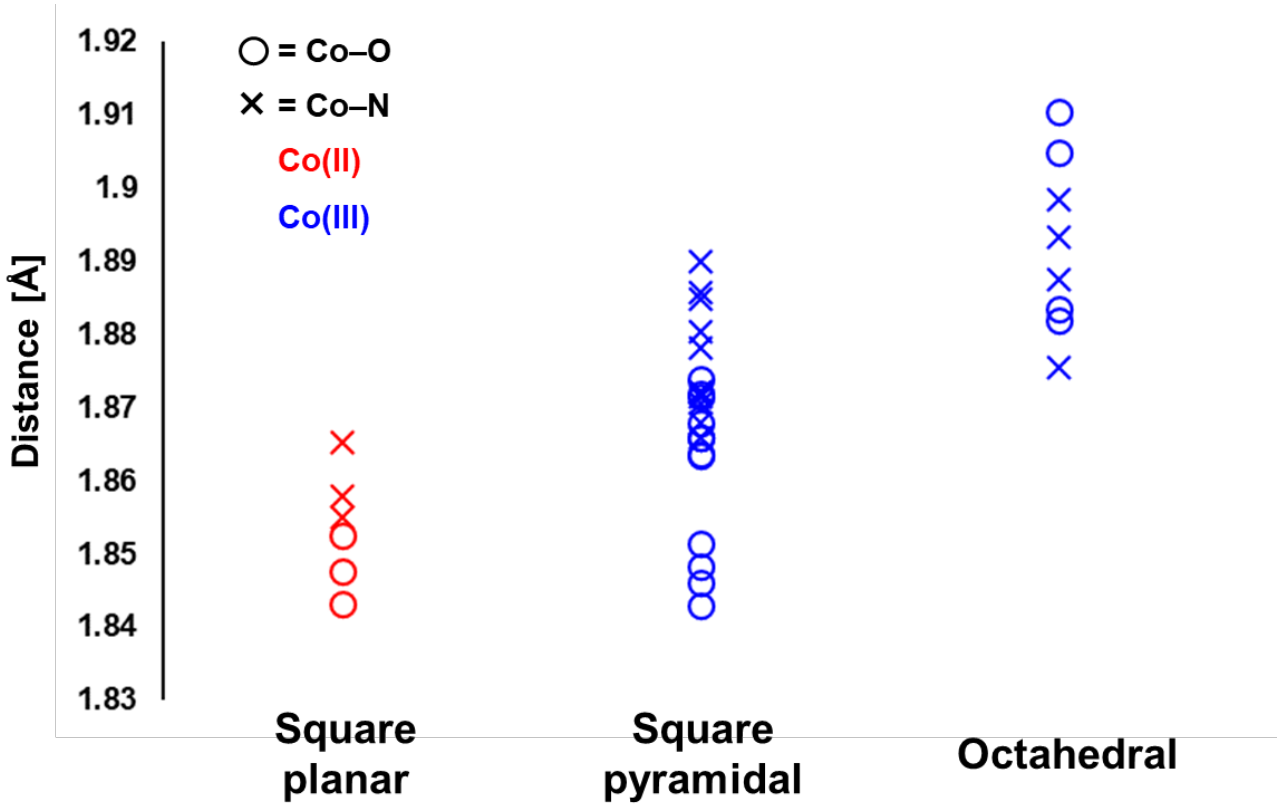

Figure S2. Diagram showing the relationship between Co–O and Co–N bond distances, coordination geometry and oxidation state for all the Co(L), Co(X)(L) and $Co(X)_2(L)$ complexes currently available in the CCDC [1–14].

