

Article

Enhanced Photo-Fenton Removal Efficiency with Core-Shell Magnetic Resin Catalyst for Textile Dyeing Wastewater Treatment

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Figure S9. SEM images of MAER **a)** before and **b)** after UV-Fenton reaction.

Figure S10. **a)** X-3B sample and **b)** actual wastewater sample collected by different reaction time; X-3B sample **c)** before and **d)** after UV-Fenton reaction; actual wastewater sample **e)** before and **f)** after UV-Fenton reaction.

Table Lists

Table S1. Properties of the selected dye.

Table S2. Properties of actual textile dyeing wastewater sample.

Table S3. Surface structural properties of MAER.

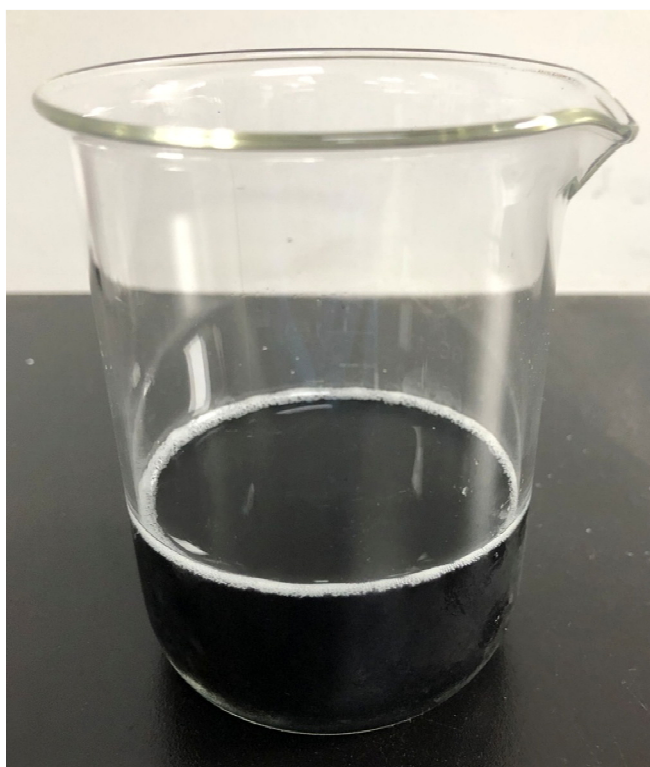


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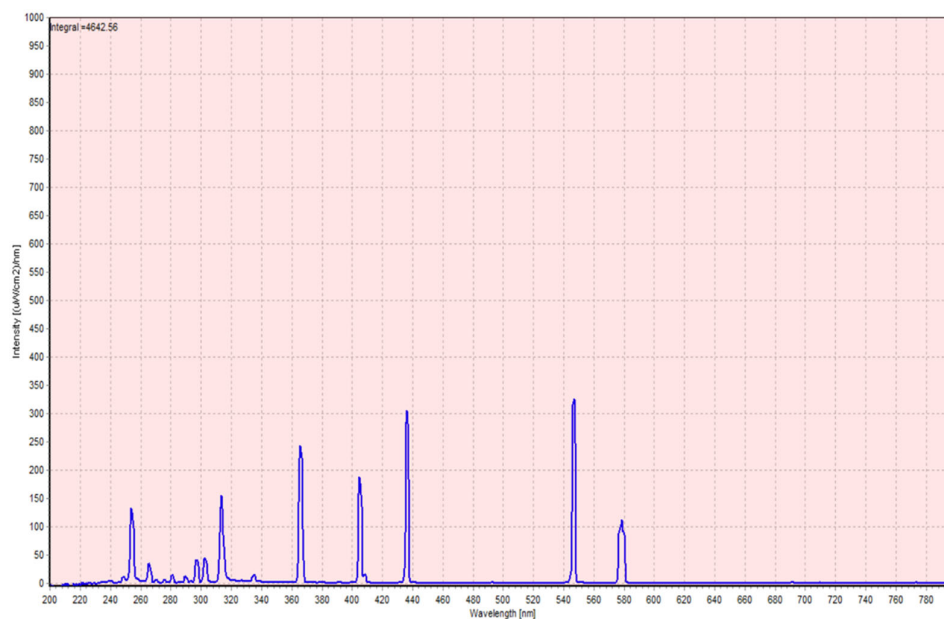


Figure S2. The UV light signal in the range of 200–800 nm.

