

## **Supporting Information**

### **Multivariate Linear Regression Models to Predict Monomer Poisoning Effect in Ethylene/Polar Monomer Copolymerization Catalyzed by Late Transition Metals**

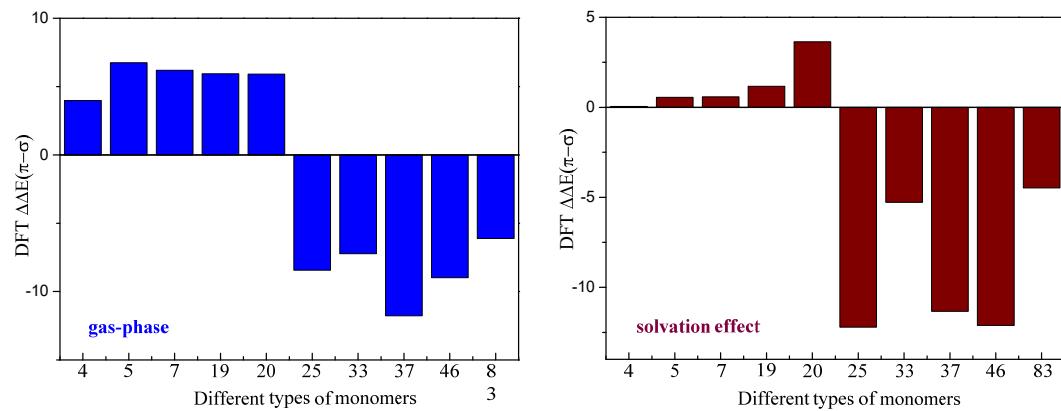
Wei Zhao,<sup>†</sup> Zhihao Liu,<sup>†</sup> Yanan Zhao,<sup>†,\*</sup> Yi Luo,<sup>†,‡,\*</sup> Shengbao He<sup>‡</sup>

<sup>†</sup>*State Key Laboratory of Fine Chemicals, School of Chemical Engineering, Dalian University of Technology, Dalian 116024, China.*

<sup>‡</sup>*PetroChina Petrochemical Research Institute, Beijing, 102206, China.*