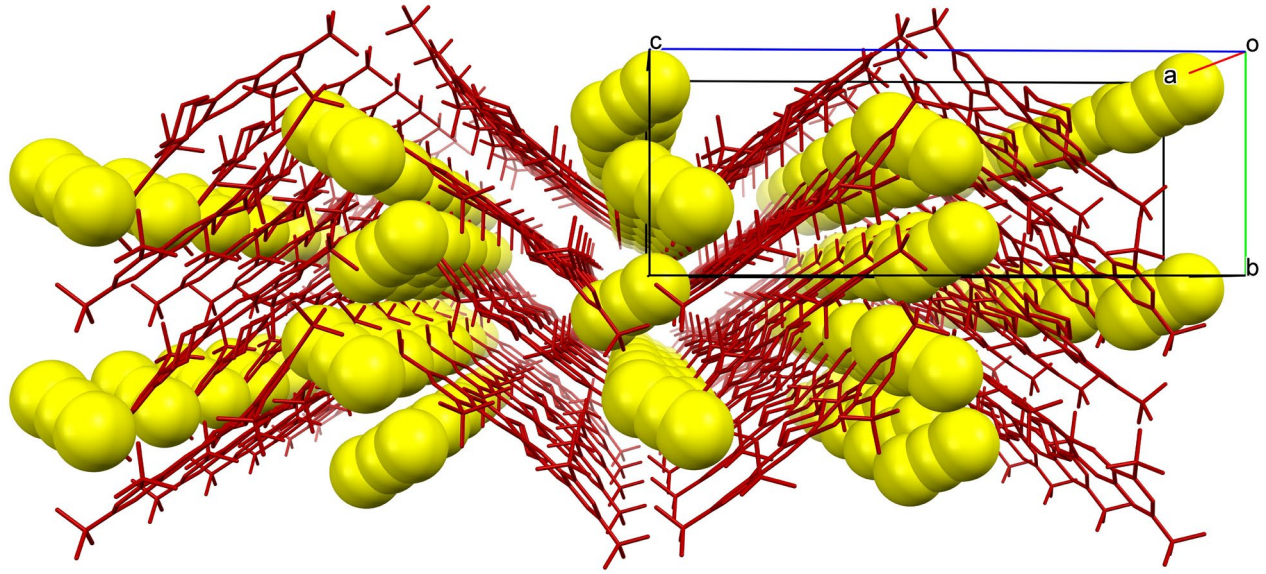

Figure S3. Packing diagram for $Co(L^{R,R})$•CS_2, with CS_2 in yellow. Hydrogen atoms are omitted for clarity.