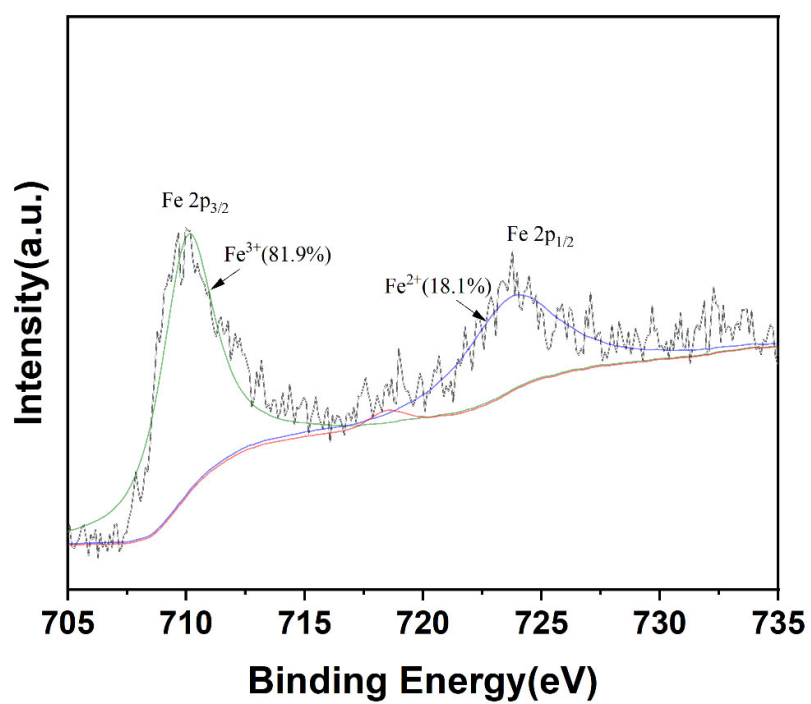


Figure S3. XPS spectra of Fe 2p for MAER catalyst.

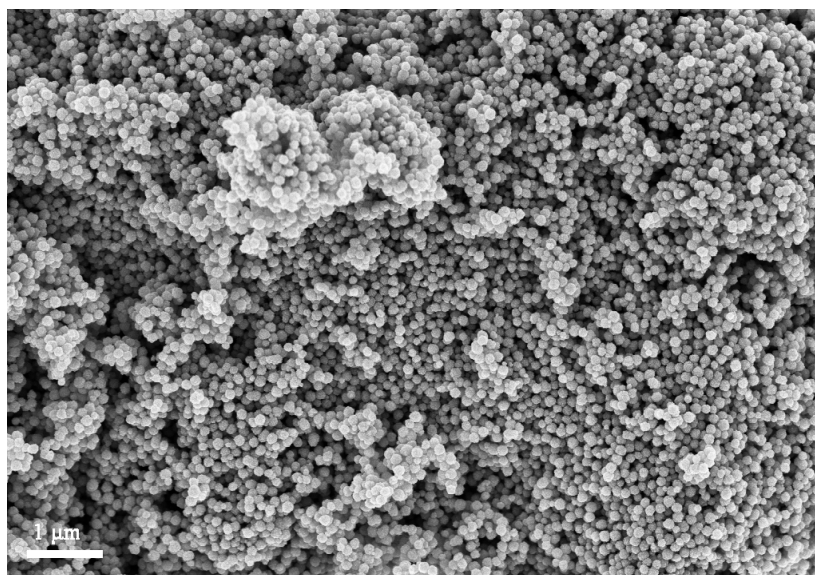


Figure S4. The SEM image of Fe₃O₄.