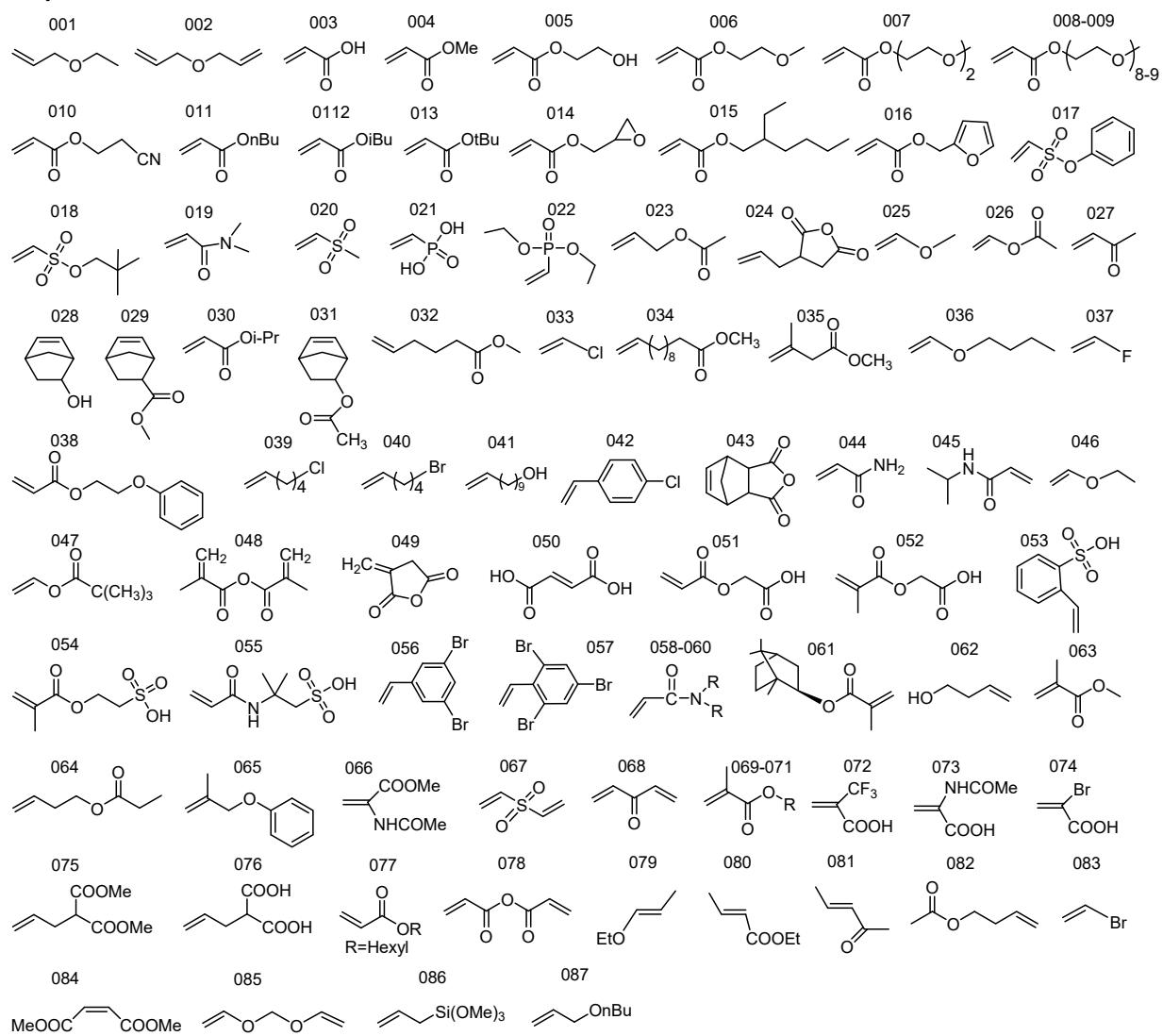
\*E-mail: yananzhao@dlut.edu.cn, luoyi@dlut.edu.cn

Taking diimide palladium as an example, the single-point calculations were further performed at the higher level by using the density functional method M06, and 6-311+G (d, p) was used for the nonmetal atoms and the basis set LANL2DZ as well as the associated pseudopotential were applied for the Pd atom. In these single-point calculations, the solvation effect of toluene ( $\epsilon = 2.37$ ) was considered through the CPCM model.



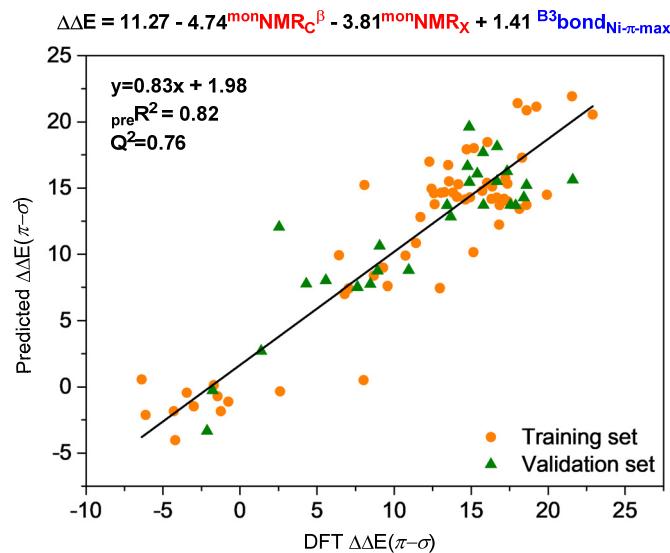
**Figure S1.** Gas-phase vs. solvation effect  $\Delta\Delta E(\pi-\sigma)$  (kcal/mol) for complex **II<sub>Pd</sub>**.