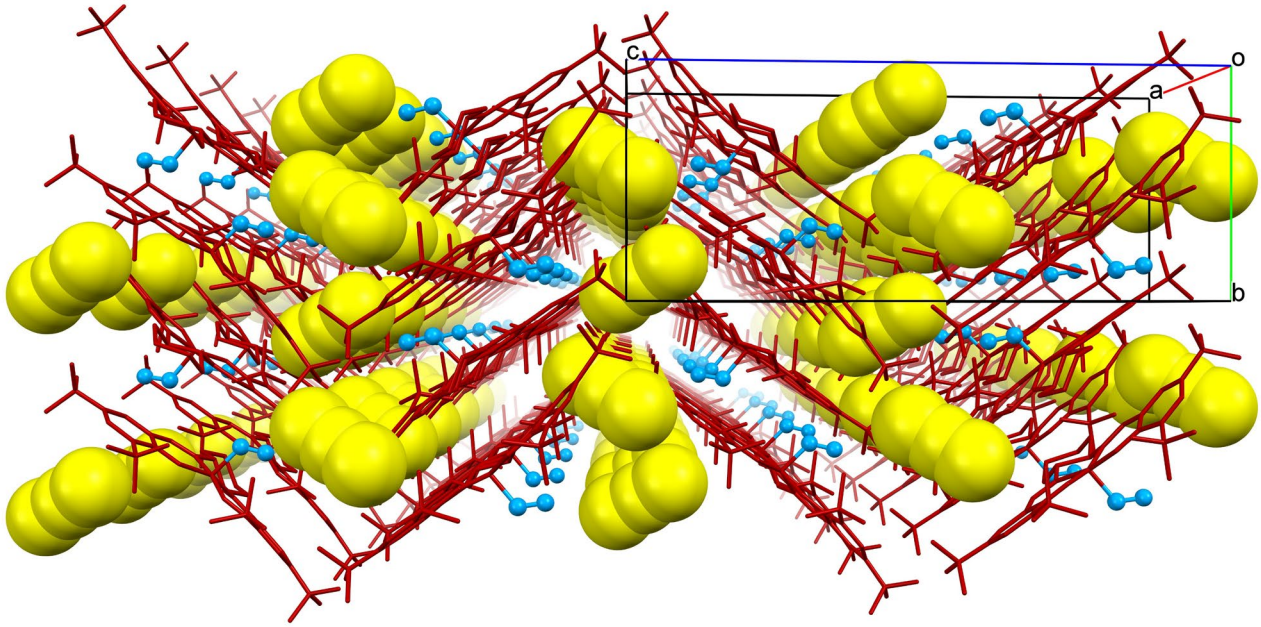

Figure S4. Packing diagram for Co(NO)($L^{R,R}$)•CS_2, with CS_2 in yellow and NO in light blue. Hydrogen atoms are omitted for clarity.

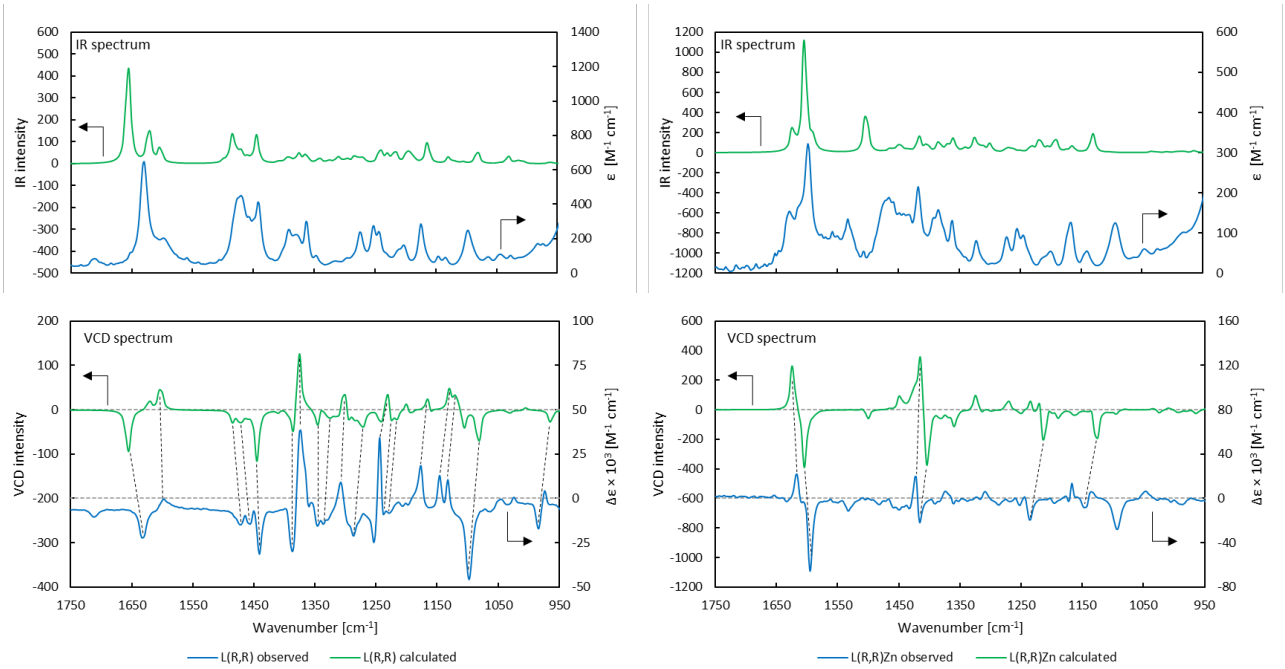

Figure S5. Comparison of observed (blue) and calculated (green) IR and VCD spectra of $L^{R,R}$ and Zn($L^{R,R}$). The calculated spectra are shifted -45 cm^{-1} and -70 cm^{-1} respectively from the originally calculated wavenumbers.