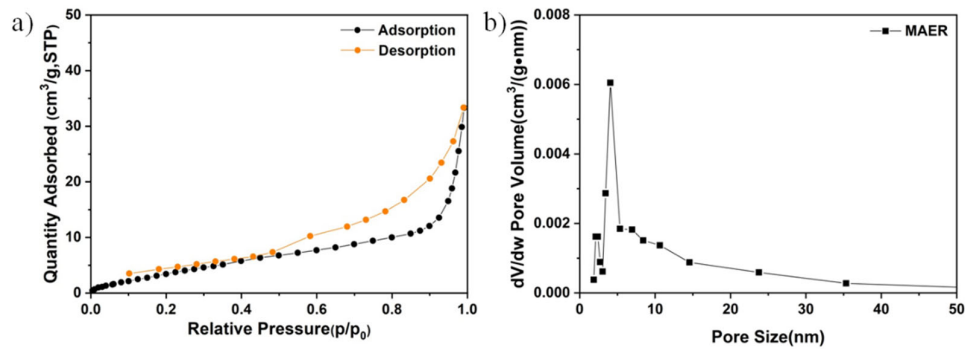


Figure S5. a) N₂ adsorption/desorption isotherm of MAER; b) The BJH (Barrett, Joyner and Halenda) pore size distribution of MAER.

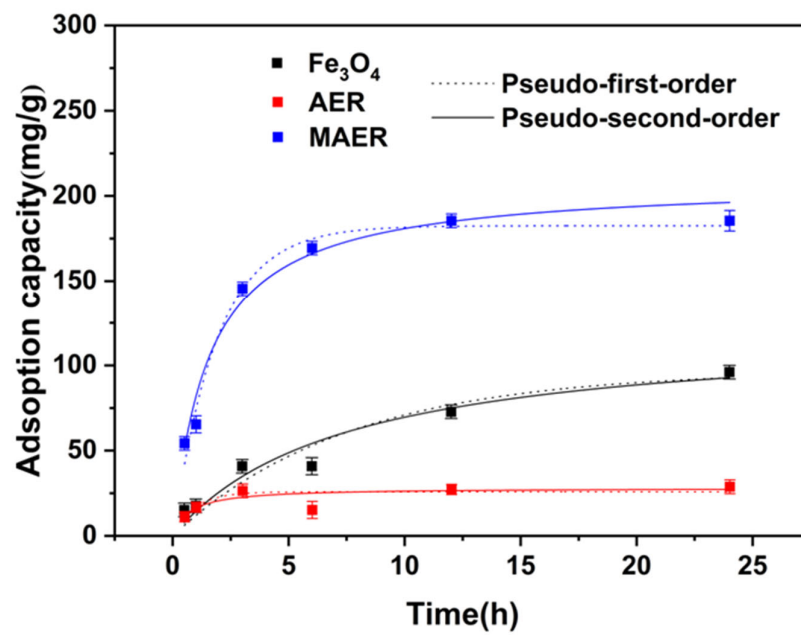


Figure S6. Adsorption behavior of X-3B on three synthesized materials.

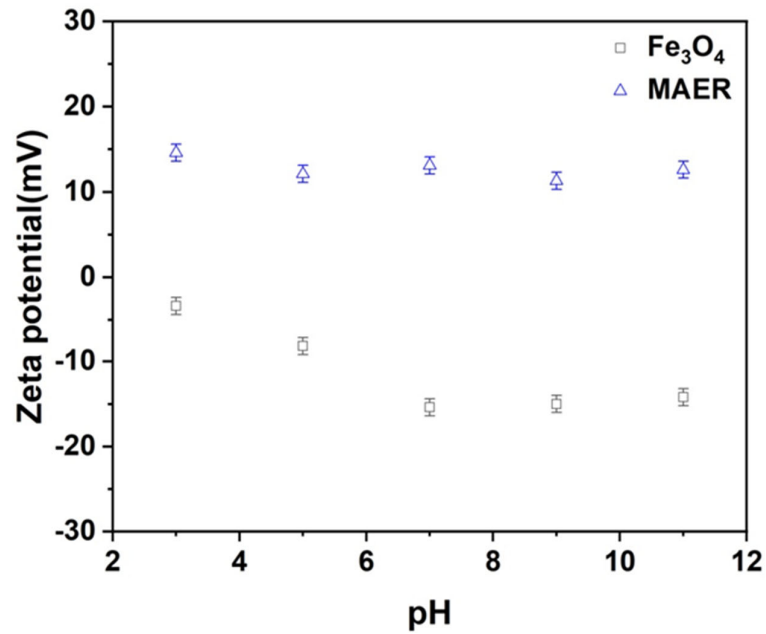


Figure S7. Zeta potentials of Fe₃O₄ and MAER in different solution pH values.