**87 polar monomers**

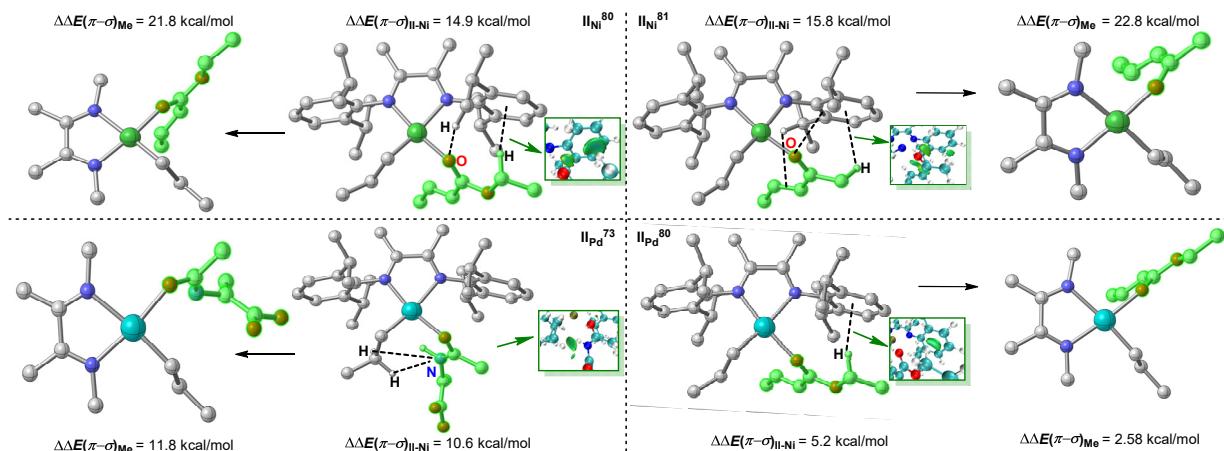


**Figure S2.** 87 polar monomers are used in the calculation of this work.

The descriptors of  $\pi$ -coordination structure **B3** (by complex **II<sub>Ni</sub>**) and polar monomers are calculated for multivariate linear regression. A total of 19 descriptors were calculated, including Sterimol values [50] (<sup>B2</sup>B1<sub>Ni-X</sub>, <sup>B2</sup>B5<sub>Ni-X</sub> and <sup>B2</sup>L<sub>Ni-X</sub>), steric hindrance of metal center (<sup>B3</sup>Steric<sub>Ni</sub>), bond length (<sup>B3</sup>bond<sub>Ni- $\pi$ -max</sub> and <sup>B3</sup>bond<sub>Ni- $\pi$ -min</sub>), Infrared freq (<sup>mon</sup>IR<sub>C=C</sub>) and Freq Intensities (<sup>mon</sup> $\nu_{C=C}$ ), NMR (<sup>mon</sup>NMR<sub>C</sub><sup>a</sup>, <sup>mon</sup>NMR<sub>C</sub><sup>b</sup> and <sup>mon</sup>NMR<sub>X</sub>), NBO (<sup>B3</sup>NBO<sub>Ni</sub>, <sup>mon</sup>NBO<sub>X</sub>, <sup>mon</sup>NBO<sub>C</sub><sup>a</sup> and <sup>mon</sup>NBO<sub>C</sub><sup>b</sup>), Polarizability (<sup>mon</sup> $\alpha$ ), HOMO (<sup>mon</sup>HOMO), volume (<sup>mon</sup>V) and Dipole Moment (<sup>mon</sup> $\mu$ ).



**Figure S3.** Plot of computed vs. predicted  $\Delta\Delta E(\pi-\sigma)$  (kcal/mol) for complex **II<sub>Ni</sub>** (base on  $\pi$ -coordination structure) using the multivariate linear regression models.



**Figure S4.** Changes of structure and  $\Delta\Delta E(\pi-\sigma)$  after substitution of aromatic ring on catalyst (complex **II<sub>Ni</sub>**) with methyl.