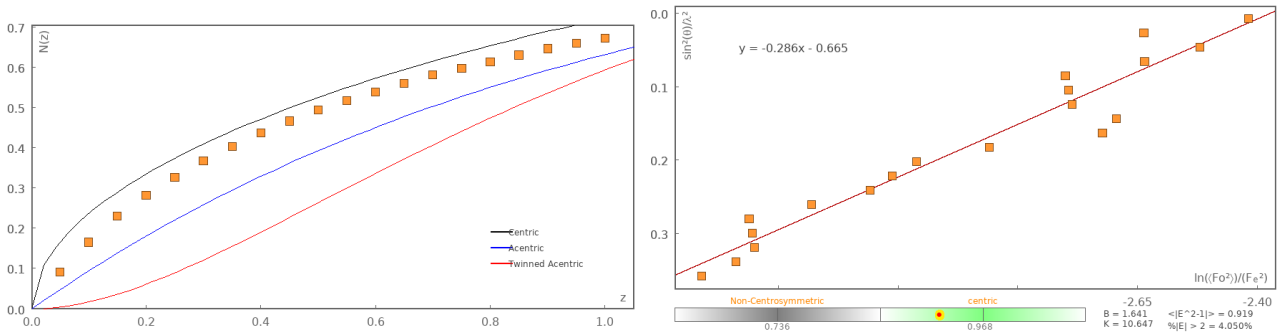

Figure S6. Cumulative intensity distribution plot and Wilson plot for Co($L^{R,R}$)•CS_2

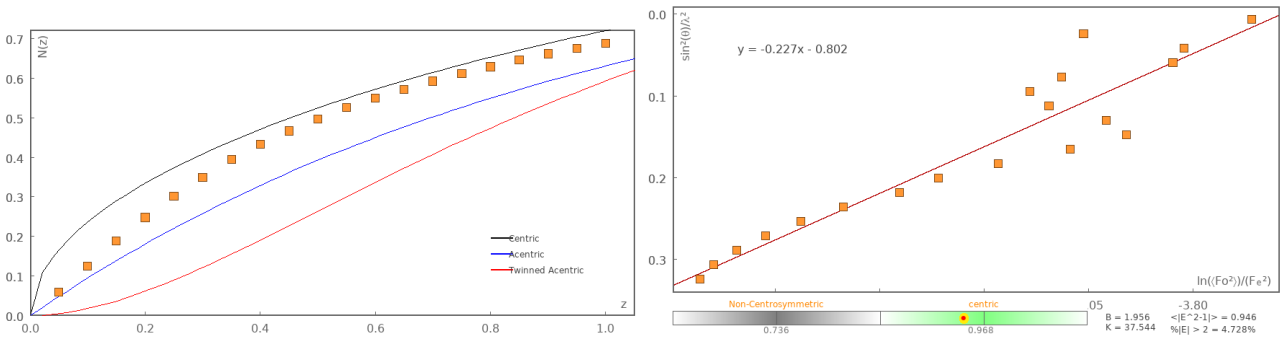

Figure S7. Cumulative intensity distribution plot and Wilson plot for Co($L^{R,R}$)

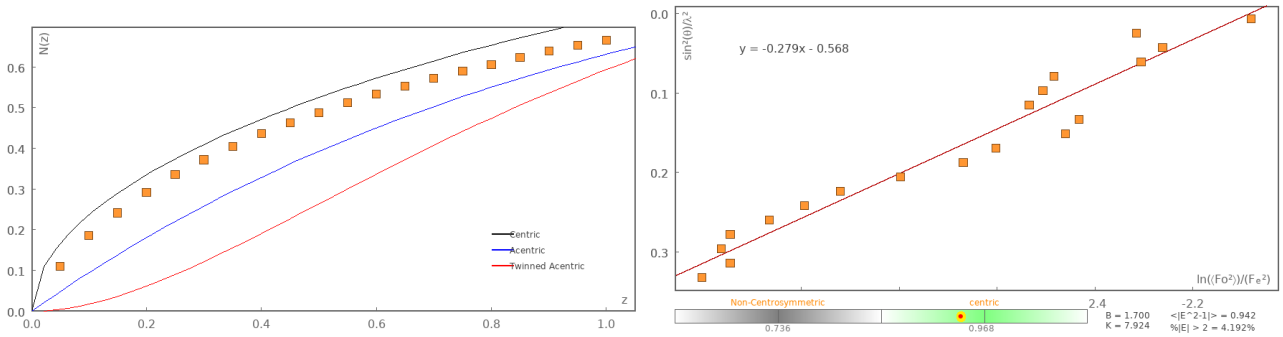

Figure S8. Cumulative intensity distribution plot and Wilson plot for Co(NO)($L^{R,R}$)•CS_2