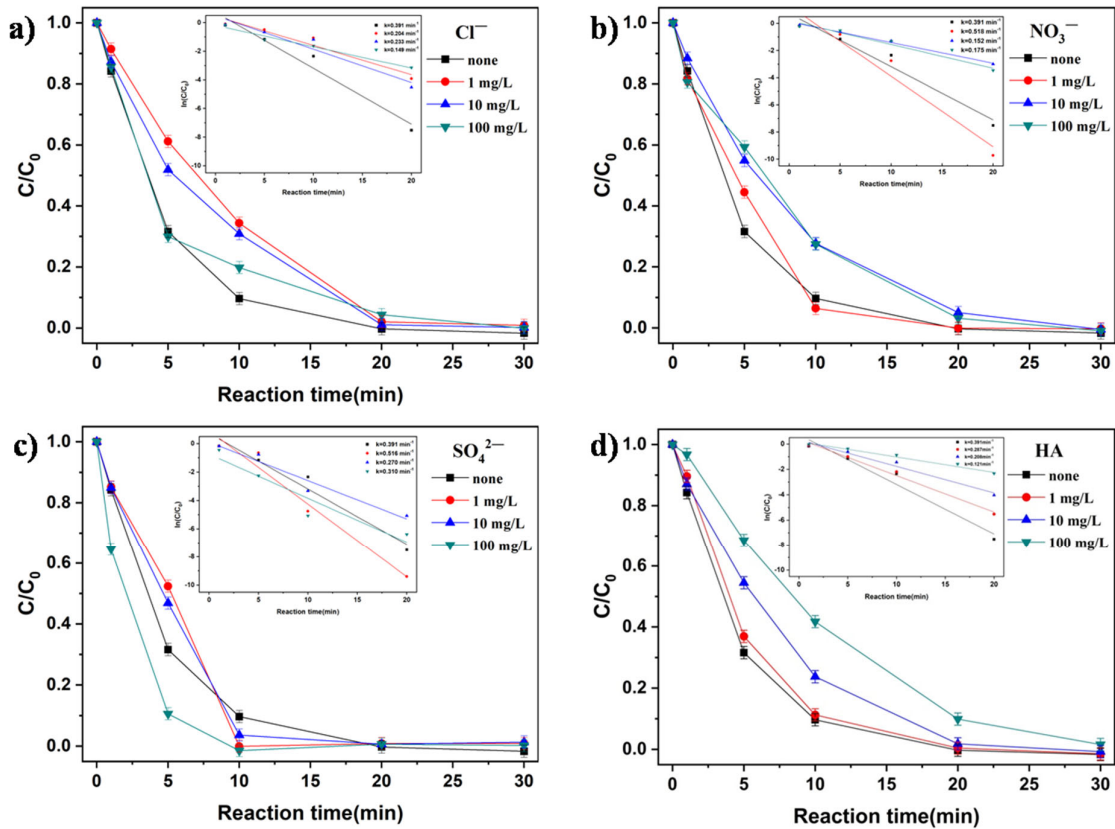


Figure S8. Matrix effects of a) Cl⁻, b) NO₃⁻, c) SO₄²⁻, d) HA on X-3B removal in MAER UV-Fenton system. Reaction conditions: C₀, B= 50mg/L, [Material]= 0.25 g/L, [H₂O₂] = 150 mM, T= 25°C, and pH with no adjustment.

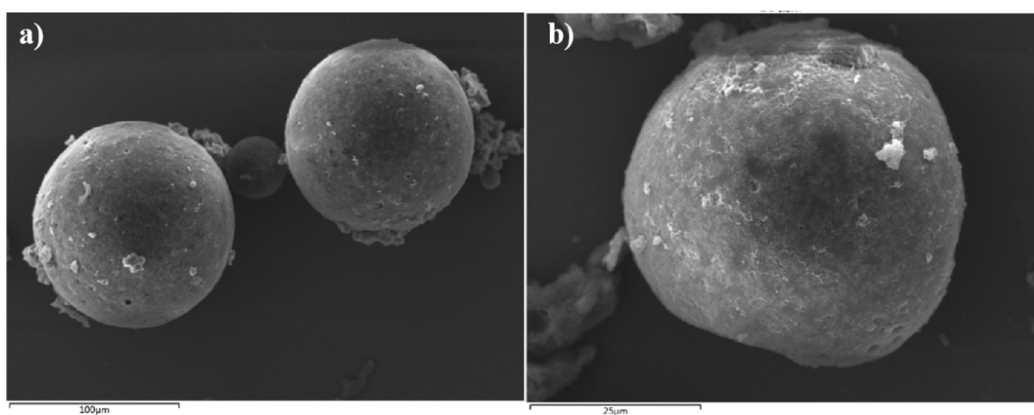


Figure S9. SEM images of MAER a) before and b) after UV-Fenton reaction.

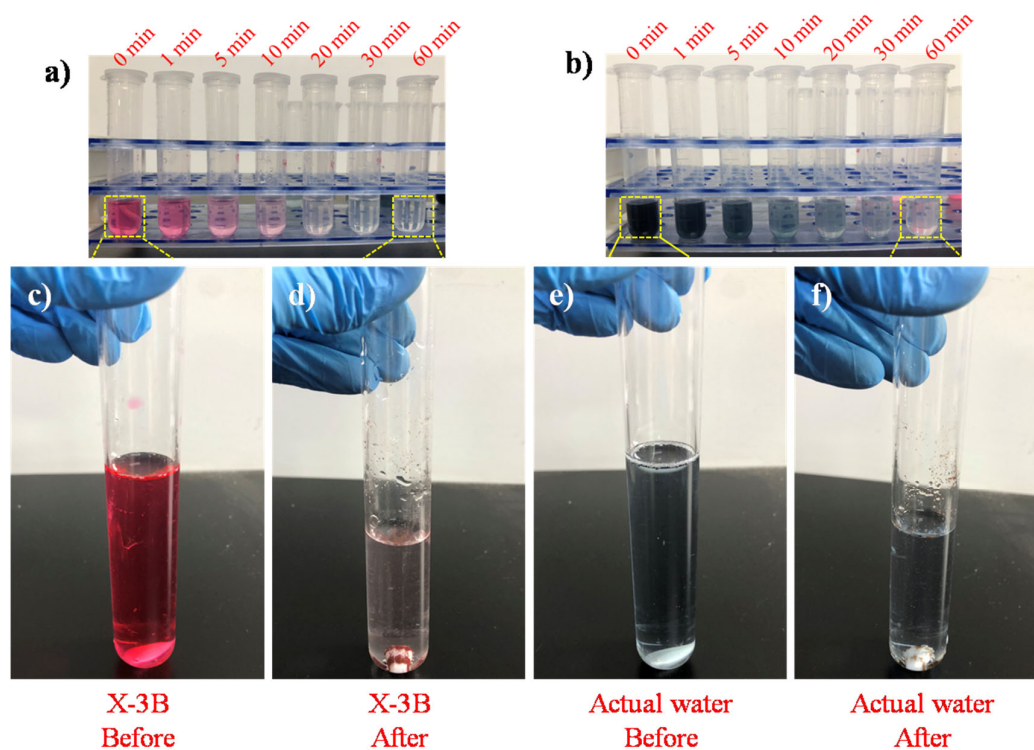


Figure S10. a) X-3B sample and b) actual wastewater sample collected by different reaction time; X-3B sample c) before and d) after UV-Fenton reaction; actual wastewater sample e) before and f) after UV-Fenton reaction.

Table S1. Properties of the selected dye.

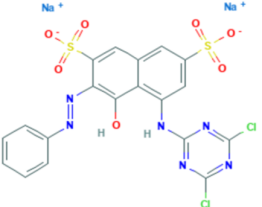
Name	Molecular structure	Molecular formula	Molecular weight	CAS
Reactive brilliant red (X-3B)		$C_{19}H_{10}Cl_2N_6Na_2O_7S_2$	615.33	17804-49-8

Table S2. Properties of actual textile dyeing wastewater sample.

Parameters	Wastewater
pH	12.64
Conductivity ($\mu S/cm$)	9.75
ORP (mV)	−333.4
UV ₄₀₀	2.62
TOC (mg/L)	77.95
IC (mg/L)	24.51

Table S3. Surface structural properties of MAER.

Sample	S_{BET} (m^2/g)	V_{pore} (cm^3/g)	Average pore size (nm)
MAER	18.67	0.051	8.69