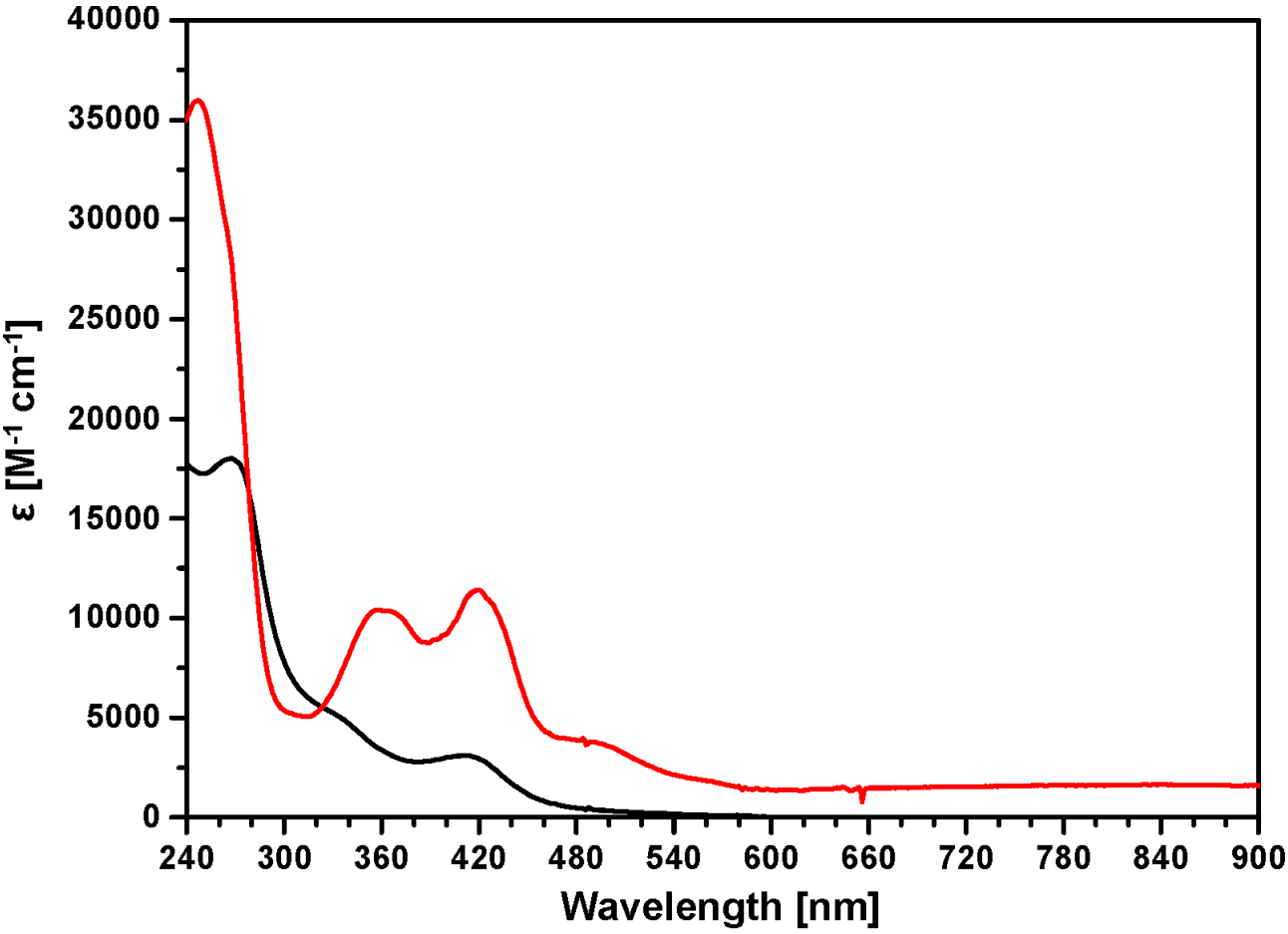

Figure S9. UV-vis spectra of Co($L^{R,R}$) (red) and Co(NO)($L^{R,R}$) (black) in dichloromethane.

References

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